2013 DOE Bioenergy Technologies Office (BETO) Project Peer Review

Development of Biofuels Using Ionic Transfer Membranes Phase III

May 20-23, 2013 Technology Area Review: Biofuels

Principal Investigator: **Dr. Kris Lipinska**, University of Nevada Las Vegas

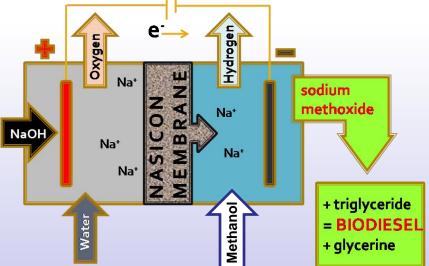
Investigators: **S. Balagopal**, Ceramatec Inc. Dr. O. Hemmers, UNLV Dr. C. Bae, Rensselaer Polytechnic Institute

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Goal Statement & Project Overview - 1

Sodium methoxide (SMO) is an effective catalyst for the transesterification of vegetable oils, animal fats and recycled greases. An important application for these materials is production of biodiesel

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Current production of SMO uses high-purity Na metal, an expensive raw material that requires a complex infrastructure for safe handling. Other methods: make SMO from Na/Hg amalgam, but final product contains toxic mercury; other produces chlorine and has higher carbon footprint.

- This project investigates an Electrochemical Membrane Process (EMP) to yield high-purity SMO from low-cost aqueous sodium hydroxide, using a Sodium Super Ionic Conductor membrane: NaSICON.
- The result is a lower-cost, high-purity SMO with no contaminants and no waste stream.

Goal Statement & Project Overview - 2

- Goal is to demonstrate an energy-efficient, low cost Electrochemical Membrane System (EMS) to produce Sodium Methoxide (SMO), on site and as needed, at a Biofuels Production Facility
- Demonstration of an integrated SMO unit with the Biofuel process will allow manufacturers the flexibility to evaluate other biomass feedstock's such as algal oil, and oils derived from other types of feedstocks
- **HIGH LEVEL** objectives of the project
 - Demonstrate the concept of selective transport Na⁺ across the NaSICON Membrane for reaction with methanol to form a final product: SMO
 - Obtain highest concentration of SMO produced versus energy consumed
 - Demonstrate long-term performance of a membrane cell to produce SMO
 - Ability to scale up size of ceramic membrane and build & operate a robust & reliable modular and scalable Multi-Tubular Unit

Quad Chart Overview - 1

Timeline

- PHASE-III Start date: 10/1/2011
- PHASE-III End date: 12/31/2013
- Percent complete: 60%

Barriers

Barriers addressed

- Biggest engineering problem to be overcome: scalability of membrane size and modularization of system footprint to meet the production demand. System operation reliability and safety (ex. handling of gases).
- Reduce commercial cost of production: the cost to produce Sodium Methoxide (SMO) by the electrochemical process is anticipated to be in the \$0.30 to \$0.35/lbs of 28 wt% SMO solution in methanol.

Budget (DOE/Cost Share)

- Funding for FY11: \$824K/\$339K
- Funding for FY12: \$726K/\$186K
- Funding for FY13: \$490k/\$0K
- Years the project has been funded:
 7 years with \$697K average annual funding.

Partners

- Key Interactions: K. Lipinska, UNLV;
 S. Balagopal, Ceramatec Inc.; O. Hemmers, UNLV; C. Bae, RPI
- Project Management: Kris Lipinska, U. of Nevada Las Vegas

Overall Approach

THE PROJECT : Development of an electrochemical process producing highpurity **Sodium Methoxide - SMO** from low-cost aqueous sodium hydroxide, with the use of **NaSICON** membranes (Sodium SuperIonic Conductors).

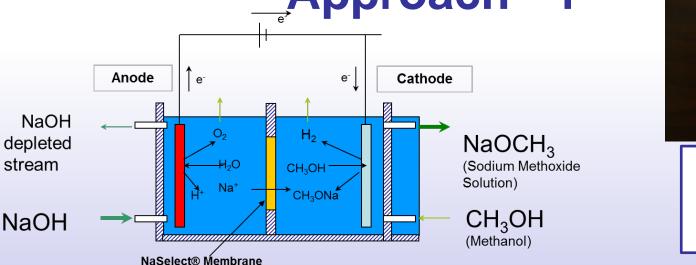
THE PROBLEM

 Our SMO production process is low-cost, but highly caustic solution reduces the stability of the NaSICON membrane and decreases its lifetime and performance.

