



“The Utility of FeVO_4 in Combination with Stabilized Titanias for Mobile SCR Application”

D. M. Chapman, M. B. Watson, Cristal
K. Schermanz, Treibacher Industrie AG

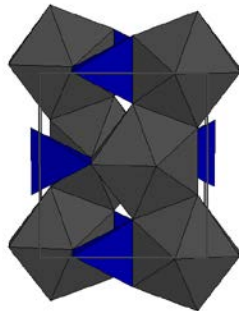
Brilliance inspired by titanium

Background

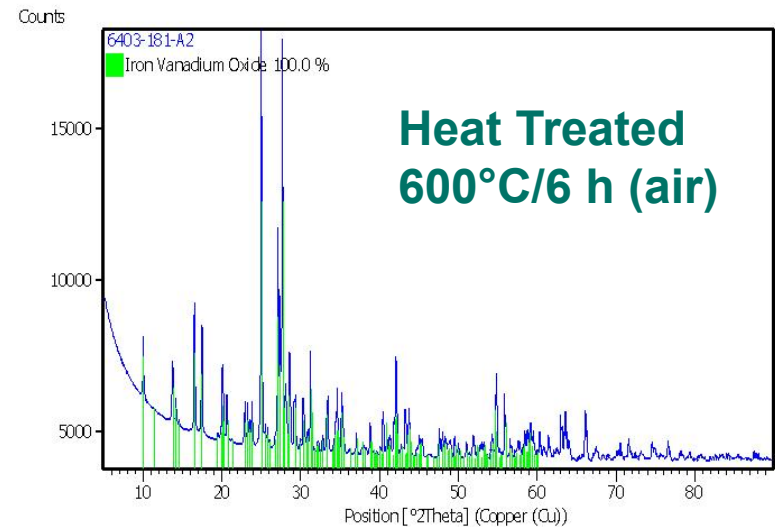
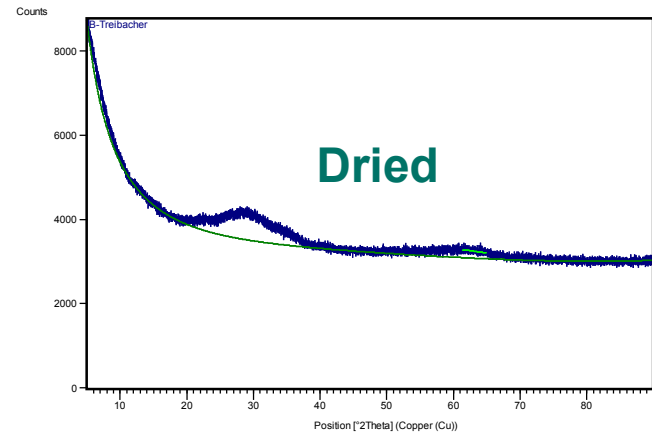
- Cristal have introduced stabilized titanias for vanadia-based mobile source SCR applications
 - Chapman, D. et al., *SAE Int. J. Fuels Lubr.*, 3(1):643-653, 2010
 - US 2011/0027154A1; US 2011/0138789A1
 - ***The new materials exhibit an attractive combination of activity and stability***
 - Treibacher have introduced transition metal vanadates as raw materials for SCR applications
 - US 2011/015073A1
 - SAE 2011-01-1331
 - ***FeVO₄ is not listed as hazardous by GHS (other vanadia precursors are)***
- Objective: evaluate stabilized titanias in combination with FeVO₄

Structure of FeVO_4

- Two samples provided
 - Dried, calcined
- Dried material is amorphous
- Calcined material is crystalline FeVO_4
 - Triclinic: P1
 - Fe^{3+} , V^{5+} valence
 - VO_4 tetrahedral
 - Not considered active form of V for SCR



