

Develop Thermoelectric Technology for Automotive Waste Heat Recovery



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by

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Energy Efficiency Renewable Energy (EERE)
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Outline

- Background
 - Objective, partnering, ...
 - Motivation: fuel economy
- Technology Development
 - Subsystem modeling
 - Cost
 - Cost-effective materials
- Summary

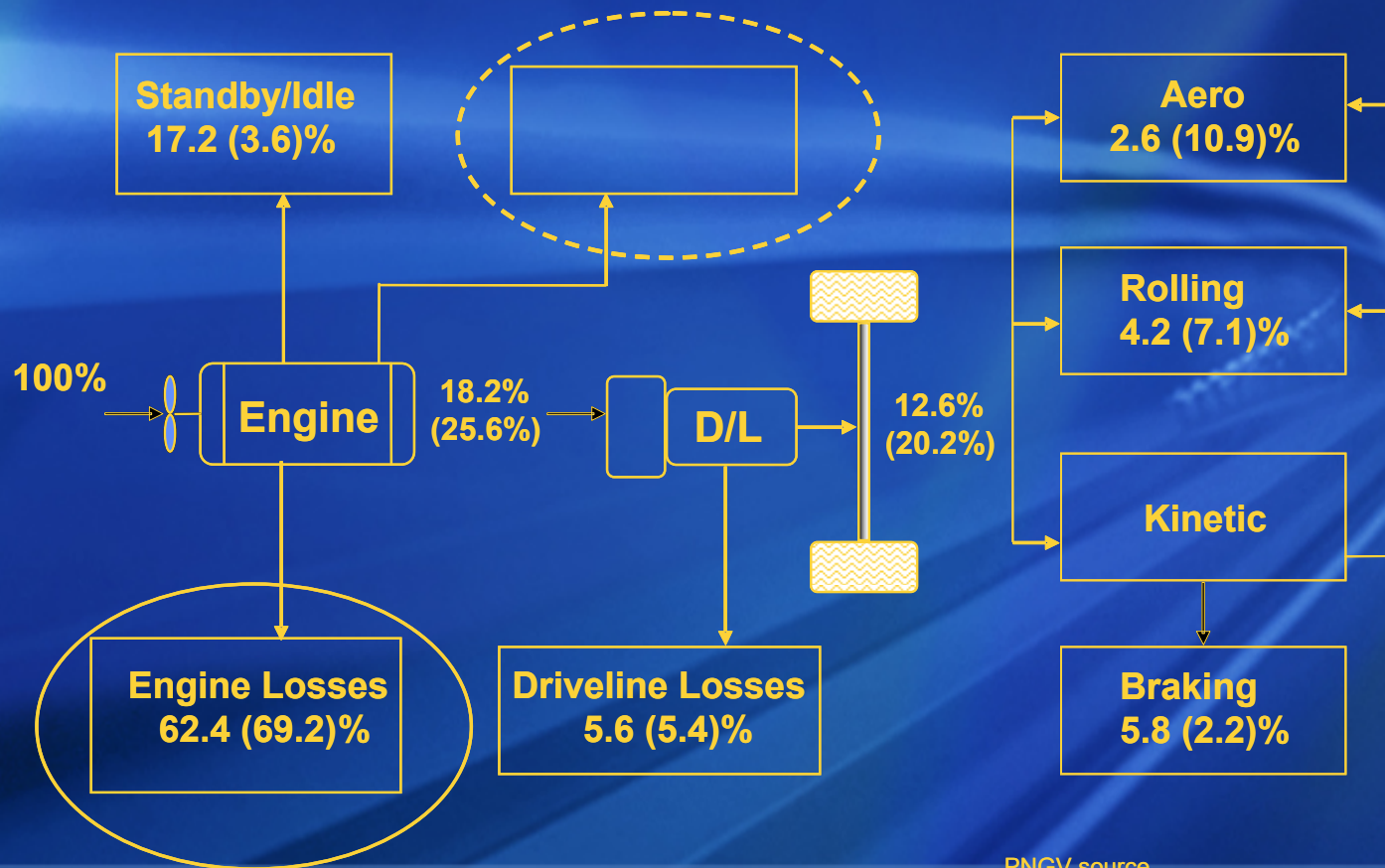
DOE Program Objective

Demonstrate a 10% fuel economy improvement using thermoelectric waste heat recovery technology, without increased emissions and at a cost-effective way.

