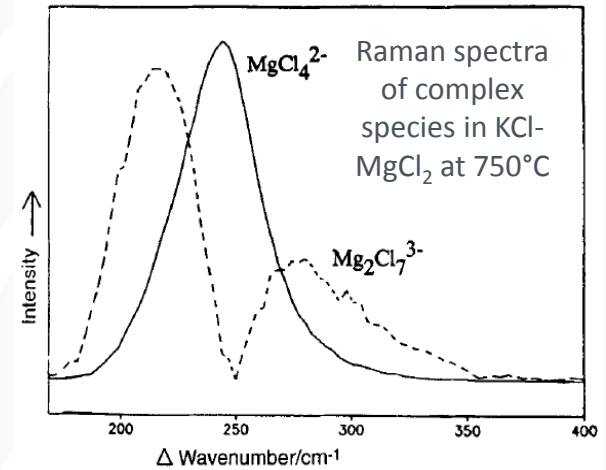


Interfacial Studies of Performance of Protecting Layer for Corrosion Inhibition



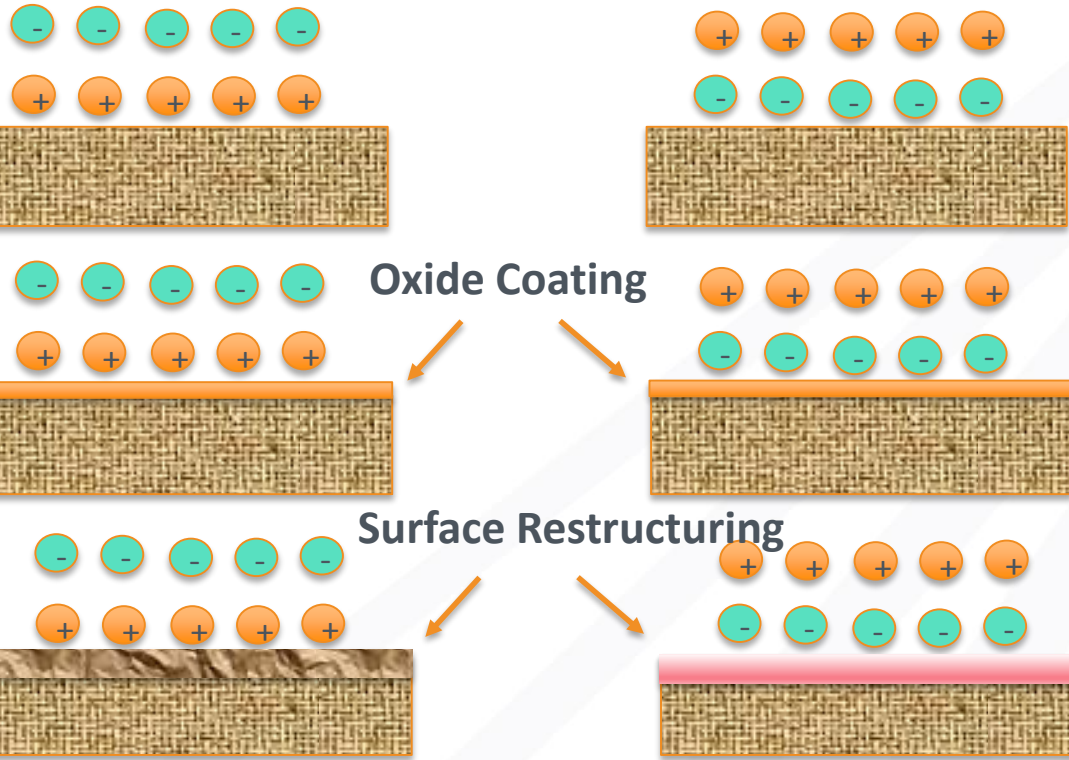
Knowledge Gap in Interfacial and Melt Structures

- Multicomponent species in equilibria
- Melts are not always ionic; can be molecular, network like