XRD Powder Patterns

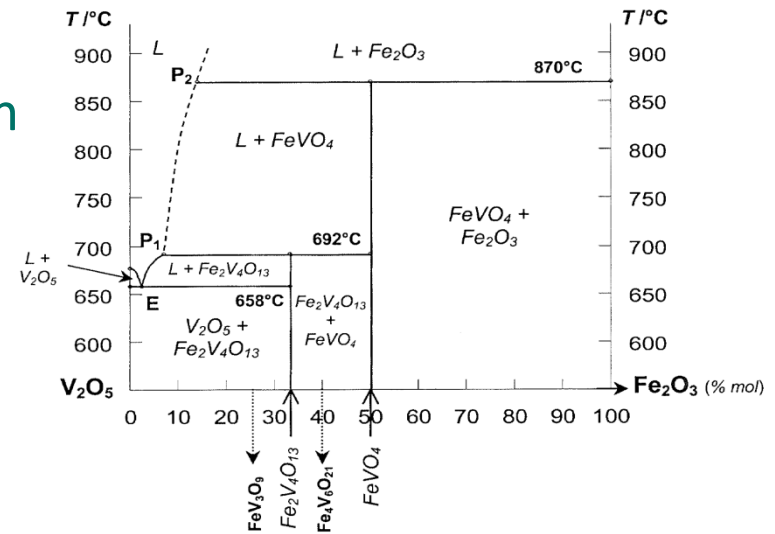


Phase and SCR Behavior of FeVO_4 -Synergy with TiO_2

P. Poizat et al. / C. R. Chimie 6 (2003) 125–134

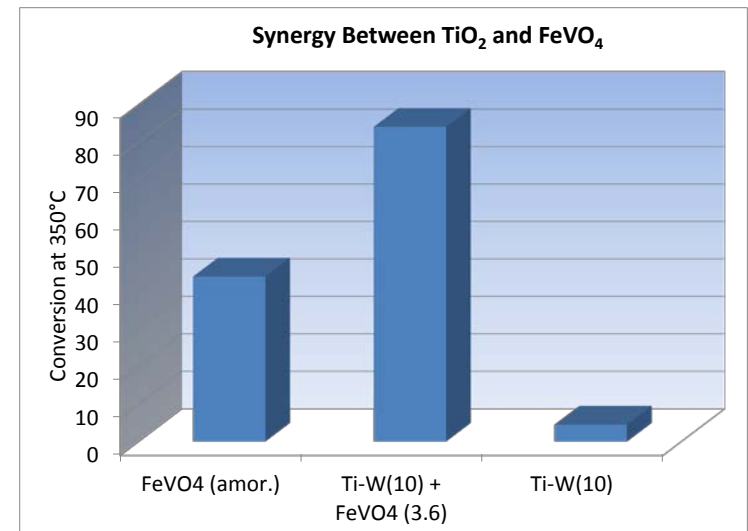
127

- FeVO_4 incongruent m.p. 870°C
- At $T < 692^\circ\text{C}$ and excess V, solid solution
- D.T.A. for pure FeVO_4 - no features $< 800^\circ\text{C}$
- D.T.A. for $\text{FeVO}_4 + \text{TiO}_2$: no unusual features $< 800^\circ\text{C}$
- Pure materials not very active for SCR
- $\text{FeVO}_4 + \text{TiO}_2$ (physical mix plus heat) is very active for SCR
- Amorphous FeVO_4 is more active than crystalline form



