

Sustainable Futures Institute



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## **Sustainability of Valorizing MSW**

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# Introduction

#### • Recycling Alternatives for Municipal Plastic Waste





Singh N. et al. (2017) Composites part B, 115: 409-422

## **Chemical Recycling of Waste Plastic: Technologies**



**Conversion**: a thermal process involving breaking bonds in the polymer to produce liquid and gaseous products such as fuels and petrochemicals.

**Decomposition**: a biological, chemical, or thermal process involving selective breaking of bonds in the polymer to produce monomers.

**Purification**: a process involving dissolving plastics in solvents to remove pigments and additives prior to separating pure resin.



Accelerating Circular Supply Chains for Plastics: A Landscape of Transformational Technologies. Closed Loop Partners, 2019

## LCA Results for Mechanical Recycling: PET, HDPE, PP



Figure 3-1. Total Energy Results for Recycled and Virgin Resins (MJ/kg)

LIFE CYCLE IMPACTS FOR POSTCONSUMER RECYCLED RESINS; PET, HDPE, PP, FRANKLIN AND ASSOCIATES DIVISION OF ERG, DEC 2018



## LCA Results for Mechanical Recycling: PET, HDPE, PP



#### Figure 3-3. Water Consumption Results for Recycled and Virgin Resins (liters water/kg resin)

LIFE CYCLE IMPACTS FOR POSTCONSUMER RECYCLED RESINS; PET, HDPE, PP, FRANKLIN AND ASSOCIATES DIVISION OF ERG, DEC 2018



## LCA Results for Mechanical Recycling: PET, HDPE, PP



#### Figure 3-6. Global Warming Potential Results for Recycled and Virgin Resins (kg CO<sub>2</sub> eq/kg resin)

LIFE CYCLE IMPACTS FOR POSTCONSUMER RECYCLED RESINS; PET, HDPE, PP, FRANKLIN AND ASSOCIATES DIVISION OF ERG, DEC 2018



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## LCA Results for Chemical Recycling: HDPE



Benavides et al., 2017, Life-cycle analysis of fuels from post-use non-recycled plastics, Fuel 203, 11–22



## Case Study: Sustainability Assessment - Thermal Conversion of Waste HDPE

#### **Research Objectives**

- Design and simulation a <u>multi-product refinery process</u> for conversion of waste High Density Polyethylene using pyrolysis.
- Evaluate the energy requirements of the refinery (Energy returned over energy invested).
- Evaluate the environmental performance of the refinery products (kg CO<sub>2</sub> eq./kg of product).
- Evaluate the economic feasibility of the project (Net present value).



## **Materials and Methods**



Gracida-Alvarez U. et al. (2018) Industrial & Engineering Chemistry Research, 57: 1912-1923



## Results – Micro-pyrolysis experiments

Composition of the two-stage micro-pyrolysis reactor outlet

(650 °C & 2.8s vapor residence time)

| Chemical class                       | Mass<br>percentage |  |
|--------------------------------------|--------------------|--|
| Hydrocarbon gases (C1-C4)            | 68.63              |  |
| Gasoline range hydrocarbons (C5-C10) | 20.68              |  |
| Diesel range hydrocarbons (C11-C15)  | 3.14               |  |
| Light waxes (C16-C20)                | 1.34               |  |
| Heavy waxes (C21-C29)                | 0.75               |  |
| Aromatics                            | 5.46               |  |

A total of 86 compounds were used in the simulation

Process Temperature range: -140 °C to 1200 °C Process Pressure range: 0.5 to 25 bar Processing Capacity: 500 tonnes/day (20.83 tonnes/hr)







#### **Results – Conceptual design**

• Process Flow Diagram (PFD)





#### **TEA Results**

90.0 84.4 83.0 80.0 fotal installed costs (MM USD) 70.0 60.0 50.0 40.0 30.0 20.0 10.0 0.0 BC HI A-100 A-200 A-300 A-400 HEN