Partnering

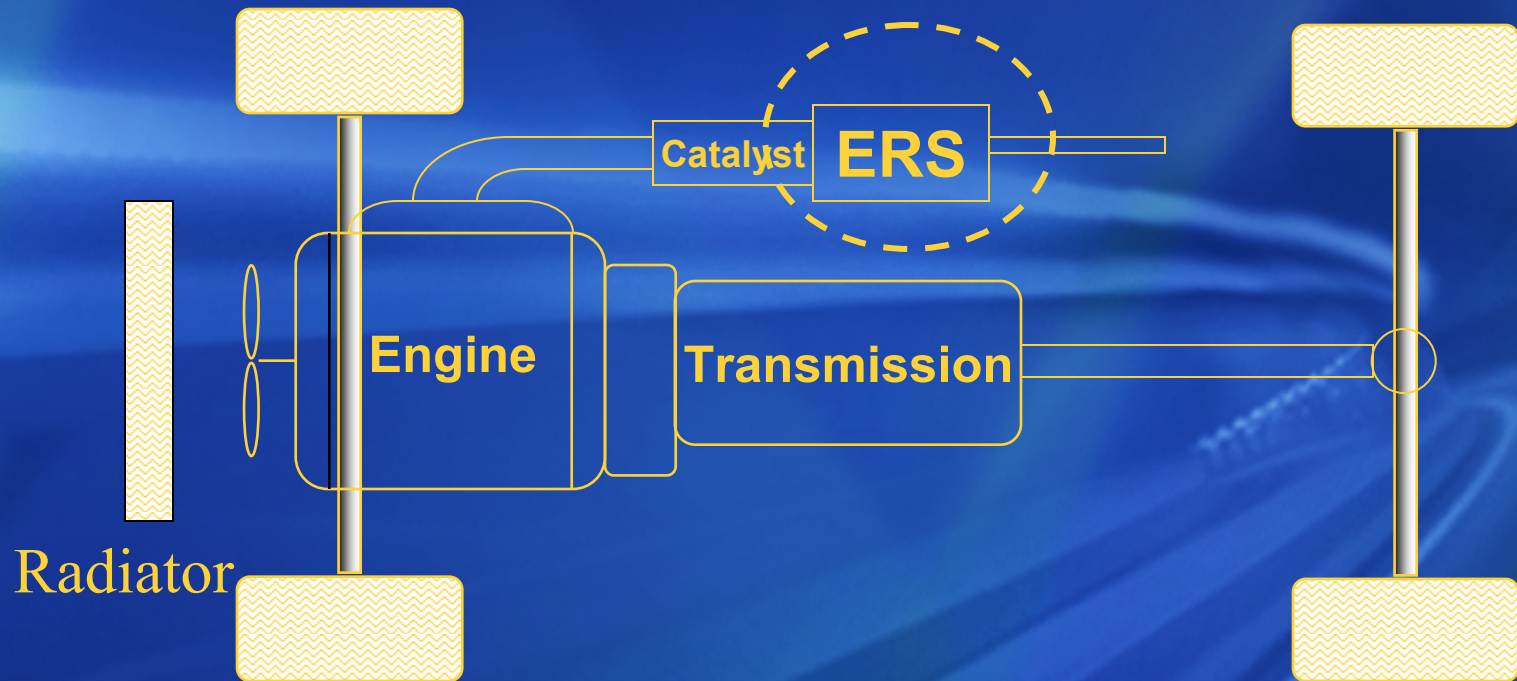
- **GM**
 - Materials Research
 - Subsystem design, integration, modeling, and validation
- **GE**
 - TE module design and construction
 - Subsystem design and construction
- **Oak Ridge National Lab**
 - High temperature materials properties measurement
- **RTI**
 - Superlattice-base materials and modules
- **University of South Florida**
 - bulk materials development: clathrates, nano-grain PbTe, ...
- **University of Michigan**
 - Bulk materials development, skutterudites, nano-composites,...

Energy Distribution – Typical Mid-Size Vehicle on Federal Test Procedure (FTP) - Urban (Highway) % energy use



