

Electrophosphorescence for Solid-State Lighting

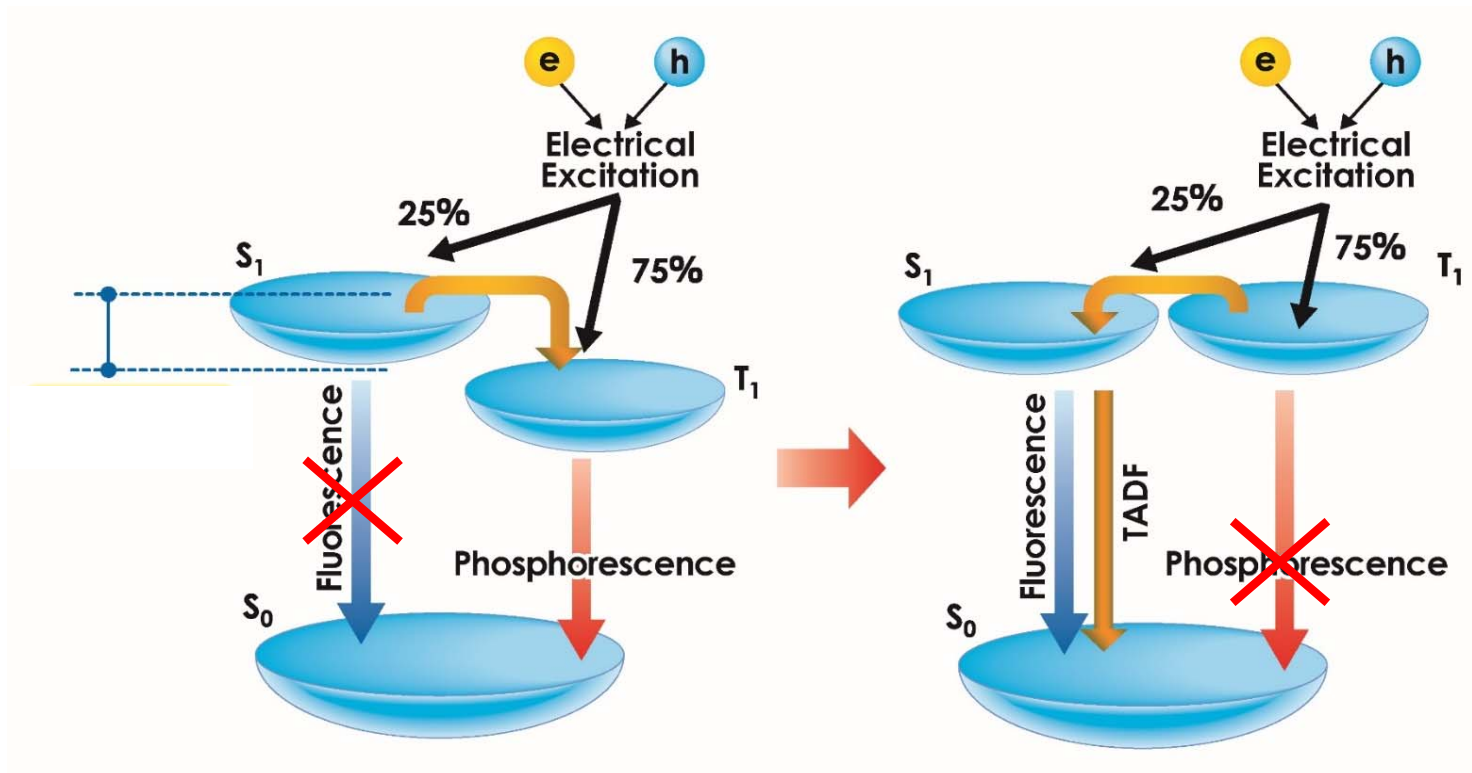
A decorative graphic consisting of three horizontal lines. The top line is red, the middle line is yellow, and the bottom line is red. The lines are slightly offset from each other, creating a layered effect.

Mark Thompson

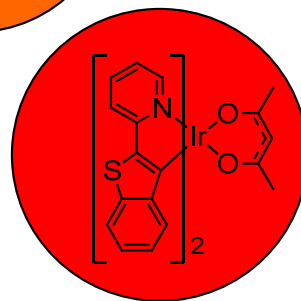
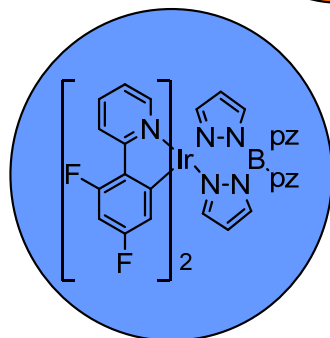
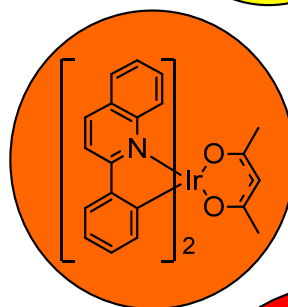
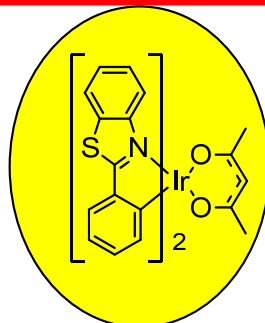
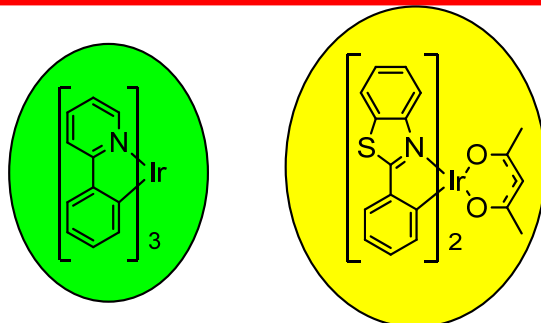
Department of Chemistry, University of Southern California

