



Water Electrolyzers and Fuel Cells Supply Chain

Supply Chain Deep Dive Assessment

U.S. Department of Energy Response to Executive Order 14017, "America's Supply Chains"

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About the Supply Chain Review for the Energy Sector Industrial Base

The report "America's Strategy to Secure the Supply Chain for a Robust Clean Energy Transition" lays out the challenges and opportunities faced by the United States in the energy supply chain as well as the federal government plans to address these challenges and opportunities. It is accompanied by several issue-specific deep dive assessments, including this one, in response to Executive Order 14017 "America's Supply Chains," which directs the Secretary of Energy to submit a report on supply chains for the energy sector industrial base. The Executive Order is helping the federal government to build more secure and diverse U.S. supply chains, including energy supply chains.

To combat the climate crisis and avoid the most severe impacts of climate change, the U.S. is committed to achieving a 50 to 52 percent reduction from 2005 levels in economy-wide net greenhouse gas pollution by 2030, creating a carbon pollution-free power sector by 2035, and achieving net zero emissions economy-wide by no later than 2050. The U.S. Department of Energy (DOE) recognizes that a secure, resilient supply chain will be critical in harnessing emissions outcomes and capturing the economic opportunity inherent in the energy sector transition. Potential vulnerabilities and risks to the energy sector industrial base must be addressed throughout every stage of this transition.

The DOE energy supply chain strategy report summarizes the key elements of the energy supply chain as well as the strategies the U.S. government is starting to employ to address them. Additionally, it describes recommendations for Congressional action. DOE has identified technologies and crosscutting topics for analysis in the one-year time frame set by the Executive Order. Along with the capstone policy report, DOE is releasing 11 deep dive assessment documents, including this one, covering the following technology sectors:

- carbon capture materials,
- electric grid including transformers and high voltage direct current (HVDC),
- energy storage,
- fuel cells and electrolyzers,
- hydropower including pumped storage hydropower (PSH),
- neodymium magnets,
- nuclear energy,
- platinum group metals and other catalysts,
- semiconductors,
- solar photovoltaics (PV), and
- wind

DOE is also releasing two deep dive assessments on the following crosscutting topics:

- commercialization and competitiveness, and
- cybersecurity and digital components.
- More information can be found at <u>www.energy.gov/policy/supplychains</u>.

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Nomenclature

AEMEC	anion exchange membrane electrolysis cell
AEMFC	anion exchange membrane fuel cell
BNEF	Bloomberg New Energy Finance
BPP	bipolarplates
DOE	U.S. Department of Energy
ESIB	Energy Sector IndustrialBase
FCEV	fuel cell electric vehicle
GDL	gas diffusion layer
IEA	International Energy Agency
IRENA	International Renewable Energy Agency
kg	kilogram
kWh	kilowatt-hour
LSC	doped lanthanum chromate ($La_{0.8}5Sr_{0.15}CrO_3$)
LSCF	lanthanum strontium cobalt ferrite
LSM	lanthanum strontium manganite
MEA	membrane electrode assembly
MMT	million metric tonnes
PEM	polymer electrolyte membrane
PEMEC	polymer electrolyte membrane electrolyzer cell
PEMFC	polymer electrolyte membrane fuel cell
PFSA	perfluorosulfonic acid
PGM	platinum group metals
PTFE	polytetrafluoroethylene
R&D	research and development
SMR	steam methane reforming
SOC	solid oxide cell
SOEC	solid oxide electrolyzer cell
SOFC	solid oxide fuel cell

WATER ELECTROLYZERS AND FUEL CELLS SUPPLY CHAIN DEEP DIVE ASSESSMENT

TW	terawatts
W	watts
yr	year
YSZ	yttria-stabilized zirconia

Executive Summary

This report is one of a series that supports the analysis of the energy industrial base called for in Executive Order 14017 on America's supply chains (Exec. Order No. 14017, 2021). Specifically, it provides a review of the supply chain for water electrolyzers and fuel cells with a focus on polymer electrolyte membrane electrolyzer cells (PEMEC), polymer electrolyte membrane fuel cells (PEMFC), solid oxide electrolyzer cells (SOEC), and solid oxide fuel cells (SOFC). Water electrolysis and fuel cells are a nascent industry with little prior information related to supply chain needs and constraints. This report provides a preliminary assessment; further industry peer review and revisions are expected.

The market basis for this effort is founded on hydrogen market sizes because electrolyzers produce hydrogen, and fuel cells use hydrogen (H₂). Today's hydrogen market is approximately 10 million metric tonnes per year (MMT/yr) in the United States and 65–100 MMT/yr globally. However, almost none of that hydrogen is electrolytic (i.e., is produced using electrolyzers). To achieve U.S. decarbonization goals, electrolytic hydrogen will be necessary, although there will likely be a role for hydrogen produced using thermal conversion processes such as today's common technology—steam methane reforming (SMR)—along with carbon capture and storage (CCS). Thus, the electrolytic hydrogen market will need to grow substantially to meet potential future demands and provide decarbonization opportunities for difficult-to-abate sectors, including synthetic fuels for air and marine transport, long-distance transport via heavy and medium duty vehicles, energy storage, and high-temperature heat. For the end point in this analysis, we build upon the *U.S. Long-Term Strategy: Pathways to Net-Zero Greenhouse Gas Emissions by 2050* and use the Princeton Net-Zero America E+RE+ scenario's U.S. market estimate of just over 100 MMT H₂/yr in 2050, which provides a more granular technology resolution. We also use the International Energy Agency's global market estimate of just over 500 MMT H₂/yr in 2050 to provide a global comparison.

To meet that U.S. market size, estimates of electrolyzer capacity required range up to 1,000 GW to meet new capacity deployments and replace existing capacity at the end of its lifetime. This is a large increase over the approximately 0.17 GW of capacity currently installed or planned in the United States and result in an approximately 20% compound annual growth rate from 2021 to 2050. We also estimate a total domestic fuel cell capacity of over 50 GW and a maximum annual manufacturing rate as high as 3 GW/yr will be needed for heavy-duty vehicles, medium-duty vehicles, and electricity generation.

The current and future electrolyzer and fuel cell supply chains include five segments: extracting the raw materials, generating processed materials, manufacturing subcomponents, manufacturing components, and recovering materials at the end-of-life. This report summarizes findings across those segments for today's supply chain and identifies key considerations for the development of supply chains to meet a 100 MMT/yr electrolytic hydrogen market.

Currently, the United States has sufficient domestic resources and imports to meet the materials demand. The United States also currently has manufacturing capabilities in most of the necessary key processed materials and subcomponent manufacturing for both polymer electrolyte and solid oxide technologies. Likewise, the United States has relatively well-positioned end product manufacturing capabilities for both technologies.

To meet the needs of a 100 MMT/yr hydrogen market, large increases in extraction and refining of many materials would be needed, with many key materials currently being addressed primarily (and exclusively, for some) by imports. Especially of concern are several materials that have both (1) larger projected electrolyzer and fuel cell demands than their current availability and (2) a currently high percentage of total market being

met via imports with no specific plans for domestic production. Those include iridium, yttrium, platinum, strontium, and graphite. The platinum group metals (PGM) catalyst report that is part of this series ("Supply Chain Review: Platinum Metal Group Catalysts" 2021) provides additional information on those metals, including vulnerabilities and opportunities. The United States appears to have sufficient resources and supply chains for many of the other key materials, including stainless steel, titanium, zirconium, and nickel.

It is difficult to exactly predict manufacturing challenges because of the extraordinary growth required in the electrolytic hydrogen market and thus the electrolyzer and fuel cell markets. Key processed materials for polymer electrolyte technologies include perfluorosulfonic acids, catalysts, graphite composites, and titanium meshes. Key processed materials for solid oxide technologies include air electrode materials, fuel electrode materials, and the electrolyte. How and where manufacturing capacity along the supply chain may grow are unknown. Thus, government support may be needed to support those industries and meet cost reduction, growth, decarbonization, and supply chain security objectives.

Key vulnerabilities in developing an electrolytic hydrogen market and the supply chains needed for that market include:

- Immature technologies that are not currently cost-competitive for both electrolytic hydrogen production and utilization
- Lack of sufficient emission reduction incentives
- Insufficient codes and standards
- Insufficient electricity generation capacity
- Electrolyzers not being compensated sufficiently in the electricity market
- Insufficient infrastructure to support hydrogen markets at their potential
- Availability of key raw materials
- Growth requirements of manufacturing capacity and supply chains
- Energy justice issues
- Environmental justice issues
- Mismatch in demand and supply of domestic workforce
- Consistent and equal standards for hydrogen production around the world.

While the United States has technology development targets and an RD&D plan, it does not currently have hydrogen deployment targets or a national plan, unlike other countries. However, the United States is developing a national plan as required by Section 40314 of the Infrastructure Investment and Jobs Act (Infrastructure Investment and Jobs Act 2021).

The overarching opportunity for electrolytic hydrogen within the United States is to capture the high value-added links of the electrolytic hydrogen supply chain for the potential market of over 100 MMT/yr for applications across the industrial, transportation, and power sectors (Department of Energy (DOE) 2020). Key opportunities to enable the growth of electrolytic hydrogen and fuel cell markets to meet the overarching opportunity include:

- Reducing cost and increasing commercialization of electrolytic hydrogen production
- Developing economically competitive applications
- Leading development of codes and standards

- Expanding the U.S. electric grid capacity
- Developing and managing bulk hydrogen storage
- Utilizing of the natural gas infrastructure for hydrogen transport and storage
- Developing domestic material supplies, including recycling and PGM-free catalysts
- Developing electrolyzer and fuel cell manufacturing capacity
- Leading energy and environmental justice issues for a new industry
- Potentially exporting hydrogen.

Find the policy strategies to address the vulnerabilities and opportunities covered in this deep dive assessment, as well as assessments on other energy topics, in the Department of Energy 1year supply chain report: "America's Strategy to Secure the Supply Chain for a Robust Clean Energy Transition."

For more information, visit www.energy.gov/policy/supplychains.

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1.Introduction

This report supports the analysis of the energy industrial base sector called for in Section (4)(a)(iv) of Executive Order 14017 on America's Supply Chains, which requires the Secretary of Energy, in consultation with the heads of appropriate agencies, to "submit a report on supply chains for the energy sector industrial base (as determined by the Secretary of Energy)" (Exec. Order No. 14017, 2021). The Secretary of Energy selected 11 specific technology areas for reporting, with electrolyzers and fuel cells being one of them. To meet the requirements in the executive order, this report considers supply chains and critical materials for electrolyzers, which split water into hydrogen and oxygen, and fuel cells, which consume hydrogen to generate electricity. Multiple electrolyzer and fuel cell systems are available, each having a different architecture, relying on different materials, and offering unique system integration opportunities; thus, the supply chains for these systems and how and where they might be deployed vary.

This section of the report summarizes the potential roles of hydrogen, electrolyzers, and fuel cells in the future, provides a technology overview, and summarizes a market size and resource requirement estimate to 2050. The next section maps supply chains for polymer electrolyte and solid oxide technologies. Then, a risk assessment is reported, and it lists key vulnerabilities. The report concludes with a section on opportunities and challenges.

1.1 Potential Roles of Hydrogen, Electrolyzers, and Fuel Cells in the Future Energy System

Hydrogen has been identified as a key energy intermediate to enable full decarbonization of the energy system because it can temporally decouple carbon-free energy production (e.g., variable renewable energy and nuclear energy) from its utilization and it can be a feedstock for independent and dispatchable energy applications and chemical processes. However, to meet that potential, new hydrogen production, transportation, storage, and utilization supply chains need to be developed and the components that support those supply chains need to be manufactured and operated. This report focuses on two critical components of this supply chain: electrolyzers and fuel cells.

Electrolyzers use electricity to split water into hydrogen and oxygen. If nuclear or renewable-generated electricity is used, the resulting hydrogen has minimal related carbon emissions. Today, electrolysis is not a common method of hydrogen production because the cost of hydrogen produced from electrolysis is greater than it is from conventional means which involve hydrocarbon reforming such as steam methane reforming (SMR). However, the U.S. Department of Energy's (DOE's) Hydrogen Shot initiative is targeting a production cost (\$1/kg) that is lower than SMR within the next decade (Department of Energy 2021).

Electrolyzers have the potential to support the energy system both by producing hydrogen for use elsewhere and at other times and by providing a controllable load for the grid. If the electrolyzer capital costs are sufficiently low, hydrogen can be produced at a lower cost by reducing or stopping production when electricity prices are high and increasing production up to the maximum load when electricity prices are low, instead of operating at all times (Badgett, Ruth, and Pivovar 2022).

Fuel cells are essentially the opposite of electrolyzers. They react hydrogen and oxygen to generate electricity with water as a byproduct. Like electrolyzers, there are both low-temperature and high-temperature fuel cells. However, the only low-temperature fuel cells with strong development support right now are polymer electrolyte membrane fuel cells (PEMFCs) also known as proton exchange membrane fuel cells. PEMFCs are expected to be used primarily for transportation applications (e.g., heavy, medium, and light-duty vehicles; material handling

equipment; and trains), and they have the potential to be used to generate electricity both as a backup power source and as a dispatchable generator for the grid. PEMFCs can produce electricity exclusively or as combined heat and power when the heat can be used. For high temperatures, solid oxide fuel cells (SOFCs) are commercialized and continue to be developed, as are molten carbonate fuel cells. Some high-temperature fuel cells can reform methane within the fuel cells and thus, use natural gas directly. High-temperature fuel cells are primarily considered for power for microgrids and the grid.

Because electrolyzers produce hydrogen and fuel cells consume it, they can be used in combination to provide energy storage for the grid where low-cost hydrogen storage is available. One benefit of hydrogen storage is that, unlike conventional batteries, the amount of stored energy can be decoupled from charging and discharging power. Thus, hydrogen storage is likely more economic for long-duration energy storage than batteries (Hunter et al. 2021). Some studies that have analyzed what would be required to reach 100% renewables on the grid have concluded dispatchable electricity generation (including long-duration storage) is required to achieve that objective (Cochran et al. 2021; Pearre and Swan 2020; Denholm et al. 2021; Kroposki et al. 2017).

1.2 Technology Overview

Several electrolysis and fuel cell technologies exist or are under development (Table 1, page 3). Electrolyzers that use electricity exclusively are referred to as low-temperature electrolyzers because they operate at temperatures lower than the boiling point of water (100°C at sea level). Low-temperature electrolyzer technologies include traditional alkaline electrolyzers, polymer electrolyte membrane electrolyzer cells (PEMECs), and anion exchange membrane electrolyzer cells (AEMECs). PEMECs are less mature than alkaline electrolyzers, but they exhibit significant potential for cost reductions and large-scale deployment. AEMECs are in early development stages, but have the potential to cost less than PEMECs while having similar performance attributes. Due to the low maturity of AEMECs, they are not considered in this analysis. PEMECs can ramp operation up and down at faster rates than traditional alkaline electrolyzers (International Energy Agency "The Future of Hydrogen - Analysis" 2019), making them favorable for directly coupling them to variable renewable energy sources such as wind or solar—one reason they could be developed at large scales.

Electrolysis technologies that use both heat and electricity and are commonly referred to as high-temperature steam electrolyzers because steam exists within their stacks. Those solid oxide cells (SOC) use high-temperature oxide-conducting ceramics as the ion-conducting membrane. SOCs are attractive for their ability to produce hydrogen at much higher efficiencies than other technologies, which they can do because the high temperatures they operate at (generally 650–800°C) reduce the minimum voltage of the water-splitting reaction (Hauch et al. 2020). Solid oxide electrolyzer cells (SOECs) have been shown to have the potential to ramp operation up and down similar to PEMECs, but they require some heat and electricity to be held in hot standby due to issues with thermal inertia (Badgett, Ruth, and Pivovar 2022).

Two types of fuelcells are considered in this analysis: PEMFC and SOFC. These systems are largely similar to their electrolyzer counterparts but have slight variations in materials used and system designs. Like SOECs, SOFCs are more efficient than their low-temperature counterparts; they can achieve 70% efficiencies when fueled with hydrogen or natural gas. PEMFCs are favorable for use in transportation applications, as they operate variably. Both systems could be developed in energy storage applications, using hydrogen generated from electrolyzers to produce electricity when needed.

Key characteristics of the electrochemical systems that are considered in this work are summarized in Table 1 (Badgett, Ruth, and Pivovar 2022). Further information regarding performance and material use assumptions for each technology considered in this analysis can be found in Appendix A.

Table 1. Summary of Material	and Performance	Characteristics	of Electrochemical	Technologies Considered
in This Analysis				

System	Summary of Materials and Performance
PEMEC	PEMEC uses platinum and iridium oxide as catalysts and perfluorosulfonic acid (PFSA) as a proton conductor and binder. Membrane electrode assemblies are separated by a titanium bipolar plate coated with a thin layer of platinum. The system operates at lower efficiencies than SOEC, but operation can be ramped up and down quickly, making PEMEC favorable for integration with variable renewable generation, such as wind and solar photovoltaics. Current systems exhibit moderate lifetimes and can operate at high current densities at moderate cell potentials.
SOEC	SOEC uses oxide ion-conducting electrolyte materials, such as yttria-stabilized zirconia (YSZ) that allow for ion transport at high temperatures and generally operate at temperatures near 600°C. High-temperature operation significantly increases the system efficiency, but it poses challenges for system durability and frequent on-off cycling.
Alkaline electrolyzers	Alkaline electrolyzers use nickel-based catalysts in an alkaline electrolyte solution such as potassium hydroxide and a diaphragm to separate electrodes and transport hydroxide ions. Alkaline systems operate at lower efficiencies than other electrolysis architectures, but they have longer lifetimes and have been deployed in large-capacity systems. Materials for alkaline electrolyte solutions are not considered in this analysis.
AEMEC	AEMEC uses an anion exchange membrane separated by nickel and nickel alloy catalysts to produce hydrogen. These systems operate at similar voltages, but lower current densities than to PEMEC systems. AEMEC systems are at a lower technology maturity level than PEMECs and alkaline electrolyzers (Miller et al. 2020).
PEMFC	PEMFC uses materials and designs that are similar to those used by PEMEC. PEMFC uses platinum and platinum-based alloys as cathode catalysts and PFSA as a proton conductor and binder. Membrane electrode assemblies are separated by a metal or carbon bipolar plate.
SOFC	SOFC uses materials, designs, and operating strategies that are similar to those of SOEC.
AEMFC	Anion exchange membrane fuel cells (AEMFCs) use an alkaline anion exchange membrane electrolyte and avoid the use of platinum catalysts that are required for PEMFCs. Similarly to their AEMEC counterparts, AEMFCs are at a lower technology maturity and are not considered in this analysis.