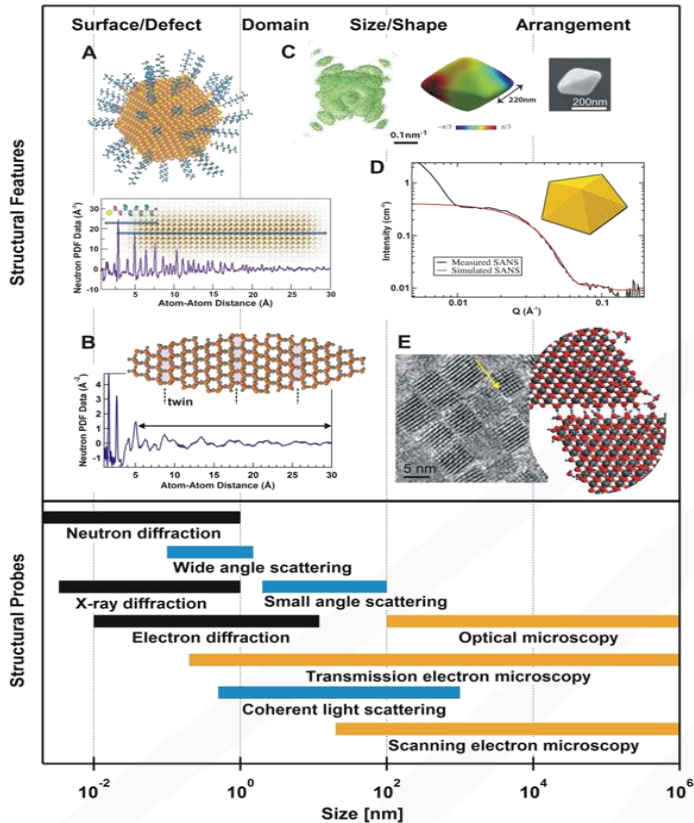
Dai, S., et. al. *J. Raman Spectr.* **1995**, 26, 929-932

Complexity in melt structures



Complexity at interfaces

X-ray and neutron scattering are ideal to probe complex molten salt media in operando



Neutrons can help in many ways:

- Neutrons easily penetrate many vessel materials, enabling *in situ* measurements
- Probe speciation in complex multi-component melts and correlate atomistic structures to thermodynamic and corrosion properties of molten salt systems
- Characterize in-situ interfacial structures between metals and molten salts

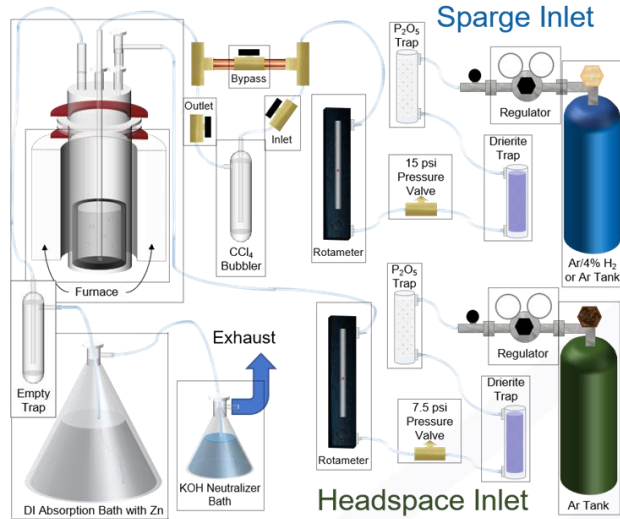
Main challenges and goals

- The extreme corrosivity of chloride-based transfer fluids for concentrated solar power hinders the use of molten salts.
- Find the mechanism of the corrosion allowing us to make educated decision on how to proceed with salt implementation in existing setups or how to create new better ones
- Employ cutting edge spectroscopy techniques, such as grazing incidence X-ray absorption, scattering, and reflectometry, allowing us to approach corrosion on the interface and study it at the molecular level.
- Design a new cell enabling *in situ* studies to see the corrosion happening step-by-step.

