



PAFC Cost Challenges

Sridhar Kanuri

Manager, PAFC Technology



UTC Power

A United Technologies Company

*Sridhar.Kanuri@utcpower.com



AGENDA

Purecell® 400 cost challenge

Cost reduction opportunities

Summary



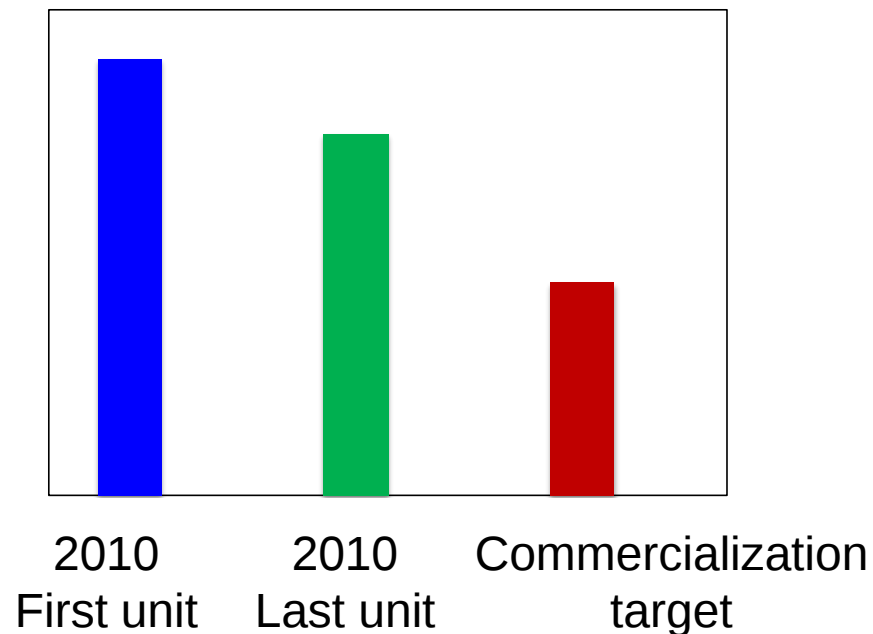
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Powerplant cost

2010 cost reduction is being accomplished by incremental changes in technology & low cost sourcing

Technology advances are required to reduce further cost and attain UTC Power's commercialization targets