This report focuses on the supply chains for two electrolysis technologies (PEMEC and SOEC) and two fuel cell technologies (PEMFC and SOFC). These technologies are anticipated to hold the largest share in the global electrolyzer/fuel cell market overall. Traditional alkaline electrolyzer cells are included in installed electrolyzer capacity estimates because they are the most mature electrolyzer technology, having been operated for years in the chemical industry and are likely to be deployed across some hydrogen applications. Though AEMECs hold potential for future applications, these systems are not included in this analysis because of their low technical maturity; thus, critical materials for alkaline and AEMEC systems are beyond the scope of this analysis.

The basic components of electrolyzers and fuel cells are illustrated in Figure 1. All fuel cell and electrolyzer systems require anode and cathode catalysts for the two half reactions occurring on either side of the cell, with the desired catalytic materials varying depending on the system architecture. These catalysts are generally supported on diffusion media such as carbon paper to facilitate liquid and gas transport to and from the catalyst layer. Anode and cathode catalysts are separated by ion exchange media, the type of which varies by system. PEMFC and PEMEC systems transport hydrogen ions through a polymer electrolyte membrane, and solid oxide systems use oxide ion-conducting ceramic materials. To form large-scale electrochemical stacks, repeat units of catalysts, support, and ion conductors are separated by bipolar plates (BPPs), which facilitate product and reactant flows and act as current collectors for the system.





The heart of a PEMFC or a PEMEC is the membrane electrode assembly (MEA), which includes the membrane, the catalyst layers, and the porous transport layers. Hardware components used to incorporate an MEA include gaskets, which provide a seal around the MEA to prevent leakage of gases, and BPPs, which are used to assemble individual cells into a fuel cell stack and provide channels for the gaseous fuel and air (DOE n.d.). The PEMEC and associated materials and subcomponents are largely similar to those of the PEMFC. Due to higher voltages on the anode side, corrosion-resistant materials like titanium, titanium alloys, and coated stainless steel are used, instead of the carbon materials commonly used in PEMFCs (e.g., for porous transport layers and BPPs). PEMEC anode catalyst compositions are different from PEMFCs, with PEMECs using iridium for the oxygen evolution reaction (HyTechCycling 2019; E4Tech 2019), while PEMFCs use platinum for both anode and cathode catalysts.

Solid oxide stacks are composed of approximately 40–60 individual ceramic cells that produce nearly 25 W each in fuel cell operations, interconnected into a single module (Bloom Energy 2019). Material sets for solid oxide electrolyzers and fuel cells are identical or very similar. Each cell is comprised of layers of different ceramometallic materials allowing for efficient ionic species and electrical charge transport at high temperature (600– 1,000°C). The most prominent cell geometry (planar) involves a thicker, fuel electrode providing support with the electrolyte deposited and sintered followed by the air electrode layers. Completed cells are connected in series or in parallel with appropriate separators, spacers, and flow fields to keep the fuel and oxygen carriers separate inside the final stack frame. Stacks can then be connected in parallel to produce a desired nominal output of electricity or fuels in a modular or fully integrated fashion for a given application.

1.3 Electrolyzer and Fuel Cell Market Size Estimates

The current domestic hydrogen market is approximately 10 MMT/yr and global hydrogen production is 65–100 MMT/yr (Connelly, Elgowainy, and Ruth 2019). Nearly all of this hydrogen is produced via conventional means, especially SMR. SMR uses natural gas as a feedstock, and the carbon dioxide that results from this reaction is usually released to the atmosphere and not captured. As a result, hydrogen production is responsible for 830 MMT/yr of carbon dioxide emissions (IEA 2019), and thus is a key contributor to total global carbon emissions. Currently, the primary applications for hydrogen are hydrocracking and hydrodesulfurization in crude oil refining and ammonia production via the Haber Bosch process (Connelly, Elgowainy, and Ruth 2019). The current U.S. hydrogen market revenue is approximately \$17.6 billion/yr (Fuel Cell and Hydrogen Energy Association 2021). The current hydrogen market includes both captive (which is hydrogen produced at the point of consumption for internal use) and merchanthydrogen (which is hydrogen sold to consumers). Approximately, 50%-75% of the current market is captive (Connelly, Elgowainy, and Ruth 2019).

With only 0.172 GW of electrolysis capacity currently installed or planned in the United States (Arjona and Buddhavarapu 2021), the maximum electrolytic hydrogen currently produced in the United States is less than 0.025 MMT/yr. Thus, the electrolytic hydrogen market is in its infancy.

To meet decarbonization goals, carbon capture and sequestration would need to be added to SMR or electrolytic hydrogen would need to displace SMR production. As of this writing, electrolyzers and fuel cells have been mostly deployed only in niche applications in the transportation and industrial chemical sectors. However, if a clean hydrogen market develops, electrolyzer and fuel cell markets will also develop.

Original work in this report estimated domestic and global electrolyzer and fuel cell market sizes for this analysis using data from recent modeling work that depicts deep decarbonization across domestic and global economies. In November 2021, the U.S. Department of State and U.S. White House released The Long-Term Strategy of the United States, which lays out how the United States can reach its goal of net-zero emissions no later than 2050 and was submitted to the United Nations Framework Convention on Climate Change (UNFCCC) at the 26th Conference of the Parties.¹ The LTS illustrates many plausible pathways through 2050 to achieve a net-zero emissions economy, and offers insights into what the overall energy system for the United States could look like between now and 2050 under a range of assumptions about the evolution of technological costs, economic growth, and other drivers to 2050. The International Energy Agency estimates global hydrogen market demand could exceed 500 MMT/yr by 2050 (IEA 2021b) and the Princeton Net-Zero America analysis (Larson et al. 2021) estimates U.S. hydrogen demand could exceed 100 MMT/yr by 2050 in the E+RE+ scenario, which we used in the analysis reported here because it extends the U.S. Long-Term Strategy (United States Department of State and United States Executive Office of the President 2021) by providing a more granular technology resolution. Figure 2a and b (page 7) show the potential global and domestic hydrogen market growth between now and 2050 based on data from IEA and Princeton University analyses. The 2020 Princeton NZA market size estimate of 5 MMT/y used in this analysis varies from the 10 MMT/yr current domestic market estimated (Connelly, Elgowainy, and Ruth 2019) due to variations in data gathering as well as conversion factors used to generate hydrogen market sizes in MMT/yr based on energy consumption data from NZA scenarios. We estimate that around 5000 TWh/yr of electricity would be required to produce 100 MMT/yr.

¹ https://www.whitehouse.gov/wp-content/uploads/2021/10/US-Long-Term-Strategy.pdf

Key results of the original analysis work conducted for this report are shown in Figures 2-6 and Figure 9. These results show possible trajectories for the growth of hydrogen markets, illustrating accompanying increases in deployment of fuel cells and electrolyzers. These projections can inform possible material requirements to produce these systems, helping to identify where raw material demands could significantly exceed current production and consumption. Assumptions and methodologies used to develop these estimates can be found in Appendix A.

The current U.S. hydrogen market revenue is estimated at \$17.6 billion/yr (Fuel Cell and Hydrogen Energy Association 2021). Assuming a current (2020) hydrogen market price of \$2/kg and hydrogen demand of approximately 5 MMT/yr from values calculated in this work, the current size of today's hydrogen market could be \$10 billion/yr. Variation between the market sizes of \$17.6 and \$10 billion/yr are subject to the same variability discussed in the prior paragraph. Assuming the Hydrogen Shot initiative meets the target of \$1/kg clean hydrogen by 2030 (Department of Energy 2021) and using the market size estimates calculated here, revenue in domestic hydrogen markets changes to \$8.6 billion/yr at 2030 and \$ 105 billion/yr in 2050 at a hydrogen price of \$1/kg.

We did not estimate impacts on U.S. jobs, but the "Road Map to a US Hydrogen Economy" estimated that if the U.S. hydrogen market grows to 17 MMT/yr by 2030, it would support 700,000 jobs and if it grows to 74 MMT/yr by 2050, it would support 3,400,000 jobs (Fuel Cell and Hydrogen Energy Association 2021).

Neither analysis provides details about the production of hydrogen to meet global and U.S. demand by sector. Therefore, we developed estimates for our analysis that estimate the amount of hydrogen and installed electrolyzer capacity by sector. We assumed the conventional technology (i.e., primarily SMR) will meet existing and near-term hydrogen demand from now to 2025 (Figure 3, c and d, page 8) and that after 2025, hydrogen market demand increases will be met by electrolyzer deployments. We also assumed that from 2030 to 2050 conventional technologies will be phased out and replaced with electrolysis, with no hydrogen being generated via conventional generation technologies by 2050.

Because the future market involves more consumers and a smaller share of industrial users, the merchant market's share in the future is likely to be similar to today's share. Assuming that 75% of new feedstock, synfuels, chemicals and existing feedstock, and energy storage are captive and the remaining applications are served by merchant hydrogen, the market share for captive hydrogen changes from the current value of 50%-75% (Connelly, Elgowainy, and Ruth 2019) to approximately 68% in 2030 and 67% in 2050.



Figure 2. Estimated global (a) and U.S. (b) hydrogen demands by economic sector from 2020 to 2050. Sources of hydrogen by type for global (c) and U.S. (d) (Original work)

Conventional supply sources include SMR. Domestic hydrogen demand is from the E+RE+ scenario in the Princeton Net-Zero America analysis. The 2020 Princeton NZA market size estimate of 5 MMT/yr used in this analysis varies from the 10 MMT/yr current domestic market estimated (Connelly, Elgowainy, and Ruth 2019) due to variations in data gathering as well as conversion factors used to generate hydrogen market sizes in MMT/yr based on energy consumption data from NZA scenarios.

Global data source: IEA 2021; U.S. data source: Larson et al. 2021

The combination of increasing hydrogen demands and phasing out conventional technologies including SMR requires many electrolyzers to be manufactured and deployed. Our estimates of cumulative (Figure 3, a and b) and annual (Figure 3, c and d) installed electrolyzer capacity are shown in Figure 3. The manufactured capacity estimates shown here include estimates resulting from both new deployment of electrolyzers and the replacement of retired systems at their end-of-life. We estimate that up to 1,000 GW of electrolyzer capacity is manufactured by 2050 to meet new capacity deployments and replacement of existing capacity at the end of its lifetime. This is a large increase over the approximately 0.172 GW of capacity currently installed or planned in the United States (Arjona and Buddhavarapu 2021). These installed capacities result in an estimated compound annual growth rate from 2021 to 2050 of 22% for PEMECs and 19% for SOECs in the United States (see Appendix A).



Figure 3. Estimated cumulative global (a) and U.S. (b) manufactured capacity of electrolyzers by type. Estimated annual global (c)^a and U.S. (d) manufactured capacity of electrolyzers by type (Original work)

Values include replacement systems manufactured to replace those at end-of-life.

^a The global manufacturing rate of electrolyzers decreases from 2025 to 2030 in Figure 3c because of how this work assumed electrolyzers phase in to the hydrogen markets and assumptions about growth in hydrogen market size. The linear hydrogen market growth for global markets creates a jump in required manufacturing capacity of electrolyzers, which slows in its rate of growth from 2025 to 2030. It is worth noting that in Figure 3c, although the capacity decreases from 2025 to 2030, the total installed capacity of the electrolyzers continues to increase and just the rate of change (derivative) is lower here than what was required to phase the electrolyzers into the market

Fuel cell manufacturing is also projected to increase both domestically and globally from current levels to 2050 (Figure 4). Most of the growth in deployment of fuel cells in these analyses is driven by applications in the transportation sector, mainly from heavy and medium duty fuel cell electric vehicles. This work also assumes fuel cells are deployed in energy storage applications for electric grid decarbonization, taking electrolytic hydrogen and generating electricity that is supplied to the power sector. In the Princeton NZA E+RE+ scenario, hydrogen for energy storage is predominantly used in combustion turbines and small portions in fuel cells. The estimated amount of fuel cell capacity for energy storage is sensitive to assumptions for combustion turbines versus fuel cells for energy storage applications.

This analysis estimated materials, subcomponents, and components needed for the construction of the fuel cell and electrolyzer stack itself, but it did not estimate any balance-of-plant material or equipment requirements. Though balance-of-plant material usage is not insignificant, the materials used for these subsystems are generally less critical than the specialty materials used in the stack itself. Additionally, power electronics used to control and condition power supplied to the system are considered in other reports that are part of this series ("Semiconductor Supply Chain Deep Dive Assessment" 2022)



Figure 4. Estimated cumulative global (a) and U.S. (b) manufactured capacity of fuel cells by type. Estimated annual global (c) and U.S. (d) manufactured capacity of fuel cells by type (Original work)

Values include replacement systems manufactured to replace those at end-of-life. Replacing existing capacity at the end-of-life drives some of the variability and peaks in annual manufacturing rates shown in Figures c and d.

As reported above, the market for these technologies is anticipated to increase through 2050, thereby increasing the demand for materials of construction. Cumulative material requirements by 2050 are shown in Figure 5, illustrating the significant number and amount of materials required to produce the electrolyzer and fuel cell systems shown in Figure 4. The large amounts of stainless steel and titanium are driven by the use of these materials in bipolar plates in PEMECs and SOECs. Bipolar plates are thicker than other components and are composed of pure metal, making the amount of material per megawatt of system capacity higher than that of other components.



Figure 5. U.S. cumulative 2050 use by material. Total use of each material across technologies is shown in yellow points.(Original work)

The y-axis of this chart is on a log-scale. For many materials, the smaller use rate is negligible compared to the larger one so the difference between the total use and the larger use may not be noticeable.

Though titanium and stainless steel are used in the highest quantities, they are not necessarily the more important materials of those shown. Both high and low-temperature systems rely on more-exotic materials, such as iridium and yttrium in their construction. Though the total amounts of these materials required are lower, they are also less abundant, and mines that produce them are more likely to be located outside the United States. These external factors, which influence the "criticality" of various materials, are discussed in detail in the following sections.

In addition to preliminary estimates of raw materials required for catalyst and supporting components of the electrolyzer stack, the amount of polymer electrolyte membrane material required by PEMFC and PEMEC systems is estimated (Figure 6). The polymer electrolyte membranes used in these systems is based on PFSA ionomers that allow for transport of protons and acting as an electrical insulator and barrier to oxygen and hydrogen. The production of PFSA membranes uses solution casting technology, where a PFSA polymer dispersion is applied to a base film that then undergoes quality control inspection and packaging (Curtin et al. 2004). This analysis uses of PFSA-based ion exchange membranes given the significant increase in demand for these materials suggested by Figure 6 and the few suppliers currently meeting the small demand; along with possible environmental concerns associated with their production warrant additional consideration (Lohmann et al. 2020; Cousins et al. 2019).



Figure 6. U.S. annual (a) and cumulative (b) use of PFSA polymer electrolyte membrane in PEMFC and PEMEC systems from current to 2050 (Original work)

Advances in the design of electrolyzers and fuel cells could reduce material use. Changes in several performance and design characteristics of these systems could result in lower material demand per kilogram of hydrogen produced. The loading rates (mg/cm²) of catalyst materials represent a key opportunity to reduce the rate at which these catalysts are used in electrochemical systems. Additionally, ensuring systems can operate over longer lifetimes reduces the need for their replacement and requires less materials. Finally, higher efficiency electrolyzers that produce more hydrogen per kilowatt-hour (kWh) of energy consumed will more effectively meet hydrogen demand, and in turn require fewer systems and materials. This analysis assumes constant system performance and catalyst loading rates, but changes in these factors can significantly impact the materials required to manufacture a system. Because reducing or even eliminating use of critical materials generally reduces capital costs as well, doing so is the subject of significant ongoing research.

In addition to advances in the technology itself, progress in recyclability and recycling infrastructure for electrochemical systems could reduce demand for new mines and materials. The ability to recycle critical materials at high recovery rates is a key opportunity to address increasing material needs as demand for these systems increases. Realizing this goal requires systems that are designed for recycling and avoid the use of coatings or designs that reduce the recovery rate of critical materials; for example, the recycling process of metal bipolar plates that are coated in a thin layer of platinum/gold requires more equipment and is likely to be more difficult than recycling plates without the coating.

2. Supply Chain Mapping

Because PEM and SOC are developing technologies, supply chains for them have yet to be established. Nonetheless, assessing the current state of supply chain elements helps identify the potential constraints and opportunities for the United States as these supply chains build up to support increased demand. For each technology, we describe the supply chain and industry structure at a high level and provide insight into current U.S. resilience and competitiveness. We also discuss recycling opportunities and current national policies and incentives.

2.1 PEMEC and PEMFC Systems

2.1.1 Supply Chain Overview by Segment

The key elements of the PEM fuel cell and PEM electrolyzer manufacturing supply chains—raw materiak, processed materials, subcomponents, and end products, a long with end-of-life material recovery opportunities—are highlighted in Figure 7.



Figure 7. Key elements of PEMFC and PEMEC supply chains

EOL is end-of-life.