High Efficiency OLED Options

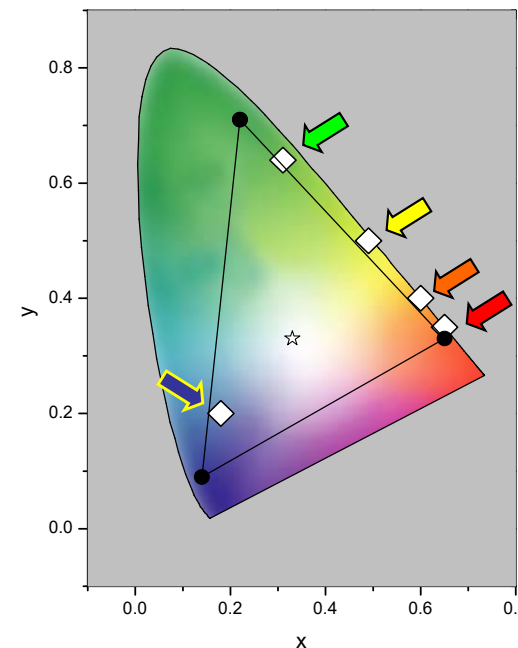
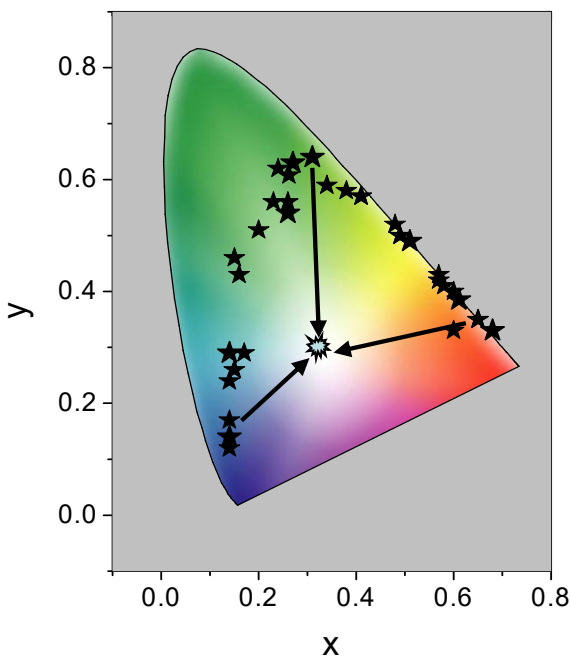
- Multiple emitters for **RGB** \Rightarrow white
- **ALL** excitons need to be utilized
- Ir-based emitters: well proven in mobile displays and television
- Thermally Assisted Delayed Fluorescence (TADF)



Organometallic Ir complexes in OLEDs

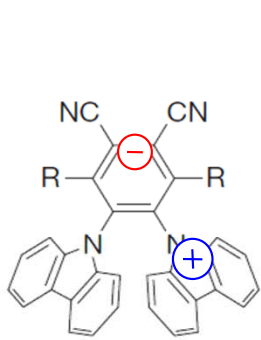


Phosphorescence efficiency of Ir emitters is ~100%

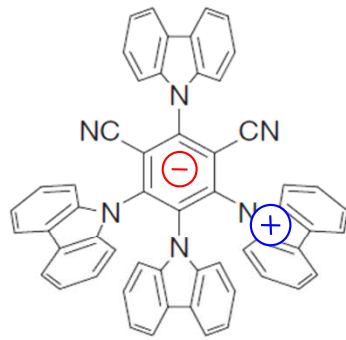


- Efficient phosphorescence with $\tau = 1-3 \mu\text{s}$ lifetime
- Optimized OLEDs give external efficiency \Rightarrow Internal efficiency up to 70-100%
- Green, Yellow, Red lifetime are very long, blue is a problem

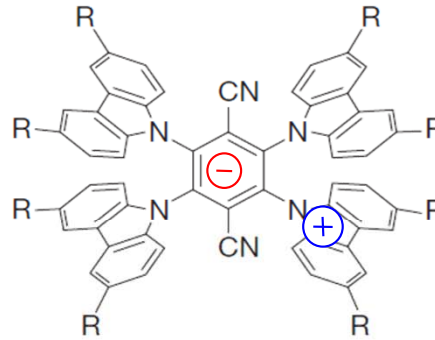
High Efficiency TADF



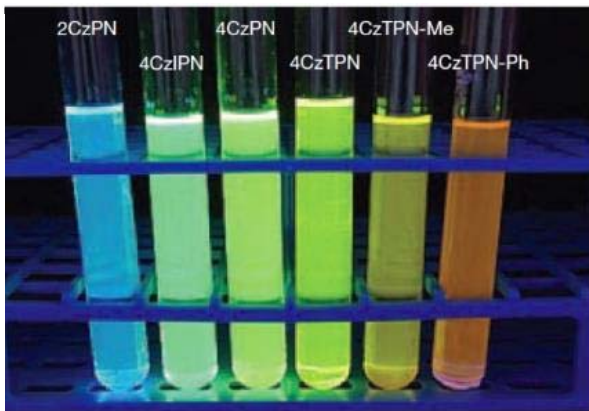
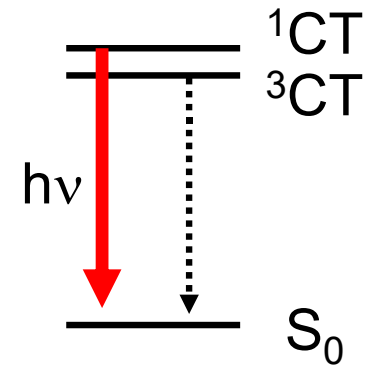
4CzPN: R = carbazolyl
2CzPN: R = H