First cost

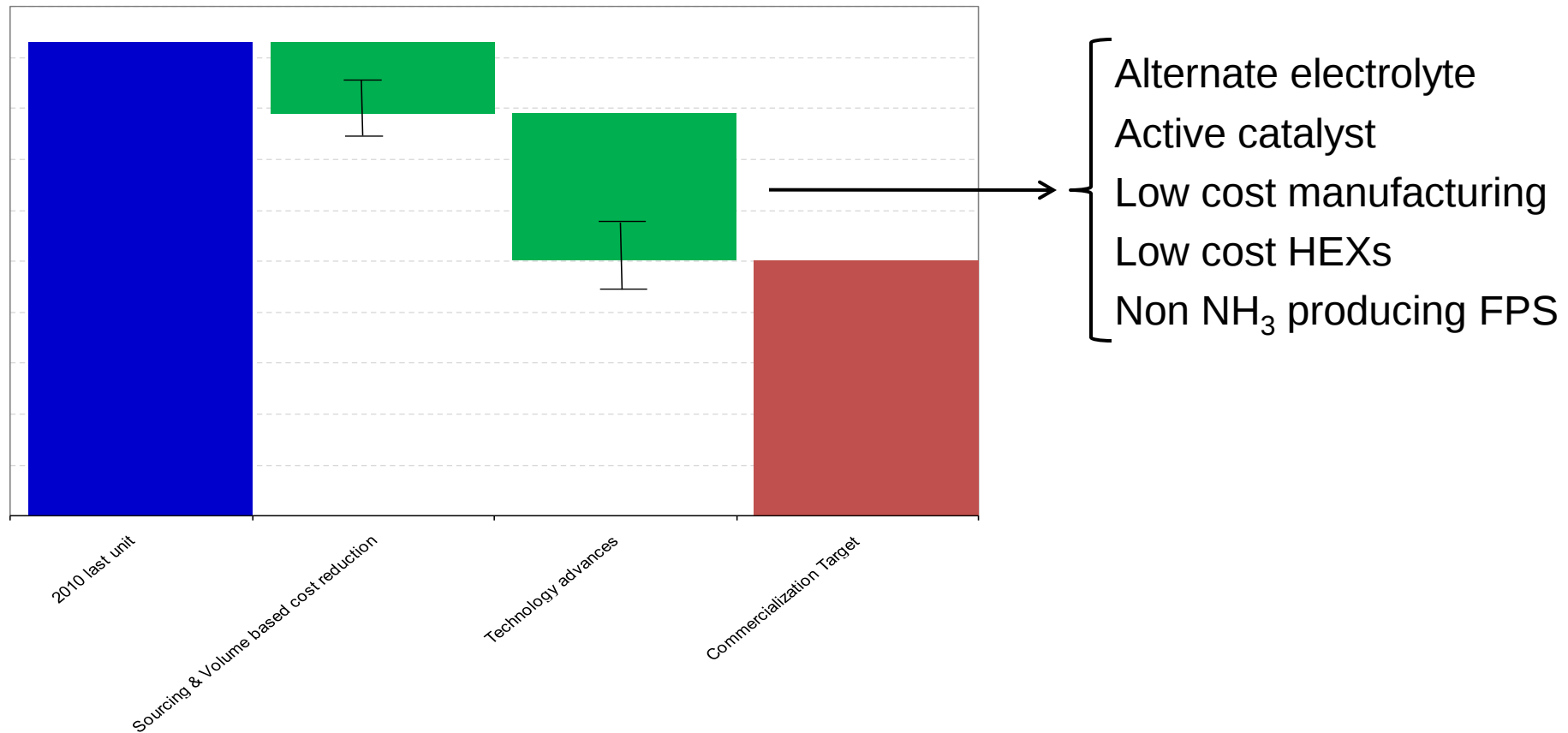




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Powerplant cost

Cost reduction



Significant technology advances are required to meet commercialization targets



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Alternate electrolyte

Property	Desired	PAFC
Operating temperature (C)	150 – 250	150 - 250
Ionic conductivity (mho/cm)	> 0.65	0.65
Vapor pressure (atm)	~ 0	4.5×10^{-7} @ 175 C
O ₂ solubility (mol/cm ³)	high	1×10^{-7} @ 175 C
Transference number	0	0.11
Anion poisoning	None	Phosphate poisons catalyst
Electrochemical stability	No decomposition products between 0 – 1.2V	H ₃ PO ₄ is very stable
Material compatibility	Compatible with fuel cell materials (Ex: Carbon, Teflon, SiC, Catalyst)	Very compatible with fuel cell materials
Freeze tolerance	-30 C – 250 C	-30 C – 250 C