3 BRANCHED APPROACH

- Approach #1: develop tubular membranes with higher burst pressure reliability (as opposed to planar) with better performance and reduced energy consumption.
- Approach #2: coating materials that conducts sodium ions and that extend membrane lifetime.
- Approach #3: composition-structure-property-synthesis route relationship studies to develop membrane materials that better withstand harsh operating conditions.

Approach - 1





- Use Na⁺ transporting NaSICON Membrane in an electro-chem. cell to make 99.99% pure SMO.
- Design Prototype Cell to transition from Planar to Tubular membrane.
- Develop a high-yield process to make large area **NaSelect[™] Membrane**.
- Conceptualize design to configure **multiple Tubular Membranes** to increase production.
- Demonstrate modular unit making up to **50 lbs/day of SMO** at highest concentration.
- Complete a techno-feasibility process analysis for integration into Biofuels production.
- Development of blocky ionic polymers with nano-structured ionic channels to coat and to protect NaSICON Membrane surface.
- Development of new NaSICON compositions with lower fabrication temperature, lower energy consumption and improved mechanical properties.
- Attempt an alternative glass-ceramic route for NaSICON synthesis by controlled crystallization of a precursor NaSICON glass in order to fabricate dense, small grain size, mostly single-phase NaSICON with good electrical conductivity.

Approach – 2 Management Approach

- The Project Management Plan (PMP) is the primary formal "guiding document" for the project and is updated as necessary and transmitted to all participants.
- Formal Reports: quarterly technical and financial status reports; (planned vs. actual expenditures for the quarter); exception reporting (explaining a major deviation from the established schedule); and semi-annual update of the PMP.
- **Project Meetings**: include bi-annual technical project accomplishment meetings, which include all project participants; and informal meetings among PIs to resolve issues and exchange information.
- UNLV lead PI, as the overall project manager, communicates with all project participants by technical accomplishment reports; by periodic project updates and by informal meetings with co-PIs and representatives of the industrial participant.

Approach – 3 Milestones for Monitoring Progress

Key Milestones Summary

- Complete flow sheet and system design
- Processing yield of NaSICON membranes: 70% of ceramic tubular membrane manufacturing yield
- Process design maturity: 1000h of cell performance in NaOH to produce sodium methoxide, cell operation at 100 mA/cm² at 50C.
- Demonstrate polymer coated NaSICON with stability over 1 week

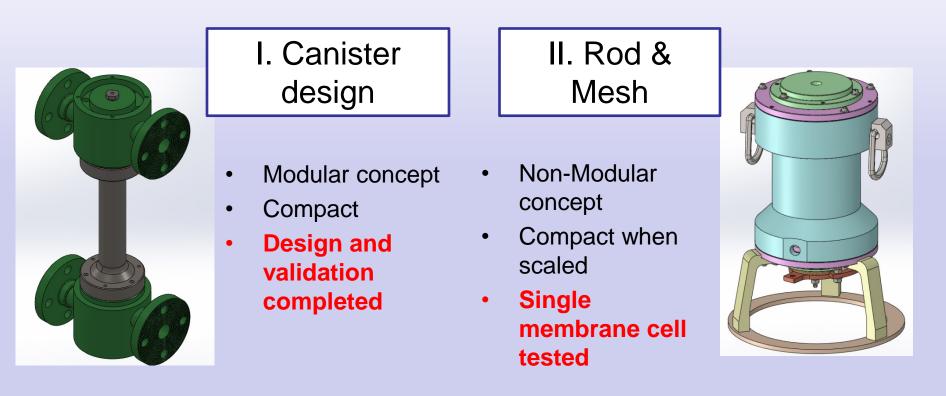
Go/No-Go Decisions are based on chemical & economic feasibility and operating parameters:

- Design single membrane cell: 500h of cell operation to establish seal reliability
- Assembly of skid: 100% dimension target to manufacture and assembly components.
- Operation single membrane cells: 1000h of cell operation at 100 mA/cm2 within 1 volts of steady performance.
- Cost target for Sodium methoxide: 100% payback and cost target to produce sodium methoxide
- Demonstrate NaSICON ceramics produced by glass technology with submicron grain size and high-crystallinity

Approach – 4 Outline of Roles & Responsibilities

- UNLV's Harry Reid Center for Environmental Studies (UNLV- HRC) manages and administers the project according to the Statement of Project Objectives.
- Ceramatec Inc is the industrial partner in the project.
- UNLV Office of Sponsored Programs (OSP) assures that all financial assistance agreements are accomplished, and is the primary POC with DOE and with the industrial partner regarding contractual and financial matters.
- Lead PI plans, directs and evaluates the research assigned to UNLV as well as oversees and evaluates the work assigned to the industrial partner. The PI reports directly to the DOE.
- **Co-PI's** are responsible for individual **Tasks**.
- Ceramatec's PI: (1) leads the development and technology maturity activities, (2) is responsible for planning, directing and evaluating the work of the sub-contract including the schedule of milestones and deliverables; (3) reports directly to the UNLV lead PI on accomplishment of the work, (4) contractual and financial aspects are the responsibility of Ceramatec's business manager whose POC is UNLV's lead PI and OSP.

Two prototype design concepts developed with Tubular NaSelect[™] membrane with electrode configuration for design qualification.



Single membrane cell

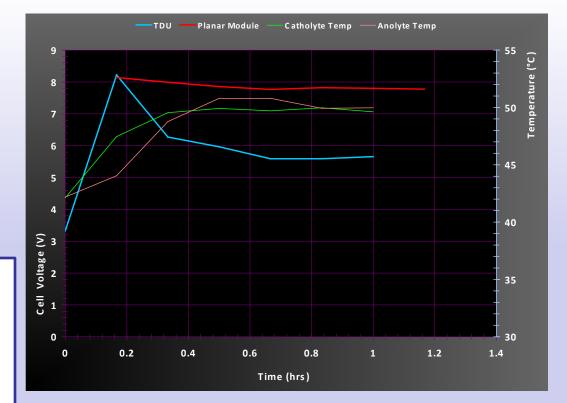
5 membrane cell

Operating conditions:

- 100 mA/cm²
- 20% NaOH
- 20% NaOCH₃

GOAL: to reduce energy consumption:

- Tubular membrane cell operates ~2.2V lower than planar membrane cell
- 0.83 kWh/lb. of SMO on 100% basis



Tubular NaSICON Membrane-based Cell Units allow operation:

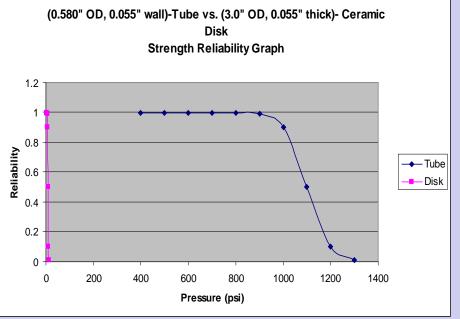
- under higher flow and pressure drop conditions
- with lower number of cell seals
- with smaller footprint of device

ACHIEVEMENTS in manufacture of large area ceramics:

- Developed process to make tubular NaSelect[™] membrane
- Transition from 0.8 inch diameter to
 1.5 inch diameter membrane
- Increased ceramic production yield from 60% to 80%
- Implemented quality control measures to make robust membranes

→ Weibull reliability: NaSelect[™]
 tubular membrane withstands >
 400 psi pressure differential





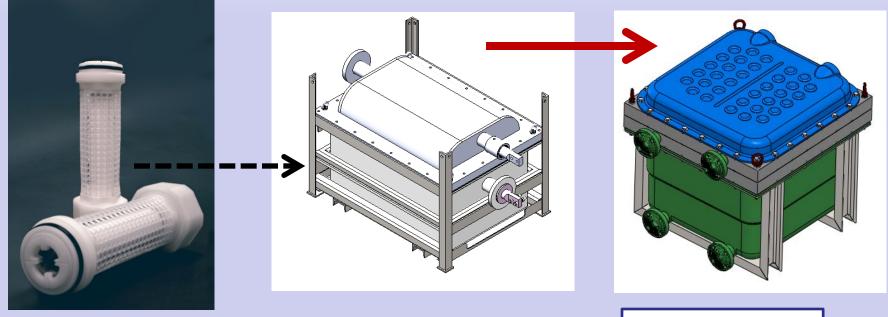
Design and manufacture **multi-Tubular Membrane-based Cell Unit** to increase processing through put.