4CzIPN



4CzTPN: R = H
4CzTPN-Me: R = Me
4CzTPN-Ph: R = Ph



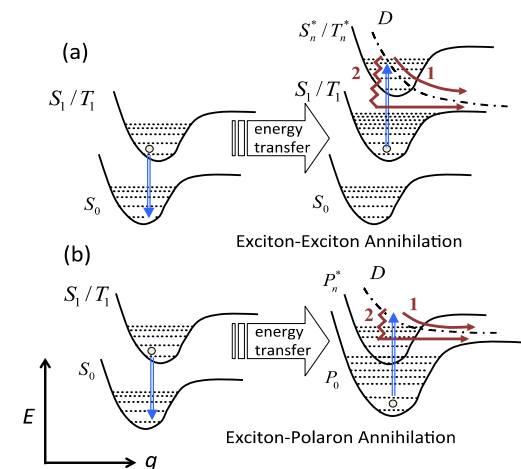
$$k_r(CT^{SOC}) \propto \frac{\langle {}^3CT | H_{SO} | {}^1CT \rangle}{\Delta E_{ST}} \cdot \langle S_0 | er | {}^1CT \rangle$$

- Color (EQE): Green (**19.3%**), Orange (11.2%), Sky Blue (8%)
– Uoyama, *et. al.*, *NATURE*, **2012**, 492
- Orthogonal donor-acceptor give CT excited state with small ΔE_{ST}
- $\tau \sim 5 \mu\text{s}$ or higher, no metal to enhance SOC, weak 1CT oscillator

Phosphors and TADF Emitters good, what's the problem?

- What about device lifetime?
 - **Green** and **Red** >30 years at display intensity 😊
 - **Blue** <2 years at display brightness 😞
- Device degradation is due to **bimolecular annihilation***

- (a) TTA: $T_1 + T_1 \rightarrow S_0 + S_n(\text{hot})$
- (b) TPA: $T_1 + H^\cdot \rightarrow S_0 + H^\cdot(\text{hot})$
- **G/R**: $S_n(\text{hot})$ or $H^\cdot(\text{hot}) \Rightarrow$ relax to GS
- **B**: $S_n(\text{hot})$ or $H^\cdot(\text{hot}) \Rightarrow$ degradation

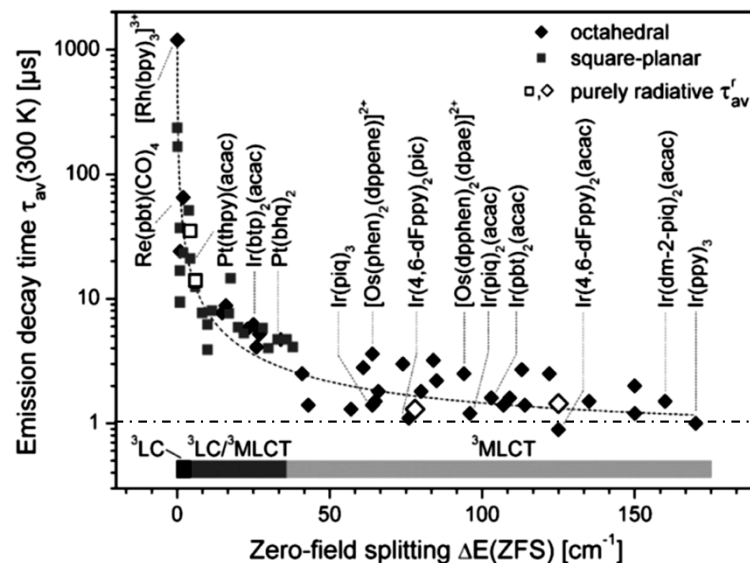


- OLED displays: Ir-based emitters for **Green**
organic fluorescent dopant for **Blue**
 - nsec lifetime of **Blue** fluorescence decreases bimolecular decay

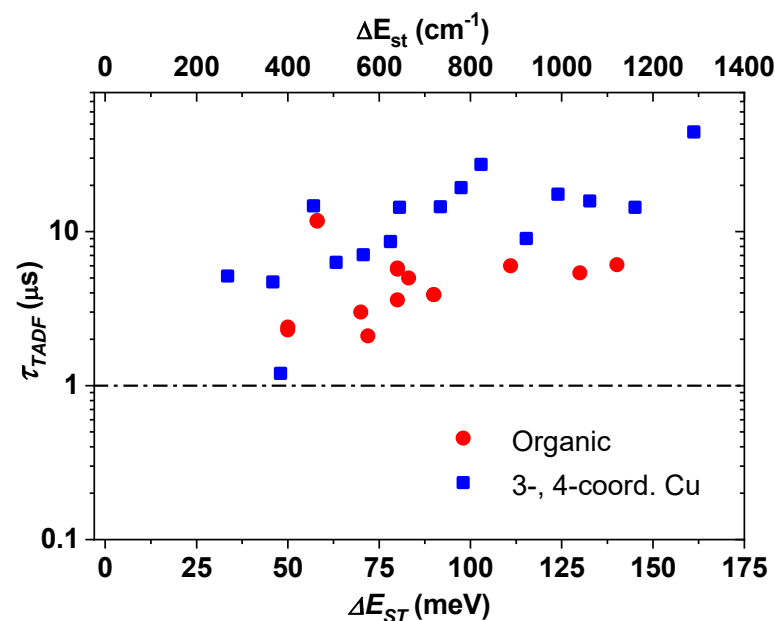
* N. Giebink, *et al.*, *J. Appl. Phys.* **2008**, 103, 044509