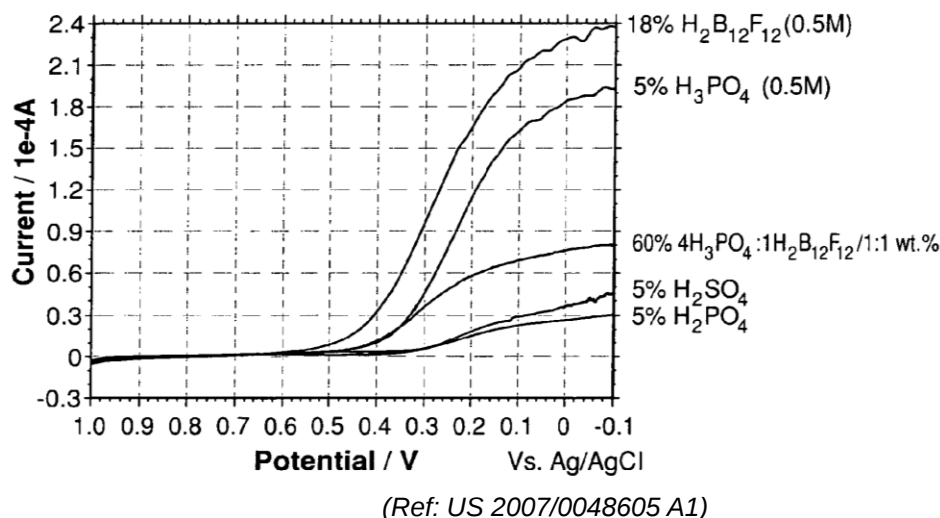




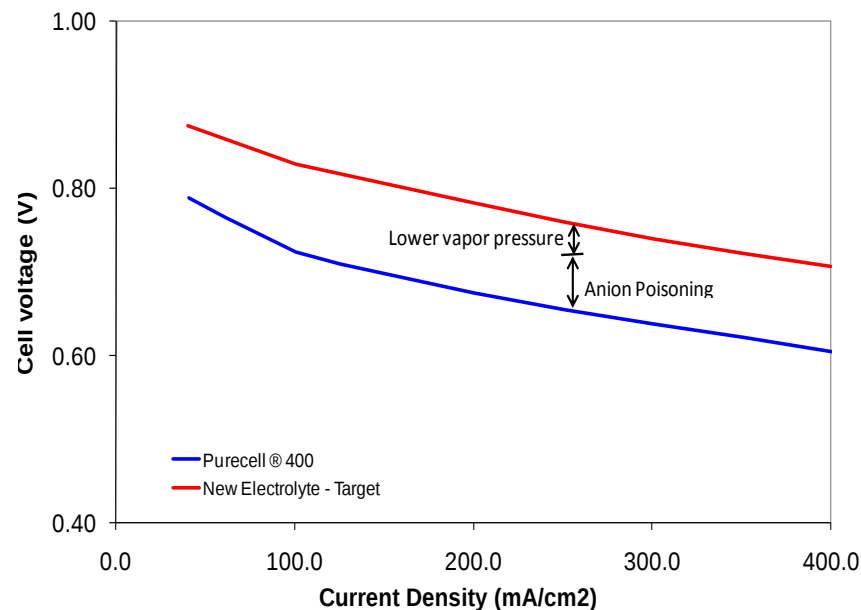
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Alternate electrolyte

RDE of aqueous acids at 1400 RPM



Performance Analysis



Literature shows significant potential for using alternate electrolytes for fuel cells

Eliminating anion poisoning and using electrolytes that have very low vapor pressure result in improving power plant efficiency by ~6 points and/or reducing power plant cost by ~ 15% - 20%.



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Alternate electrolyte – H_3PO_4 additives

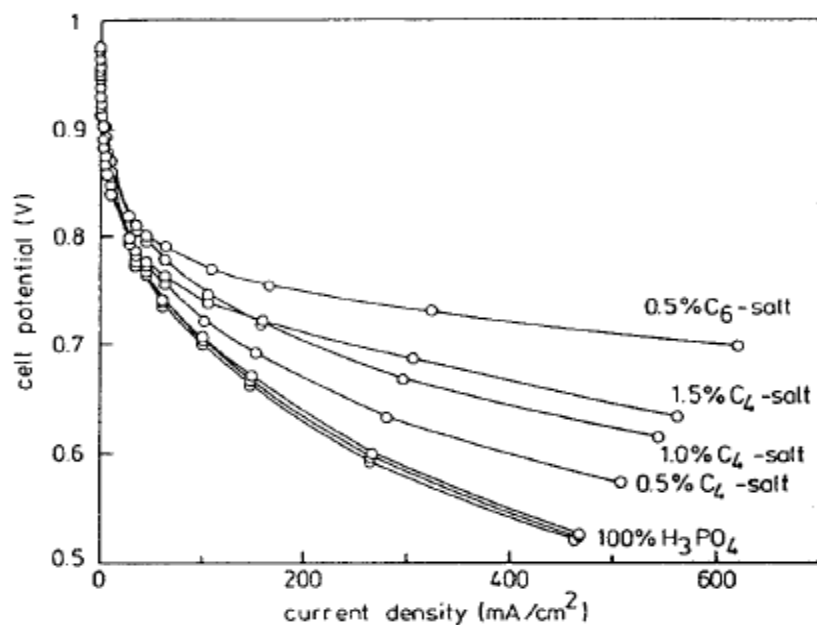


Fig. 8. Polarization curves for electrolyte modified by addition of C_6 -salt and C_4 -salt (fuel, H_2 ; oxidant, O_2 ; electrolyte, 100% H_3PO_4 ; and temperature, 190°C).