MODULAR DESIGN: 50 LBS/DAY SMO UNIT

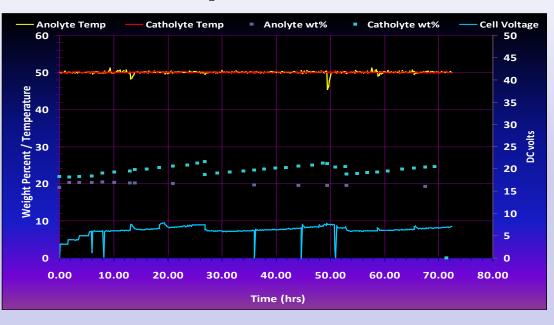
Membrane canister in unit

Generation-1 UNIT: LARGE footprint

Generation-2 UNIT: SMALL footprint

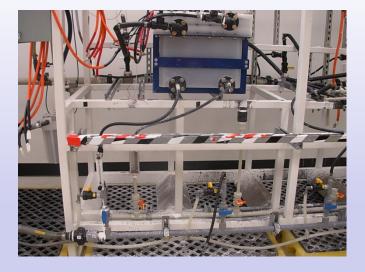


Generation 2 UNIT: Module Operational Results



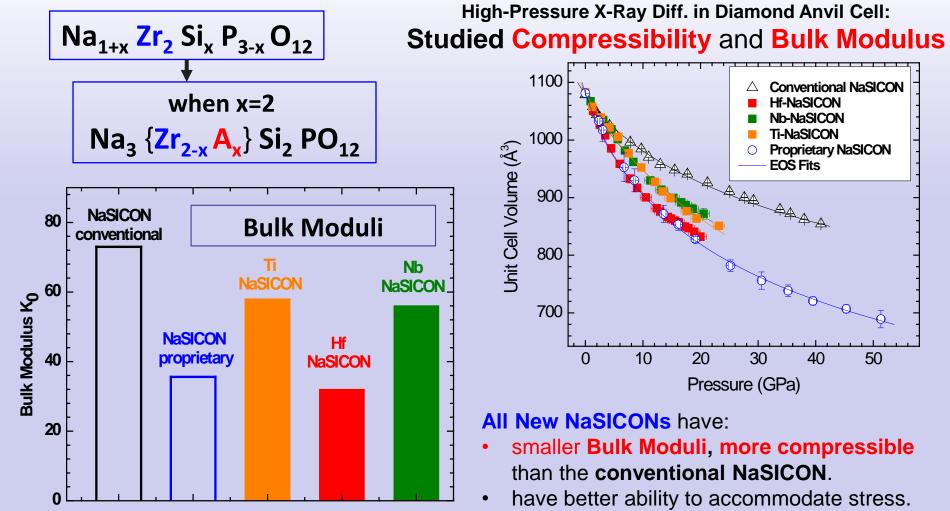
Operational Milestone:

- Demonstrated batch unit operation at 100 mA/cm² to produce SMO
- Cell voltage increase from allowing concentration of SMO to increase from 18 to 25 wt%.



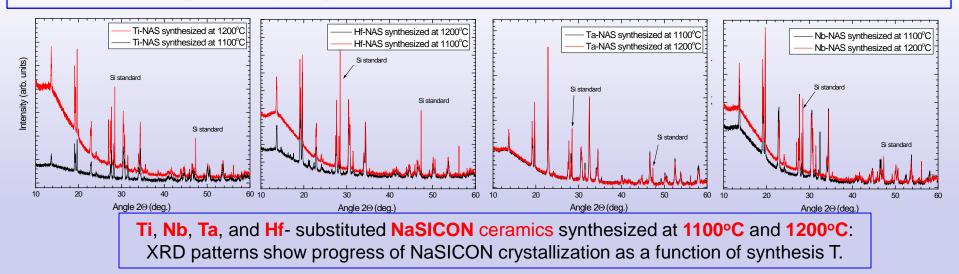


Fabricated New NaSICON compositions: single-doping in the Zr-site



Fabricated NaSICONs

Successfully Fabricated New NaSICON compositions single-doping in the Zr-site: Na₃ {Zr_{2-x} A_x} Si₂ PO₁₂