Best phosphor and TADF Lifetimes $\sim 1 \mu\text{s}$

- Ir and Pt based phosphors seem to flat-line at $\sim 1 \mu\text{s}$ (H. Yersin, *et al.*, *Coord. Chem. Rev.* **2011**)



- TADF based emitters hit 1-2 μs baseline, most are in neat solids not doped films (Organic: Liu, Y., *et al.* *Nature Rev. Mat.*, **2018**) (Copper: R. Czerwieniec, *et al.*, *Coord. Chem. Rev.* **2016**)



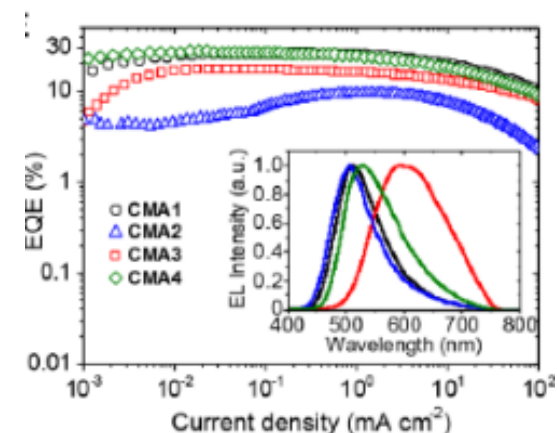
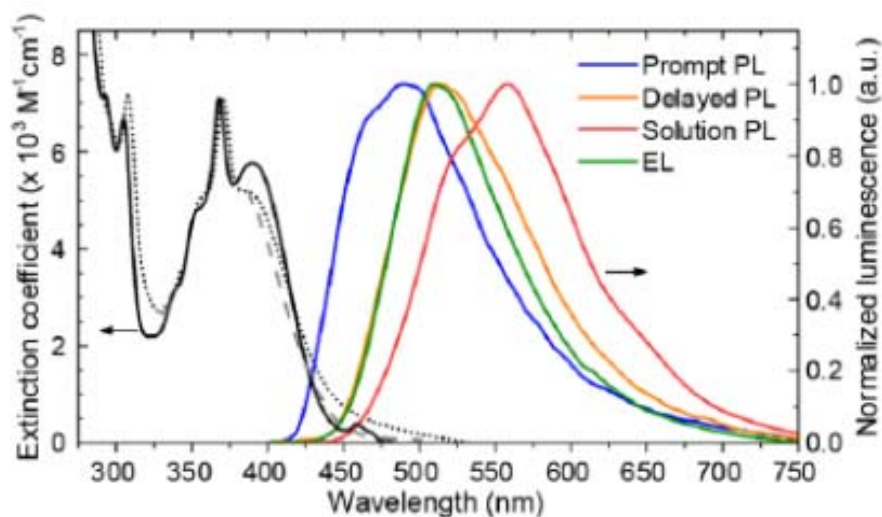
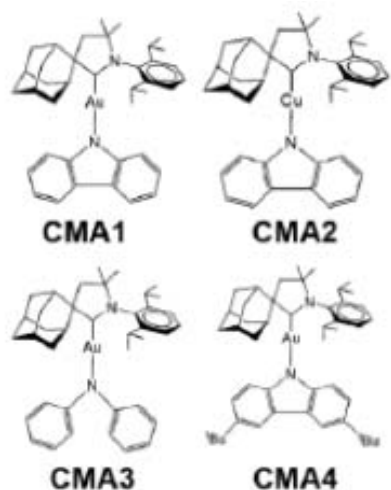
2-coordinate CAAC-M-Cz

Science

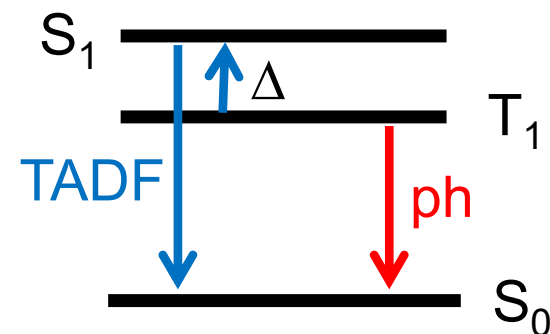
REPORTS

Cite as: D. Di *et al.*, *Science*
10.1126/science.aah4345 (2017).

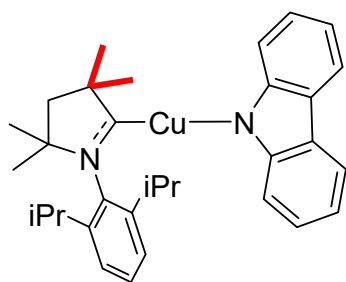
High-performance light-emitting diodes based on carbene-metal-amides



- Clearly both singlet and triplet are utilized
- Are these “normal” phosphors or TADF?
- Very little photophysical characterization