2.1.1.1 Industry Structure

Today's nascent PEMFC and PEMEC industry is made up of fairly few suppliers across the supply chain (James et al. 2018). Many key players are large companies (e.g., 3M, DuPont, and Cummins), but the fuel cell/electrolyzer business is only a small portion of their business profiles (BNEF 2021). One of the largest PEMFC manufacturers, Ballard Power, produces most of the subcomponents (i.e., bipolar plates, gas diffusion layer, and electrolyte membrane) in-house. Other suppliers typically produce one or two subcomponents in the supply chain (Table 4 on page 16 and Appendix B), but none currently has the capacity to produce fuel cell systems and components at high manufacturing rates (James et al. 2018). To a large extent, the PEMEC industry

has benefited from progress in PEMFC R&D and it expects to be able to leverage PEMFC manufacturing supply chains as they develop.

2.1.2 Current U.S. Resilience

Because these technologies are nascent and markets have yet to grow, global demand for PEMECs and PEMFCs is fairly low. Consequently, global manufacturing capacity for these technologies is low and supply chains to support manufacturing have not yet developed as is discussed in the Electrolyzer and Fuel Cell Market Size Estimates section above. The United States has the potential to build domestic capacity as demand for hydrogen and subcomponents technologies grows. Current U.S. resilience is summarized in Table 2 in terms of strengths and weaknesses in production capabilities, innovation and technology, workforce, policy, and infrastructure. Since water electrolysis and fuel cells are a nascent industry, the table focuses on the current status and will change as the industry evolves and policies and initiatives are established. The table will need to be updated as those occur.

	Strengths	Weaknesses
Existing U.S. production	Sufficient U.S. manufacturing capacity to meet current demand	U.S. manufacturing capacity may not be sufficient to meet growing demands.
capabilities	for PEM electrolyzers and fuel cells, catalyst, membrane, gas diffusion layer (GDL), and bipolar	Reliance on imports of key materials especially platinum, iridium, and graphite.
	plate subcomponents	High manufacturing cost for fuel cell/electrolyzer components and lack of high-throughput assembly processes
Emerging U.S. production	Presence of high-technology domestic industries including	Reliance on imports platinum, iridium, and graphite
capabilities	automotive, electrolysis, and chemical processing	Meeting expanded demand for PEM electrolyzer and fuel cell manufacturing in the United States requires technical advancements, capital investment, and demonstration of higher production volumes of fuel cells, electrolyzers and subcomponents.
		Growth in Asian markets is likely to outpace the rest of the world. Demand will most likely be met by Asian suppliers, who can leverage manufacturing economies of scale to outcompete U.S. suppliers. European investment and targets (e.g., 40 GW of electrolysis) are driving investment in several GW-scale manufacturing plants in Europe, including by U.S. companies.
Innovation and technology	United States' leadership in innovation and strong innovation ecosystem	Increasing R&D investments outside the United States. However, the United States is involved in international collaborations including the International Partnership for Hydrogen and
	Robust R&D funding at national laboratories and academia (often	Fuel Cells in the Economy

Table 2. Current U.S. Resilience (Strengths and Weaknesses) of PEMFC and PEMEC Supply Chain

	Strengths	Weaknesses
	in partnership with private industry)	(https://www.iphe.net/) and European Collaborations to leverage international knowledge and support partnerships.
Workforce	Skilled labor Access to educated workforce	Limited pool of trained workers with expertise in hydrogen and fuel cell technologies
U.S. policy	Import/export policies (no tariffs)	Lack of coordinated incentives/facilitation
	Buy America incentivizes domestic components	Lack of tax liability and reduced value of tax credits for emerging industry
	Support from federal and state programs including development of a National Hydrogen Strategy and Roadmap as required by section 40314 of the Infrastructure Investment and Jobs Act (Infrastructure Investment and Jobs Act 2021)	Without coordinated implementation of domestic manufacturing content requirements, insufficient access to domestic industrial supplies may increase costs and reduce competitiveness in global markets
U.S. infrastructure	Reliable low-cost electricity supply and growth of renewables U.S. road, rail, and coastal port infrastructure for moving freight	Lack of U.S. hydrogen infrastructure Limited dedicated hydrogen transmission and storage infrastructure including interregional connections
Sources: (Mayyas	and Mann 2019); (Fullenkamp et al. 2017)	

2.1.3 U.S. Competitiveness

Available information on PEMECs is limited due to the nascent markets and the focus on fuel cells during recent decades. Thus, we expect the U.S. status with electrolyzers to be similar to that of fuel cells.

Based on a recent EU fuel cell supply chain assessment of all fuel cell types (European Commission 2020), the United States and Asia are the current leaders in fuel cell and fuel cell subcomponent manufacturing. Moving up the supply chain, manufacturing is less concentrated. For processed materials, the United States is relatively well-positioned. However, the United States depends almost completely on other countries for raw materials used to manufacture fuel cells. A related study by the European Commission Joint Research Centre (Blagoeva et al. 2020) provided details on some key processed materials in the current fuel cell supply chain in the Directorate-General assessment. This JRC analysis shows that the United States is a global supplier for some portion of the fuel cell processed materials: 60% of ionomer (Nafion) (for electrolyte membrane); 20% of carbon cloth/paper (for GDLs); and 27% of stainless steel and 30% of carbon fiber (for bipolar plates). Unlike China, which relies on imports across the supply chain, the United States and Japan can currently supply most of their fuel cell subcomponents with domestic manufacturing (Xun et al. 2021). As discussed in the Market Size Estimate section above, the market is likely to grow dramatically. Thus, the supply chain in the future may differ from the current supply chain; the current status may no longer be applicable and the United States may become increasingly dependent upon imported systems, subcomponents, processed materials, and/or raw materiaks. Policies may need to evolve rapidly to support manufacturing capacity as the market grows.

2.1.3.1 Key Manufacturers

Key global players—major companies and their locations (both headquarters and manufacturing locations if known)—in the PEMEC and PEMFC manufacturing supply chains are summarized in Table 3 and Table 4, respectively. Unlike mature solar technology, PEMEC and PEMFC are not yet widely commercialized and do not have established manufacturing supply chains which are tracked in industry market reports that provide standardized information on manufacturing locations and capacities. However, by cross referencing (1) information found on company websites, reports, and press releases with (2) high-level market highlight webpages with information from Bloomberg New Energy Finance (BNEF), we identified major manufacturers of systems and components. Public information on PEMEC manufacturing is limited and provides little insight into upstream manufacturing of components, although many of the fuel cell supply chain companies also offer components for electrolyzers. These tables identify key sellers of the end product and specific components. Some companies are vertically integrated, at least to a certain extent, and produce many of the components but sell only the end product. Intermediate components are not listed for those companies except where noted. An expanded list of manufacturers and estimates of company manufacturing capacities are provided in Appendix B. Not all companies are listed and the data represents only one snapshot in time. Given the rapid pace of growth and the number of acquisitions, emerging companies, and joint ventures in numerous countries, the data in these tables will change frequently.

Company	Headquarters	Manufacturing Location	Product
Elogen (subsidiary of GTT, recently rebranded from Areva H2Gen) ("GTT Group"; "Elogen")	Les Ulis, France	France	PEMEC
Hydrogenics – subsidiary of Cummins (Cummins 2019)	Mississauga, Ontario, Canada	Ontario, Canada; Indiana and California, US	PEMEC
lon Power	New Castle, DE, US	Delaware and Pennsylvania, US	Membrane
ITM	Sheffield, UK	UK	PEMEC
Kobelco Eco-solutions	Kobe, Japan	Japan	PEMEC
Nel ASA (Løkke 2021)	Oslo, Norway	Connecticut, US; Norway; Denmark	PEMEC
Plug Power ("Plug Power Green Hydrogen & Fuel Cell Solutions")	Latham, NY, US	New York, US	PEMEC
Siemens Energy (Siemens Energy 2020)	Munich, Germany	Unavailable	PEMEC

Table 3. Select PEMEC Manufacturers

Included companies were mentioned either by at least three market reports or by two market reports and BNEF. Companies for which the electrolyzer type is unavailable are not included (see Appendix B for details). Note that manufacturing locations listed are specifically for PEMECs, and the list may not be a complete for a given company. Details, select capacities/upcoming developments, and citations are in Appendix B.

Table 4. Select PEMFC and Component Manufacturers

Company	Headquarters	Manufacturing Locations	Products
3M	St. Paul, MN, US	Minnesota and Wisconsin, US	Membrane, lonomer
Advent	Athens, Greece; Boston, MA, US	Greece; MA, US	Membrane, MEA
Ballard Power Systems	Burnaby, British Columbia, Canada	British Columbia, Canada; Denmark	PEMFC, MEA*, BPP*
BASF	Ludwigshafen, Germany	Germany	MEA, Catalyst
Chemours	Wilmington, DE, US	South Carolina, US	Membrane, lonomer
Horizon Fuel Cell	Singapore	Singapore; China	PEMFC
Hydrogenics – subsidiary of Cummins (Cummins 2019)	Mississauga, Ontario, Canada	Ontario, Canada	PEMFC
Intelligent Energy	Loughborough, UK	UK	PEMFC
lon Power	New Castle, DE, US	Delaware and Pennsylvania, US	MEA
Johnson Matthey	London, UK	Pennsylvania, US; UK	MEA, Catalyst
Plug Power, Inc. ("Plug Power Green Hydrogen & Fuel Cell Solutions")	Latham, NY, US	New York, US	PEMFC, MEA*
Solvay	Brussels, Belgium	NJ, US; Italy	Membrane, lonomer
TANAKA	Tokyo, Japan	Japan	Catalyst
Umicore	Brussels, Belgium	China, US, Germany, Denmark and Korea	Catalyst
W. L. Gore	Newark, DE, US	Delaware, US; Japan	Membrane

Included companies were mentioned by at least four market reports and could be found in the BNEF database, and/or they are known to the DOE's Hydrogen and Fuel Cell Technologies Office as a manufacturer of fuel cell components. Note that manufacturing locations listed are specifically for fuel cell components, and the list may not be complete for a given company. Details, select capacities/upcoming developments, and citations are in Appendix B.

*Component is produced for use in company end-products but is not sold directly.

2.1.3.2 U.S. Position: U.S. Competitiveness in PEMFC Subcomponents

The DOE Hydrogen and Fuel Cell Technologies Office funded a PEMFC supply chain study (Fullenkamp et al 2017) that reviewed U.S. competitiveness for key subcomponents in the MEA. Findings from that study, which was published in 2017, are summarized here in alphabetical order:

Bipolar Plates: Because the BPP component is ultimately expected to be manufactured close to the fuel cell system assembly site, BPPs are expected to be produced in the United States as long as demand continues. Currently, Europe and Asia hold the lead in BPP technology. However, there is a substantial opportunity for the United States to innovate in plate formation, coatings, and joining.

Catalyst: Europe (Umicore, Johnson Matthey) and Asia (Tanaka) are currently the world leaders in fuel cell catalyst technology. Given the long development lead time and other barriers to market entry, this is likely to continue for many years. Overall prospects for U.S. catalyst production competitiveness are low in the near term and low to moderate in the far term. U.S. innovation competitiveness is moderate.

GDL: Four main competitors predominate and are located in Europe (SGL, Freudenberg), Asia (Toray), and the United States (AvCarb). The United States does not seem to enjoy any clear advantages over other regions.

Membrane: The United States currently holds the global lead in membrane technology and will likely continue to innovate. The ionomer is likely to be produced in the future in large quantities at foreign sites (probably China). U.S.-based W.L. Gore Inc. is currently the world leader in expanded polytetrafluoroethylene (ePTFE) membrane support, although fuel cell membrane production currently occurs in Japan for the Asian market. Other U.S.-based companies (e.g., 3M and Giner) have development efforts in non-ePTFE supports. Roll-to-roll/casting membrane fabrication techniques are expected to be used in the future. Though the United States is competitive in this general field, Europe and Asia are also strong. Additionally, localized production of the catalyst-coated membrane/MEA may be favored over remote centralized production with shipping of value-added components.

Additional information on key global players and their expansion plans is provided in Appendix B although the information reported there will change often and thus it may be inaccurate.

2.2 Solid Oxide Electrolysis and Fuel Cells

2.2.1 Supply Chain Overview by Segment

The supply chain for solid oxide electrolysis and fuel cells consists of five key segments: the necessary raw materials, processed functional materials, subcomponents, the end product, and end-of-life recovery. Figure 8 highlights the function of each of the critical materials in the production of SOC cells and stacks as well as end-of-life material recovery opportunities.

Raw Materials		rocessed Aaterials	Subcompo	nents	End Product	EOL Recovery
Lanthanum (La) Strontium (Sr) Cobalt (Co) Iron (Fe) Manganese (Mn)	LSM LSM-YSZ LSCF	Air Electrode Materials	Solid Air Electrode			SOC Nickel from fuel
Yttrium (Y) Zirconium (Zr)	8-YSZ	Electrolyte	Solid Electrolyte			electrodes, interconnects
Nickel (Ni)	Ni-YSZ	Fuel Electrode Materials	Solid Fuel Electrode	Solid Oxide Cell	Solid Oxide Cell Stack	YSZ from spent electrolyte
Stainless Steel	SS-441	Frame, Separator Plate, Electrode Flow Fields, End Plates	Cell			
Borosilicate Glass	Boric Oxide Silica Glass	Spacers	- Interconnects			

Figure 8. Key elements of the SOC supply chain

2.2.1.1 Industry Structure

The current industry structure for development of SOC-processed materials, subcomponents, and end products is centered on a select few commercial developers. Several materials in solid oxide systems, including yttrium, strontium, and manganese, come entirely from imports, and these are typically obtained directly by developers as needed. Other SOC critical materials of which a significant portion are obtained as imports include nickel and cobalt. The typical cell geometry is fuel electrode supported, with the fuel electrode being comprised of Ni-YSZ. The air electrode of the cell relies on the availability of materials such as lanthanum, strontium, cobalt, manganese, and iron, which are needed in significantly lower quantities. The U.S. Geological Survey's "Mineral Commodity Summaries 2021" discusses the production and current use of several of these materials (USGS "Mineral Commodity Summaries 2021" 2021).

Nexceris is a primary producer of processed material that is either sold commercially or used internally to produce individual cells and stacks. Developers, such as Bloom Energy, FuelCell Energy, and Cummins, typically purchase or generate their own processed materials as part of their cell, stack, and system development process. As part of that process, stainless steel materials are used for bipolar plate material, borosilicate glass is used for spacers, and other similar materials are used as needed based on particular designs. Bloom Energy and FuelCell Energy are currently the largest suppliers of end-product SOC technology in the United States, and several smaller companies are working to increase production.

2.2.2 Current U.S. Resilience

The U.S. commercial development structure is organized for converting critical materials into processed materials, individual cells, cell stacks, and finally the system through a tailored manufacturing process. The process begins with combining the critical materials into processed materials to make the appropriate subcomponents. The cell electrodes and electrolyte processed material, which is produced by each developer, is tailored to meet specific composition and microstructure requirements. Processed materials such as LSM and LSCF are used to reduce oxygen molecules supplied from air into oxygen ions in SOC electrodes. The electrolyte, which is typically a YSZ material, transports the oxygen ions to active reaction sites. The fuel

electrode is typically Ni-YSZ, which serves as a catalyst and electron transport material for the electron exchange reaction.

Industry has developed advanced manufacturing methods for producing the needed quantity of SOCs, stacks, and systems (including incorporation of balance-of-plant equipment) that is tailored to their production processes, resulting in a final SOC system. Balance-of-plant equipment can include, but is not limited to air blowers, heat exchangers, fuel and product storage, and inverters. There is significant R&D investment in the development of these balance-of-plant materials specifically for solid oxide system technologies, and the investment is especially geared toward distributed generation. Current U.S. resilience is summarized in Table 5.

Table 5. Current U.S. Resilience (Strengths and Weaknesses) of Solid Oxide Electrolyzer and Fuel Cell Supply	
Chain	

	Strengths	Weaknesses
Existing U.S. production capabilities	Sufficient production capacity to meet current demand	Significant volume expansion needed to meet anticipated demand; reliance on imports of yttrium, strontium, manganese, nickel, and cobalt
		High manufacturing cost for complex ceramic materials
Emerging U.S. production capabilities	U.S. development structure's focus on scaling up current manufacturing process for higher volume production of cells and stacks	Scale up in production not fully realized in United States, could be challenged due to regulations, component availability, or other unforeseen challenges
		U.S. production outpaced by Europe and Asia
Innovation and technology	United States' leadership in innovation and strong innovation ecosystem	Potential commercial reliance on government funding.
	Robust R&D funding at national laboratories, academia, and commercial developers that is well- supported by U.S. government agencies including a Clean Hydrogen Electrolysis Program as required by section 40314 of the Infrastructure Investment and Jobs Act (Infrastructure Investment and Jobs Act 2021).	
Workforce	Access to educated and skilled labor from a global perspective	Limited pool of trained workers with expertise in hydrogen and fuel cell technologies
U.S. policy	Support from federal and state programs including development of a National Hydrogen Strategy and Roadmap as required by section	Expiring subsidies resulting in a high-cost solution; lack of coordinated incentives/facilitation

	Strengths	Weaknesses
	40314 of the Infrastructure Investment and Jobs Act (Infrastructure Investment and Jobs Act 2021) Localized subsidies	Lack of tax liability and reduced value of tax credits for emerging industry
U.S. infrastructure	Reliable, low-cost electricity supply Global leadership of U.S. road, rail, and coastal port infrastructure for moving freight	Limited access to excess renewable energy and nuclear/thermal sources for electrolysis operation, hydrogen fuel for fuel cell operation Limited dedicated hydrogen transmission and storage infrastructure including interregional connections

2.2.3 U.S. Competitiveness

SOFC and SOEC technologies are not yet widely commercialized and do not have established manufacturing supply chains. By cross referencing (1) information found on company websites, reports, and press releases with (2) high-level market highlight webpages with information from BNEF, we identified major manufacturers of systems and components. Some information is given here, and a high-level summary is provided in Table 6 (page 21). Details are provided in Appendix B. Not all companies are listed and the data represents only one snapshot in time. Given the rapid pace of growth and the number of acquisitions, emerging companies, and joint ventures in numerous countries, the data in these tables will change frequently.