Total Installed Costs (MM USD)

Gracida-Alvarez, U.R., et al., ACS Sustainable Chemistry and Engineering, DOI: 10.1021/acssusch emeng.9b04763









# **Results – Environmental evaluation**

- Functional unit: 1 kg of product
- Scope: Cradle to gate
- Allocation: Mass allocation
- US grid electricity



MRF: Materials Recovery Facility

- Heat from Flue gas is utilized internally
- Electricity generated in turbines is utilized internally

General inventory

Basis: Processing of 20.83 tonnes of HDPE (Plant capacity for 1 hr of operation)

| area  | input                           | BC     | HI-1   | HI-2   |
|-------|---------------------------------|--------|--------|--------|
| A-100 | electricity (kWh)               | 333.54 | 330.70 | 330.70 |
|       | natural gas (GJ)                | 2.20   | 0.50   | 0.50   |
|       | purge combustion (GJ)           | 3.10   | 3.10   | 3.10   |
|       | cooling water (m <sup>3</sup> ) | 3.95   | 1.44   | 1.44   |
|       | helium (kg)                     | 0.04   | 0.04   | 0.04   |
| A-200 | electricity (kWh)               | 954.55 | 791.57 | 791.57 |
|       | natural gas (GJ)                | 2.42   | 0.59   | 1.49   |
|       | cooling water (m <sup>3</sup> ) | 7.37   | 5.75   | 5.75   |
| A-300 | natural gas (GJ)                | 0.60   | 0.51   | 0.51   |
|       | cooling water (m <sup>3</sup> ) | 0.79   | 0.63   | 0.63   |
|       | sulfolane (kg)                  | 0.27   | 0.27   | 0.27   |
|       | process water (kg)              | 0.17   | 0.17   | 0.17   |
| A-400 | natural gas (GJ)                | 0.25   | 0.00   | 0.00   |
|       | cooling water (m <sup>3</sup> ) | 0.09   | 0.09   | 0.09   |
|       | hydrogen (kg)                   | 15.82  | 15.82  | 15.82  |

Note: Recycled inputs (Helium, sand, and refrigerants) were not considered in the inventory.

Fitzgerald G. et al. (2012) Resources, Conservation, and Recycling, 69: 50-56



#### LCA Results – Carbon Footprint of Products



1.6 kg CO<sub>2</sub> eq. / kg of broduct 3.0 d. / kg of broduct 3.0 0.4 0.2 0.4 1.34 1.31 1.27 1.25 1.10 1.02 0.99 0.80 0.0 вс HI-1 HI-2 F BC HI-1 HI-2 F Low MWHCs **High MWHCs** CST A-100 A-200 A-300 ■A-400

#### **GHG Savings**

| HI-2 vs Fossil (F): |        |
|---------------------|--------|
| Ethylene:           | 32.5%  |
| Propylene:          | 11.6%  |
| Aromatics:          | 29.3%  |
| Low MWHCs:          | -11.1% |
| High MWHCs:         | -27.5% |

Gracida-Alvarez, U.R., et al., ACS Sustainable Chemistry and Engineering, DOI: 10.1021/acssusch emeng.9b04764



#### LCA Results – Regional Electricity Grid Effects



Figure 9. Effect of state mixture composition on the GHG emissions of the refinery products. (A) Scenario HI-1 and (B) scenario HI-2.



## Systems Analysis Framework

#### Sustainability Assessments of Plastics in a Global Circular Economy



Shonnard, et al., 2019, Systems analysis for PET and olefin polymers in a circular economy, *Procedia CIRP*, 80, 602-606, 26th CIRP Life Cycle Engineering (LCE) Conference.