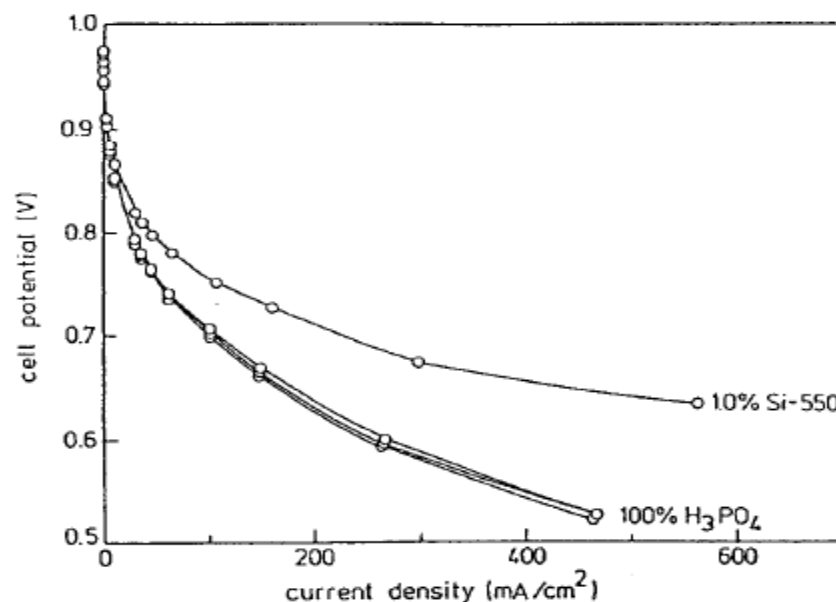


Fig. 10. Polarization curves for electrolyte modified by addition of Si-550 (fuel, H_2 ; oxidant, O_2 ; electrolyte, 100% H_3PO_4 ; temperature, 190°C).

H_3PO_4 modifications using additives have been shown to improve performance

Ref: Xiao Gang et. Al, "Electrolyte additives for phosphoric acid fuel cells, J. Electrochem. Soc., Vol. 140, No. 4, April, 1993



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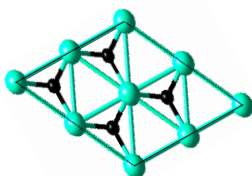
Anode catalysts

Primary requirements for PAFC anode:

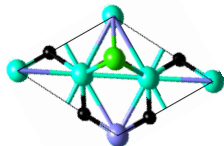
- Stable in operating temperatures between 150 C & 225C
- CO tolerant
- Stable surface area
- Low Pt usage / kW: (PAFC: 2.4 g/kW vs. PEM: 0.03 g/kW)

Opportunities:

- High mass activity catalysts & low Pt loaded anodes
- Research indicates transition metal carbides are significantly active toward HOR and are CO tolerant



● W ● C



● W ● C ● Pt ● Ta

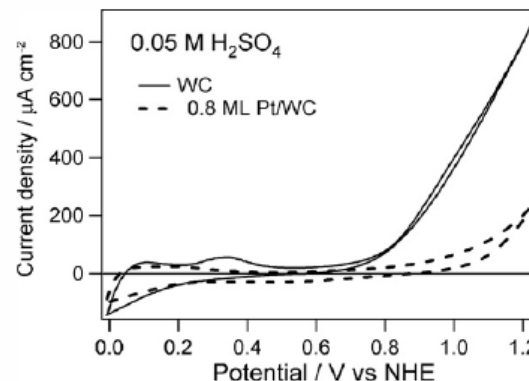
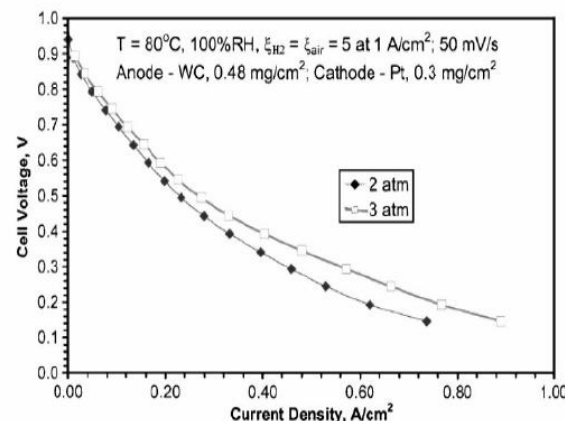