Currently, Europe is leading in solid oxide electrolysis development, and U.S. domestic commercial developers are competing with several European and Asian SOC developers (Hauch et al. 2020). Primary domestic developers include Bloom Energy, FuelCell Energy, and Cummins, and several smaller companies are increasing production to meet anticipated future demand. Their competition includes Mitsubishi Power, Kyocera, Hitachi, and SOLIDPower.

Domestic manufacturers are at a disadvantage because the global market for higher-cost, but more-efficient generators of electricity and fuel is much more prominent than the domestic market, where energy is available at lower cost. Additional information on key global players and their expansion plans is provided in Appendix B.

The United States is most competitive in the production of the processed materials needed to produce SOC components, the production of the necessary interconnect components, and the final construction of the stacks and systems (including balance-of-plant equipment). One primary developer of the processed materials is Nexceris; it sells processed materials and individual cells for commercial uses and for R&D needs. Domestic commercial developers of SOC stacks and systems either purchase processed materials from a company such as Nexceris or develop those materials in-house.

For raw material production, there is continued opportunity for the United States to be a leader in the production of stainless steels (especially SS-441), zirconium, and nickel. Borosilicate glass is another opportunity for U.S. production. Additional opportunity is available in the resumption of previous domestically produced materials for use in SOC technology, including cobalt, strontium, and manganese.

Company	Headquarters	Manufacturing Location	Products
Bloom Energy	San Jose, CA, US	California and Delaware, US	Bloom Energy Server (SOFC)
FuelCell Energy	Danbury, CT, US	Alberta, Canada	Molten carbonate fuel cells, SOFC, SOEC
Cummins, Inc.	Columbus, IN, US	United States, Canada	PEMFC, SOFC, engines
Nexceris	Lewis Center, OH, US	United States	Processed SOC materials, SOFC, SOEC
Mitsubishi Power, Ltd.	Yokohama, Japan	Japan	SOFC, hydrogen turbines
Kyocera	Kyoto, Japan	Japan	SOFC, solar cells, advanced ceramics
SOLIDPower	Mezzolombardo, Italy	Germany	BlueGen SOFC
Hitachi	Toshiaki, Japan	Japan	SOFC, SOEC
Robert Bosch	Gerlingen, Germany	Germany	SOFC
Special Power Sources	Alliance, OH, US	United States	Tubular SOFC
Ceres Power Holdings	West Sussex, UK	UK	SOFC, SOEC
OxEon Energy	North Salt Lake, UT, US	United States	SOEC, SOFC

Table 6. Maior Solid Oxide Fuel	Cell and Solid Oxide Electrolysis	Cell and Stack Manufacturers

2.3 Recycling Potential for Electrolyzers and Fuel Cells

End-of-life strategies for fuel cells and electrolyzers focus primarily on recovery of precious metals used in the catalyst subcomponent. Improvements in the technologies currently used to recover spent catalysts (hydrometallurgical and pyrometallurgical processes) can help reduce costs and mitigate regional supply constraints associated with these materials. Other high-value and hazardous materials in fuel cells and hydrogen technology components also offer the potential for recycling, reuse, or both (HyTechCycling 2018b). Additional details on recycling platinum group metals (PGM) from fuel cells and electrolyzers is provided in Supply Chain Review: Platinum Metal Group Catalysts ("Supply Chain Review: Platinum Metal Group Catalysts" 2021).

PEMFC Recycling Opportunities include (HyTechCycling 2018a):

- **Bipolar Plates:** Alteration of the physical and chemical structure of bipolar plate materials (gold, platinum, graphite, stainless steel and carbon composites) during operation precludes reuse or recovery for the same original application. Materials recovered from bipolar plates could potentially be used as insulation raw material for electronic devices or in steel manufacturing (i.e., open-loop recycling).
- MEA: Existing technologies (hydrothermal and hydrometallurgical) and novel technologies (acid process, transient dissolution, and selective electrochemical dissolution) can be used to recover PGM from the electrodes along with other valuable materials, such as ionomers from the membrane or the carbon support of the noble catalyst. Alcohol dissolution can also be used as pretreatment for the recovery of ionomers from the membrane before the catalyst recovery process. The recovery of MEA's critical materials allows recycling in a closed-loop scheme, which means a potentially higher benefit with respect to open-loop recycling.

PEMEC Recycling Opportunities include (HyTechCycling2018a):

- **Bipolar Plates:** Though titanium can generally be recovered through conventional methods based on physical separation (size reduction and magnetic separation), recovering it from the titanium alloys used for bipolar plates requires more-complex processes (e.g., hydrometallurgical processes).
- MEA: Similar to PEMFC stacks, novel and existing end-of-life technologies can be used to recover the critical materials of the PEMEC MEA. The anode electrocatalyst is commonly iridium (although ruthenium is being researched as an alternative), which can be recovered through existing (pyrometallurgical and hydrometallurgical processes) or novel (transient dissolution) methods. As with PEMFCs, these end-of-life technologies would facilitate closed-loop recycling schemes.

SOC Recycling Opportunities include (HyTechCycling2018a):

- Fuel Electrode Materials/Interconnects: Existing end-of-life technologies (hydrothermal and hydrometallurgical technologies) are applicable for nickel recovery from the fuel electrode and interconnects.
- Electrolyte: YSZ can be recovered from the spent electrolyte using hydrothermal technologies for use in open-loop recycling—for example, for its application in electrical/electrochemical sectors with a lower purity requirement than the SOC fuel electrode or electrolyte. Alternatively, the ceramic composite material can—after grinding and mechanical separation—be recycled, still in an open-loop scheme, for use in construction applications.
- Air Electrode Material/Interconnects: For lanthanum compounds (LSM and LSCF), no recovery process is currently available and due to their hazardous nature, they are disposed in hazardous waste landfills.

Regarding stack components, no novel technologies are found to be applicable for the recovery of critical materials.

Table 7 summarizes existing and novel recovery technologies applicable to critical materials of fuel cell and electrolyzer stacks. Metals belonging to the platinum group (Pt, Ru, and Ir), whose reserves are depleting, and which are associated with high economic costs, are the materials for which most of the end-of-life technologies

have been found. In contrast, information that is as detailed as it is for PGMs is unavailable for novel end-of-life technologies applicable to rare earth elements (e.g., lanthanum compounds and YSZ) and nickel-based materials.

Device	Component	Meterial	Critical	Recovery Te	Technologies	
Device	Component	Material	Aspects	Existing	Novel	
	Anode	Pt	Cost, supply risk	HMT; PMT	SED; TD; AP	
PEMFC	Cathode	Pt	Cost, supply risk	HMT; PMT	SED; TD; AP	
	Electrolyte	lonomer	Cost, hazard ^a	n/a	AD; AP	
	Anode	lr; Ru	Cost, supply risk, hazard	HMT; PMT	TD	
PEMEC	Cathode	Pt	Cost, supply risk	HMT; PMT	SED; TD; AP	
	Electrolyte	lonomer	Cost, hazard ^a	n/a	AD; AP	
	Bipolar plate	Ti	Cost	HMT	n/a	
SOC	Fuel electrode	YSZ	Cost, supply risk	HDT	n/a	
		Ni; NiO	Hazard	HDT; HMT	n/a	
	Air Electrode	LSM, LSCF	Hazard, supply risk	n/a	n/a	
	Electrolyte	YSZ	Cost, supply risk	HDT	n/a	
		Ni; NiO	Hazard	HDT; HMT	n/a	
	Interconnects	LSC	Hazard, supply risk	n/a	n/a	
HDT: hydrothermal technology; HMT: hydrometallurgical technology; PMT: pyrometallurgical technology; TD:						

Table 7. Materials in Fuel Cells and Electrolyzers With Recovery Potential (HyTechCycling 2018a)

HDT: hydrothermal technology; HMT: hydrometallurgical technology; PMT: pyrometallurgical technology; TD: transient dissolution: AP: acid process; SED selective electrochemical dissolution: AD: alcoholdissolution; PEMEC: PEMWE (proton exchange membrane water electrolyzer) is the acronym used in the cited HyTechCycling Report.

^a Concerns linked to hydrogen fluoride (HF) emissions if the membrane is incinerated

2.4 Current National Policies and Incentives for Electrolyzers and Fuel Cells

The development and deployment of hydrogen and fuel cell technologies to date has been driven primarily by government policies and incentives focused on addressing larger goals of addressing climate change, reducing emissions, and advancing clean energy technologies. A number of nations and the European Union have developed policies and strategies for the hydrogen economy. The IEA listed a number of targets and policies and their list is included in Table 8. Note that Table 8 includes nonbinding road maps, strategies, and targets as well as investments that are, in some cases, legally binding incentives. While the United States has technology development targets and an RD&D plan (Department of Energy (DOE) 2020), it does not currently have hydrogen deployment targets or a national plan, unlike other countries. However, the United States is developing
a national plan as required by section 40314 of the Infrastructure Investment and Jobs Act (Infrastructure Investment and Jobs Act 2021). Appendix C lists existing U.S. policies.

Origin	Document, year	Deployment targets (2030)	Production	Uses	Public investment committed
Australia	<u>National</u> <u>Hydrogen</u> <u>Strategy</u> , 2019	• None specified	 Coal with CCUS Electrolysis (renewable) Natural gas with CCUS 	 Building Electricity Exports Industry Shipping Transport 	AUD 1.3B (~USD 0.9B)
Canada	<u>Hydrogen</u> <u>Strategy for</u> <u>Canada</u> , 2020	 Total use: 4 MMT H₂/yr 6.2% TFEC 	 Biomass By-product H₂ Electrolysis Natural Gas with CCUS Oil with CCUS 	 Buildings Electricity Exports Industry Mining Refining Shipping Transport 	CAD 25M by 2026 ¹ (~USD 19M)
Chile	National <u>Green</u> <u>Hydrogen</u> <u>Strategy</u> , 2020	• 25 GW electrolysis ²	 Electrolysis (renewable) 	 Buildings Exports Industry (chemicals) Mining Refining Transport 	USD 50M for 2021
Czech Republic	<u>Hydrogen</u> <u>Strategy</u> , 2021	 Low-carbon demand: 0.097 MMT H₂/yr 	Electrolysis	 Industry (chemicals) Transport 	n/a
European Union	EU Hydrogen Strategy, 2020	• 40 GW electrolysis	 Electrolysis (renewable) Transition role of natural gas with CCUS 	IndustryRefiningTransport	EUR 3.77B by 2030 (~USD 4.3B)
France	Hydrogen Deployment Plan, 2018 National Strategy for Decarbonised Hydrogen Development, 2020	 6.5 GW electrolysis 20%–40% industrial H₂ decarbonized³ 20,000–50,000 light-duty FCEVs³ 800–2,000 FC heavy-duty FCEVs³ 400–1,000 HRSs³ 	• Electrolysis	IndustryRefiningTransport	EUR 7.2B by 2030 (~USD 8.2B)

 Table 8. Hydrogen Targets and Policies Identified by the International Energy Agency (IEA 2021a)

Origin	Document, year	Deployment targets (2030)	Production	Uses	Public investment committed
Germany	<u>National</u> <u>Hydrogen</u> <u>Strategy</u> , 2020	• 5 GW electrolysis	 Electrolysis (renewable) 	 Aviation Electricity Industry Refining Shipping Transport 	EUR 9B by 2030 (~USD 10.3B)
Hungary	<u>National</u> <u>Hydrogen</u> <u>Strategy</u> , 2021	 Production: 0.02 MMT/yr of low-carbon H₂ 240 MW electrolysis Use: 0.034 MMT/yr of low-carbon H₂ 4,800 FCEVs 20 HRSs 	 Electrolysis Fossil fuels with CCUS 	Electricity	n/a
Japan	Strategic Roadmap for Hydrogen and Fuel Cells, 2019 Green Growth Strategy, 2020, 2021 (revised)	 Total use: 3 MMT H₂/yr Supply: 420 kT low-carbon H₂ 800,000 FCEVs 1,200 fuel cell buses 10,000 fuel cell forklifts 900 HRSs 3 MMT NH₃ fuel demand⁴ 	 Electrolysis Fossil fuels with CCUS 	 Buildings Electricity Industry (steel) Refining Shipping Transport 	JPY 699.9B by 2030 (~USD 6.5B)
Korea	Hydrogen Economy Roadmap, 2019	 Total use: 1.94 MMT H₂/yr 2.9 million fuel cell cars⁵ 1,200 HRSs⁵ 80,000 fuel cell taxis⁵ 40,000 fuel cell buses⁵ 30,000 fuel cell trucks⁵ 8 GW stationary fuel cells (plus 7 GW exported)⁵ 2.1 GW of micro- cogeneration fuel cells⁵ 	 By-product H₂ Electrolysis Natural gas with CCUS 	BuildingsElectricityTransport	KRW 2.6T in 2020 (~USD 2.2B)

Origin	Document, year	Deployment targets (2030)	Production	Uses	Public investment committed
Netherlands	National Climate Agreement, 2019 Government Strategy on Hydrogen, 2020	 3–4 GW electrolysis 300,000 fuel cell cars 3,000 heavy-duty FCEVs⁶ 	 Electrolysis (renewables) Natural gas with CCUS 	 Aviation Buildings Electricity Industry Refining Shipping Transport 	EUR 70M/yr (~USD 80M/yr)
Norway	Government Hydrogen Strategy, 2020 Hydrogen Roadmap, 2021	● n/a ⁷	 Electrolysis (renewables) Natural gas with CCUS 	IndustryShippingTransport	NOK 200M for 2021 (~USD 21M)
Portugal	<u>National</u> <u>Hydrogen</u> <u>Strategy</u> , 2020	 2.0–2.5 GW electrolysis 1.5%–2% TFEC 1%–5% TFEC in road transport 2%–5% TFEC in industry 10%–15% by volume H₂ in gas grid 3%–5% TFEC in maritime transport 50100 HRSs 	• Electrolysis (renewables)	 Electricity Industry Transport 	EUR 900M by 2030 (~USD 1B)
Russia	<u>Hydrogen</u> <u>roadmap,</u> 2020	• Exports: 2 MMT H ₂	 Electrolysis Natural gas with CCUS 	 Electricity Industry Refining Exports 	n/a
Spain	<u>National</u> <u>Hydrogen</u> <u>Roadmap</u> , 2020	 4 GW electrolysis 25% industrial H₂ decarbonized 5,000–7,500 FCEVs 150–200 fuel cell buses 100–150 HRSs 	 Electrolysis (renewables) 	 Aviation Electricity Industry (chemicals) Refining Shipping Transport 	EUR 1.6B (~USD 1.8B)

Origin	Document, year	Deployment targets (2030)	Production	Uses	Public investment committed
United Kingdom	<u>UK Hydrogen</u> <u>Strategy</u> , 2021	 5 GW low-carbon production capacity 	 Natural gas with CCUS Electrolysis 	 Aviation Buildings Electricity Industry Refining Shipping Transport 	GBP 1B (~USD 1.3B)

Note: TFEC = total final energy consumption, HRS = hydrogen refueling station. For investments, M = million, B = billion, T = trillion.

¹In addition to CAD 25M, Canada has committed over CAD 10B to support clean energy technologies, including H₂.

²This target refers to projects that at least have funding committed, not to capacity installed by 2030.

³Target for 2028.

⁴From the interim Ammonia Roadmap.

⁵Target for 2040.

⁶Target for 2025 from the National Climate Agreement, 2019 (currently under revision).

⁷Norway's strategy defines targets for the competitiveness of hydrogen technologies and project deployment.

3. Supply Chain Risk Assessment

Because the supply chain for electrolyzers and fuel cells is still to be established as demand grows, the risk assessment we performed for this report focused on the key materials required for electrolyzers and fuel cell production and does *not* assess subcomponents. The materials required to manufacture PEMFCs and PEMECs that meet future electrolytic hydrogen demand of around 100 MMT/yr could be significantly higher than 2020 levels of apparent consumption² for many of these materials (Figure 3 and Figure 4, pages 8 and 9) (USGS 2021). In this report, apparent consumption is defined in a manner consistent with the USGS, where it equals primary production plus secondary production plus imports minus exports for a given material. Projected apparent consumption greater than current rates suggests that securing and expanding the supply chains for these materials is critical to enabling the high rates of deployment of electrolyzers and fuel cells shown here.

Materials whose demand is projected to increase and which are currently imported into the United States at high percentages can be considered most critical. These two parameters are plotted for key electrolyzer and fuel cell materials in Figure 9, which shows the relationship between these variables for key materials. The y-axis value in Figure 9 depends on manufactured capacity and material usage for electrolyzers and fuel cells and is reported as a range between baseline and low material use scenarios. The upper bound baseline material use scenario is consistent with the state of technology and market sizes discussed throughout this report. The low material use scenario assumes lower market penetration of electrolyzers and lower usage rates of key materials. These two scenarios provide a plausible range of demand growth for various materials, illustrating the influence of electrolyzer deployment rates and material usage on total demand. For a detailed summary of the low material use scenario assumptions please see Appendix A.

² Apparent Consumption is defined as domestic primary metal production + recovery from scrap + net import reliance (USGS 2021)



Figure 9. Ranges of projected material demand as a percentage of annual U.S. consumption and U.S. import reliance shown for key fuel cell and electrolyzer materials (Original work)

In this report, apparent consumption is defined in a manner consistent with the USGS, where apparent consumption equals primary production plus secondary production plus imports minus exports for a given material. Ranges of possible demand are shown for baseline and low material usage scenarios (see Appendix A).

Demand estimates were calculated internally based on hydrogen market sizes from (Larson et al. 2021) for baseline and low material use scenarios. Please see Appendix A for a detailed summary.

Import percentages and apparent consumption are from USGS (2021).

Some of the key risks associated with the materials required for PEM and solid oxide electrolyzers and fuel cells are highlighted here. They include U.S. import reliance, key sources and suppliers, and competing uses. The information was extracted from the USGS Mineral Commodity Summaries 2021 unless otherwise noted (USGS 2021). A comprehensive evaluation table with preliminary estimates across the supply chain elements is included in Appendix D.