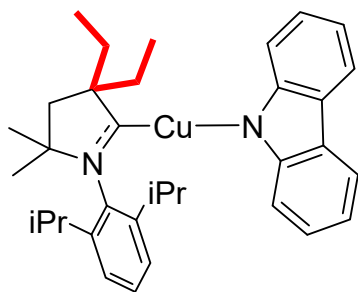


CAAC-Cu-Cz complexes



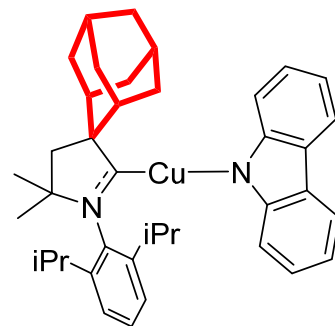
Me₂

$\Phi_{PL} = 11\%$



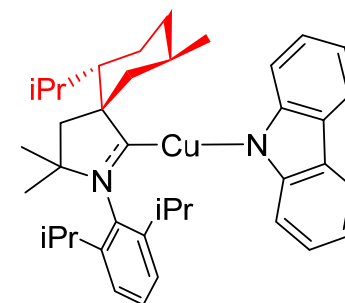
Et₂

$\Phi_{PL} = 56\%$



Ad

$\Phi_{PL} = 68\%$



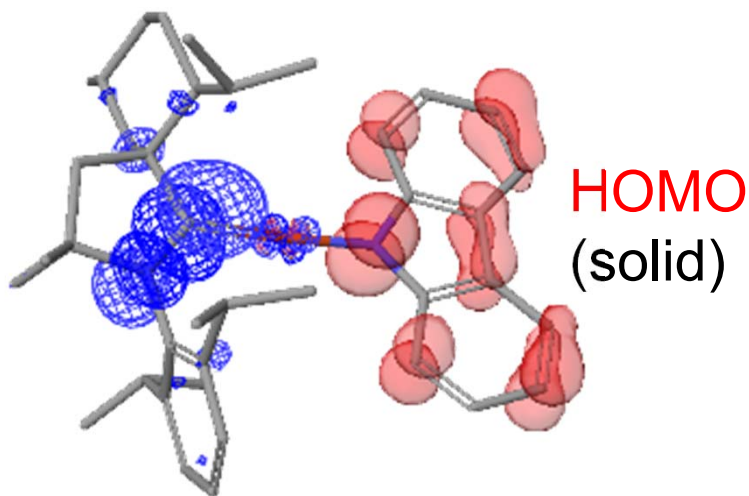
Men

$\Phi_{PL} = 100\%$

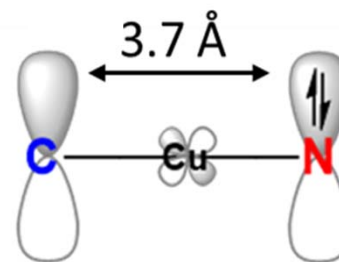
in solution and
doped thin film

Increasing Φ_{PL} IN SOLUTION

LUMO
(mesh)



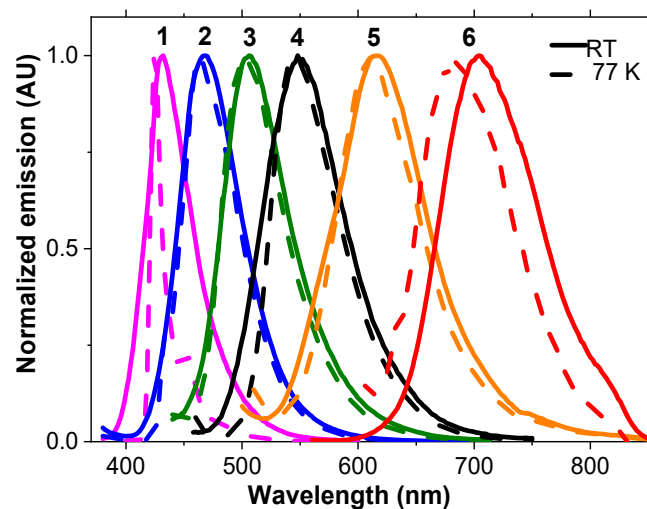
HOMO
(solid)



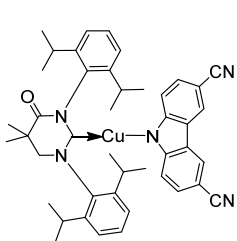
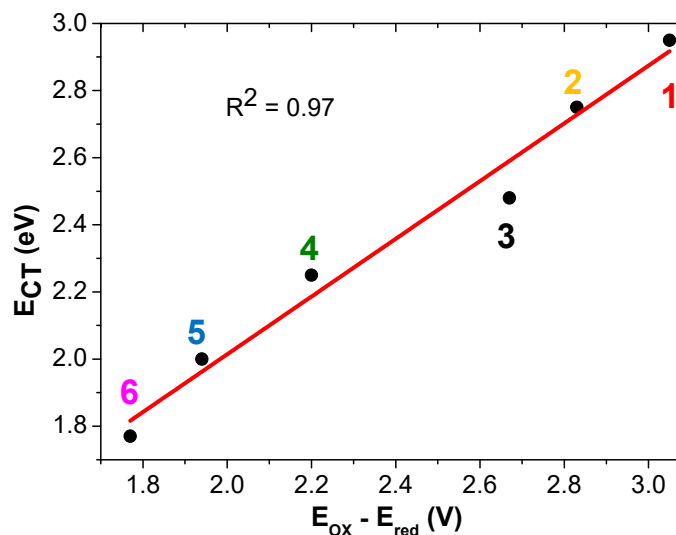
Excited state is
interligand charge
transfer (ICT),
NOT MLCT