Partial Substitution of Zr-sites	Ionic Radius (Å)	Unit Cell Volume (Å ³)	Cell Param. <i>a</i> (Å)	Cell Param. <i>b</i> (Å)	Cell Param. c (Å)	Angle (deg.)
$Na_3Zr_2Si_2PO_{12}$	0.72	1084.72	15.64	9.04	9.22	123.74
Ti ⁴⁺	0.61	1082.79	15.62	9.04	9.21	123.65
Hf ⁴⁺	0.71	1081.56	15.62	9.03	9.22	123.75
Nb ⁵⁺	0.64	1080.33	15.61	9.02	9.22	123.63
Ta⁵⁺	0.64	1076.24	15.60	9.01	9.22	123.73

NaSICON ceramics synthesized by solid state reaction at 1100°C: unit cell volume- V and cell parameters- a, b, c in function of Zr substitution by Ti, Nb,Ta, Hf based on Rietveld structural refinements of XRD patterns.

Successfully Fabricated New NaSICON compositions by single-doping in the Zr-site: Na₃ {Zr_{2-x} A_x} Si₂ PO₁₂

Partial Substitution of Zr-sites	Synthesis Temperature: 1100°C			Synthesis Temperature: 1200°C			Hf-doped
	% NaSICON Phase	% Zirconia Secondary Phase	% Other	% NaSICON Phase	% Zirconia Secondary Phase	% Other	NaSICON is the best for further
Na ₃ Zr ₂ Si ₂ PO ₁₂ (baseline comp.)	89.3	10.7		90.7	9.3		development: ✓ lowest bulk
Ti ⁴⁺	91.8	4.9	3.3	73.7	6.8	19.5	modulus
Hf ⁴⁺	87.5	12.5		92.0	8.0		✓ largest % of
Nb ⁵⁺	86.0	3.8	10.2	84.1	14.4	1.5	NaSICON
Ta⁵⁺	59.1	14.1	26.8	69.5	9.5	21	phase

Comparison of NaSICON ceramics synthesized at 1100°C and 1200°C:

- quantity of NaSICON primary crystalline phase and
- quantity of Zirconia secondary crystalline phase as a function of Zr substitution by Ti, Nb, Ta, and Hf.

(All data based on Rietveld structural refinements of XRD patterns).

New Glass-Ceramic route for NaSICON synthesis

ROUTE

- First step: fabrication of NaSICON glass.
- Second step: controlled crystallization of NaSICON glass.
- Route alternative to solid state synthesis (=sintering of polycrystalline powders).

