



A Novel Flash Ironmaking Process

DE-EE0005751

American Iron and Steel Institute/University of Utah

09/01/2012 – 8/31/2015

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U.S. DOE Advanced Manufacturing Office Peer Review Meeting

Washington, D.C.

May 6-7, 2014





Project Objective

- Develop a new ironmaking process w/ significant reduction in energy consumption and CO₂ generation
- Blast furnace requires pelletization and/or sintering of iron ore concentrate
- Consumes large amounts of energy and carbon → CO₂ emissions
- Alternative ironmaking processes must have:
 - Large production capacities (e.g., ~1,000,000 tpy of iron)
 - Use the main raw material (i.e., iron ore) with minimal pretreatment



Technical Approach

Current practice

- Blast furnace (BF) produces >90% iron for steelmaking
- BF needs large capital investments
- High energy consumption in raw materials preparation and CO₂ emissions
- Use of special coals for cokemaking

New Approach – A Novel Flash Ironmaking Process

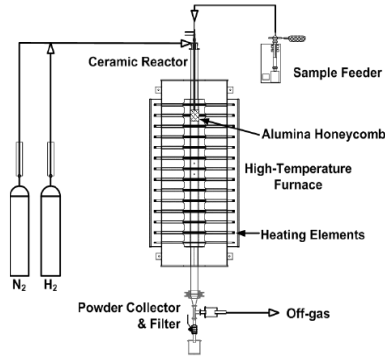


- Direct use of iron concentrate (~30 μm)
 - Bypass pelletizing and sintering
- Use of inexpensive, abundant natural gas [or hydrogen, coal gas]
 - No cokemaking required
 - Lower energy consumption
 - Less CO₂ emissions
- Rapid reaction rate and favorable Net Present Value (NPV)

Technical Approach

- Install, commission & conduct test on a new large bench reactor at the University of Utah
- Multidisciplinary team:
 - American Iron and Steel Institute
 - ArcelorMittal USA
 - The Timken Company
 - United States Steel Corporation
 - University of Utah (Lead Research Organization)
 - Berry Metal Company (Bench reactor fabrication)



Transition and Deployment

Project Objectives	Kinetic Feasibility Technology Road Map (2005-2007)	Proof of Concept at Lab Scale AISI CO ₂ Breakthrough (2008-2011)	Process Validation/ Scale-up Innovative Manufacturing Initiative (2012-2015)	Industrial Pilot TBD (2016+)
Experimental Apparatuses				<p>Approaches</p> <ol style="list-style-type: none"> 1. Large scale <u>75-100k tpy</u> 2. Modest-scale: <u>10-25k tpy</u> 3. Expand U of Utah work: <u>Similar to bench reactor but larger</u>
Funding	<p>Federal, \$350k Industry, \$150k Total, \$500k</p>	<p>Federal, \$ 0 Industry, \$ 4.8million Total, \$4.8million</p>	<p>Federal, \$ 7.1million Industry, \$ 1.8million Total, \$8.9million</p>	<p>\$10 – 75million Funding TBD</p>



Transition and Deployment

- Benefits steel users and steel-related industry
- U.S. Steel industry would be the end user
- To be used to produce iron as a raw material for steelmaking resulting in:
 - Direct use of iron ore concentrate
 - Low capital cost
 - Scalable to large capacities
 - Avoidance of cokemaking
- Commercialization through licensing & royalty
- Sustainable as a more energy efficient and green ironmaking step

Measure of Success

- If successful, iron will be produced at a lower cost, using less energy, and emitting less CO₂
- Potential energy savings: ~3.5 GJ/ton Fe vs. avg. BF
- CO₂ emission: Less than 36% vs. avg. BF process

Metric	H ₂ -based process	Reformerless natural gas process	Blast Furnace process
Energy Requirement (GJ/ton of hot metal)	11.3	14.5	18.0
CO ₂ emission (tons/ton of hot metal)	0.04	1.02	1.60

- NPV for standard case (15 year period): \$401M (2010)/(1 M tpy) Natural gas cost: \$5/M (2010) BTU HHV

Project Management & Budget

Task	Description	Milestones		
		Key Inputs	Criteria	Date
1	Bench Scale Reactor - Installation - Commissioning	Go/No Go Decision # 1:		1/31/2015
		Operating Temperature	1400°C	
		Solid feed rate	>1 kg/hr	
		Operation time	>6 hr	
2	Testing Program - Existing lab flash reactor - Drop-tube reactor - Bench reactor - CFD model	Go/No Go Decision # 2:		7/31/15
		Metallization	95%	
		Min. amt. reducing gas	3.0x	
		Go/No Go Decision # 3:		1/31/16
		Metallization	95%	
		Min. amt. reducing gas	1.5x	
Milestone # 4:		7/31/16		
Metallization	95%			
Solid feed rate		>5 kg/hr		
3	Industrial pilot reactor - Design - Cost estimate			
4	Program Administration	1/31/17		

Total Project Budget	
DOE Investment	\$7,120,000
Cost Share	\$1,780,000
Project Total	\$8,900,000

Results and Accomplishments

- Fabricated New bench reactor and ancillary equipment
- Designed/fabricated main burner – key component
- Prepared site for bench reactor installation

- Achieved 80-95% metallization in existing lab reactor*
- Fuel/reductant: Hydrogen
 - [Natural gas tests are planned]
- Reaction time: 4-6 seconds
- Temperature: ~1200°C
 - [less than 1300°C in bench reactor]



Existing Lab reactor at the University of Utah

**Different from New Bench Reactor with respect to the size and material of construction, which limits the operating temperature. Solid feed rate is limited (~ 0.5 kg/hr).*



Results and Accomplishments

- Measurement of reduction kinetics with natural gas using existing Lab reactor (2014)
- Computational Fluid Dynamic model development (2014-2016)
- Installation of new bench reactor (1Q15)
- Testing with new bench reactor (2015-2016)
- Industrial pilot plant design (2016)