Fig. 4. Cyclic voltammetry curves of WC and 0.8 ML Pt/WC surfaces exposed to 0.05 M H₂SO₄.



H₂ / Air Polarization curves of WC based anode in 5 cm² cell

Ref: D.J.Ham & J.S.Lee, "Transition metal carbides and nitrides as electrode materials for low temperature fuel cells"



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Cathode catalysts

Primary requirements for PAFC cathode:

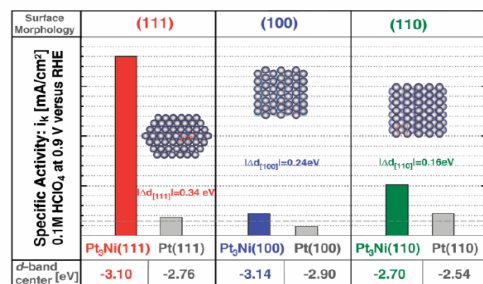
Stable in operating temperatures between 150 C & 225C

Stable surface area

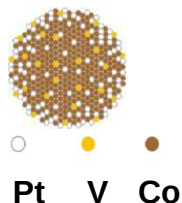
High mass activity

Low Pt usage / kW: (PAFC: 5.2 g/kW vs. PEM: 0.13 g/kW)

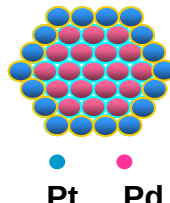
Concepts for Improved mass activity



New synthesis
(3X mass activity)



New structure
(10X mass activity)



Concepts for surface area stabilization

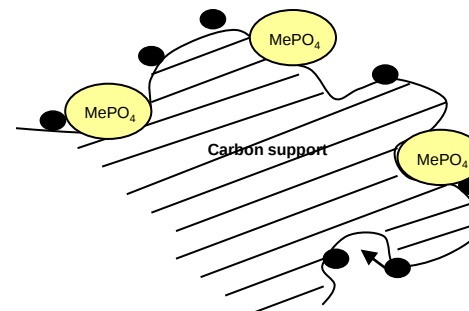


Fig. 3. Influence of the surface morphology and electronic surface properties on the kinetics of ORR. RRDE measurements for ORR in HClO₄ (0.1 M) at 333 K with 1600 revolutions per minute on Pt₃Ni(hk) surfaces as compared to the corresponding Pt(hk) surfaces (a horizontal dashed gray line marks specific activity of polycrystalline Pt) are shown. Specific activity is given as a kinetic current density i_k , measured at 0.9 V versus RHE. Values of d-band center position obtained from UPS spectra are listed for each surface morphology and compared between corresponding Pt₃Ni(hk) and Pt(hk) surfaces.

Ref: Vojislav R. Stamenkovic, et. al, "Improved oxygen reduction activity on Pt₃Ni(111) via increased surface site availability."