Absorption (THF) and emission (PS thin film) spectra

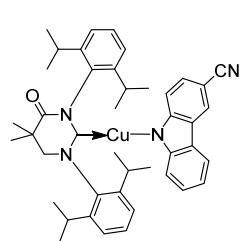
1% doped polystyrene thin film



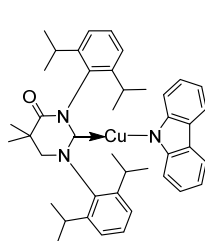
THF Solution



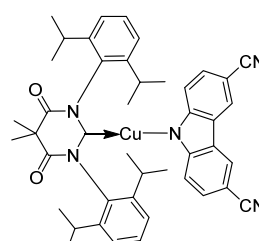
MAC*CuCzCN2 (1)



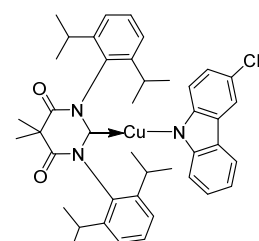
MAC*CuCzCN (2)



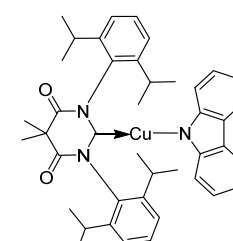
MAC*CuCz (3)



DAC*CuCzCN2 (4)



DAC*CuCzCN (5)



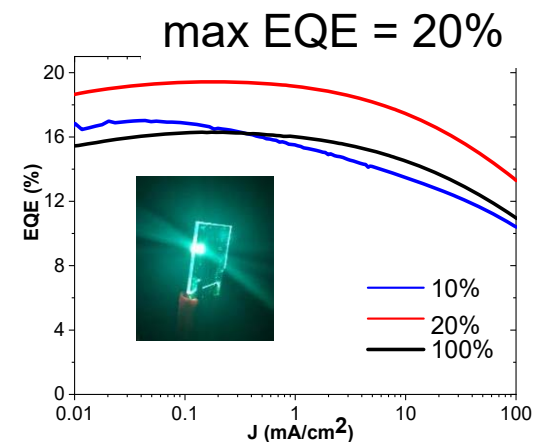
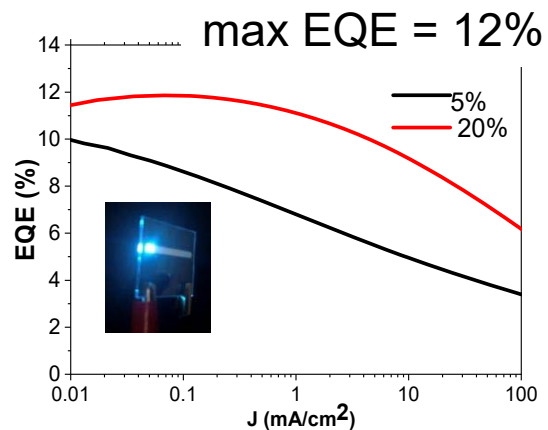
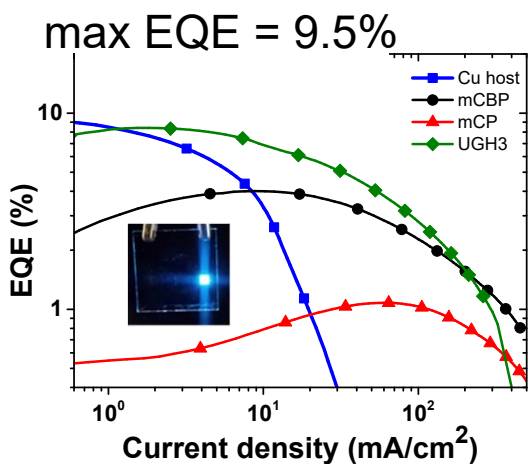
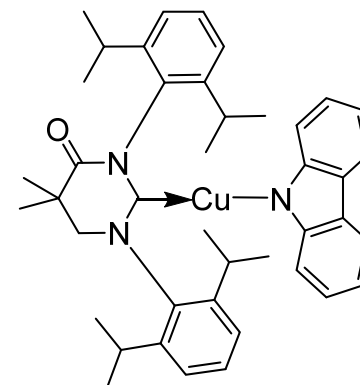
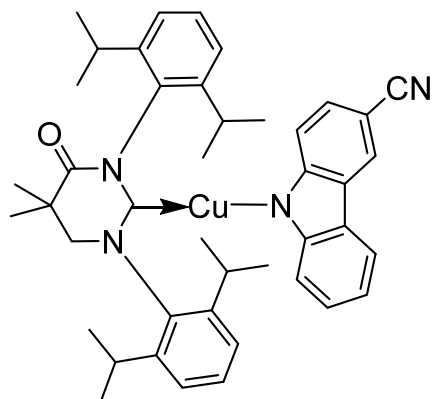
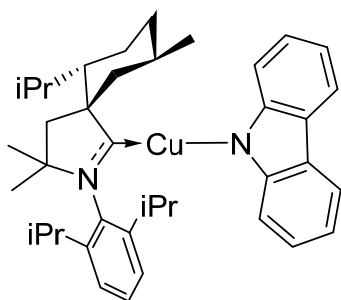
DAC*CuCz (6)