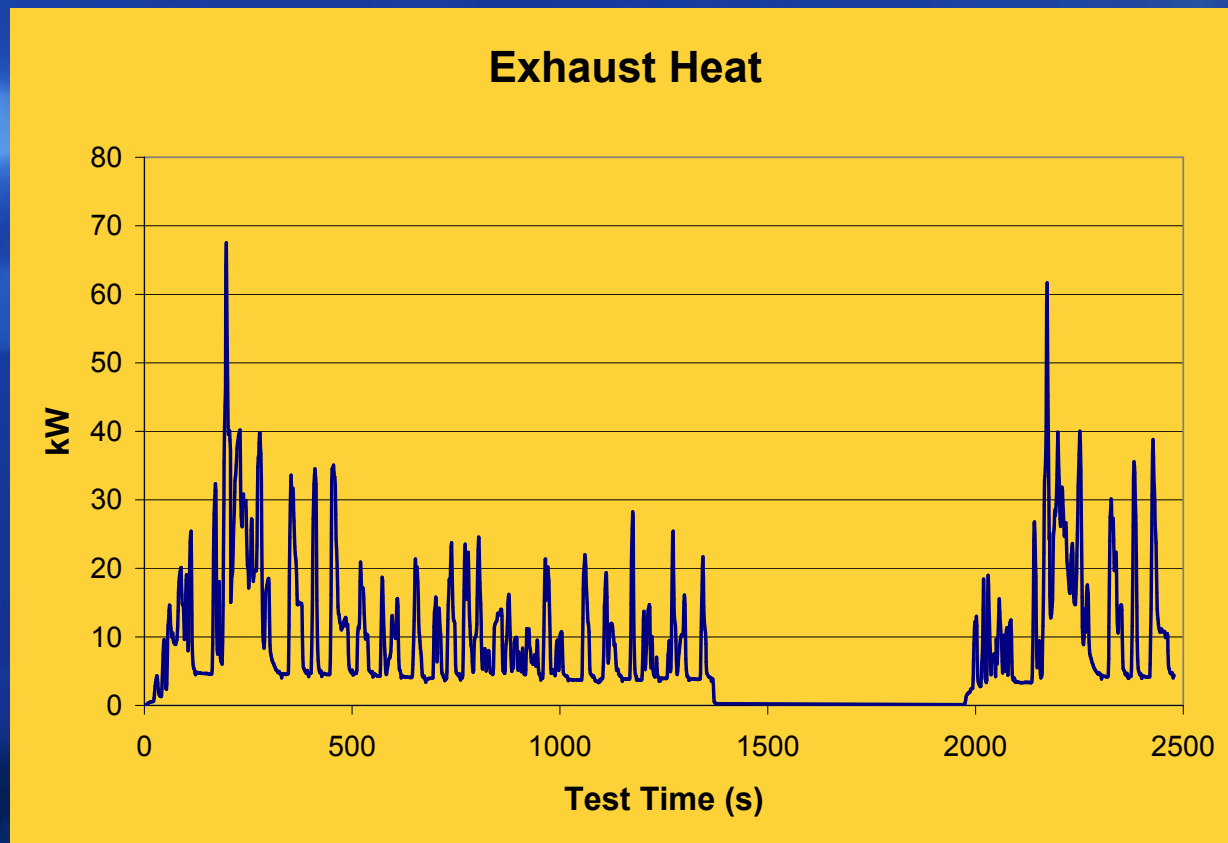
PNGV source

Example of an Energy Recovery System

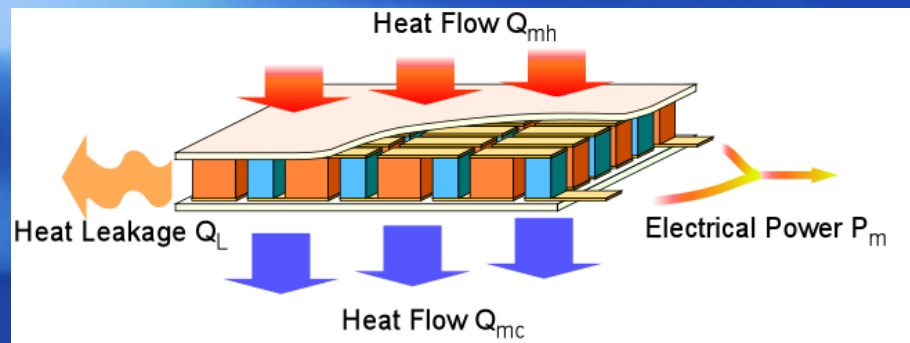


**Alternate: make the radiator
into an energy recovery device.
(smaller ΔT)**

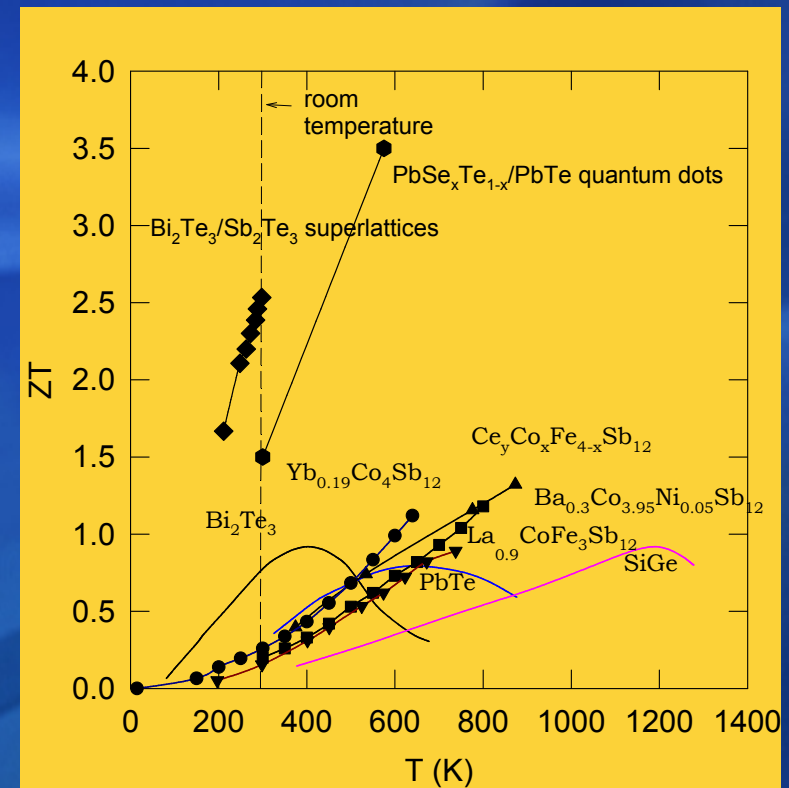