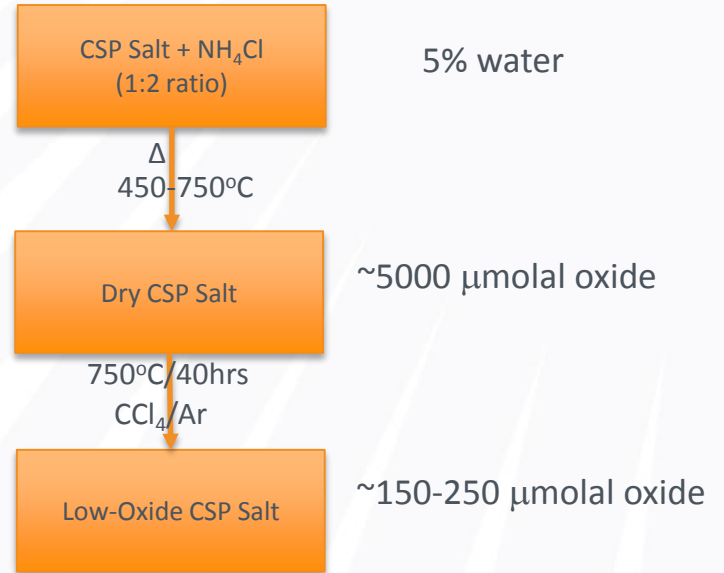
Key activities

- Purification of the candidate salt for transfer fluid: MgCl_2 and KCl
- Film preparation by sputtering of the pure alloy and then the salt on a substrate
- Annealing of the films for *ex situ* studies
- Grazing Incidence X-ray spectroscopy and scattering measurements
- *In situ* measurements and *in situ* cell design

Salt purification

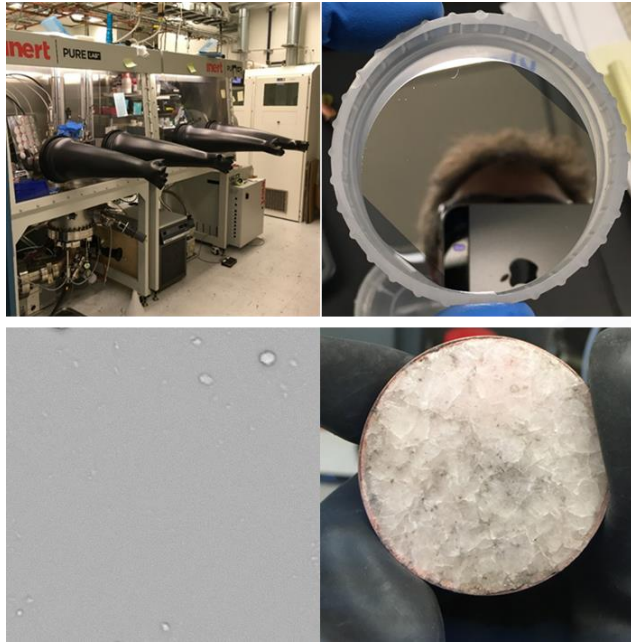


- Ultra High purity argon
- P_2O_5 filled traps to further dry gas
- Salt quantity dictates sparging time
- Final purge with H_2/Ar mixture to remove residual chlorination products

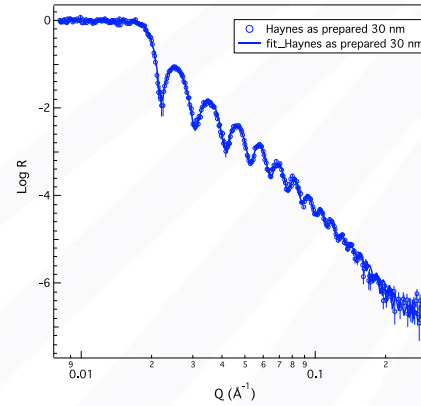


→ Oxide content measured via acid/base titration (~50ppm)

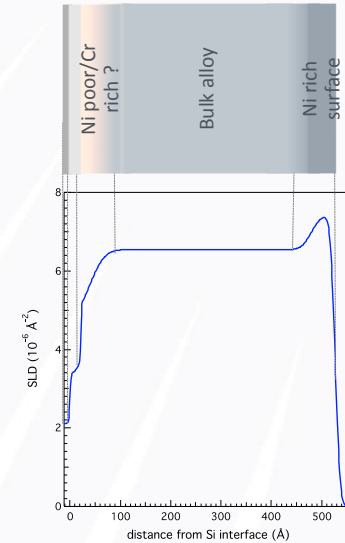
Alloy and salt deposition



Deposition of alloys and salt



Haynes "30 nm" Film			
	Thickness (Å)	SLD ($\times 10^{-6} \text{ \AA}^{-2}$)	Roughness/Interfacial Width (Å)
Surface layer	47.5	7.42	7.6
Bulk layer	440.5	6.54	15.6
Interfacial layer	25.0	5.19	9.7
SiOx	21.7	3.25	1.2
Si Substrate	--	--	1.6