Equivalent Concentrations

wt%		
V_2O_5	V	FeVO_4
1.5	0.8	2.8
2.0	1.1	3.8
2.5	1.4	4.7

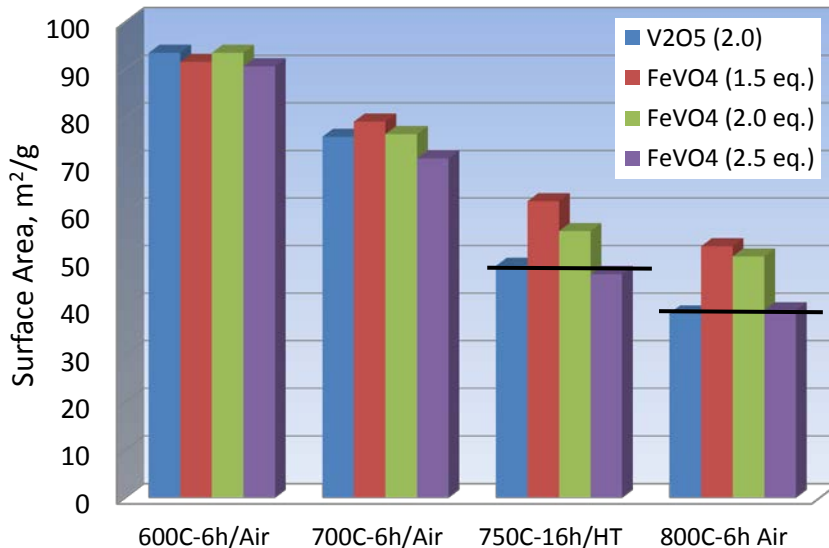


Screening Studies

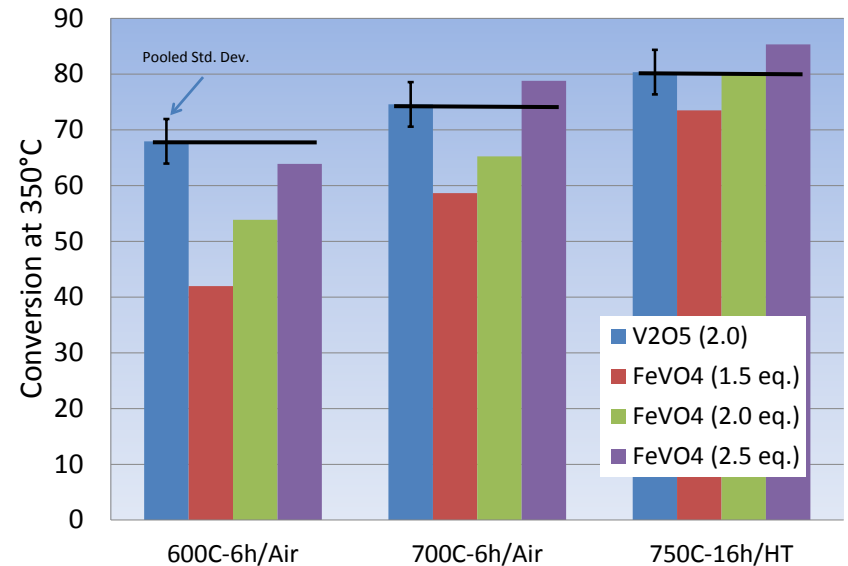
FeVO₄ Level and Thermal History

- Commercial Ti-Si(10)-W(9)
- Amorphous FeVO₄
 - Powders slurried, sonicated, dried and heat-treated
 - “FeVO₄ (2.0 eq.)” corresponds to 2.0 wt% V₂O₅
- DeNO_x 650 L/(h*g); 500 ppm NO, $\alpha = 1.0$

Effect of FeVO₄ Level on Surface Area



Effect of FeVO₄ Level on Conversion at 350°C

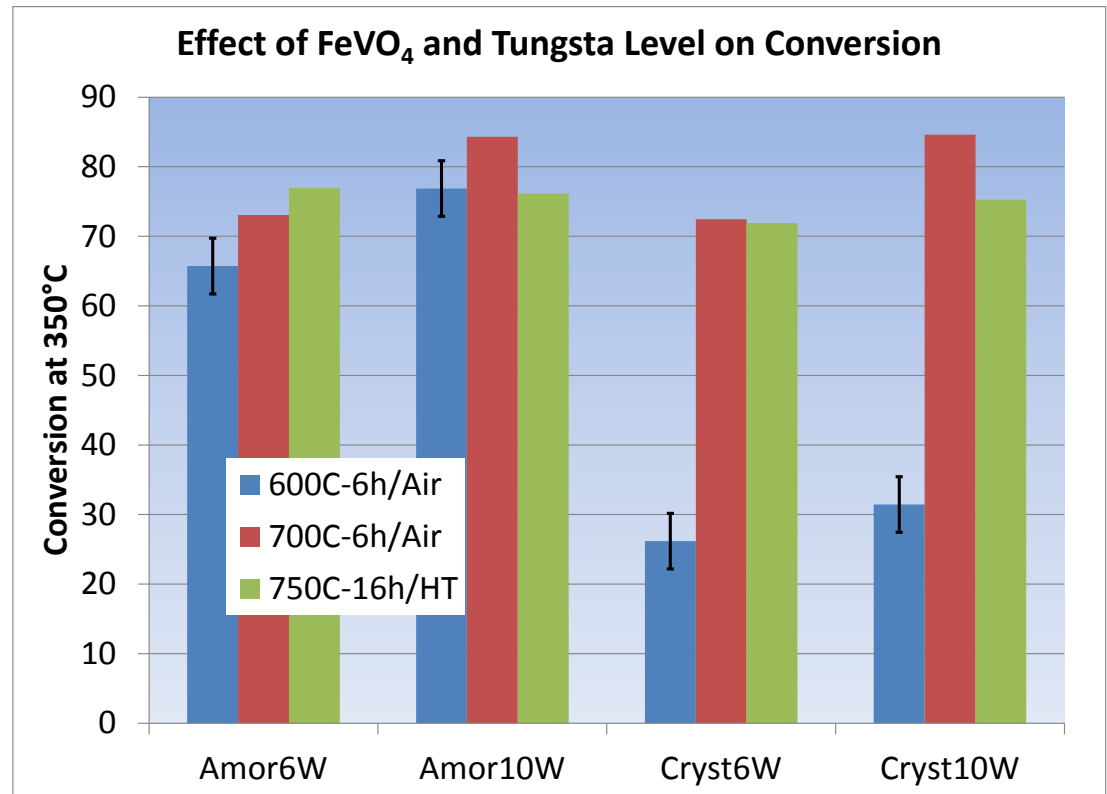


- Similar performance
 - Activity and stability benefit for FeVO₄ after severe aging

Screening Studies

FeVO₄ Source, Tungsta Level and Thermal History

- *Developmental Ti-Si(4.5)*
- Amorphous or crystalline FeVO₄
 - Powders slurried, sonicated, dried and heat-treated
 - 3.2 wt% FeVO₄ (1.7 eq.)
- DeNO_x 650 L/(h*g); 500 ppm NO, $\alpha = 1.0$