$^3Cz + ^1,^3CT$
 $\Phi_{PL}(RT) = 80\%$

$\Phi_{PL}(RT) = 80-100\%$
 $\tau = 1-1.5 \mu s$
 $(k_r = 7 \times 10^5 s^{-1})$

Energy gap law problem
 $\Phi_{PL}(RT) = 30, 2\%$

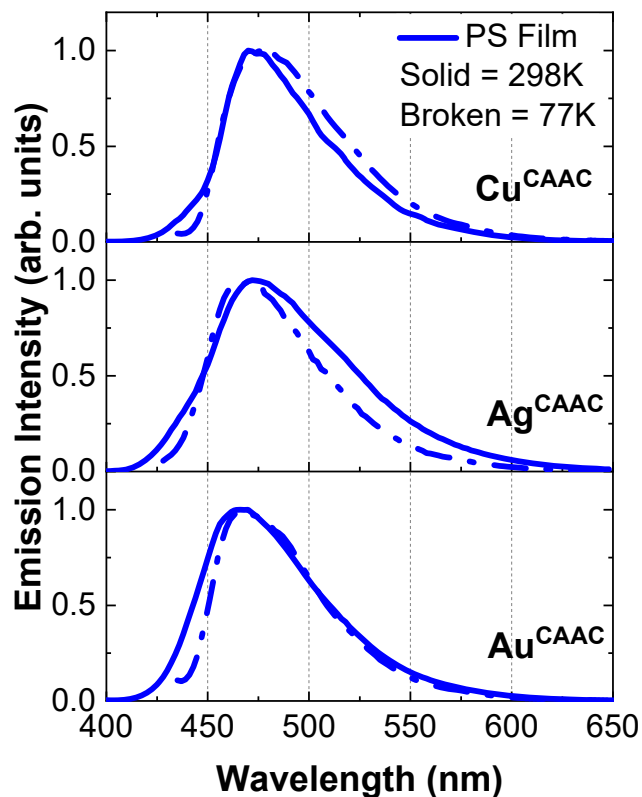
(carbene)Cu(Cz) based OLEDs



R. Hamze, *et al.*, *Science*, 2019

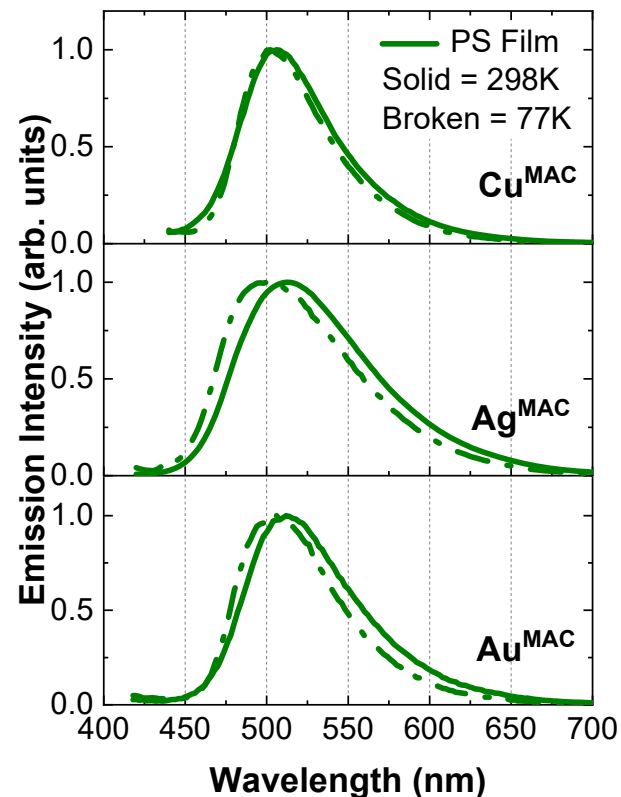
S. Shi, *et al.*, *JACS*, 2019

Copper, Silver, Gold Based Phosphors



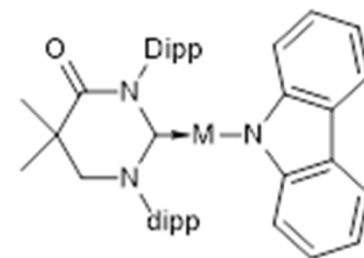
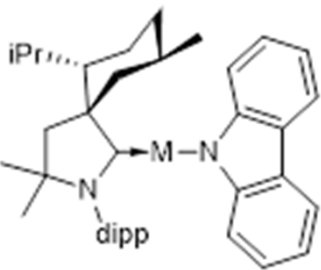
Blue Emission

	Φ_{PL}	τ (μs)
Cu	1.0	2.8
Ag	1.0	0.5
Au	1.0	1.1



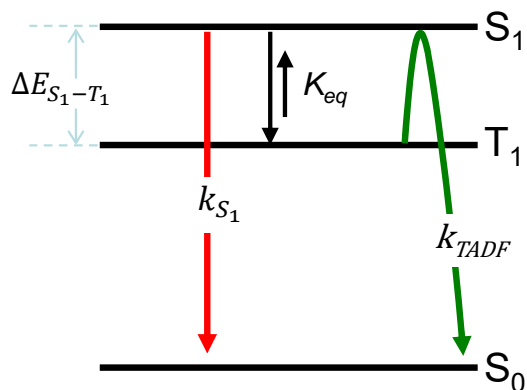
Green Emission

	Φ_{PL}	τ (μs)
Cu	0.9	1.4
Ag	0.8	0.33
Au	0.9	0.83



Where is the best place to look for lower τ_{TADF} ?

- Intersystem crossing here is very fast, τ (ISC) = 20-200 ps
- Ag shows the fastest TADF rate primarily due to small $\Delta E_{S_1-T_1}$
- In the limit of very fast ISC: $\tau_{TADF} = \tau_{S_1}/K_{eq}$; $\dagger K_{eq} \propto \exp(\Delta E_{S_1-T_1})^*$



- Is decreasing $\Delta E_{S_1-T_1}$ further useful?