Exhaust as Potential TE Heat Source



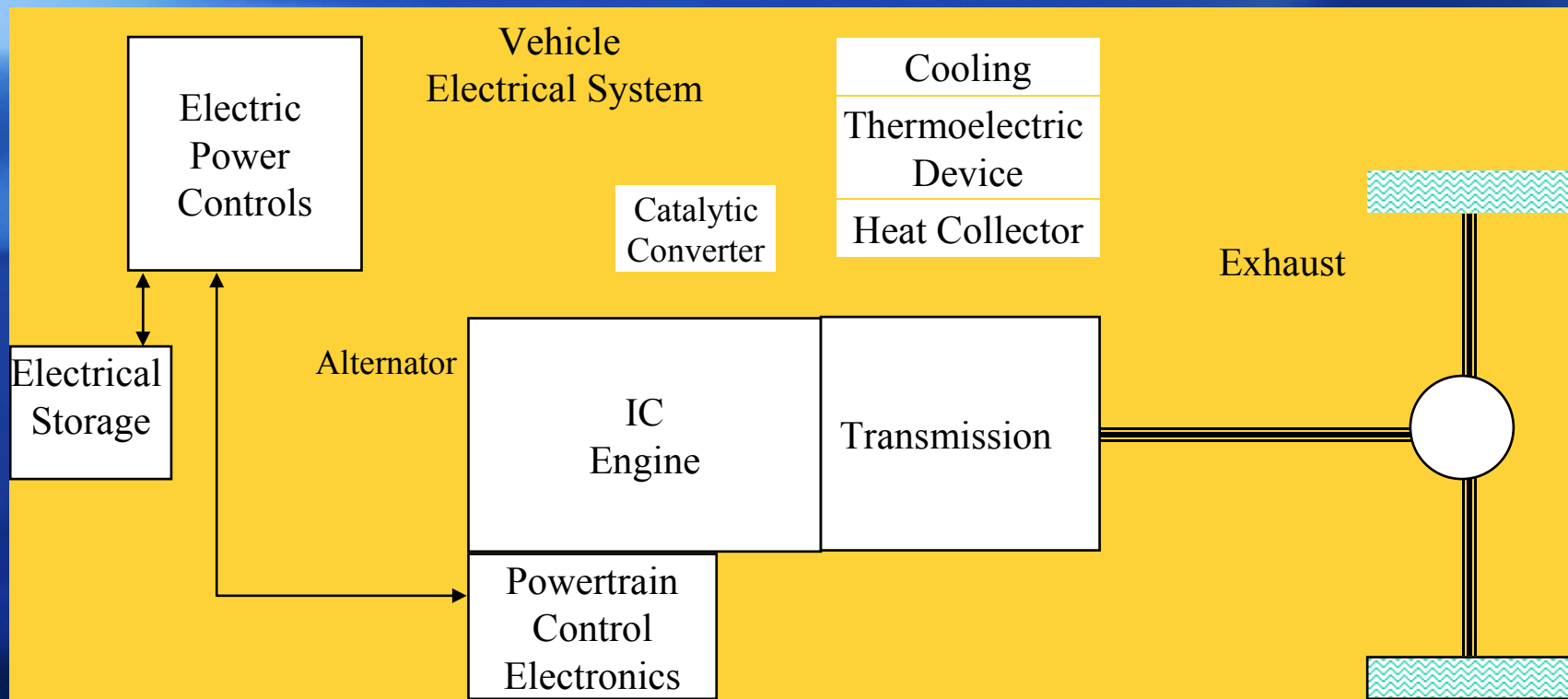
Thermoelectric Energy Conversion



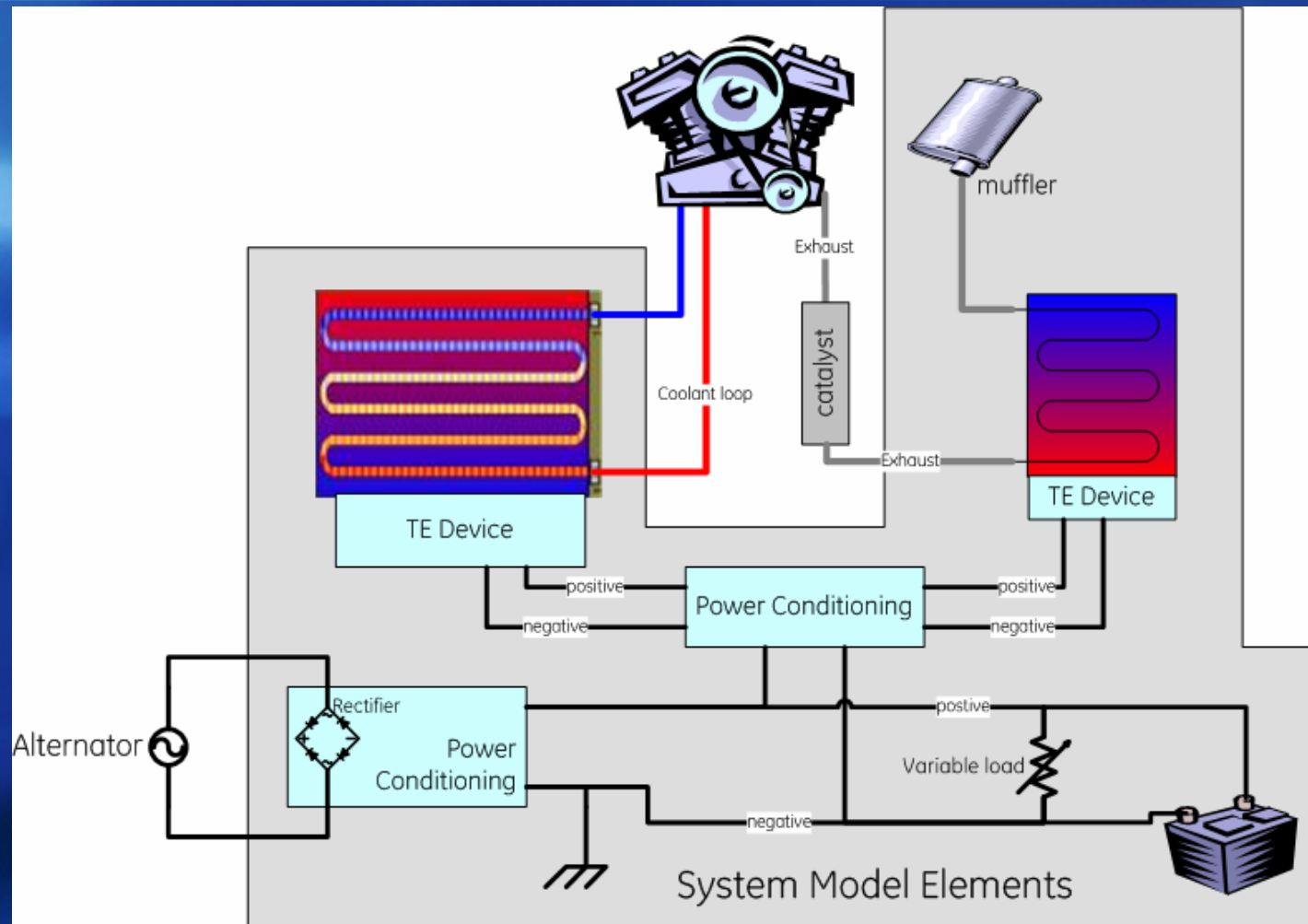
$$\text{Efficiency: } \varepsilon = \frac{T_H - T_C}{T_H} \frac{\sqrt{1 + ZT} - 1}{\sqrt{1 + ZT} + \frac{T_C}{T_H}}$$