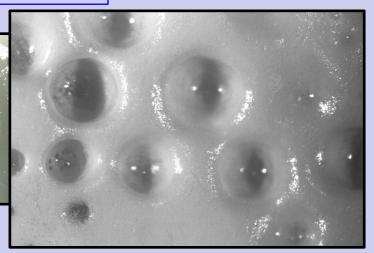
EXPECTED BENEFITS

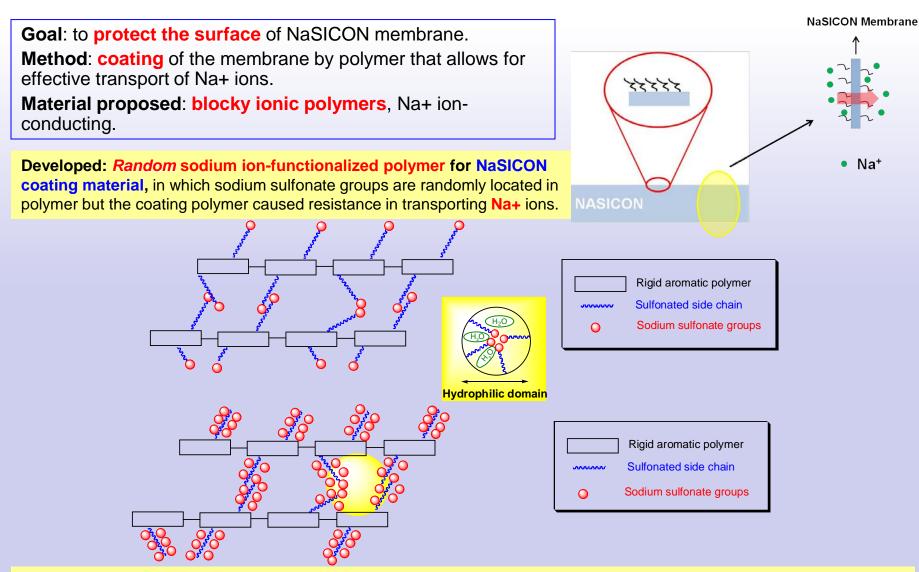
- High-density NaSICON ceramics.
- Smaller NaSICON ceramics grain size.
- Single-phase NaSICON ceramics.
- Lower synthesis temperature.
- Better electrical conductivity.

Developed New NaSICON compositions with double-atom doping in the Zr-site: $Na_3{Zr_{2-(x+y)}(A_x B_y) Si_2 PO_{12}}$

First Fabricated NaSICON melts – glasses



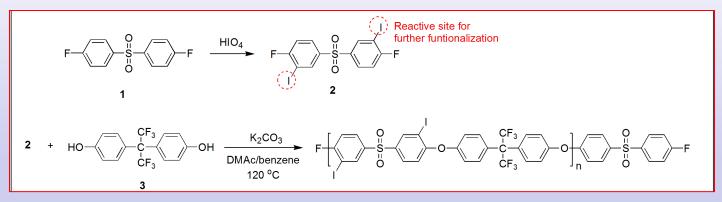




Developed: *Blocky* sodium ion-functionalized graft copolymer for more efficient Na⁺ transport. **Under Development**: *Blocky* copolymers where sodium sulfonate groups are more localized in nano-scale domains, creating more favorable morphology for better transport of Na+ ions in the coating material.

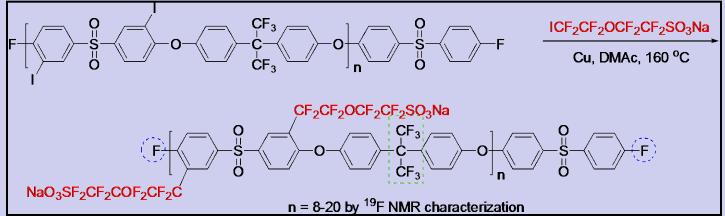
Graft Side Chain Precursors

- iodonated oligosulfones with different chain lengths in order to attach hydrophilic units to hydrophobic polysulfone chains.
- ooligosulfones with average molecular weight 6.1-17.3 kg/mol were prepared by adjusting the ratio of compounds 2 and 3 in polycondensation reaction.



Sodium Sulfonated Graft Side Chains

□ The iodine group in the oligomer was converted to sodium sulfonate groups using copper-catalyzed coupling reaction



Relevance to Advancement of Renewable Energy – 1

- **SMO** is an **effective catalyst to produce biofuels** by the transesterification of oils, animal fats, recycled greases, and oils from edible and non-edible feedstock.
- Commonly used methods to produce SMO are all energy intensive and environmentally hazardous.
- Onsite storage of **SMO** presents a safety hazard with stringent requirements for nitrogen blanketing for moisture control and basins for waste spill containment.
- This project **Develops**, **Demonstrates** and **Deploys** a **NEW**, inexpensive method to produce **SMO** at a desired concentration in methanol, on-site, and on-demand with very low concentration of impurities.
- Our process has a 21-26% lower carbon footprint than other methods based on lifecycle analysis.
- NaSICON fabrication route based on glass-ceramic technology will produce glassderived NaSICON with pore-free texture, very small grain size, and negligible grain boundary effects with reduced energy barrier for Na⁺ transport and enhanced the material's ionic conductivity.
- Successful polymer coating of NaSICON surface will allow manufacturing of thinner membranes and thereby reduce power consumption of the electrochemical process, promoting more energy-efficient and cost-effective SMO catalyzed biodiesel production.
- This novel technology **benefits Biofuels**' and other industries that use **SMO**.
- Once the maturity of the SMO technology is demonstrated at Biofuel facility, the plants can then independently integrate the SMO technology for processing non-edible oils feeds derived from plants (*ex. jatropha, algae..*)