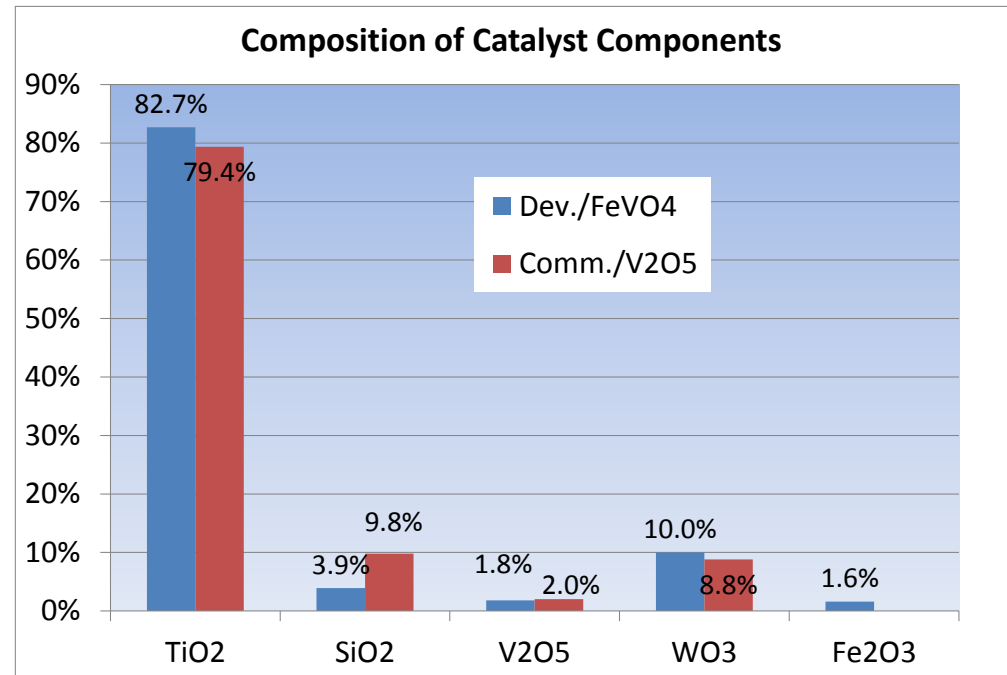
- *Amorphous FeVO₄ gives highest activity at lowest activation T*
- *Higher tungsta gives higher activity*

IRMA Study

Catalytic Evaluation of Wash-coated Parts



- *Test: Dev. Ti-Si(4.5) + WO₃ + FeVO₄*
 - Nominal 3.4% FeVO₄
- *Ref: Comm. Ti-Si(10)-9(W) + V₂O₅*
 - Nominal 2.0% V₂O₅
- 80% Catalyst, 20% binder
 - Nominal 150 g/L loading
 - 2" l. x 1" dia. cores
- Fast SCR (NO, NO₂ = 200 ppm; NH₃ = 400 ppm)
- Std. SCR (NO = NH₃ = 400 ppm)
- GHSV = 30,000 h⁻¹
- Fresh: Calcined 500°C
- Age: 700°C, 10 h, 10% H₂O

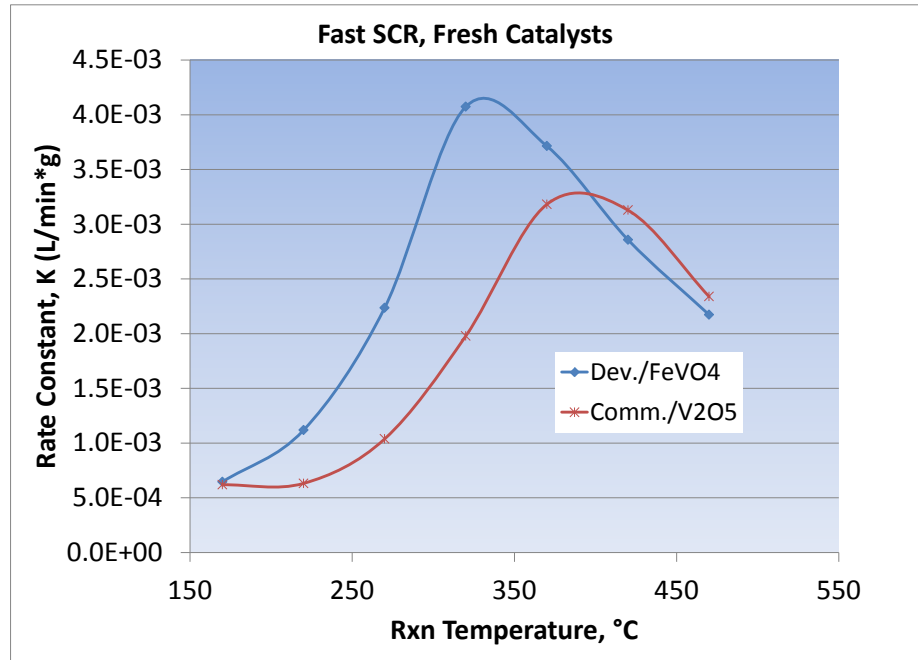
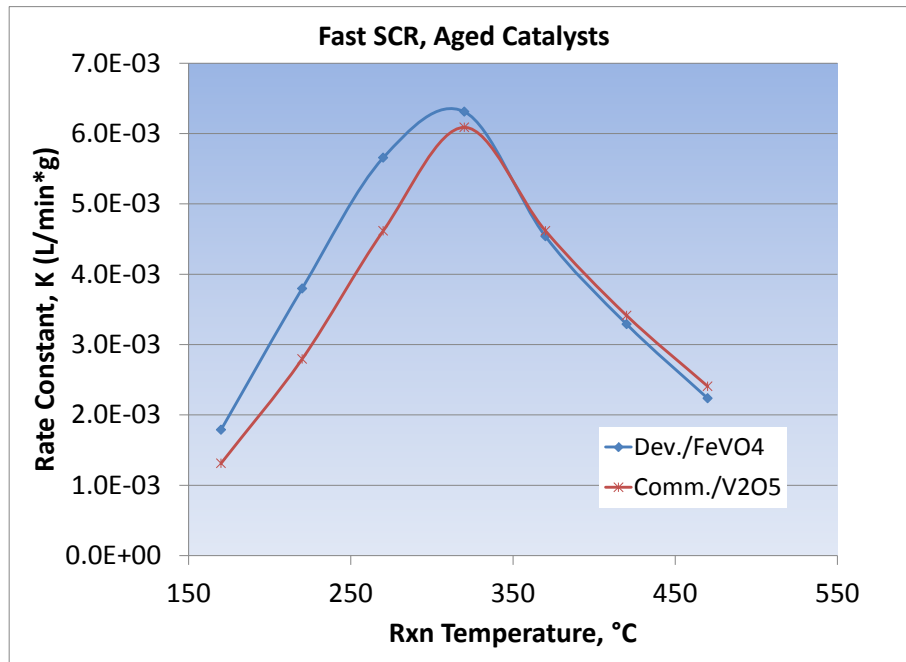