The PGM catalyst report that is part of this series provides additional information on some of these materials ("Supply Chain Review: Platinum Metal Group Catalysts" 2021).

High-Risk Electrolyzer and Fuel Cell Materials (in alphabetical order)

• Graphite/Activated Carbon: The United States is 100% dependent on foreign sources, mainly China, to meet domestic demand of natural graphite. During 2020, China was the world's leading graphite producer, producing an estimated 62% of total world output. Other key producers of natural graphite

are Mozambique, Brazil and Madagascar. Competing uses include batteries (including for electric vehicles), brake linings, lubricants, powdered metals, refractory applications, and steelmaking.

- Iridium: Iridium is one of the scarcest elements on earth. Iridium mining is highly concentrated in South Africa, and production is coupled with the mining rate of the primary PGMs (i.e., platinum and palladium). The United States is 100% reliant on imports to meet iridium demand. Hence, the presence of an iridium recycling infrastructure, end-of-life recycling rates of at least 90%, and low catalyst loading targets of 0.05 g/kW are crucial to meet future iridium demands for PEMEC. Competing uses for iridium are similar to platinum. (Minke et al. 2021).
- Platinum: Platinum mining is mainly concentrated in South Africa (72% of worldwide production), followed by Russia and Zimbabwe. The United States is 79% reliant on imports to meet its platinum demand. The main U.S. import sources of platinum are South Africa, Germany, Italy, and Switzerland. Competing uses include automobile catalytic converters, catalysts for chemical production and petroleum refining, medical devices, electronic applications (e.g., hard disk drives), jewelry, glass manufacturing, and laboratory equipment.
- Strontium: The United States imports 100% of the strontium it requires domestically, although significant domestic strontium deposits do exist across the United States. Domestic consumption of strontium is primarily associated with ceramic ferrite magnet manufacturing and pyrotechnics. The main import sources of strontium for the United States in 2020 were Mexico and Germany, which makes the supply somewhat secure.
- Yttrium: Currently, the United States does not have a significant domestic demand for yttrium. Primary end uses include catalysts, ceramics, lasers, metallurgy, and phosphors. The United States is currently 100% reliant on imports for yttrium, and 94% of the supply comes from China.

Moderate-Risk Electrolyzer and Fuel Cell Materials (in alphabetical order)

- Aluminum: Aluminum is primarily produced from alumina extracted from bauxite; bauxite resources are concentrated in Africa, Oceania, South America and the Caribbean, and Asia. The United States relies on imports to meet ~50% of demand. U.S. import sources include Canada (50%); the United Arab Emirates (10%), Russia (9%), China (5%), and other nations (26%). Domestic resources of bauxite are insufficient to meet long-term U.S. demand, but other subeconomic resources (other than bauxite) are widely available. Competing uses include transportation applications aerospace and automotive (40%), packaging (21%), building (14%), electrical (8%), consumer durables (7%), machinery (7%), and other uses (3%).
- Chromium: Chromium is supplied mainly by South Africa, where 41% of chromium is produced globally. As of 2020, the United States was 75% reliant on imports to meet its chromium demand. The main global suppliers of chromium to the United States in 2020 were South Africa, Kazakhstan, Mexico, and Russia. The main use of chromium is in stainless steel and heat-resisting steel manufacturing.
- **Cobalt:** The United States was 76% reliant on imports to meet its cobalt demand as of 2020. In the United States, cobalt is used mainly to produce superalloys for aircraft gas turbine engines. Globally, cobalt mining is concentrated in the Democratic Republic of Congo, and China is the largest supplier of refined cobalt. The main U.S. suppliers of cobalt intermediates (e.g., cobalt powders) in 2020 were Norway, Canada, Japan, and Finland (Igogo et al. 2019).

- **Copper:** Domestic import reliance of the United States for refined copper is fairly low: around 37% in 2020. The United States mines, smelts, refines, and recycles copper, and it has significant copper reserves. The main U.S. suppliers of refined copper are Chile, Canada, and Mexico.
- Iron: The United States was a net exporter of iron in 2020; other major suppliers of iron included Brazil, Canada, Sweden, and Chile.
- Lanthanum: There is virtually no domestic production of lanthanum in the United States; it relies for 100% of its domestic lanthanum demand on imports from China. This reliance will lessen once the primary domestic rare earth mine, Mountain Pass in California, starts separating light rare earths at its mining facility in 2022 (MP Materials n.d.).
- Manganese: In 2020, the United States was 100% reliant on imports, including imports from Gabon (69%), South Africa (17%), Mexico (8%), and Australia (4%). There is no significant domestic supply of manganese. Steel production—either directly in pig iron manufacturing or indirectly through upgrading the ore to ferroalloys—is the main competing domestic processes that consume manganese.
- Nickel: The domestic nickel demand is for stainless and alloy steels (≈85%), nonferrous alloys and superalloys, electroplating, and other uses. It is currently 50% met by imports, mainly from Canada, Finland, Norway, and Russia. In the United States, the leading uses for primary nickel are stainless and alloy steels, nonferrous alloys and superalloys, electroplating, and other uses, including catalysts and chemicals.
- **Titanium:** Production of titanium mineral concentrates is mainly concentrated in China, South Africa, and Australia. In 2020, The United States imported 88% of its titanium mineral concentrates demand and more than 50% of titanium sponge demand. That year, the main U.S. suppliers of titanium mineral concentrates were South Africa, Australia, Madagascar, and Mozambique and those of titanium sponge were Japan, Kazakhstan, and Ukraine. Competing uses include aerospace applications, chemical processing, marine hardware, medical implants, as well as pigments and coatings.
- Zirconium: The United States was a net exporter of zirconium before 2020 and now is regarded to be cost competitive among domestic and global suppliers. In years when the United States did import zirconium, the major import sources were Australia and South Africa. The primary uses of zirconium are ceramics, foundry sand, opacifiers, and refractories. Current leading consumers of zirconium metal are chemical processing and the nuclear energy industry.

3.1 Key Vulnerabilities

Key U.S. vulnerabilities with respect to electrolyzer and fuel cell supply chains include the immaturity of electrolytic hydrogen markets, the need for electricity to produce hydrogen and market structures to access that electricity, a lack of sufficient hydrogen infrastructure to support market growth, a lack of electrolyzer and fuel cell manufacturing capacity, energy and environmental justice issues for key materials, a need for workforce development, and a need to consider international competitiveness.

Electrolytic Hydrogen Markets

The electrolytic hydrogen market is miniscule today, but as is shown in Figure 2 (page 7), it needs to grow to over 100 MMT/yr by 2050 to meet climate goals. To do so, technologies will need to improve and become more cost-effective, and all aspects of hydrogen production, utilization, and the transmission-delivery-storage infrastructure will need to grow.

Immature Technologies: PEMEC and SOEC systems are not at cost-parity with conventional hydrogen production technologies (e.g., natural gas reforming) and will require technology development to achieve widespread deployment and commercialization. Likewise, PEMFCs and SOFCs will require significant technology advancement to be cost-competitive with conventional combustion-based technologies currently used in stationary and vehicular applications. Without development, these technologies are unlikely to economically support the hydrogen market size assessment shown in Figure 2. R&D support for electrolyzers and fuel cells, as established in the Infrastructure Investment and Jobs Act (Infrastructure Investment and Jobs Act 2021) would need to be sustained and focused to ensure that clean hydrogen technologies are cost competitive with incumbent technologies.

Lack of Sufficient Emission Reduction Incentives: Currently, electrolytic hydrogen is more expensive than conventional hydrogen (primarily produced via SMR) and current emission reduction incentives are insufficient to enable an electrolytic hydrogen market to grow.

Insufficient Codes and Standards: Regulations, codes, and standards—which provide information needed to safely build, maintain, and operate facilities and systems—are a major institutional barrier to deploying hydrogen technologies and are likely to delay, if not suspend the growth of hydrogen markets. Codes and standards are in place for refining, the chemical industry, and ammonia production (DOE Hydrogen Program n.d.). However, modelbuilding codes and technical standards that are recognized by federal, state, and local governments may be insufficient for certain application such as for bulk electricity and nonindustrial applications of hydrogen. NFPA 2, the Hydrogen Technologies Code, undergoes regular revision cycles to address insufficiencies as needed. Efforts within standards development organizations (e.g., International Electrotechnical Commission, International Organization for Standardization, Society of Automotive Engineers) are ongoing to address gaps which may limit the near-term deployment of hydrogen for certain applications. Specific gaps in federal regulations relating to hydrogen technologies have been identified by Sandia National Laboratories (Baird et al. 2021).

Electricity Resources and Markets

Insufficient U.S. Electric Grid Generation Capacity: Electrolytic hydrogen production could increase the needed electricity generation capacity in the United States by 2050 by 2.5 terawatts (TW) or 1.8 TW assuming all additional capacity is solar or wind. For reference, the total installed generation capacity in the United States was 1.12 TW in 2020 (IEA 2020).

Electrolyzers Not Sufficiently Compensated in Electricity Markets: Flexible electrolytic hydrogen production can use electricity during periods of oversupply on the grid and reduce power spikes. By operating flexibly, the electrolyzers' net electricity prices could be lower than market purchase prices (Ruth et al. 2019). Some markets allow electrolyzers to receive value for providing ancillary services. However, current electricity markets are not structured so that electrolyzers realize other values of flexible operations such as real-time electricity prices and avoided dispatchable capacity requirements.

Hydrogen Infrastructure

Insufficient Infrastructure to Support Hydrogen Markets: Hydrogen delivery and storage infrastructure will need to grow to support a large hydrogen market. Pipelines are the most energy-efficient approach to transporting hydrogen. However, their deployment is challenged by their high capital costs. Blending of hydrogen into existing pipelines comingled with natural gas or other products is also possible as the economy builds demand. Some applications can use blends of hydrogen, while other applications may require separation of hydrogen and natural gas at the end use. Also, technology advancement is needed to reduce the cost and ensure the safety of

gaseous hydrogen storage, especially in areas not having large-scale geological capabilities (Department of Energy (DOE) 2020).

Electrolyzer and Fuel Cell Material Supply and Manufacturing Capacity

Availability of Key Raw Materials: Many key resources (notably platinum, iridium, graphite, titanium, lanthanum, strontium, manganite, cobalt, yttrium, gadolinium, and samarium) are almost entirely imported and thus introduce supply chain vulnerabilities. Of note, iridium is present in lower concentrations than can support operations and capital investment for platinum mining and refining. Due to this vulnerability, the anticipated iridium supply is likely to be inadequate to meet the demands estimated in the Market Size Estimate section above ("Supply Chain Review: Platinum Metal Group Catalysts" 2021). In addition, many of these materiaks also have competing uses that are also growing and will require larger quantities to supply multiple applications.

Manufacturing Capacity Growth Requirements: A very large and sustained growth rate is needed to achieve targets; specifically, U.S. PEMEC and SOEC installed capacity needs to grow from the current market sizes of approximately 0.17 GW and 0.023 GW, respectively, to approximately 12.66 GW and 4.85 GW by 2030 and by 396 GW and 144 GW by 2050 as discussed in the Electrolyzer and Fuel Cell Market Size Estimates section above. These installed capacities result in an estimated compound annual growth rate from 2021 to 2050 of 22% and 19% for PEMECs and SOECs, respectively (see Appendix A) (Arjona and Buddhavarapu 2021).

Growth Requirements for Manufacturing Supply Chains: To meet the electrolyzer and fuel cell manufacturing capacity, the ability to supply processed materials and subcomponents will need to grow. In addition, reliance on the few manufacturers that produce some key processed materials (e.g., PFSA) today represents a potential supply chain risk as deployment expands, and it highlights the need to engage multiple suppliers for each supply chain element (James et al 2018).

Energy and Environmental Justice Issues for Key Materials

Energy Justice Issues: Many key materials are extracted and/or refined in nations that use forced labor or have minimal environmental protections.

Environmental Justice Issues: PFSA ionomers and membranes may not meet future environmental regulations due to concerns about possible health hazards associated with the production of PFSAs (Lohmann et al. 2020; Cousins et al. 2019) and perfluorinated compound emissions if the membrane is incinerated (Feng et al. 2015). Some governments are moving to ban per- and polyfluoroalkyl substances (PFAS) which may include PFSAs due to these and other environmental concerns – impacts on human health and their high global warming potential.

Workforce development

Mismatch in demand and supply of domestic workforce: A large workforce will be needed that is capable of manufacturing and operating electrolyzers and fuel cells.

International Competitiveness

Consistent and Equal Emission Standards for Hydrogen: International markets for hydrogen are emerging with production in countries rich in low-carbon energy resources planning to export to countries with strong demands for clean hydrogen. Without consistent and equal standards, which would also suppress hydrogen production with carbon-intensive means, energy and environmental justice issues may arise around traded hydrogen.

4.U.S. Opportunities and Challenges

The overarching opportunity for electrolytic hydrogen within the United States is a potential market of over 100 MMT/yr for applications across the industrial, transportation, and power sectors. By taking advantage of the sub-opportunities below, the United States could capture high value-added links of the electrolytic hydrogen supply chain.

An electrolytic hydrogen market of that size powered with clean electricity would provide decarbonization opportunities for difficult to abate sectors including synthetic fuels for air and marine transport, long-distance transport via heavy and medium duty vehicles, energy storage, and high-temperature heat. If the electrolytic hydrogen market does not develop, many of those technologies would need to use fossil fuels and carbon sequestration would be necessary to achieve net zero emissions. Within that overarching opportunity, there are multiple opportunities along the supply chain and challenges that would need to be overcome to reach that level of production.

Electrolytic Hydrogen Markets

Commercialization of Electrolytic Hydrogen Production: Electrolytic hydrogen offers an opportunity to the United States to decarbonize heavy industry and meet climate goals, but electrolytic hydrogen is currently more expensive than SMR-produced hydrogen which is currently widely used by industry. Thus, to commercialize and scale this technology, the cost will need to be addressed. This can be achieved through R&D that reduces electrolyzer capital costs both directly and through at-scale manufacturing technologies. R&D could also increase electrolyzer durability and efficiency. The recently passed Infrastructure Investment and Jobs Act, through sections 40313 and 40314, provides a starting point by initiating the Clean Hydrogen Research and Development Program, as well as support for the development of hydrogen production, there are opportunities to support manufacturing development, scale, and ensure market pull for electrolytic hydrogen.

Applications for Electrolytic Hydrogen: Many of the hydrogen applications in the transportation, electricity, industrial, commercial building, and residential building sectors are more expensive than alternatives, but they provide benefits such as emissions-minimal long-duration storage and chemical reduction that alternatives do not. For example, PEMFCs and SOFCs are more expensive than conventional combustion-based technologies currently used in transportation and stationary applications. Customers and industry are reluctant to invest in hydrogen technologies that are more expensive than incumbent technologies. There is the opportunity for R&D to reduce the cost of fuel cells and other hydrogen applications so that they are competitive if low-cost electrolytic hydrogen is available. Under sections 40313 and 40314 of the Infrastructure Investment and Jobs Act, funding is provided to address this need through the Clean Hydrogen R&D Program and the development of hydrogen demonstration hubs (Infrastructure Investment and Jobs Act 2021).

Leadership in Developing Codes and Standards: For some applications, standards prevent hydrogen from being used to its potential (e.g., the natural gas system does not allow for more than very low hydrogen concentrations). Improved, science-based standards could safely increase hydrogen's potential. Leadership by the United States in setting global hydrogen codes and standards could enable the United States to capitalize on the green hydrogen economy in tandem with other countries as the world transitions to a carbon-free future, including manufacturing of equipment that meets those standards. Efforts have begun to develop updated standards in one area – hydrogen blended with natural gas (Infrastructure Investment and Jobs Act 2021).

Electricity Resources and Markets

Expanding the U.S. Electric Grid Capacity: Achieving the overall opportunity of 100 MMT/yr hydrogen would require around 5000 TWh electricity annually—approximately doubling the grid's annual generation. That additional need could drive economic and job growth where clean electricity generation resources are the most abundant—especially in rural areas. Electrolysis can also benefit the grid by acting as a dispatchable load that absorbs excess generation when it is available and shutting off when the electrical load is high or generation is low. However, adding 5000 TWh/yr electricity will require much more generation capacity. Recently established efforts for the development of regional hydrogen hubs in section 40314 of the Infrastructure Investment and Jobs Act could serve as an opportunity to investigate regional needs and conduct an economic assessment where electricity prices are prohibitive or balancing markets create opportunities for services; this could help identify opportunities where the grid and electrolysis can grow symbiotically (Infrastructure Investment and Jobs Act 2021).

Hydrogen Infrastructure

Development and Management of Bulk Hydrogen Storage: Large-scale hydrogen storage systems provide supply chain value (e.g., steady supply, increased resilience, and predictable prices) to multiple end-use industries. The geology necessary for large-scale hydrogen storage (e.g., salt caverns, saline aquifers, depleted natural gas or oil reservoirs, and engineered hard rock reservoirs) are available and can provide the physical conditions for this type of storage. However, financial performance of storage built early in the evolution of an electrolytic hydrogen market will be poor because most of the expenses are in construction and initial income will be low due to low initial utilization, although they are likely to grow over time.

Utilization of the Natural Gas Infrastructure for Hydrogen Transport and Storage: The natural gas infrastructure could be converted to transport and store hydrogen instead, reducing the cost of hydrogen transmission and storage as larger volumes of clean hydrogen are produced and used. However, the technical requirements and methods to convert it (and the subsequent costs) are unknown. In addition, the supply and demand locations are likely to be different so the network will likely need to be modified. Current efforts have begun to address technical barriers to hydrogen blending into natural gas pipelines (DOE Hydrogen and Fuel Cell Technologies Office 2021) and quantify the benefit of using the existing natural gas infrastructure (Infrastructure Investment and Jobs Act 2021).