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Low cost manufacturing – Electrodes

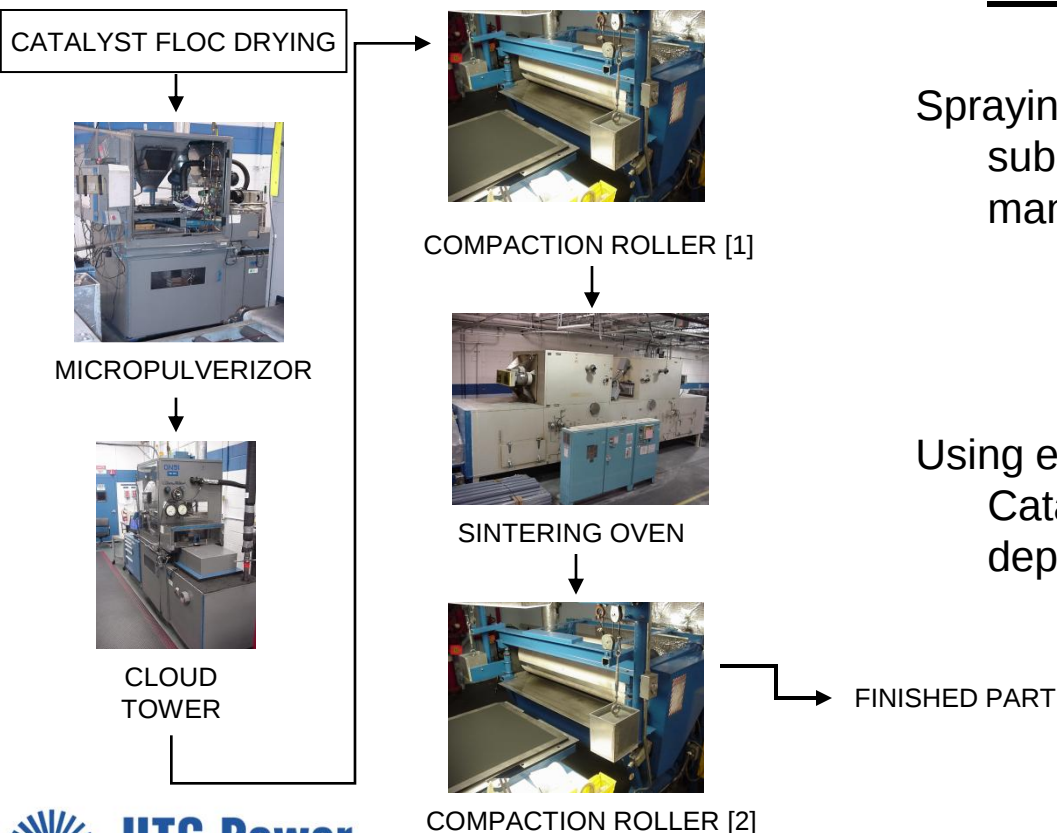
Current process of PAFC electrode manufacture is capital intensive

PAFC electrode manufacturing

Low cost electrode manufacturing concepts

Spraying mix of Catalyst, Teflon® directly onto a substrate with subsequent heat treatment to manufacture electrodes

Using emulsions to obtain a uniform dispersion of Catalyst & Teflon® with subsequent wet deposition of catalyst layer and drying





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Low cost materials / manufacturing – GDLs

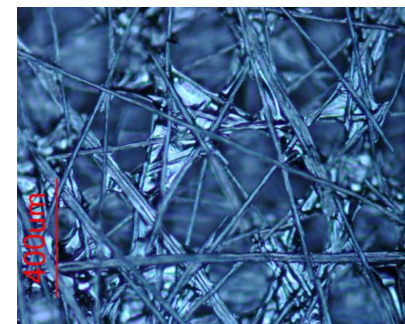
Primary requirements for PAFC GDL

Density	~ 0.6 g/cc
Electrical resistivity:	< 0.08 mV/mil
Thermal conductivity	> 400 BTU/hr-ft ² -°F
Flex strength	> 2200
Corrosion resistance	< 40 mA/g

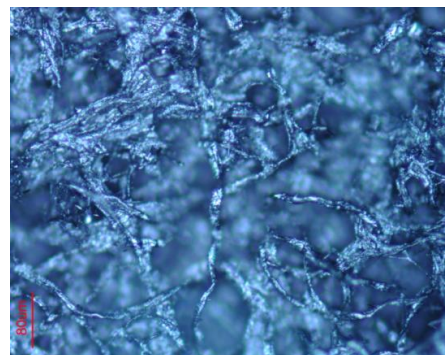
Opportunities:

Low cost Carbon composite fibers, Carbonized cellulose based fibers & cotton based fibers, low cost manufacturing techniques etc. represent significant cost reduction opportunity for H₃PO₄ fuel cells

PAN-based (long)—Today



Cellulose-based



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Low cost HEXs

Brazed plate HEXs and automotive radiators offer a significant BOP cost reduction opportunity for stationary power plants

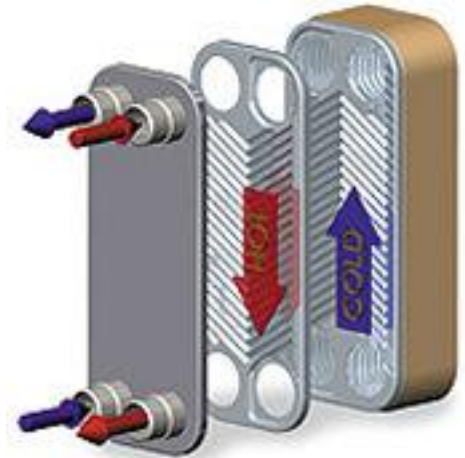
Technology challenges involve thermal cycle stability and coating development to endure high temperature H_3PO_4 environment

Advantages:

Light weight and compact

Low pressure drop for the gas side

Low cost due to volume production and simple structure



Brazed plate HEX



Low cost radiators



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Non NH₃ producing FPS

Requirement: NH₃ < 250 ppb

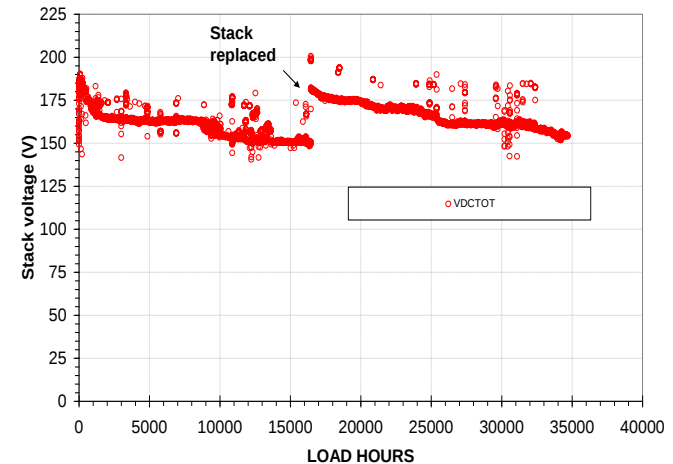
N₂ in natural gas produces NH₃ which poisons cathode catalyst

Purecell[®] 400 employs a scrubber to remove NH₃ from fuel stream

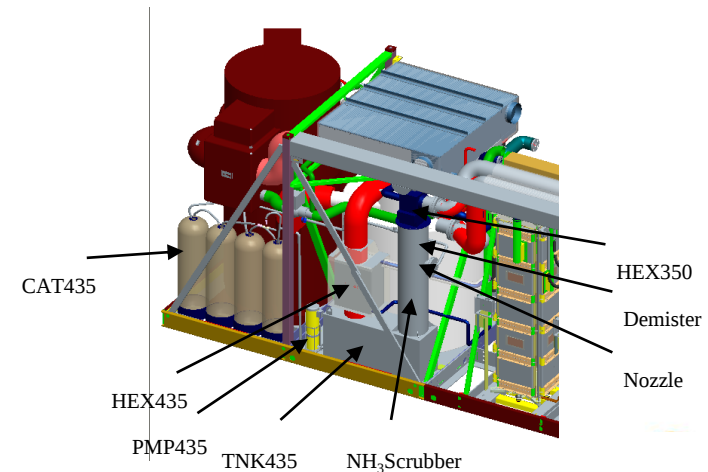
Future concepts:

Investigate non NH₃ producing FPS catalysts

Impact of NH₃ on performance



Purecell[®] 400 ARD





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Summary

First cost is a significant challenge for all stationary fuel cell power plants

Incremental improvements in technology are currently being worked / implemented by industry

Accelerated commercialization requires increased investment in technology