IRMA Study Fast SCR



- Rate constants calculated from conversion:

$$K = -((\text{total flow} * \text{NO conc}) / \text{cat mass}) * \text{LN}(1 - \text{conv}/100)$$



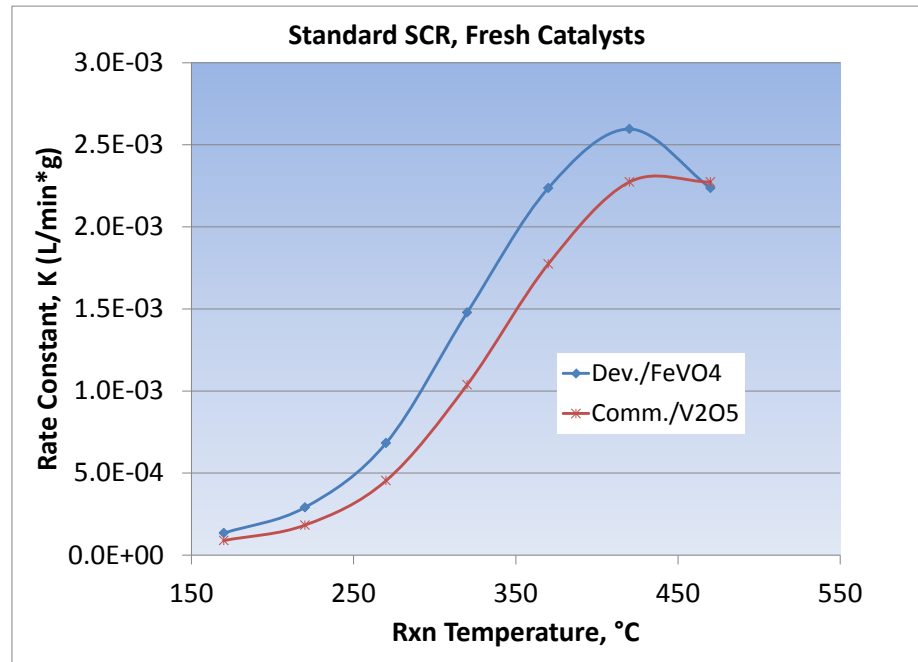
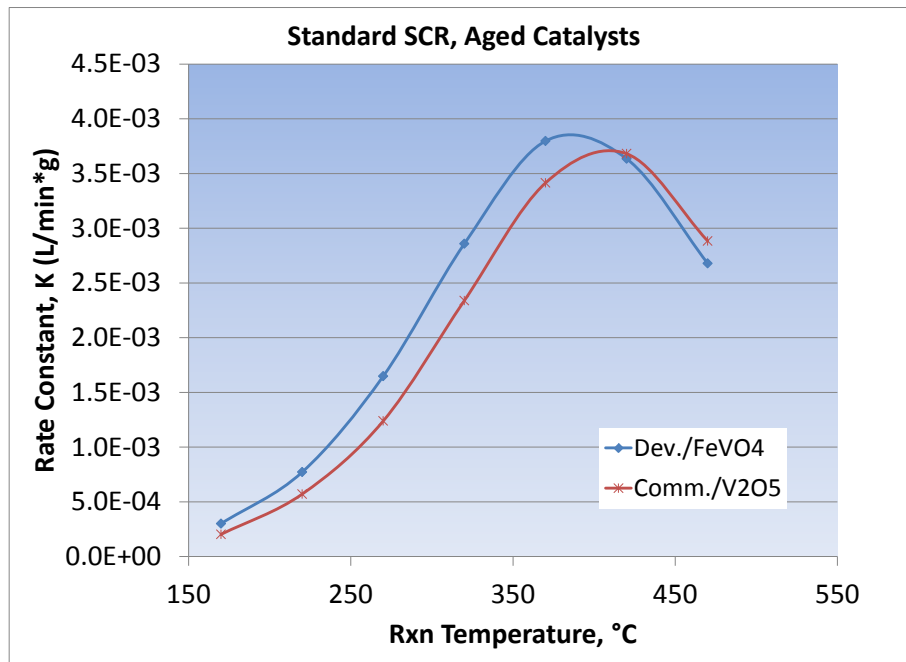
- In fast SCR, Dev./FeVO₄ has higher activity at low temperature fresh and aged