Electrolyzer and Fuel Cell Material Supply and Manufacturing Capacity

Domestic Material Supplies: There is an opportunity for domestic supplies to meet at least a portion of the demand for platinum, iridium, graphite, lanthanum, and yttrium, including exploration, extraction, and processing and refining infrastructure. Additionally, recovery and recycling of valuable materials from end-of-life products, including the technology and processes to do so economically, represent a potential area of leadership and domestic sourcing. Challenges exist in the long lead times for permitting and mitigating of environmental impact. In addition, there are opportunities to further reduce PGM-content and even develop alternatives to PGM catalysts in electrolyzers and fuelcells. AdditionalR&D will be necessary to develop those advanced low-PGM and PGM-free catalyst options.

Electrolyzer and Fuel Cell Manufacturing Capacity: Manufacturing of electrolyzers and other hydrogen production technologies as well as fuel cells has significant growth potential and the opportunity for economic leadership by countries that are early adopters. Stakeholder (developers, suppliers and end-user) coordination at the regional level, as envisioned in the support of hydrogen hubs established in section 40314 of the Infrastructure Investment and Jobs Act (Infrastructure Investment and Jobs Act 2021), can strengthen the business structures around hydrogen technology manufacturing.

Energy and Environmental Justice Issues for Key Materials in a New Industry

Leadership on Equity and Environmental Justice Issues: A developing supply chain for electrolyzers and fuel cells is an opportunity to lead equity and environment in a growing industry, instead of addressing them after commercialization. Global challenges regarding the environmental, community, and labor impacts of the manufacturing of energy technologies present an opportunity for collaboration and demonstration of positive environmental, social, and governance performance as the manufacturing and supply chains for electrolytic hydrogen are built.

International Competitiveness

Potential Electrolytic Hydrogen Exports: Some countries recognize the need for electrolytic hydrogen to decarbonize, but they lack the electricity generation resources to address the need. The United States is resource-rich and may be able to export low-carbon, electrolytic hydrogen, and related chemicals, as well as the electrolyzers, fuel cells, and related equipment. An international standard could limit global deployment of carbon-intensive produced hydrogen and enhance U.S. competitiveness.

5. Conclusions

This report summarizes potential supply chain and materials issues for electrolyzers and fuel cells in a decarbonized future. The overarching opportunity for electrolytic hydrogen within the United States is a potential market of over 100 MMT/yr for applications across the industrial, transportation, and power sectors. An electrolytic hydrogen market of that size powered with clean electricity would provide decarbonization opportunities for difficult-to-abate sectors, including synthetic fuels for air and marine transport, long-distance transport via heavy- and medium-duty vehicles, energy storage, and high-temperature heat. For the domestic electrolytic hydrogen market to grow to 100 MMT/yr, the electrolyzer capacity required ranges up to 1,000 GW to meet new capacity deployments and replace existing capacity at the end of its lifetime. This is a large increase over the approximately 0.17 GW of capacity currently installed or planned in the United States, resulting in an approximately 20% compound annual growth rate from 2021 to 2050. In addition, over 50 GW of domestic fuel cell capacity is required in the decarbonization scenario with an annual manufacturing requirement of over 3 GW/yr. This level of growth represents a significant opportunity for the United States as electrolytic hydrogen markets and supply chains rapidly grow and develop globally.

Large increases in extraction and refining of many materials would be needed, with many key materials currently being addressed primarily (and exclusively, for some) by imports. Especially of concern are several materials that have both (1) larger projected electrolyzer and fuel cell demands than their current totals and (2) a currently high percentage of total market being met via imports with no specific plans for domestic production. Those include iridium, yttrium, platinum, strontium, and graphite. The United States appears to have sufficient resources and supply chains for many of the other key materials, including stainless steel, titanium, zirconium, and nickel.

It is difficult to exactly predict manufacturing challenges because of the extraordinary growth required in the electrolytic hydrogen market and thus the electrolyzer and fuel cell markets. Key processed materials for polymer electrolyte technologies include perfluorosulfonic acid ionomers, catalysts, graphite composites, and titanium meshes. Key processed materials for solid oxide technologies include air electrode materials, fuel electrode materials, and the electrolyte. The United States currently has manufacturing capabilities in most of the necessary key processed materials and subcomponent manufacturing for both polymer electrolyte and solid oxide technologies. Likewise, the United States has relatively well-positioned end product manufacturing capabilities for both technologies. However, how and where manufacturing capacity along the supply chain may grow are unknown. Thus, government support may be needed to support those industries and meet cost reduction, growth, decarbonization, and supply chain security objectives.

Key vulnerabilities in regard to developing an electrolytic hydrogen market and the supply chains needed for that market include:

- Immature technologies that are not currently cost-competitive for both electrolytic hydrogen production and utilization
- Lack of sufficient emission reduction incentives
- Insufficient codes and standards
- Insufficient electricity generation capacity
- Electrolyzers not being compensated sufficiently in the electricity market
- Insufficient infrastructure to support hydrogen markets at their potential

- Availability of raw materials
- Growth requirements for manufacturing supply chains
- Energy justice issues
- Environmental justice issues
- Mismatch in demand and supply of domestic workforce
- Consistent and equal standards for hydrogen production around the world.

While the United States has technology development targets and an RD&D plan (Department of Energy (DOE) 2020), it does not currently have hydrogen deployment targets or a national plan, unlike at least 15 other countries. However, the United States is developing one as required by section 40314 of the Infrastructure Investment and Jobs Act (Infrastructure Investment and Jobs Act 2021).

Key opportunities to enable the growth of electrolytic hydrogen and fuel cell markets to meet the overarching opportunity of 100 MMT/yr and associated supply chains include:

- Reducing cost and increasing commercialization of electrolytic hydrogen production
- Developing economically competitive applications
- Leading development of codes and standards
- Expanding the U.S. electric grid capacity
- Development and management of bulk hydrogen storage
- Utilization of the natural gas infrastructure for hydrogen transport and storage
- Development of domestic material supplies, including recycling and PGM-free catalysts
- Development of electrolyzer and fuel cell manufacturing capacity
- Leadership on energy and environmental justice issues
- Potentialhydrogen exports.

Recommended policy actions to address the vulnerabilities and opportunities covered in this report may be found in the Department of Energy 1-year supply chain review policy strategies report, "America's Strategy to Secure the Supply Chain for a Robust Clean Energy Transition." For more information, visit www.energy.gov/policy/supplychains.

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Appendix A: Performance and Material Use Assumptions

A-1. Technology Assumptions

Tables A-1 through A-3 outline electrolysis and fuel cell specific design criteria and performance values that are used to estimate material requirements.

Table A-1. Water Electrolyzer and Fuel Cell Design and Performance Assumptions

Туре	Capacity Factor (%)	Cell Voltage (V)	Current Density (A/cm^2)	Efficiency (kWh/kg)	Anode Catalyst	Anode Catalyst Loading (mg/cm^2)	Cathode Catalyst	Cathode Catalyst Loading (mg/cm^2)	Lifetime (thousand h)
PEMEC	90% (a)	1.9 (b)	2 (b)	55 (b)	Iridium oxide (b)	2 (b)	Platinum (b)	1 (b)	40 (b)
Alkaline electrolyzer	90% (a)	2.1 (c)	0.3 (c)	64 (d)	Nickel (d)		Nickel (d)		80 (c)
SOEC	90% (a)	1.28	1.0	36.8	LSCF		Ni-YSZ		35
AEMEC	90% (a)	2 (c)	0.4 (c)	63 (e)	Ni-Fe-Ox (f)	2.5 (f)	Ni-Fe-Co (f)	2.5 (f)	5 (e)
PEMFC	90% (a)	0.55 (g)	1.5 (g)	23.64 (d)	Platinum (g)	0.05 (g)	Platinum (g) and Pt alloys (PtNi, PtCo)	0.1 (g)	80
SOFC	90% (a)	0.8	0.4	52% higher heating value (natural gas)	Ni-YSZ		LSCF, LSM-YSZ		40

^a High capacity factor operation assumed

^bH2NEW baseline model

^c Miller, Hamish Andrew, Karel Bouzek, Jaromir Hnat, Stefan Loos, Christian Immanuel Bernäcker, Thomas Weißgärber, Lars Röntzsch, and Jochen Meier-Haack. 2020. "Green Hydrogen from Anion Exchange Membrane Water Electrolysis: A Review of Recent Developments in Critical Materials and Operating Conditions." Sustainable Energy and Fuels 4 (5): 2114–33. https://doi.org/10.1039/c9se01240k.

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^g Memorial Institute. 2016. "Manufacturing Cost Analysis of PEM Fuel Cell Systems for 5- and 10-KW Backup Power Applications."

Table A-2. Low-Temperature Water Electrolyzer and Fuel Cell Bipolar Plate and Gas Diffusion Layer Material Assumptions

Туре	Bipolar Plate Material	Bipolar Plate Thickness (cm)	Cathode GDL Material	Cathode GDL Thickness (cm)	Cathode GDL Porosity (%)	Anode GDL Material		Anode GDL Porosity (%)
PEMEC	Titanium (a)	0.15 (a)	Carbon paper (a)	n/a	n/a	Titanium mesh (a)	0.025 (a)	0.3 (a)

Туре	Bipolar Plate Material	Bipolar Plate Thickness (cm)	Cathode GDL Material	Cathode GDL Thickness (cm)	Cathode GDL Porosity (%)	Anode GDL Material	Anode GDL Thickness (cm)	Anode GDL Porosity (%)
Alkaline electrolyzer	Stainless steel (b)	0.15	n/a	n/a	n/a	n/a	n/a	n/a
AEMEC	Stainless steel (b)	0.15	n/a	n/a	n/a	n/a	n/a	n/a
PEMFC	Graphite	0.15	n/a	n/a	n/a	n/a	n/a	n/a

^a H2NEW baseline model

^b IRENA. 2020. Green Hydrogen Cost Reduction: Scaling up Electrolysers to Meet the 1.5 C Climate Goal./Publications/2020/Dec/Green-Hydrogen-Cost-Reduction.https://www.irena.org/-

/media/Files/IRENA/Agency/Publication/2020/Dec/IRENA_Green_hydrogen_cost_2020.pdf.

Table A-3. SOC Electrolyzer and Fuel Cell Material Use Assumptions

Туре	LSM (kg LSM/MW)	LSCF (kg LSCF/MW)	LSM-YSZ (kg LSM-YSZ/MW)		8YSZ (kg 8YSZ/MW)	Ni-YSZ (kg Ni- YSZ/MW)	441-SS (kg 441 SS/MW)	Borosilicate Glass (kg borosilicate glass/MW)
SOEC	5.14	12.33	9.54	3.06	3.98	228	2579	79.05
SOFC	19.35	46.45	35.94	11.53	14.99	861	9709	297.47

LSM: Lanthanum strontium magnatite

LSCF: Lanthanum strontium cobalt ferrite

LSM-YSZ: Lanthanum strontium magnetite-yttria stabilized zirconia

441-SS: Stainless steel type 441

A-2. Market Scenario Assumptions

Tables A-4 and A-5 summarize hydrogen demand projections for global and domestic hydrogen by sector from current to 2050.

International Energy Agency (IEA) Net Zero by 2050 Scenario

The following tables outline market sizes and electrolysis supply mixes for various sectors outlined in a recent report by the International Energy Association (IEA 2021b).

Table A-4. Ambitious Scenario Hydrogen Market Demand Assumptions by Sector (MMT/yr)

Year	Units	2020	2040	2050
Total	MMT/yr	87.4	390.1	528.1
Blended in gas grid	MMT/yr	0	37.9	59.9
Energy storage	MMT/yr	0	110	100
Buildings	MMT/yr	0	11.4	16.2

Year	Units	2020	2040	2050
Road transportation	MMT/yr	0	46	93.2
Aviation	MMT/yr	0	22.1	51.9
Shipping	MMT/yr	0	35.1	57.3
Industry	MMT/yr	1	20.5	28.3
Refineries	MMT/yr	13.8	5	4.6
Chemicals: onsite	MMT/yr	45.9	58.4	60.3
Iron and steel: onsite	MMT/yr	4.7	28.9	40.4
Refineries: onsite	MMT/yr	22	8.1	3.8
Other	MMT/yr	0	6.7	12.2

Princeton Net-Zero America: E+RE+ scenario

The following tables outline market sizes and electrolysis supply mixes for various sectors outlined in the Princeton University Net-Zero America E+RE+ scenario (Larson et al. 2021). Hydrogen market sizes for subsectors are provided in annual energy consumed, and were converted from these units to MMT/yr as shown in Table A-5. An efficiency of 50.51 kWh/kg hydrogen was assumed. Estimates for installed electrolyzer capacity provided in this report vary slightly from those in the Princeton NZA scenarios due to differences in assumptions regarding the types of electrolyzers (e.g., PEMEC, SOEC, etc.) used to meet hydrogen demand and the rate at which these electrolyzers replace conventional sources of hydrogen over time (Table A-8).

Table A-5. E+RE+ Scenario Hydrogen Market Demand Assumptions by Sector (MMT/yr)

Year	Units	2020	2025	2030	2035	2040	2045	2050
Total	MMT/yr	4.9	5.2	8.6	14.7	28.1	54.9	106.0
Bulk chemicals manufacturing	MMT/yr	4.9	5.1	5.1	5.2	5.6	5.8	5.9
Direct reduced iron production	MMT/yr	0.0	0.0	0.1	0.8	1.5	1.9	2.4
Gas turbine fuel	MMT/yr	0.0	0.0	0.0	0.0	0.0	0.2	1.4
Gaseous fuel synthesis	MMT/yr	0.0	0.0	0.0	0.0	0.0	0.0	5.6
Hythane	MMT/yr	0.0	0.0	0.0	0.0	0.2	0.6	1.2

Year	Units	2020	2025	2030	2035	2040	2045	2050
Industrial boilers	MMT/yr	0.0	0.0	0.0	0.0	0.0	0.0	0.2
Liquid fuels synthesis	MMT/yr	0.0	0.0	2.5	6.1	14.6	37.0	78.0
MD/HD FCEVs	MMT/yr	0.0	0.1	0.7	2.4	4.6	6.4	7.5
Other industry	MMT/yr	0.0	0.0	0.0	0.2	0.8	1.4	1.5
Other transportation	MMT/yr	0.0	0.0	0.1	0.1	0.9	1.6	2.4

A-3. Sectoral Aggregation and Market Shares for Electrolyzers and Fuel Cells

The following tables summarize sector aggregation between global and domestic hydrogen demand projections. Both projections are aggregated into like sectors for comparison between the two reports. Tables A-6 and A-7 summarize the market shares between hydrogen production technologies and fuel cells over time, respectively.

Table A-6. Sector Aggregation Between Global and U.S. Hydrogen Market Projections

Aggregated Sectors in this report	IEA: Global	Princeton: Net-Zero America
Natural gas blending	Blended in gas grid	Hythane
Energy storage	Energy storage	Gas turbine fuel
Buildings	Buildings	
FCEVs	Road transportation	MD/HD FCVs
	· · · · · · · · · · · · · · · · · · ·	Other transportation
Synfuels	Aviation	Gaseous fuel synthesis
	Shipping	Liquid fuels synthesis
Industry	Industry	Industrial boilers
Chemicals and existing feedstock	Refineries	
	Refineries: onsite	Bulk chemicals manufacturing
	Chemicals: onsite	
New feed stock	Iron and steel: onsite	Other industry
	Other	Direct-reduced iron production

Table A-7. Estimated U.S. and Global Hydrogen Market Share for Electrolyzers for New Installed Capacity Used in this Analysis

Year	2020	2025	2027	2050
Natural gas blending	100% SMR	Growth in	SMR phase-out	100% PEM
Energy storage	100% SMR	hydrogen	begins, completed	60% PEM
Energy storage	100% SIVIR	demand	by 2050	40% SOEC

Year	2020	2025	2027	2050
Buildings	100% SMR	met with		100% PEM
		electrolysis		50% PEM
FCEVs	100% SMR			20% alkaline
				30% SOEC
				50% PEM
Synfuels	100% SMR	MR		20% alkaline
				30% SOEC
Industry	100% SMR			30% PEM
industry				70% SOEC
				35% PEM
Chemicals and existing feedstock	100% SMR			30% alkaline
				35% SOEC
Newfeedstock	100% SMR			100% PEM

Table A-8. Estimated U.S. and Global Hydrogen Market Share for Fuel Cells Used in this Analysis

Year	2020	2030	2040	2050
Natural gas blending	n/a	n/a	n/a	n/a
Energy storage	40% PEM	40% PEM	40% PEM	40% PEM
	60% SOFC	60% SOFC	60% SOFC	60% SOFC
Buildings	n/a	n/a	n/a	n/a
FCEVs	100% PEM	100% PEM	100% PEM	100% PEM
Synfuels	n/a	n/a	n/a	n/a
Industry	n/a	n/a	n/a	n/a

Year	2020	2030	2040	2050
Chemicals and existing feedstock	n/a	n/a	n/a	n/a
Newfeedstock	n/a	n/a	n/a	n/a

This analysis uses projections for the size of hydrogen markets as a basis for estimating the installed capacity of electrolyzers E_{PEMEC} , which is estimated using Equation A-1. Equation A-1 shows values for PEMEC systems, but this methodology extends to all electrolyzer types. In this equation, D_{H2} is the annual hydrogen demand in MMT/yr, S_{PEMEC} is the market share for PEMEC systems, η_{PEMEC} is the efficiency of the PEMEC system in kWh/kg H2, and CF is the capacity factor for the electrolyzer.

$$E_{PEMEC} = \frac{D_{H2}S_{PEMEC}\eta_{PEMEC}}{CF(8,760)}$$
(A-1)

Two types of components are included in the model for material use in electrolyzers and fuelcells: components that directly scale with the catalytic active area of the system as a function of loading rate (e.g., catalyst coatings) and components that scale based on the total area of each cell in a stack and thickness of the component (e.g., bipolar plates).