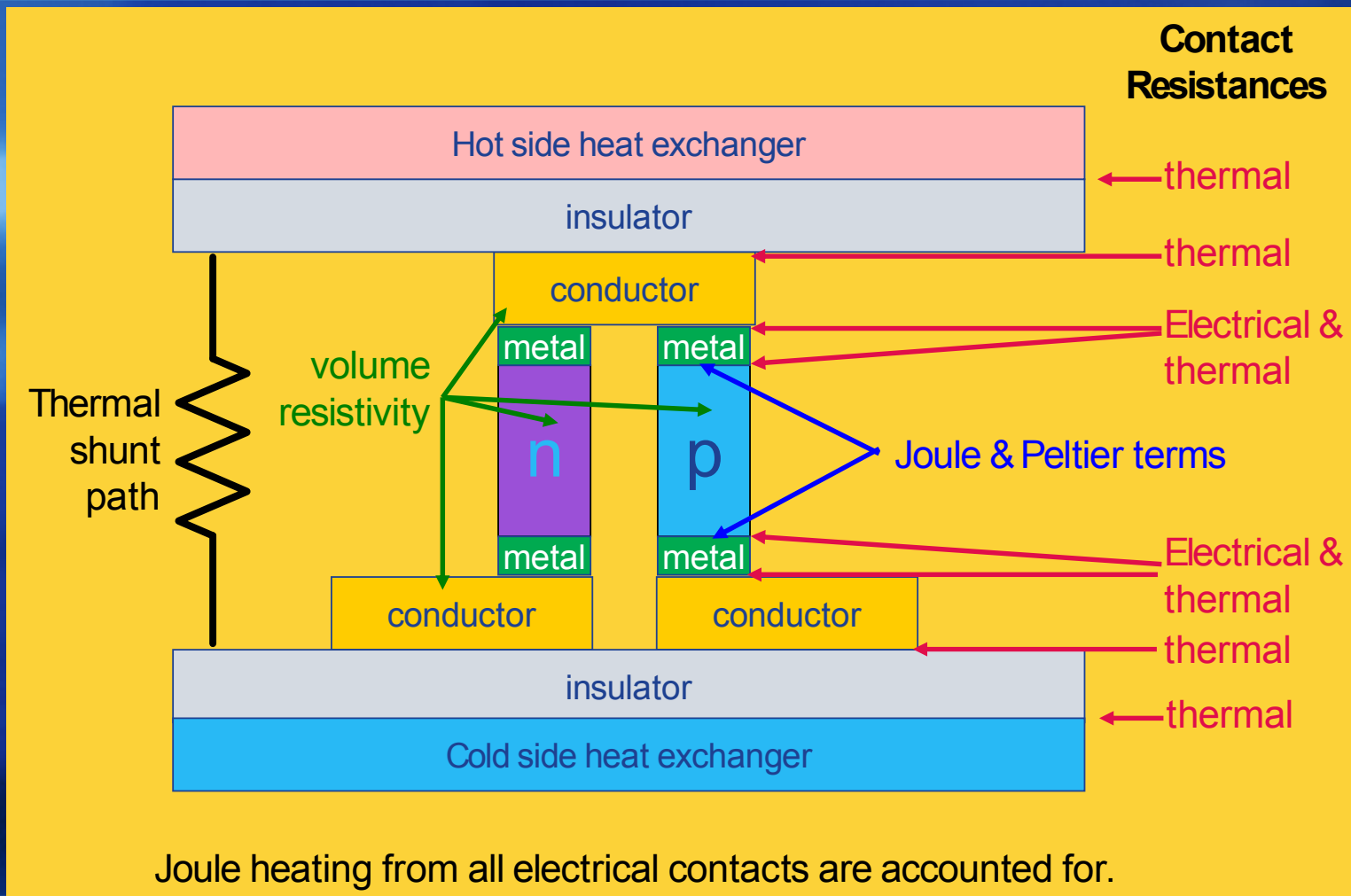
Generic Concept for a Thermoelectric Energy Recovery Augmented Electrical System