IRMA Study Standard SCR



- Rate constants calculated from conversion:

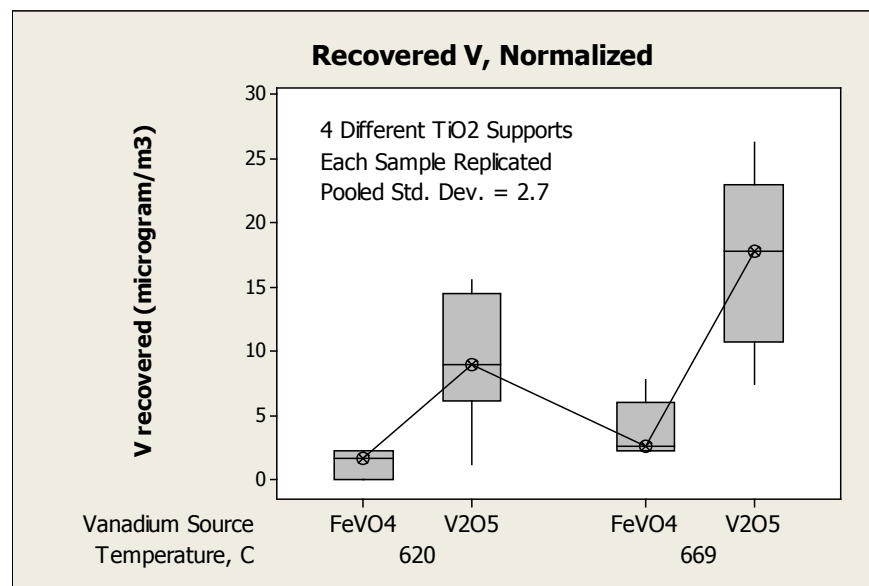
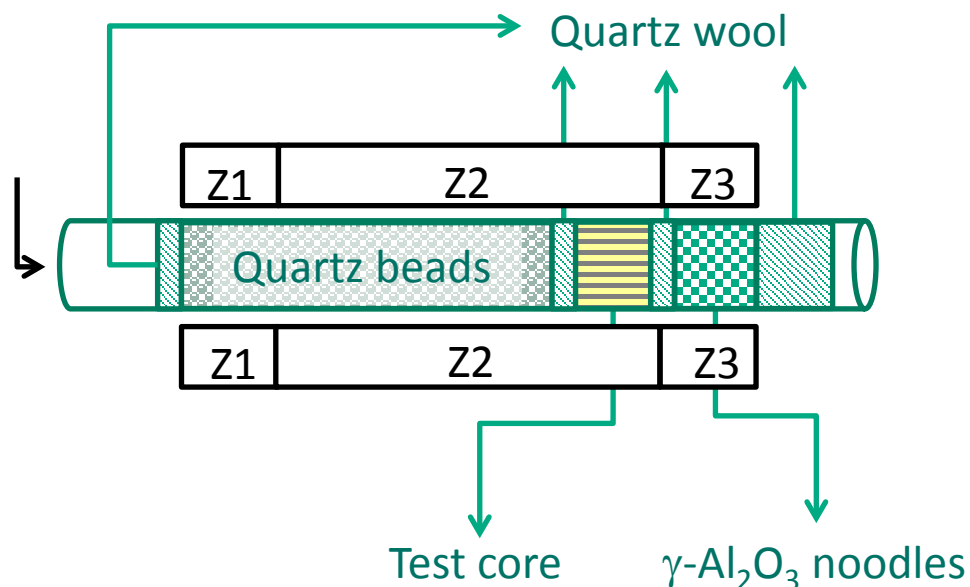
$$K = -((\text{total flow} * \text{NO conc}) / \text{cat mass}) * \text{LN}(1 - \text{conv.} / 100)$$