Equation A-2 calculates the material required for material j per MW of PEMEC capacity, where LR_j is the loading rate of material j in mg/cm², AA_{PEMEC} is the catalytic active area per cell within the stack in cm²/cell, K_{PEMEC} is the number of cells per PEMEC stack (cells/stack), and P_{PEMEC} is the power rating of the stack (MW/stack).

$$m_j = \frac{LR_j A A_{PEMEC} K_{PEMEC}}{P_{PEMEC}}$$
(A-2)

For components such as bipolar plates whose size is quantified using a thickness rather than loading rate, the material demand per MW estimate is slightly modified (Equation A-3). Bipolar plates extend beyond the catalytic active area, covering buffer areas outside the cell, therefore the dimension of this component is based on the total area in the PEMEC cell TA_{PEMEC} . The thickness of material *j* is included in variable T_j , and the density of the material used in the component converts the volume to mass ρ_j .

$$m_j = \frac{T_j T A_{PEMEC} K_{PEMEC} \rho_j}{P_{PEMEC}}$$
(A-3)

The compound annual growth rate for installed electrolyzer capacity is estimated with Equation A-4, where C_2 and C_1 signify installed electrolyzer capacity in 2050 and 2021, respectively. In this equation t is the number of years between the installed capacity values (29 in this case).

$$CAGR = (C_2 - C_1)^{\frac{1}{t}} - 1$$
 (A-4)

A-4. Low Material Use Scenario Assumptions

This section describes assumptions used to generate low scenario material estimates that are shown in Figure 9 (page 29). The upper bound of these ranges is given by the baseline scenario material estimates documented in this appendix and throughout the remainder of the report.

The low material use scenario assumes lower market penetration of electrolyzer and a large portion of hydrogen comes from bioenergy with carbon capture and sequestration (BECCS) rather than electrolysis (Table A-9). These assumptions are based on the Princeton Net Zero America E+ scenario, while the baseline estimates use the E+RE+ scenario (Larson et al. 2021). While the total hydrogen market sizes between these two scenarios are similar, the market shares of hydrogen production technologies varies. In the E+RE+ scenario nearly all hydrogen is produced from electrolysis by 2050, while nearly 60% of hydrogen is produced with BECCS in the E+ scenario.

Table A-9. Estimated U.S. Hydrogen Market Share for Electrolyzers for Low Material Use Scenario Used in this Analysis

Year	2020	2025	2027	2050
Natural gas blending	100% SMR			40% PEM
Natural gas biending				60% BECCS
				20% PEM
Energy storage	100% SMR			20% SOEC
				60% BECCS
Buildings	100% SMR			40% PEM
Dulutings		Growth in		60% BECCS
	hydrogen demand	hydrogen	SMR phase-out	15% PEM
FCEVs			begins, completed by 2050	10% alkaline
				15% SOEC
				60% BECCS
				15% PEM
Synfuels	100% SMR	MD		10% alkaline
				15% SOEC
				60% BECCS
Industry	100% SMR			20% PEM
inducty				20% SOEC

Year	2020	2025	2027	2050	
				60% BECCS	
				15% PEM	
Chemicals and existing feedstock	100% SMR			10% alkaline	
					15% SOEC
				60% BECCS	
Newfeedstock	100% SMR			40% PEM	
NewTeedSlock	100% SIVIR			60% BECCS	

Additionally, usage rates for key materials in electrolyzers and fuel cells are assumed to decrease 80% by 2035 in the low material use scenario (Table A-10) (Pivovar and Boardman 2021). Decreasing the use rates of these materials lowers the rates that they are used in newly manufactured systems. Any systems manufactured prior to 2035 use materials at the same rates outlined in Tables A-1 through A-3.

Table A-10. Estimated Adjusted Material Usage Rates for Key Materials in Low Material Use Scenario Used in this Analysis

Values are expressed in percentage of baseline material demand

Year	2020	2035	2050
PEMEC platinum usage	100%	20%	20%
PEMEC iridium usage	100%	20%	20%
SOEC LSCF usage	100%	20%	20%
SOEC LSM-YSZ usage	100%	20%	20%
SOEC Ni-YSZ usage	100%	20%	20%
PEMFC platinum usage	100%	20%	20%
SOFC LSCF usage	100%	20%	20%
SOFC LSM-YSZ usage	100%	20%	20%
SOFC LSCF usage	100%	20%	20%

Appendix B: Electrolyzer and Fuel Cell Manufacturers

B-1. PEMEC and PEMFC Technology

We surveyed market report pages, company websites and reports, press releases, and Bloomberg New Energy Finance (BNEF) databases to compile information on PEMEC and PEMFC manufacturers, locations, and capacities, shown in Tables B-1 through B-3. Manufacturers of electrolyzers using a kaline technology are included because most market reports were not specific to PEMEC technology. Additionally, a kaline technology is used significantly today and is worth considering in the short term (see source 43 in list below). Some listed companies are vertically integrated, at least to a certain extent, and produce many of the components but sell only the end product. Intermediate components are not listed for those companies except where noted.

Not all companies are listed and the data represents only one snapshot in time. Given the rapid pace of growth and the number of acquisitions, emerging companies, and joint ventures in numerous countries, the data in these tables will change frequently.

Table B-1. Examples of PEMEC and Other Electrolyzer Manufacturers

Company ¹	Headquarters	Manufacturing Location(s) ²	Technology	Additional source(s) ¹
Angstrom Advanced ⁴	Stoughton, MA, US	MA, US	PEMEC, Alkaline	88,122,123
Asahi Kasei Corporation ³	Japan	Japan	Alkaline	124,125
Cockerill Jingli Hydrogen	Suzhou, China	China	Alkaline	22, 37
Electric Hydrogen ⁴	Boston MA, San Francisco, CA, US	Boston MA, San Francisco, CA, US	components of PEMEC	93
Elogen (subsidiary of GTT, recently rebranded from Areva H2Gen)	Les Ulis, France	Les Ulis, France	PEMEC	13, 14
Erredue ³	Livorno, Italy	Italy	PEMEC, Alkaline	16, 115
Green Hydrogen	Kolding, Denmark	Kolding Denmark	Alkaline, maybe PEMEC	33, 35, 36
GTA, Inc. ⁴	Atlanta, GA, US	Atlanta, GA, US	PEMEC	91
Heraeus	Hanau, Germany	Hanau, Germany; Nanjing, China	Catalyst	81, 82, 136
Hydrogenics (subsidiary of Cummins)	Mississauga, Ontario, Canada	Ontario, Canada; IN, US; CA, US	PEMEC, Alkaline	5, 15
ldroenergy ³	Livorno, Italy	Italy	Alkaline	17, 126
lon Power	New Castle, DE, US	DE, US; PA, US	Membrane	83, 84

Company ¹	Headquarters	Manufacturing Location(s) ²	Technology	Additional source(s) ¹
ITM	Sheffield, UK	UK	PEMEC	126
Kobelco Eco-solutions	Kobe, Japan	Harima, Japan	PEMEC, Alkaline	105
Longi ^{3,4}	Xi'an, China	Jiangsu, China	Alkaline	38, 39, 43, 116
McPhy	La Motte-Fanjas, France	San Miniato, Italy; Wildau, Germany	Alkaline	18
Millennium Reign Energy ⁴	Dayton OH, US	Unavailable	Alkaline	95
Nel ASA	Oslo, Norway	Wallingford, CT, US; Notodden/Herøya, Norway; Denmark	PEMEC, Alkaline	19, 80
Next Hydrogen ³	Mississauga, Ontario, Canada	Mississauga, Ontario, Canada	Alkaline	20, 121
Ohmium ⁴	Incline Village, NV, US	Bengaluru, India	PEMEC	89,90
Plug Power	Latham, NY, US	Rochester, NY, US	PEMEC	9
Pochari Technologies ⁴	Bodega Bay, CA, US	China	Alkaline	92
Siemens Energy ³	Munich, Germany	Germany	PEMEC	21
Teledyne Energy Systems	Hunt Valley, MD, US	MD, US	Alkaline	117, 135
Tianjin Mainland Hydrogen Equipment ³	China (specific city/region unavailable)	Unavailable	Unavailable	
Toshiba Energy Systems	Kanagawa, Japan	Kawasaki, Japan	Alkaline	118

¹Included companies were mentioned by at least two market reports and BNEF except where noted (sources 45–70). Additional sources were used when location and technology information was unavailable from market reports and BNEF.

²Manufacturing locations may not be a complete list for a given company.

³Company was not mentioned in BNEF regarding electrolyzers.

⁴Company was not mentioned by market reports but is believed to be a significant manufacturer of electrolyzers. Longi specifically is a major manufacturer of solar cell technology that recently expanded into electrolyzer manufacturing (see source 38).

Table B-2. Examples of PEMFC and Component Manufacturers

Company ¹	Headquarters	Manufacturing Location(s) ²	Product(s)	Additional source(s) ¹
3M	St. Paul, MN, US	MN, US; Menomonie, WI, US	Membrane, lonomer	32, 106, 107, 108
Advent	Athens, Greece; Boston, MA, US	Patras, Greece; MA, US	Membrane (high- temperature), MEA	85
Altergy ⁴	Folsom, CA, US	Folsom, CA, US	PEMFC	100, 101
AvCarb	Lowell, MA, US	Lowell, MA, US	GDL	134
Ballard Power Systems	Burnaby, British Columbia, Canada	Burnaby, British Columbia, Canada; Hobro, Denmark	PEMFC, MEA ⁵ , BPP ⁵	1
BASF	Ludwigshafen, Germany	Germany	MEA, Catalyst	127,128
Bosch and PowerCell Sweden (strategic partnership) ⁴	Gerlingen, Germany (Bosch) & Gothenburg, Sweden (PowerCell)	Bamberg, Germany	PEMFC (automotive applications)	96, 97, 98, 99
Cell Impact	Karlskoga, Sweden	Karlskoga, Sweden	BPP	2
Chemours (2015 spin-off from DuPont)	Wilmington, DE, US	Fayetteville, SC, US	Membrane, lonomer	71, 72, 109
Dana	Maumee, OH, US	Neu-Ulm, Germany	BPP	119
ElringKlinger	Dettingen, Germany	Dettingen, Germany	PEMFC, BPP	3, 129
Freudenberg	Weinheim, Germany	Weinheim, Germany	GDL	120
Greenerity ³ (subsidiary of Toray)	Alzenau, Germany	Alzenau, Germany	MEA	4
Heraeus	Hanau, Germany	Hanau, Germany; Nanjing, China	Catalyst	80, 81, 82
Horizon Fuel Cell	Singapore	Singapore, China	PEMFC	130, 86
Hydrogenics (subsidiary of Cummins)	Mississauga, Ontario, Canada	Ontario, Canada	PEMFC	5, 74
HyPlat ³	Cape Town, South Africa	Cape Town, South Africa; Johannesburg, South Africa	MEA, Catalyst	6

Company ¹	Headquarters	Manufacturing Location(s) ²	Product(s)	Additional source(s) ¹
Hypoint ⁵	Menlo Park, CA, US; Sandwich, UK	None (still in product development)	High-temperature PEMFC (aviation applications)	103
Hyzon Motors ⁴	Rochester, NY, US	Rochester, NY, US (PEMFC); Chicago, IL, US (MEA)	PEMFC (automotive applications), MEA ⁵	102
Intelligent Energy	Loughborough, UK	Loughborough, UK	PEMFC	110
lon Power	New Castle, DE, US	DE, US; PA, US	MEA	83, 84
IRD Fuel Cells Technology	Denmark	Denmark; Albuquerque, NM, US	MEA, BPP	7, 111
Johnson Matthey	London, UK	Swindon, UK; Sonning Common, UK	MEA, Catalyst	31, 112
Mitsubishi Chemical Corporation	Tokyo, Japan	Unavailable	GDL	
Nedstack Fuel Cell Technology B.V.	Arnhem, The Netherlands	Arnhem, The Netherlands	PEMFC	131
Nisshinbo	Tokyo, Japan	Japan	BPP, Catalyst	41
Nuvera Fuel Cells, LLC	Billerica, MA, US	Billerica, MA, US; Fuyang, China	PEMFC	8
Plug Power	Latham, NY, US	Rochester, NY, US	PEMFC, MEA ⁵	9
POCO Materials (subsidiary of Entegris)	Billerica, MA, US	Decatur, TX, US; Russellville, AR, US	BPP	10
Proton Motor Fuel Cell ^{3,4}	Puchheim, Germany	Nuremburg, Germany	PEMFC	29, 42
Renewable Innovations ⁵	Salt Lake City, UT, US	Salt Lake City, UT, US	PEMFC	104
SGL	Wiesbaden, Germany	Germany	BPP, GDL	132
Shanghai Hongfeng Industrial ³	Shanghai, China	China	BPP	12, 94
Solvay ⁴	Brussels, Belgium	West Deptford NJ, US; Italy	Membrane, lonomer	75, 76

Company ¹	Headquarters	Manufacturing Location(s) ²	Product(s)	Additional source(s) ¹
Symbio ⁴	Lyon, France	France	PEMFC	28
TANAKA ⁴	Tokyo, Japan	Kanagawa, Japan	Catalyst	77, 78, 79
Toray	Tokyo, Japan	Japan, France, US, Korea	GDL	113, 114
Toshiba Energy Systems	Kanagawa, Japan	Kawasaki, Japan	PEMFC	11, 118
Umicore	Brussels, Belgium	China, US, Germany, Denmark and Korea	Catalyst	87
W. L. Gore	Newark, DE, US	Newark, DE, US; Germany, UK, Japan and China	Membrane	73, 133
location and technology information	on was unavailable from marke			irces were used when
² Manufacturing locations are spe	cifically for fuel cell components	and may not be a complete list for a given compar	ny.	
³ Company was not mentioned in	BNEF regarding fuel cells.			
⁴ Company was not mentioned by	market reports but is believed t	o be a significant manufacturer of fuel cells.		
⁵ Component is produced for use	in company end-products but is	not sold directly.		

Table B-3. Selected Manufacturing Capacities and Upcoming Developments for PEMEC, Alkaline Electrolyzer, and PEMFC Manufacturers

	Company	Manufacturing			
		Location	Capacity	Upcoming developments	Source(s)
Electrolyzers	Plug Power	Rochester, NY, US	500 MW/yr		9
	Nel ASA	Herøya/Notodden, Norway	500 MW/yr	Room to expand Herøya to capacity of 2 GW/yr	19
		Wallingford, CT	50+ MW/yr		
	Cummins	Ontario	Unavailable	Planning to build a PEM electrolyzer manufacturing facility in Guadalajara, Spain, in 2023 with capacity of 0.5–1.0 GW/yr	30
		San Miniato, Italy	300 MW/yr		13

	Company	Manufacturing			
		Location	Capacity	Upcoming developments	Source(s)
	McPhy (alkaline)	Wildau, Germany	Unavailable		
	Elogen	Les Ulis, France	160 stacks/yr (40 electrolyzers/		14
			yr)		
	Next Hydrogen	Mississauga, Ontario, Canada	20 MW/yr		20
	Green Hydrogen	Kolding, Denmark	75 MW/yr	Planning to scale to 400 MW/yr in 2023, with ability for eventual expansion to 1 GW/yr	34
	Cockerill Jingli Hydrogen (alkaline)	Suzhou, China	350 MW/yr	Planning to scale up to 500 MW/yr	37
	ITM	UK	GW scale		40
Fuel Cells	Plug Power	Rochester, NY, US	60,000 stacks/yr		9
			7 million MEAs/yr		
	Doosan ¹ (phosphoric acid fuel cell)	lskan, South Korea	63 MW/yr (144 units/yr)	Increase Iksan plant capacity to 275 MW/yr by 2022; Commercialize PEMEC/PEMFC technology 2023–2025	23, 24
		South Windsor, CT, US	Unavailable	PEINEC/PEINFC lectinology 2023-2025	
	Ballard	Burnaby, British Columbia, Canada	10,000 stacks/yr		
			1 million MEAs/yr	Expand MEA production in Burnaby to 6+ million/yr	1, 25, 26
		Hobro, Denmark	40 MW/yr		
Compony	Manufacturing		Uncoming developments	Source(e)	
---------------------------	--------------------	---------------------	---	-----------	
Company	Location	Capacity	Upcoming developments	Source(s)	
Nuvera	Fuyang, China	5,000 stacks/yr		8	
Cummins	Herten, Germany	10 MW/yr		27	
Symbio Lyon, France		Unavailable	Opening Lyon factory in 2023 with plans to scale to 200,000 stacks/yr by 2030	28	
Proton Motor Fuel Cell	Nuremburg, Germany	10,000 stacks/yr	Nuremberg facility could be scaled up to 30,000–50,000 stacks/yr	29	
Greenerity	Alzenau, Germany	Unavailable	Eventual capacity at Alzenau of 10 million MEAs/yr	44	
Horizon Fuel Cell	Rugao, China	30 MW/yr		86	

Sources Used for Information on PEMEC and PEMFC Manufacturers

The following websites, press releases, company reports, and other sources were used to compile the information above in Tables B-1 through B-3. All web pages were accessed November 5,2021, unless otherwise noted.