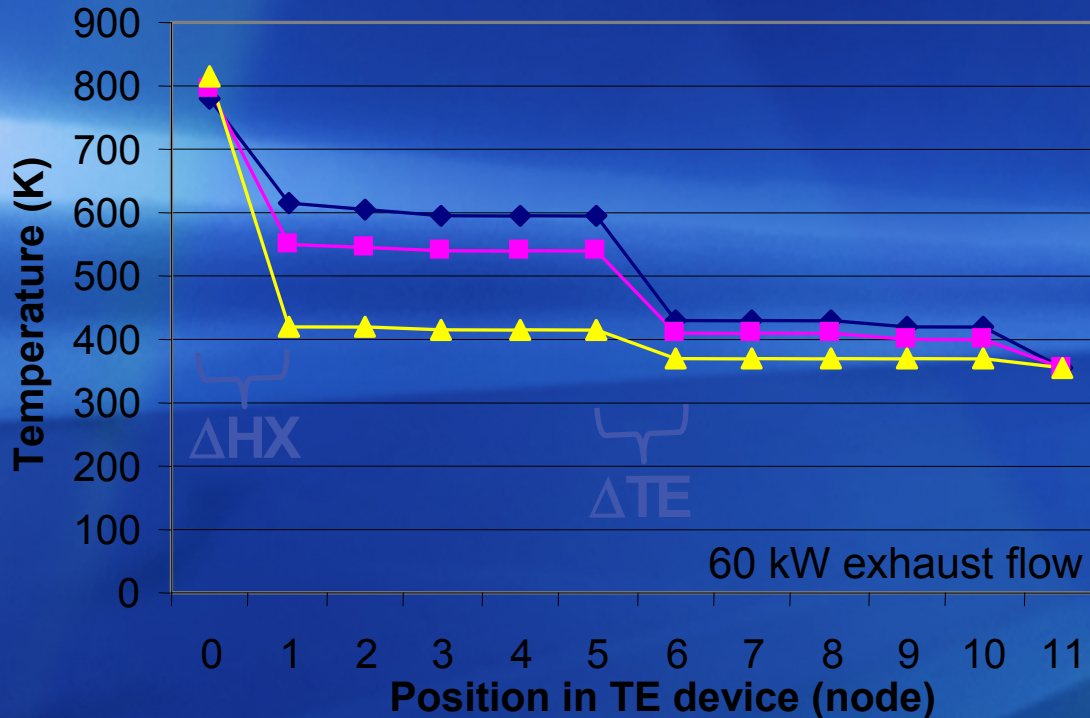
Subsystem Modeling



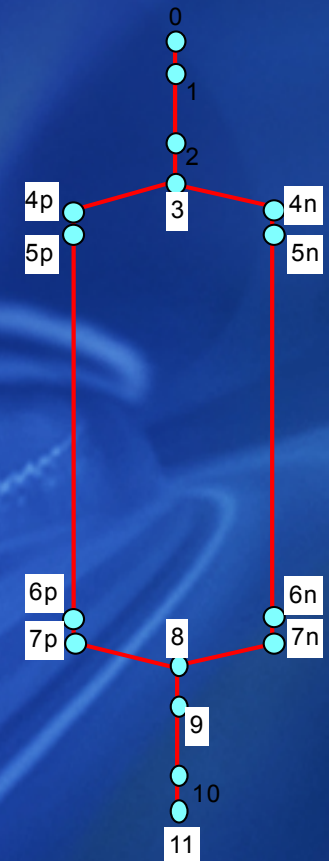
TE Module Modeling



Preliminary Temperature Profile



- ◆ 20%
- ◻ 15%
- ▲ 5%

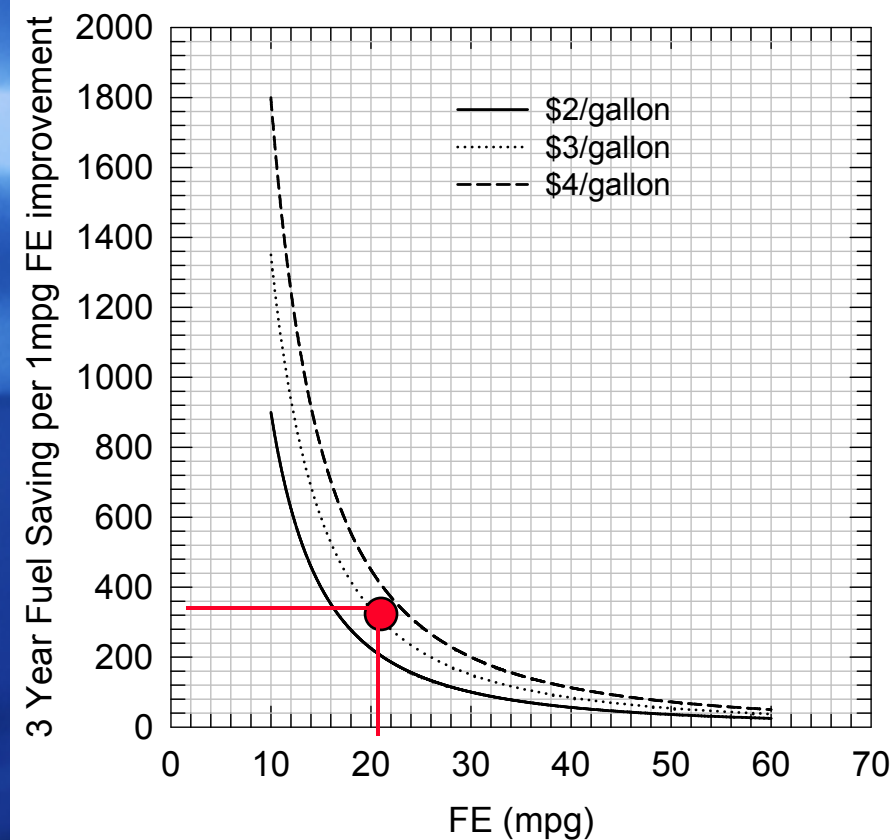


GM Validation of GE's subsystem Model with GEN I TE Generator



	Experiment	Model
Output Power	109W	121W
V_{load}	16.5V	17.4V
Current, I	6.6A	7.0A
Hot side temperature	183C	189C
Cold side temperature	35C	35C

On the Cost of Fuel Economy



- ❑ Cost of fuel economy - $\$/\Delta\text{mpg}$
 - This kind of calculation can be used to balance technology options
- ❑ $\$/\Delta\text{mpg} \leq \text{Savings}/\Delta\text{mpg}$ is necessary to provide consumer value

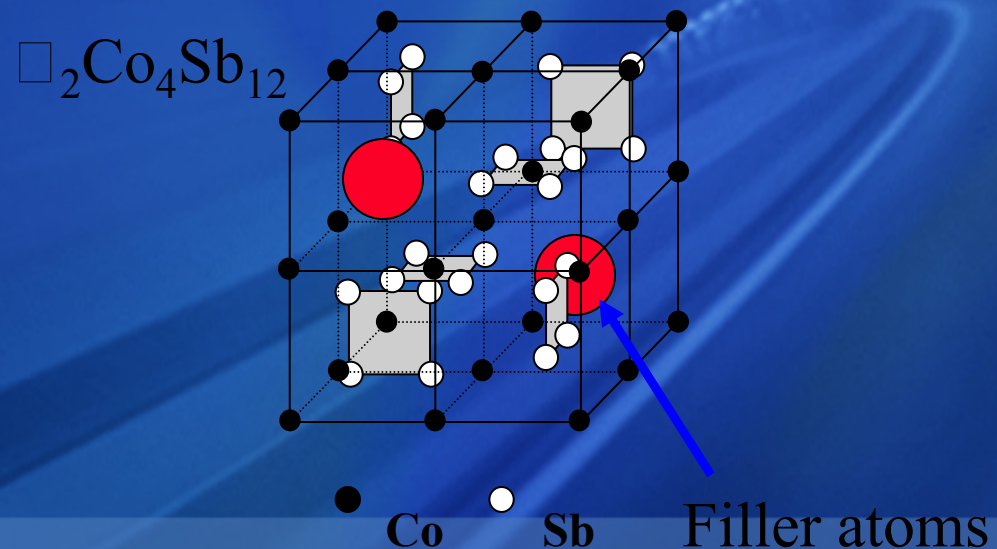
Consumer Fuel Savings/ $\Delta\text{mpg} \approx \$300\text{-}400/\Delta\text{mpg}$ (15000 mile/yr., 3yrs., 18-20 mpg)