- In std. SCR, Dev./FeVO₄ has higher activity at low temperature fresh and aged

Vanadium Volatility

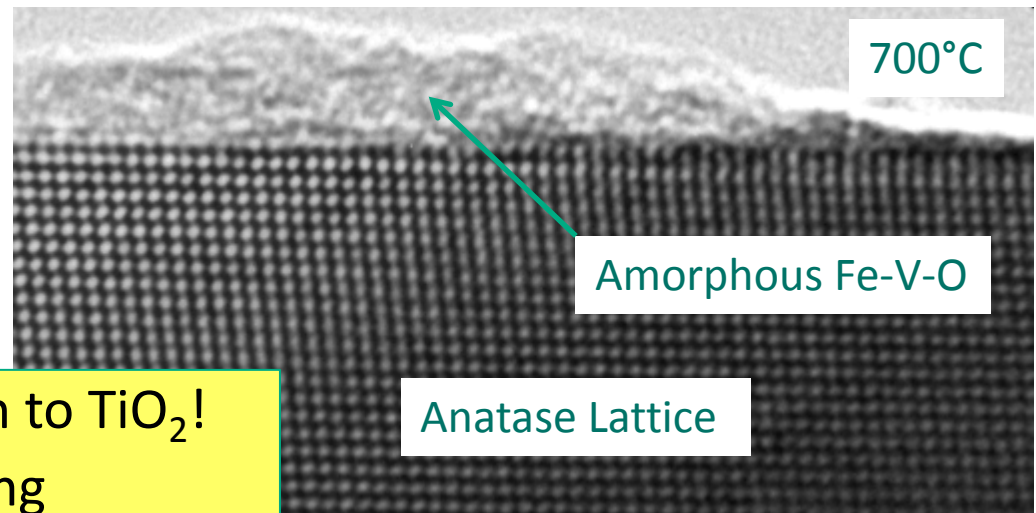
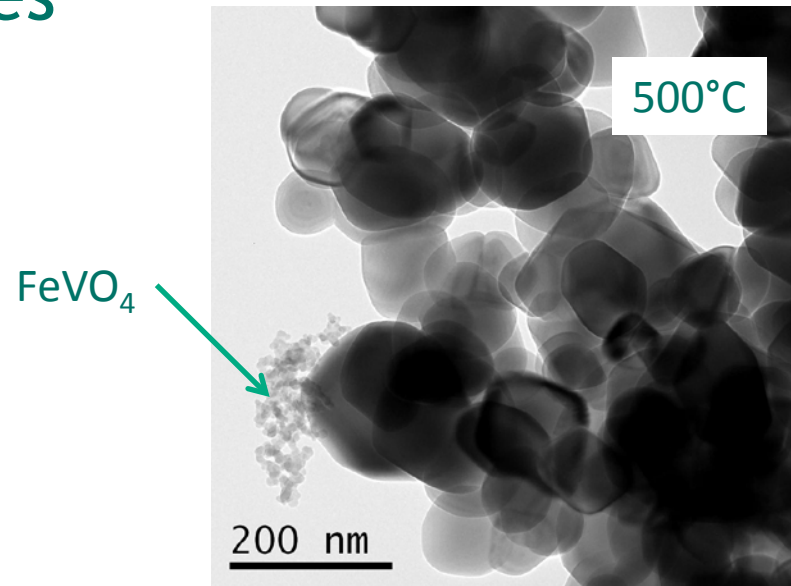
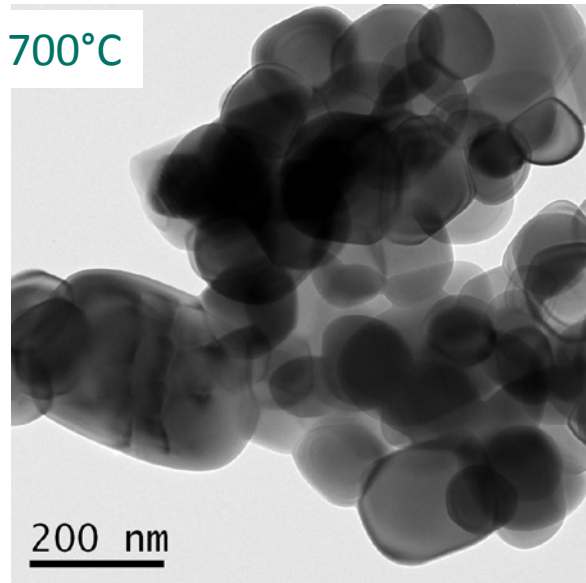
- 3-zone clamshell furnace
- 1" or 2" I core at end of Z2
- 4.5 g $\gamma\text{-Al}_2\text{O}_3$ 1/8" extrudates
- Gas: 5% O_2 , H_2O bal. N_2
- 7.5 L/m, 18 h, GHSV $\sim 35,000 \text{ h}^{-1}$ (1" core)
- Analysis:
 - Digest Al_2O_3 and downstream Quartz Wool
 - ICP-OES quantification
 - < 1 ppm detection limit
 - Blank < $0.3 \mu\text{g}/\text{m}^3$
- Reference V_2O_5 and test FeVO_4 cores from IRMA study
 - < $1 \mu\text{g}/\text{m}^3$ V at 620°C