Relevance to Advancement of Renewable Energy – 2

- NEW! Electrochemical Membrane Process (EMP) with NASICON membrane: high-purity SMO output from low-purity low-cost NaOH input + no hazardous waste generation.
- Over 400,000 MT (Measurement Ton = 40 cubic feet) of SMO is imported into the U.S. Main supplier is China.
- 26% less energy: when using the EMP to produce SMO compared to current fabrication methods.
- 40% less expensive, based on a detailed cost analysis: to produce SMO at 500 kg/day compared to current SMO fabrication methods.
- Extended lifetime, 14 months demonstrated, 24 months projected: EMP cell lifetime is surpassing expectations in continuous cell operation.
- On-site production of SMO catalyst using EMP projected: scalable process, reduces transportation costs, avoids inventory and shelf-life costs, and reduces safety hazards.
- Ultimately a "one-step process" or "zero inventory option" is envisioned by integrating the on-site SMO generation with the downstream Biodiesel process.

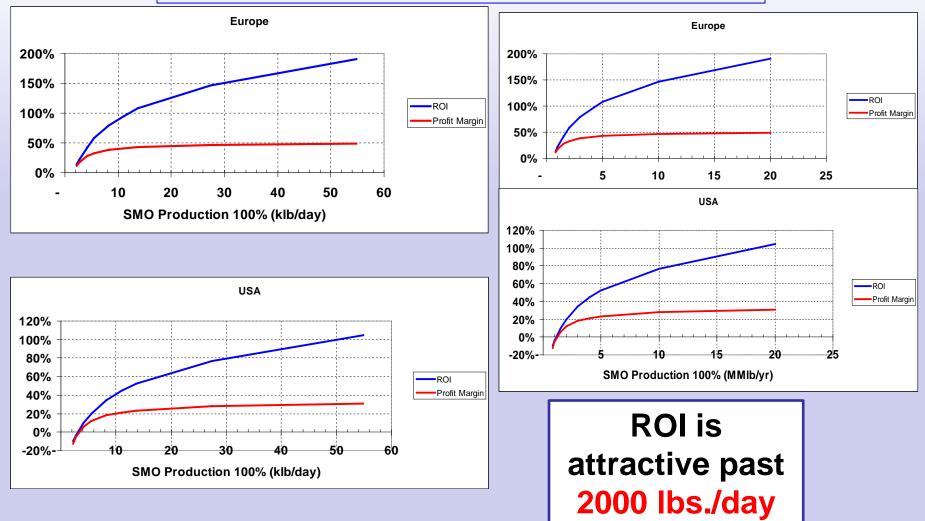
Critical Success Factors – 1

Techno-Economics: CAPEX and OPEC Costing for a 3300 lbs/day SMO of solids basis and higher system for payback target of < 2 years and return of investment of > 30%

- Replacement items:
 - NaSelect[™] membranes replacement : once in 24 months
 - Cell Components: electrode and plastics once in 5 yrs.
- Device Operating Conditions:
 - Current density: 100 mA/cm², Cell Voltage: 3.38 volts at 50°C
- Cost breakdown of equipment including cell lines (Tubular Membranes) and site services are lumped