Skutterudites Nano-101

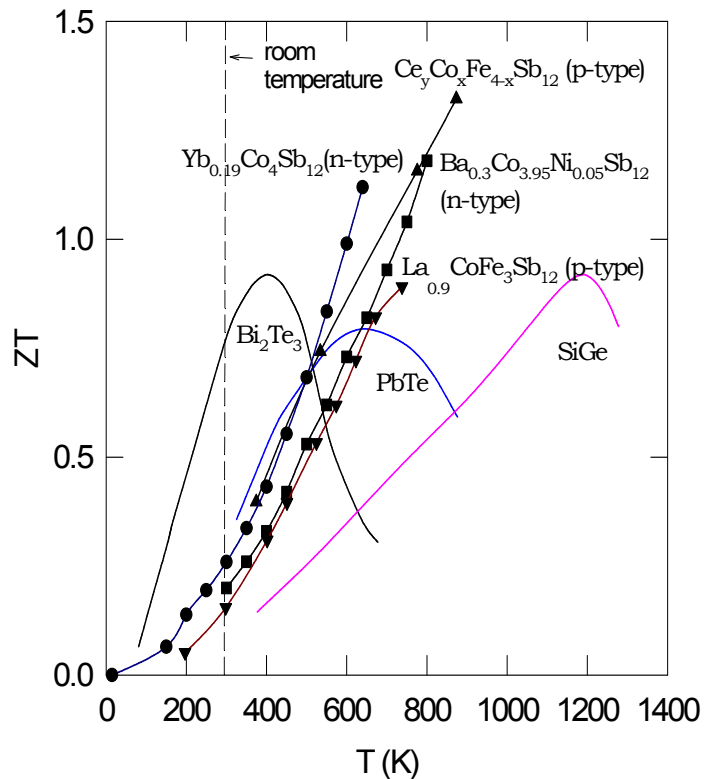
- ❑ CoAs₃ -based minerals found in region of Skutterud, Norway
- ❑ Compounds with the same crystal structure are known as “skutterudites”
- ❑ Filled skutterudites are electron-crystal-phonon-glass materials



$$Z = \frac{S^2}{\kappa_T \rho} = \frac{S^2}{(\kappa_L + \kappa_e) \rho}$$



Filled skutterudites as Candidate Materials for Exhaust Heat Recovery



- High ZT values near the exhaust temperature
- Both high performance n- and p-type exist – suitable for TE module

Mischmetal

- ❑ Mischmetal: from German - “mixed metals”; a naturally occurring alloy of rare earth elements
 - ❑ Composition (wt%):
Ce : 50-55 La : 30-35 Nd : 5-10 Pr : 5-10
 - ❑ Cost Comparison*:

Mischmetal (99.0%):	\$0.19/g
Cerium rod (99.9%):	\$8.37/g
Lanthanum rod (99.9%):	\$6.40/g
- * source: Alfa Aesar

Mischmetal Starting Material

electron probe
microanalysis

Element	Wt. %
Ce	52.4
La	23.5
Nd	17.1
Pr	5.9
Si	0.6
Fe	0.3
Al	0.1
Sum	99.9

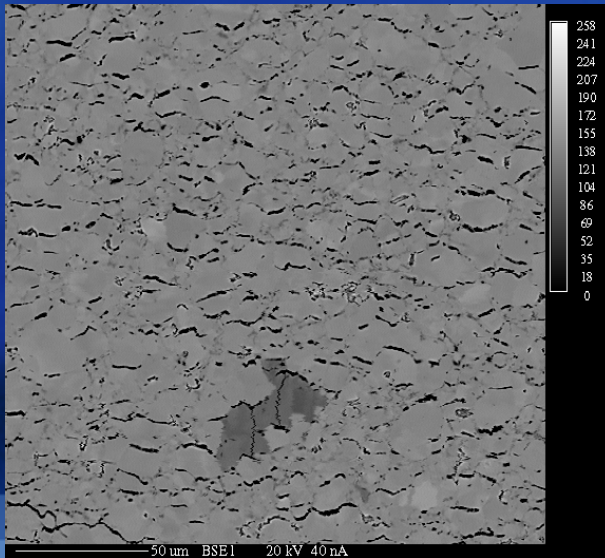
wet chemistry

%

Element	Ce	La	Nd	Pr
Sample	53	23	18	5

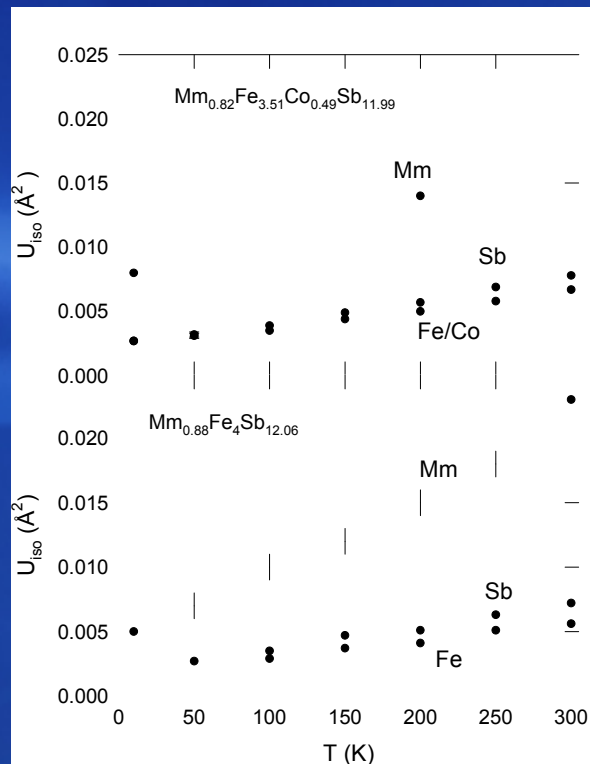
Electron Probe Microanalysis of Mischmetal-filled Skutterudites