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B-2. SOEC and SOFC Technology

Table B-4. Selected SOEC Companies' Manufacturing Capacity and Future Development Plan

	Manufacturing			
Company	Locations	Capacity	Upcoming Developments	Source(s)
Bloom Energy	San Jose, CA, US	500MW	1GW within 1 year	1,2
FuelCell Energy	Danbury, CT, US	100 MW	200MW when second phase of expansion is done	5
Nexceris	Lewis Center, OH, US	Subsidiary company Fuelcellmaterials offers their own tailored powders SOEC Materials, SO Cell Stacks, catalysts to customer		6

	Manufacturing			
Company	Locations	Capacity	Upcoming Developments	Source(s)
OxEon	Salt Lake Valley, UT, US	Various Testing project with external partners	Under development to commercialize	7
Sunfire	Germany, Norway, and Switzerland	Operated and tested 0.25MW	3MW electrolyzer in the scope of the EU-funded MultiPLHY Project	21
Haldor Topsoe	Denmark		Topsoe will invest in a manufacturing facility producing highly efficient solid oxide electrolyzers (SOEC) with a total capacity of 500 MW/yr with the option to expand to 5 GW/yr. The facility is expected to be operational by 2023.	22
H2E Power	India	Facility in Winterthur, Switzerland and Pune, India to produce 1.5kW, 4kW, 10kW and 50kW SOFC systems.		23

Table B-5. Selected SOFC Companies' Manufacturing Capacity and Future Development Plan

Company	Manufacturing		Upcoming Developments	
Company	Locations	Capacity		Source(s)
BloomEnergy	San Jose, CA	132.6MW sales in 2020	Contracted additional 500 MW of power between 2022 and 2025 with Korea. Its Korean partner SK E&C are building factory in Gumi to manufacture 50 MW of Bloom's SOFC systems.	2,3,4
FuelCell Energy	Danbury, CT	100 MW	200MW when second phase of expansion is done.	5
Nexceris	Lewis Center, OH	Nexceris claims to have a strong distribution network to ensure global reach		6

Company	Manufacturing		Upcoming Developments	
Company	Locations	Capacity		Source(s)
WATT Fuel Cell Corporation	Mount Pleasant, PA. Offices in Port Washington and the Hampton Bays, NY	Capacity to support high volume batch or continuous production		8
Cummins	Columbus, IN, US	Cummins signs long-term agreement to ensure highest supply chain performance when purchases externally. And designs and/or manufactures their strategic components in FC technology.		9,10
OxEon	Salt Lake Valley, UT,US	Various testing project with external partners	Under development to commercialize	7
Ceres Power	UK	Bosch and Ceres Power strengthen partnership to prepare for full-scale production. Ceres Power is also working with AVL (Austria) to further strengthen competencies for SOFC technology	Multiple sites in Germany are aiming to produce an initial aggregate 200MW capacity in 2024	11
Convion	Finland	Testing project with Lempäälän Energia's energy community	Under development to commercialize	12
Elcogen	Tallinn, Estonia (SO Cells); Vantaa, Finland (SO Stacks)	In the process of delivering mass produced SOFC and SOEC	Elcogen plans to expand its European cell manufacturing capacity to 50 MW by 2021/22.	13
SolidPower	Italy	In 2020, 16,000 BlueGen power plants came off the production line.	Over the next three years, 18.9 million euros will be invested into the expansion of the factories and modern production machinery.	14
Mitsubishi Power	Japan	Business alliance with NGK Spark Plug Co., Ltd. To mass produce cell stacks		15

Compony	Manufacturing		Uncoming Developments	
Company	Locations	Capacity	Upcoming Developments	Source(s)
Aisin Seiki Co	Japan	In 2020, 47,000 units.		16
Kyocera	Japan	Demonstrated a tubular SOFC 250kW fuel cell also targeting power generation at the MW level	Double productivity by developing production technologies that make full use of Al, robots and loT, Kyocera will expand these automation technologies and systems to each business in order to improve the productivity of the Group as a whole.	17,18
Posco Energy	South Korea	SOFC under development to commercialize		19
Doosan	South Korea	63MW	Doosan plans to invest 72.4 billion won to build SOFC cell stack manufacturing line and an SOFC system assembly line, Iksan plant will increase capacity to 260MW	20

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Appendix C: Applicable Existing U.S. Policies

Existing policies provided here are summaries of information available from the Database of State Incentives for Renewables (DSIRE) and an Alternative Fuels Data Center online database.

C-1. Federal Fuel Cell Investment Tax Credit

The federal fuel cell investment tax credit:

- Applies to new stationary fuel cells or material handling fuel cell equipment rated at 500 W or greater
- Requires the electricity generating efficiency to be greater than 30%
- Can be claimed for \$3,000/kW or 30% of project cost, whichever is less
- Began to be phased out starting in 2020
- Is claimed by businesses through Internal Revenue Service Form 3468, Residential installations through Internal Revenue Service Form 5695

For additional information, see "Federal Tax Credits: Fuel Cells (Residential Fuel Cell and Microturbine System)," https://www.energystar.gov/about/federal_tax_credits/fuel_cells.

C-2. Stationary Fuel Cells

(information from Database of State Incentives for Renewables (DSIRE) database: https://www.dsireusa.org/)

At the state level, 46 of 50 states include stationary fuel cells in clean energy financial incentive programs or rules, regulations, and policies:

- 39 states and the District of Columbia include fuel cells in rules, regulations, and policies, such as Renewable Portfolio Standards, Public Benefits Fund, Net Metering, Interconnection, and Green Power Purchasing
- 34 states include fuel cells in financial incentive programs, such as tax incentives, grants, loans, rebates, performance-based incentives, feed-in tariffs, and renewable energy credits (RECs)
- 27 states include fuel cells in both financial incentives programs and rules, regulations, and policies.

Federal programs include tax credits and grants, loan guarantees or grants, and manufacturing assistance programs. Federal programs include:

- Business Energy Investment Tax Credit
- Residential Renewable Energy Tax Credit
- U.S. Department of Agriculture (USDA) Biorefinery, Renewable Chemical, and Biobased Product Manufacturing Assistance Program
- U.S. Department of Agriculture (USDA) RuralEnergy for America Program (REAP) Loan Guarantees or Grants
- U.S. Department of Agriculture (USDA) Rural Energy for America Program (REAP) Energy Audit and Renewable Energy Development Assistance Program
- DOE Loan Guarantee Program.

C-3. Hydrogen Fuel or Fuel Cell Vehicles

(information from the Alternative Fuels Data Center database: https://afdc.energy.gov/)

At the state level, 29 of 50 states include hydrogen fuel or fuel cell vehicles in renewable energy incentive programs or rules, regulations, and policies:

- 21 states and the District of Columbia include hydrogen fuel or fuel cell vehicles in rules, regulations, and policies, such as renewable fuel standards or mandates; air quality or emissions Mandates; Climate Change or renewable energy initiatives
- 19 states and the District of Columbia include fuel cells in financial incentive programs, such as tax incentives or vehicle rebates
- 9 states and the District of Columbia include fuel cells in both incentives programs and rules, regulations, and policies.

Federal programs include tax credits for fuel cell vehicles, fuels, and fueling infrastructure and targeted collaborative grant programs that include alternative fuel vehicles, such as the U.S. Environmental Protection Agency Diesel Emissions Reduction Act and U.S. Department of Transportation Congestion Mitigation and Air Quality programs. Federal programs include:

- Alternative Fuel Excise Tax Credit
- Alternative Fuel Tax Exemption
- Alternative Fuel Infrastructure Tax Credit
- Clean Cities Coalition Network
- Clean Construction and Agriculture
- Clean School Bus
- Congestion Mitigation and Air Quality (CMAQ) Improvement Program
- Diesel Emissions Reduction Act
- Low or Zero Emission Ferry Program
- National Multimodal Cooperative Freight Research Program
- Port Infrastructure Development Program
- Ports Initiative
- Public School Energy Program
- State Carbon Reduction Program
- State Energy Program (SEP) Funding
- Voluntary Airport Low Emission (VALE) Program Alternative Fuel Excise Tax Credit.

Appendix D. Summary of Material Risks

Table D-1. Electrolyzer and Fuel Cell Evaluation Table

(prepared for DOE Office of Policy)

Supply Chain Element	Product/Components	Significant Domestic Suppliers	Significant Domestic Demand	Projected Significant Domestic Demand	Significant Global Market	Projected Significant Global Demand	Cost Competitive among U.S. Suppliers	Cost Competitive between U.S. Suppliers vs. Global Suppliers	Is foreign supply source significant secure?	Is there sufficient effort to address environmental concerns?	Is there sufficient effort to address human rights concerns?	Does it make sense to build a domestic capability for this product/ component?
	PGM-containing ore	Yes	Yes	Maybe	Yes	Yes	Maybe	Yes	No	No	No	Maybe
	PGM concentrate	Maybe	Yes	Maybe	Yes	Yes	Maybe	Yes	Maybe	No	No	Maybe
	PGM (Pt) (catalyst)	No	Yes	Yes	Yes	Yes	n/a	n/a	Yes	Yes	Yes	Maybe
	Pt-based catalyst	No	Yes	Yes	Yes	Yes	No	Yes	Maybe	Yes	Maybe	Yes
	PGM (Ir) (catalyst)	No	No	Yes	No	Yes	n/a	n/a	Yes	Yes	Yes	Yes
Raw materials	lr	No	Yes	Yes	Yes	Yes	No	No	No	No	No	Yes
	Graphite (BPP)	No	Yes	Yes	Yes	Yes	n/a	n/a	Maybe	Maybe	Unknown	Maybe
	Titanium (ore, metal TiCl)	No	Yes	Yes	Yes	Yes	n/a	n/a	Yes	Unknown	Unknown	
	Aluminum (housing)	Yes	Yes	No	Yes	Unknown	Yes	Yes	No	Unknown	Unknown	
	Chromium (SS)	No	Yes	No	Yes	No	n/a	n/a	Yes	Unknown	Unknown	
	Silicone Elastomer (Seal)	Unknown	Unknown	Unknown	Unknown	No	Unknown	Unknown	Unknown	Unknown	Unknown	
	Viton Elastomer (Seal)	Unknown	Unknown	Unknown	Unknown	Unknown	Unknown	Unknown	Unknown	Unknown	Unknown	

Supply Chain Element	Product/Components	Significant Domestic Suppliers	Significant Domestic Demand	Projected Significant Domestic Demand	Significant Global Market	Projected Significant Global Demand	Cost Competitive among U.S. Suppliers	Cost Competitive between U.S. Suppliers vs. Global Suppliers	Is foreign supply source significant secure?	Is there sufficient effort to address environmental concerns?	Is there sufficient effort to address human rights concerns?	Does it make sense to build a domestic capability for this product/ component?
	Ethylene Propylene Diene Monomer Elastomer (Seal)	Unknown	Unknown	Unknown	Unknown	Unknown	Unknown	Unknown	Unknown	Unknown	Unknown	
	Borosilicate glass (HT)	Unknown	No	Maybe	No	Maybe	Unknown	Unknown	Unknown	Unknown	Unknown	
	Lanthanum (HT)	No	No	Maybe	No	Maybe	n/a	n/a	No	Unknown	Unknown	
	Strontium (HT)	No	No	Maybe	No	Maybe	n/a	n/a	Yes	Unknown	Unknown	Yes
	Cobalt (HT)	No	Yes	Yes	Yes	Yes	n/a	n/a	Yes	Maybe	Maybe	
	Yttrium (HT)	No	No	Yes	No	Yes	n/a	n/a	No	Unknown	Unknown	
	Zirconium (HT)	Yes	No	Maybe	No	Maybe	Yes	Yes	Yes	Unknown	Unknown	
	Manganese (HT)	No	No	Maybe	No	Maybe	n/a	n/a	Maybe	Unknown	Unknown	
	Cerium (HT)	No	Unknown	Unknown	Unknown	Unknown	n/a	n/a	Maybe	Unknown	Unknown	
	Iron (HT)	Yes	Yes	Maybe	Yes	Maybe	Unknown	Unknown	Yes	Unknown	Unknown	
	Nickel (SS)	No	Yes	No	Yes	No	Unknown	Unknown	Yes	Yes	Unknown	
	Pt-based catalyst	No	Yes	Yes	Yes	Yes	No	Yes	Maybe	Yes	Maybe	Yes
Drocossed	Ir-based catalyst	No	Yes	Yes	Yes	Yes	No	No	Maybe	Yes	Maybe	Yes
Processed material	Other PGM-based catalysts	Maybe	Yes	Yes	Yes	Maybe	Maybe	Maybe	Maybe	Yes	Maybe	Yes
	Perfluorosulfonic acid; PFSA (Nafion) (electrolyte)	Unknown	Yes	Yes	Yes	Yes	Unknown	Unknown	Unknown	Unknown	Unknown	

Supply Chain Element	Product/Components	Significant Domestic Suppliers	Significant Domestic Demand	Projected Significant Domestic Demand	Significant Global Market	Projected Significant Global Demand	Cost Competitive among U.S. Suppliers	Cost Competitive between U.S. Suppliers vs. Global Suppliers	Is foreign supply source significant secure?	Is there sufficient effort to address environmental concerns?	Is there sufficient effort to address human rights concerns?	Does it make sense to build a domestic capability for this product/ component?
	Sulfonated polyether ether ketone (s-PEEK) (alternative electrolyte)	Unknown	No	Maybe	No	Maybe	Unknown	Unknown	Unknown	Unknown	Unknown	
	Polystyrene sulfonic acid (PSSA) (alternative electrolyte)	Unknown	No	Maybe	No	Maybe	Unknown	Unknown	Unknown	Unknown	Unknown	
	Pt alloys (catalyst)	Unknown	No	Yes	Yes	Yes	Unknown	Unknown	Unknown	Unknown	Unknown	
	carbon metal oxide, carbides, etc.) (catalyst support)	Unknown	No	Yes	No	Yes	Unknown	Unknown	Unknown	Unknown	Unknown	
	PAN (polyacrylonitrile) - based carbon fiber (GDL)	Unknown	No	Yes	No	Yes	Unknown	Unknown	Unknown	Unknown	Unknown	
	Polytetraflouroethylene (PTFE) (CF GDL coating, membrane support)	Yes	No	No	Unknown	No	Unknown	Unknown	Unknown	Unknown	Unknown	Yes
	Stainless Steel (end plate)	Unknown	Yes	No	Yes	No	Unknown	Unknown	Unknown	Unknown	Unknown	
	Cathode Contact Layer (LSM)	No	No	Maybe	No	Maybe	n/a	n/a	Unknown	Unknown	Unknown	
	Cathode Current Collector (LSCF)	No	No	Yes	No	Yes	n/a	n/a	Unknown	Unknown	Unknown	

Supply Chain Element	Product/Components	Significant Domestic Suppliers	Significant Domestic Demand	Projected Significant Domestic Demand	Significant Global Market	Projected Significant Global Demand	Cost Competitive among U.S. Suppliers	Cost Competitive between U.S. Suppliers vs. Global Suppliers	Is foreign supply source significant secure?	Is there sufficient effort to address environmental concerns?	Is there sufficient effort to address human rights concerns?	Does it make sense to build a domestic capability for this product/ component?
	Cathode Active Layer (LSM-YSZ)	No	No	Maybe	No	Maybe	n/a	n/a	Unknown	Unknown	Unknown	
	Electrolyte (8YSZ)	No	No	Maybe	No	Maybe	n/a	n/a	Unknown	Unknown	Unknown	
	Anode Active Layer and Support (Ni-YSZ)	No	No	Yes	No	Yes	n/a	n/a	Unknown	Unknown	Unknown	
	Interconnect - Frame, Separator Plate, Anode and Cathode Flow Fields, End Plates (SS- 441)	No	No	Yes	No	Yes	n/a	n/a	Unknown	Unknown	Unknown	
	Spacers (Borosilicate glass)	No	No	Maybe	No	Maybe	n/a	n/a	Unknown	Unknown	Unknown	
	Membrane Electrode Assembly	Unknown	No	Yes	No	Yes	Unknown	Unknown	Unknown	Unknown	Unknown	Yes- PEM
Subcomponents	Electrolyte Membrane	Yes (Support)	No	Yes	No	Yes	Unknown	Yes	Yes	Unknown	Unknown	Yes - PEM Support
		No (lonomer)							Maybe			No (PEM ionomer)
	Supported Catalyst	Unknown	No	Yes	No	Yes	Unknown	Yes	Unknown	Unknown	Unknown	
	Gas Diffusion Layer	Unknown	No	Yes	No	Yes	Unknown	Maybe	Yes	Unknown	Unknown	Yes - PEM
	Bipolar Plates	Unknown	No	Yes	No	Yes	Unknown	Yes	Unknown	Unknown	Unknown	Yes - PEM

Supply Chain Element	Product/Components	Significant Domestic Suppliers	Significant Domestic Demand	Projected Significant Domestic Demand	Significant Global Market	Projected Significant Global Demand	Cost Competitive among U.S. Suppliers	Cost Competitive between U.S. Suppliers vs. Global Suppliers	Is foreign supply source significant secure?	Is there sufficient effort to address environmental concerns?	Is there sufficient effort to address human rights concerns?	Does it make sense to build a domestic capability for this product/ component?
	Solid Anode (Active + Support)	No	No	Yes	No	Yes	n/a	n/a	Yes	Unknown	Unknown	No
	Solid Electrolyte	No	No	Yes	No	Yes	n/a	n/a	Yes	Unknown	Unknown	No
	Solid Cathode (Active + Support)	No	No	Yes	No	Yes	n/a	n/a	Yes	Unknown	Unknown	No
	Cell Interconnects	No	No	Yes	No	Yes	n/a	n/a	Yes	Unknown	Unknown	No
	PEM Fuel Cells and Electrolyzers	Yes	No	Yes	No	Yes	Unknown	Yes	Yes	Unknown	Unknown	Yes - PEM
	PEM Electrolyzers	Yes	No	Yes	Yes	Yes	Maybe	Yes	No	Yes	Maybe	Yes
End Product	PEM FCs	Maybe	Maybe	Yes	Yes	Yes	No	Yes	Maybe	Yes	Maybe	Yes
	SOEC Electrolyzers	Yes	No	Yes	Yes	Yes	No	No	Maybe	Yes	Maybe	Maybe
	SOFCs	Maybe	Maybe	Yes	Yes	Yes	Maybe	Maybe	Maybe	Yes	Maybe	Yes
	Solid Oxide Fuel Cell and Electrolyzers	Yes	No	Yes	No	Yes	Unknown	Yes	Yes	Unknown	Unknown	Maybe

Color-coding: Yes = Green; No = Red; Maybe = Yellow; Unknown = Gray; n/a = White Last column: Yes/Maybe = Blue; No = Dark Gray; All others are white.



For more information, visit: energy.gov/policy/supplychains