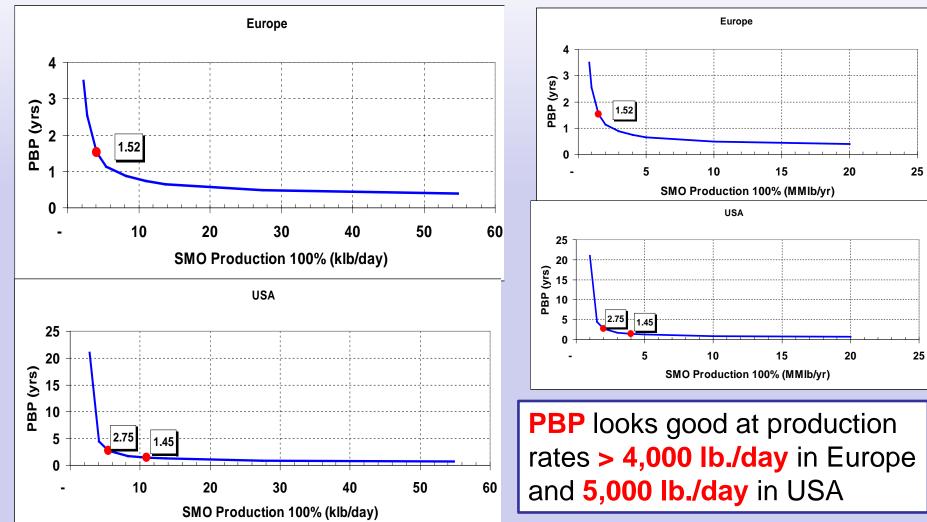
Critical Success Factors-2

Analysis ROI & Profit Margin



Critical Success Factors-3

Cost Analysis Pay Back Period (PBP)



Future Work

- Demonstrate up to 1000 hours of operation of Generation-2 Unit after design qualification to make sodium methoxide.
- Modify design according to lessons learned from the demonstration to maximize reliability and low-cost operation of Unit.
- Synthesize a final blocky ionic polymer, by connecting Na-sulfonated oligomer and non-sulfonated high-molecular polymer and use it for coating the surface of NaSICON ceramics.
- Evaluate the NaSICON fabrication route based on glass-ceramic technology to produce glass-derived NaSICON from selected new compositions in light of Na-conductivity and working lifetime in the Unit.
- Develop reliable and robust Unit to produce a maximum concentration of Sodium Methoxide-SMO with near zero leak in Unit system and meet product (SMO) requirements and moisture specification for product.
- Carry out Field Analysis of Technology at a **Biodiesel Facility** and complete a feasibility analysis for installation

SUMMARY

Approach:

 Developed and attained target specifications in an unique concept that uses an Electrochemical Process with a selective Sodium Transport NaSICON Membrane to produce SMO as well as other Alkali Alcoholates.

Technical accomplishments:

• Developed a low-cost process to synthetize **NaSelect**[™] ceramic material and produce large-sized Membranes that allow the scale up of devices.

Relevance:

 Successfully designed Prototype Cell to increase active working surface area of the Unit and developed multiple NaSelect[™] Membrane-based Unit configurations.

Critical Success factors and challenges:

 In the process of demonstrating Modular Cell-Unit capable to produce up to 50 lbs./day of SMO which is scalable to large production size.

Future Work:

• Long term operation of **Cell-Unit** planned in facility.

Technology Transfer:

• In discussion with Industrial customers to evaluate this technology in the field.

ABREVIATIONS USED

- NaSICON: ceramic membrane, sodium (Na) SuperIonic CONductor
- NaSelect[™]: proprietary composition of NaSICOn ceramic
- SMO: Sodium MetOxide
- EMP: Electrochemical Membrane Process
- DCP: Down Cell Process
- TDU: Technological Demonstration Unit

Publications, Presentations, and Commercialization

- **PATENTS**: Two **U.S. Patents** have been received for this technology
- PRESENTATIONS:
 - First Annual STEM Summit: Science, Technology, Engineering and Mathematics, Las Vegas, NV, Jan 14-15, 2013 Development of Biofuels Using Ionic Transfer Membranes; K. Lipinska *et al.*
 - 2011 DOE Biomass Program Review Integrated Biorefineries Platform IBR & Infrastructure, Washington D.C. Feb. 1-3, 2011: Development of Biofuels Using Ionic Transfer Membranes, Phase II; K. Lipinska, O. Hemmers, C. Bae and S. Balagopal
 - American Physical Society APS March Meeting, Dallas, TX, March 21-25, 2012: Synthesis, Microstructure and Properties of Complex Nasicon-Type Ceramics; K.
 Lipinska, O. Hemmers, J. Romin, P. Kalita, S. Sinogeikin, S. Balagopal, A. Nickens
- COMMERCIALIZATION: Currently in communication with three major industrial users in the field of Chemicals, Biodiesel and Pharmaceutical for evaluation of this technology.

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