Nominal composition	Actual composition	Room temperature lattice parameter (Å)
$\text{Mm}_{0.71}\text{Fe}_{2.5}\text{Co}_{1.5}\text{Sb}_{12}$	$\text{Mm}_{0.55}\text{Fe}_{2.44}\text{Co}_{1.56}\text{Sb}_{11.96}$	9.109(1)
$\text{Mm}_{0.82}\text{Fe}_3\text{CoSb}_{12}$	$\text{Mm}_{0.65}\text{Fe}_{2.92}\text{Co}_{1.08}\text{Sb}_{11.98}$	9.117(1)
$\text{Mm}_{0.93}\text{Fe}_{3.5}\text{Co}_{0.5}\text{Sb}_{12}$	$\text{Mm}_{0.72}\text{Fe}_{3.43}\text{Co}_{0.57}\text{Sb}_{11.97}$	9.126(1)
$\text{MmFe}_4\text{Sb}_{12}$	$\text{Mm}_{0.82}\text{Fe}_4\text{Sb}_{11.96}$	9.146(1)
$\text{Mm}_{0.93}\text{Fe}_{3.5}\text{Co}_{0.5}\text{Sb}_{12}^*$	$\text{Mm}_{0.82}\text{Fe}_{3.51}\text{Co}_{0.49}\text{Sb}_{11.99}$	9.1294(2)
$\text{MmFe}_4\text{Sb}_{12}^*$	$\text{Mm}_{0.88}\text{Fe}_4\text{Sb}_{12.06}$	9.1433(2)



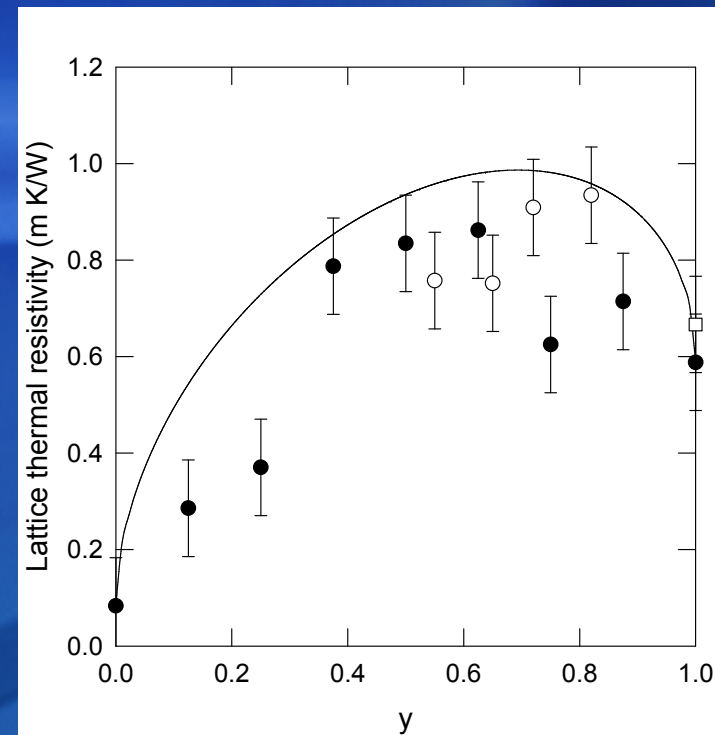
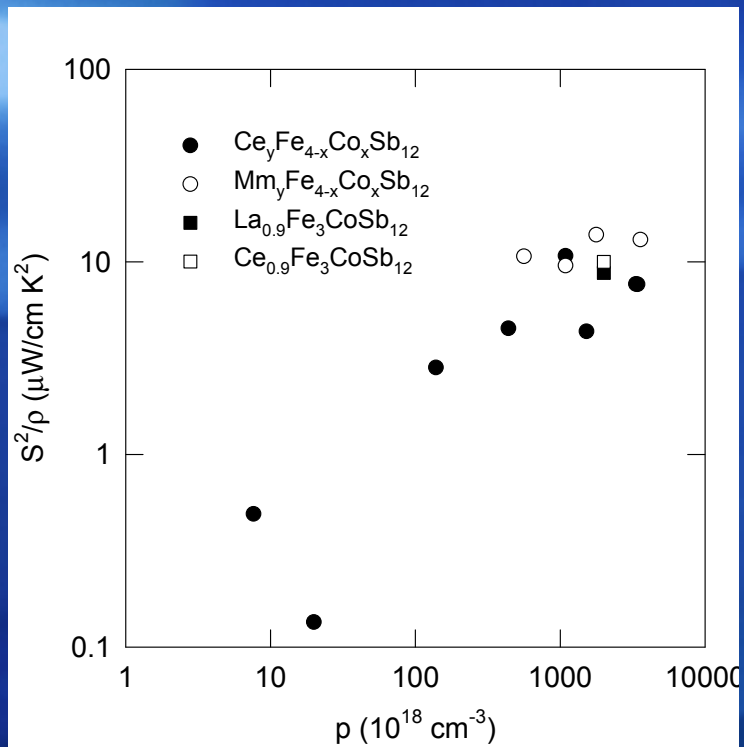
- ❑ nominal comp. chosen according to the optimal filled skutterudites $\text{Ce}_y\text{Fe}_{4-x}\text{Co}_x\text{Sb}_{12} \sim \text{PRL } 80, 3551 (1998)$
- ❑ Mm actual concentration < nominal due to high Mm vapor pressure
- ❑ all samples are nearly phase pure
- ❑ typical secondary phase: MnSb_2 or $\text{CoSb}_2 < 1 \text{ vol. } \%$

Isotropic Atomic Displacement Parameter – Evidence of Rattling



- low T intercepts of U_{iso} represent a combination of zero-point vibration and static disorder at the corresponding crystallographic sites
- zero-point vibration $\propto 1/M$, expected contribution $\sim Mn < Sb < Fe \rightarrow$ static disorder at the Sb sites, more so at the Mn sites (Mn vs. □)
- $\theta_E \sim 71.5 K$ (La: 80 K, Ce: 78 K, Eu: 83 K, and Yb: 65 K)

Room Temperature Properties Comparison



Summary of Challenges for Automotive TE Waste Heat Recovery

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- Motivation: fuel economy

□ Technology Development

- Subsystem modeling
- Cost
- Cost-effective materials

□ Summary

Need better materials, heat exchanger, contact joining, thermal & mechanical stability, ...

Consumer focus: consider $\$/\Delta\text{mpg}$ (also valuable for balancing tech. options)

Low cost materials is a must

Thank you ! Questions ?