- Clear trend for lower volatile V with FeVO_4 as source

Microscopy (I)- Powder Mixes

- SEM/EDS and TEM/EDS on $\text{TiO}_2/\text{FeVO}_4$
 - Calcined 500°C/6 h
 - Calcined 700°C/6 h



- Sub-liquidus Fe, V migration to TiO_2 !
- Heterogeneous TiO_2 sintering

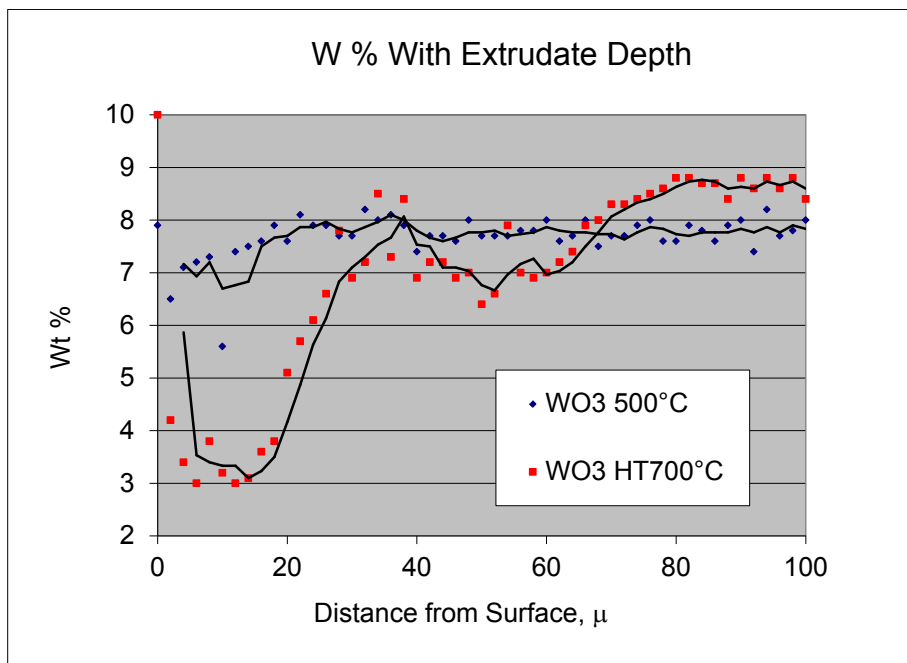
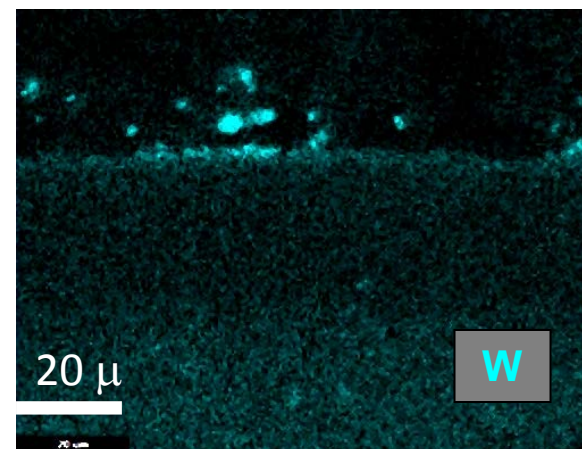
XRD and Microscopy (II)

- XRD on $\text{TiO}_2/\text{FeVO}_4$
 - Calcined 700°C/6 h
 - Hydrothermal age 750°C/16 h
- SEM/EDS on $\text{TiO}_2\text{-WO}_3(10)$ extrudates coated with FeVO_4 shell
 - Calcined 500°C/6 h
 - Hydrothermal age 700°C/16 h

XRD Rietveld Analysis

Tungsta Level	Age Condition	Fe_2O_3	Fe_2WO_6
6% WO_3	700°C-6h/Air	Y	N
	750°C-16h/HT	Y	N
10% WO_3	700°C-6h/Air	Y	N
	750°C-16h/HT	N	Y

HT 700°C



- After high temperature age
 - Evidence for V, Fe migration <100 μ into extrudate
 - Evidence of W migration to FeVO_4

Conclusions

- For intimate mixtures of FeVO_4 and TiO_2 that are heated at high temperature ($\sim 700^\circ\text{C}$ and above)
 - FeVO_4 separation into constituent phases
 - V (and some Fe) migration onto TiO_2 surface
 - Fe-rich phase crystallization to Fe_2O_3 or Fe_2WO_6
- Tungsta appears to migrate to and interact with FeVO_4
- Fe(III) believed to be mostly a harmless spectator
- ***Vanadia volatility is suppressed***
- ***Developmental Ti-Si(4.5) support with 10% WO_3 and $\sim 3.4\%$ FeVO_4 shows excellent stability, activity and selectivity***
 - ***At least as good as commercial TiO_2 -V catalyst of comparable V concentration***

Thank you!