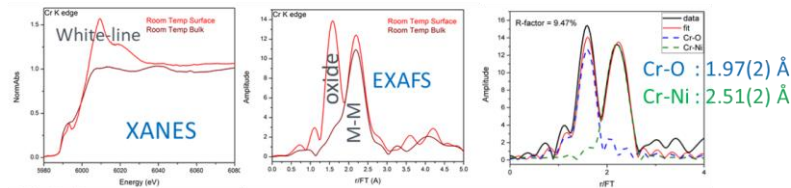


- As-deposited film thickness of 51.3 nm
- Film segregates into 3 regions
 - Ni rich surface
 - Bulk alloy
 - Ni poor or Cr rich substrate interface

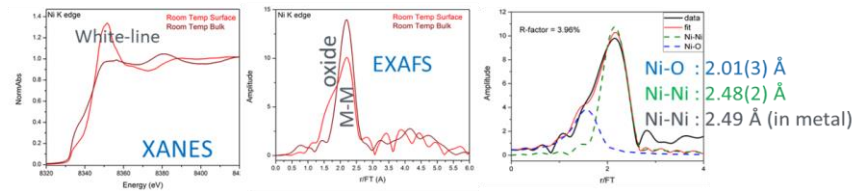
Grazing Incidence X-ray Absorption measurements

Room temperature

Chromium

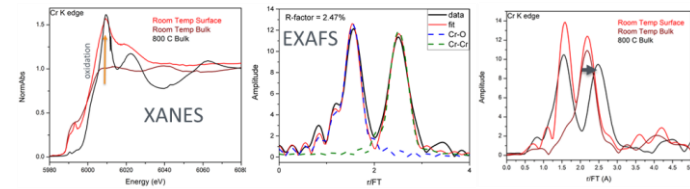


Nickel

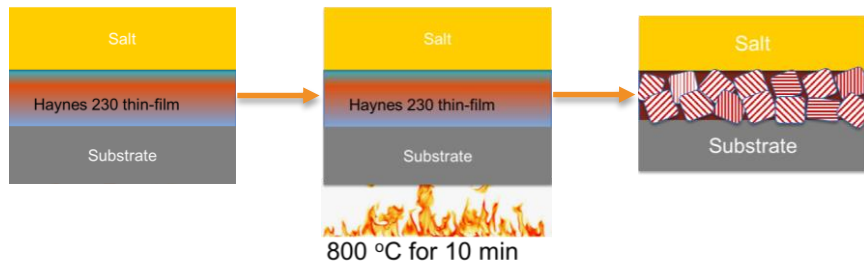
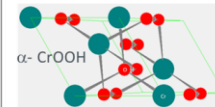
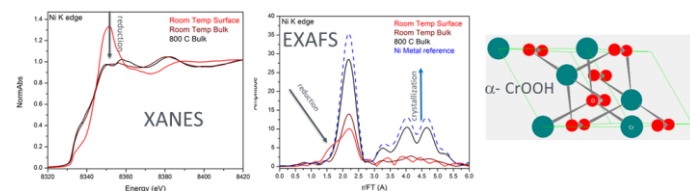


After annealing

Chromium



Nickel



- Loss of planar interface
- 111 Textured Haynes film transformed to grain-coarsened untextured FCC Ni
- Evidence for α -CrOOH and other phases

Summary

- Successfully purified MgCl_2 and KCl and lowered **oxide concentration down to 50 ppm**, which allows more systematic study of corrosion
- Deposited **smooth alloy and salt films**, which are necessary for grazing incidence studies
- Grazing incidence X-ray absorption shows **oxidation of Cr in the alloy to $\alpha\text{-CrOOH}$** indicating pathway for the corrosion
- Grazing incidence diffraction also shows **texturing of Ni metal** after Cr leaves the alloy structure showing what happens to the corroded alloy
- Neutron and X-ray scattering can be done ***in situ* up to 600°C and 900°C**, respectively
- Neutron scattering cell for **higher temperatures is designed to allow *in situ* studies** at operating temperatures