#### **REMADE Project 18-01-SA-04**



#### **Research issues and questions**

- Will a plastics <u>circular economy</u> improve performance compared to the current plastics <u>linear economy</u>
  - environmental, economic, and societal impacts?
- How would the prevalence of chemical versus mechanical recycling versus incineration for energy affect system performance?
- If renewable (i.e. plant-derived) feedstocks increase vs fossil, what affect would this have on system performance?
- What could be the impacts of biodegradable plastics on system performance?
  - Including ocean debris effects
- External effects beyond the plastics pathways
  - Indirect economic multipliers
  - Impacts to the petroleum, gas, and petrochemical industries



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# Thank you for your attention! **Contact Information:** David R. Shonnard: drshonna@mtu.edu

Advancing the Bioeconomy: From Waste to Conversion-Ready Feedstocks

#### **Extra Slides**



## **Plastics Challenge**

 Global Situation of Plastics **PRODUCTION:** 322 million tonnes /year (2015) **WASTE GENERATION:** 300 million tonnes /year (2015) **RECYCLING RATE:** 9 % (2015)



#### World Plastics Production 1950 – 2015

World

1977: 50

2015: ~322

2011:~280

2009: 250

2002: 200

1989: 100

in Mio. t 350

300

250

200

150

100

50

2020

## **Cumulative Plastic Production/Use Data**

#### **Production/Use**

- 4% of petroleum (feedstocks)
- 4% of petroleum (process energy)
- Additional inputs in Natural Gas
- Non-fiber plastics (88%)
- Packaging (39%) is largest consumption sector (PE, PP, PET) with the shortest in-use lifetime (<1 yr)

#### **End of Life**

- Landfilling (79%)
- Incineration (12%)
- Recycled (9%)

RRS, Ann Arbor, MI, 2017

![](_page_22_Figure_12.jpeg)

![](_page_22_Picture_13.jpeg)

### Projections to 2050

• Cumulative plastic waste generation and disposal

![](_page_23_Figure_2.jpeg)

2014

20

1:5

2025 >1:1

- Health risk to aquatic and terrestrial life.
- Displacing primary plastic production.
- Use of emerging technologies.

Plastics Europe (2018) Plastics – the Facts 2017

Geyer R. et al. (2017) Science Advances, 3: e1700782

World Economic Forum et al. (2016) The New Plastics Economy. Rethinking the future of plastics.

![](_page_23_Picture_9.jpeg)

![](_page_23_Picture_10.jpeg)

## Linear vs Circular Economy for Plastics

![](_page_24_Figure_1.jpeg)

- 80% of plastics is landfilled or lost to the environment.
- Economic losses between 80 to 120 billion USD/year.
- Consumption of virgin fossil resources.

World Economic Forum et al. (2016) The New Plastics Economy. Rethinking the future of plastics. Arena U. et al. (2011) Waste Management, 31, 1494-1504. European Commission (2016) A European Strategy for Plastics in a Circular Economy.

- Reduce the use of virgin materials.
- Eliminate mismanagement and leakage.
- Build up recycling infrastructure.

Shonnard, D.R., Tipaldo, E., Thompson, V., Pearce, J., Caneba, G., Handler, R.M., 2019, Systems analysis for PET and olefin polymers in a circular economy, *Procedia CIRP*, 26th CIRP Life Cycle Engineering (LCE) Conference.

![](_page_24_Picture_10.jpeg)

#### **Linear Economy: Production Inputs**

![](_page_25_Figure_1.jpeg)

![](_page_25_Picture_2.jpeg)

## **Circular economy: production inputs**

![](_page_26_Figure_1.jpeg)

![](_page_26_Picture_2.jpeg)

#### **Conceptual Design – Mass and Energy Balances**

#### **Refinery Mass Balances**

| Streams | Product              | Amount<br>(tonnes/hr) |
|---------|----------------------|-----------------------|
|         | Waste HDPE           | 20.83                 |
|         | Helium               | 0.001                 |
|         | Sulfolane            | 0.006                 |
| INLEI   | Water                | 0.004                 |
|         | Hydrogen             | 0.33                  |
|         | TOTAL                | 21.17                 |
|         | Flue gas purge       | 1.32                  |
|         | Ethylene             | 3.91                  |
|         | Propylene            | 2.80                  |
| OUTLET  | Aromatic mixture     | 0.77                  |
|         | Low MW HC (C4-C12)   | 11.25                 |
|         | High MW HC (C12-C29) | 1.12                  |
|         | TOTAL                | 21.17                 |

#### **Refinery Energy Balances**

| Streams | Energy Source    | Amount<br>(GJ/hr) |
|---------|------------------|-------------------|
|         | Process Energy   | 223.03            |
| INLET   | Materials Energy | 931.31            |
|         | TOTAL            | 1154.35           |
|         | Process Energy   | 180.78            |
| OUTLET  | Products Energy  | 966.43            |
|         | TOTAL            | 1147.20           |

![](_page_27_Picture_5.jpeg)

#### **Conceptual Design - Results**

#### **Refinery Products Specifications**

| Product              | Recovery (%) | Purity (%) |
|----------------------|--------------|------------|
| Ethylene             | 89.51        | 97.22      |
| Propylene            | 99.70        | 97.85      |
| Aromatics<br>mixture | 57.15        | 84.27      |
| Low MWHCs            | 56.10        | 97.74      |
| High MWHCs           | 76.43        | 83.33      |

#### **Energy Returned over Invested (EROI)**

Base Case (BC): 2.2 Heat Integrated (HI): 3.0 Petroleum Refining: 9

#### Primary Energy Requirements

![](_page_28_Figure_6.jpeg)

#### **Primary Energy Savings**

HI vs BC: 35% reduction

![](_page_28_Picture_9.jpeg)

#### **TEA Methods**

#### Parameters for Discounted Cash Flow Analysis

| Parameter                     | Value                    |
|-------------------------------|--------------------------|
| Internal rate of return (%)   | 10                       |
| Project economic life (years) | 20                       |
| Depreciation method           | 7-year MACRS             |
| Tax rate (%)                  | 21                       |
| Working capital (WC)          | 15% FCI                  |
| Base year                     | 2017                     |
| Operating days per year       | 350                      |
| Investment year 1             | 30% FCI                  |
| Investment year 2             | 50% FCI                  |
| Investment year 3             | 20% FCI + WC + FOC + 50% |
|                               | VOC                      |
| Investment year 4             | FOC + 90% VOC            |
| Investment year 5             | FOC + VOC                |

#### Prices for Discounted Cash Flow Analysis

| Product                             | Price |
|-------------------------------------|-------|
| Waste HDPE (USD/tonne)              | 22.0  |
| Electricity (USD/kWh)               | 0.069 |
| Natural gas (USD/GJ)                | 3.95  |
| Cooling water (USD/m <sup>3</sup> ) | 0.053 |
| Hydrogen (USD/kg)                   | 2.83  |
| Helium (USD/kg)                     | 42.81 |
| Ethylene (USD/kg)                   | 0.61  |
| Propylene (USD/kg)                  | 0.97  |
| Aromatics mixture (USD/kg)          | 1.02  |
| Low MWHC mixture (USD/kg)           | 0.86  |
| High MWHC mixture                   | 0.84  |
| (USD/kg)                            |       |
| LP steam (USD/kg)                   | 0.021 |

![](_page_29_Picture_5.jpeg)

## **Results – Environmental evaluation**

Multi-product mass allocation

![](_page_30_Figure_2.jpeg)

E: Ethylene product, P: Propylene product, A: Aromatics mixture product, L: Low MW HC mixture product, H: High MW HC mixture product

• Allocation was product-based, therefore, trace amounts of different chemical species included in a particular product were also included on its allocation factors.

![](_page_30_Picture_5.jpeg)

## **Results – Environmental evaluation**

• Carbon Footprint: Process Sections

![](_page_31_Figure_2.jpeg)

![](_page_31_Picture_3.jpeg)