- Consider K_{eq} at 300K for Ag^{CAAC}

- $\Delta E_{ST} = 150 \text{ cm}^{-1} \Rightarrow K_{eq} = 0.16$

- $\Delta E_{ST} = 50 \text{ cm}^{-1} \Rightarrow K_{eq} = 0.26$

$\tau_{TADF} = 330 \text{ ns}$

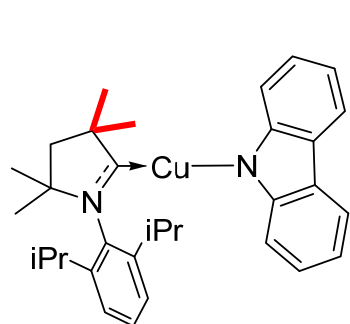
- Decreasing τ_{TADF} further will require a decrease in τ_{S_1}

	τ_{meas} s	τ_{rad}	<u>Full kinetic scheme</u> 200-325K	
	(μs)	(μs)	$\Delta E_{S_1-T_1}$ (cm^{-1})	τ_{S_1} (ns)
Cu^{CAAC}	2.8	2.8	590	73
Ag^{CAAC}	0.50	0.50	150	85
Au^{CAAC}	1.1	1.1	570	25
Cu^{MAC}	1.4	1.6	570	28
Ag^{MAC}	0.33	0.42	180	46
Au^{MAC}	0.83	0.98	570	24

\dagger D. Sylvinson M.R., M.E. Thompson, *et al.*, *Mat. Horiz.*, **2020**

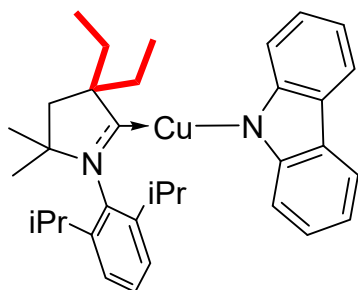
* P. F. Jones, A. R. Calloway, *Chem. Phys. Lett.*, **1971**

What makes (carbene)Cu-Cz complexes work?



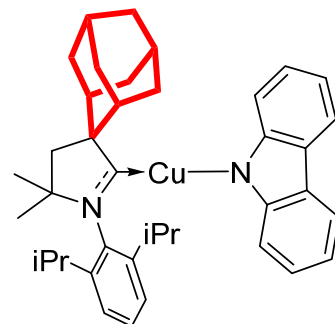
Me₂

$\Phi_{\text{PL}} = 11\%$



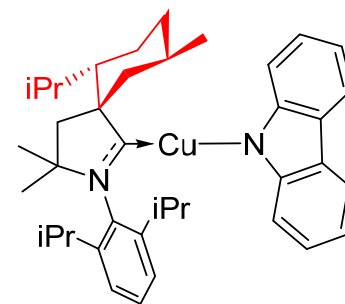
Et₂

$\Phi_{\text{PL}} = 56\%$



Ad

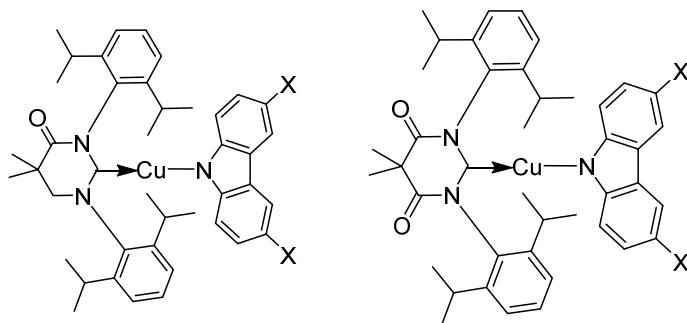
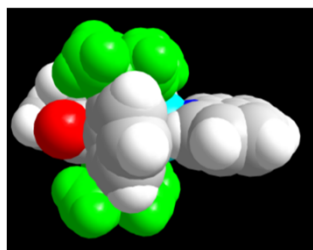
$\Phi_{\text{PL}} = 68\%$



Men

$\Phi_{\text{PL}} = 100\%$

in solution and
doped thin film



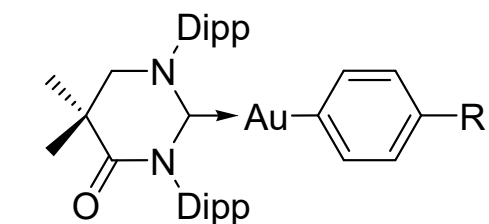
High efficiency for violet→orange emitters
Low efficiency if Dipp replaced with phenyl

Are sterically bulky ligands required? What do they do?

- Keep C:→M-N linear
- Prevent rotation about the C-M or M-N bonds
- Prevent the formation of exciplexes and ligand rotation

Origin of high efficiency for two-coordinate complexes

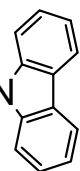
T₁ Geometries



MLCT
excited
state
(C⁻-Au⁺-N)

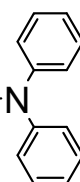
H: R=H

Cz: R=

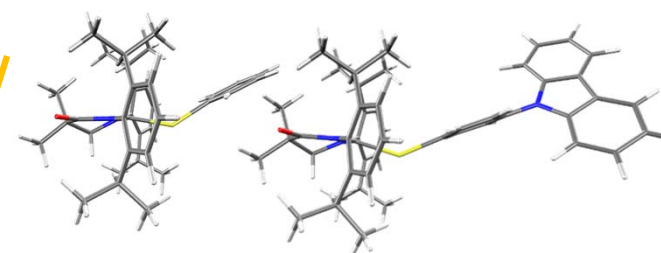


ICT
excited
state
(C⁻-Au-N⁺)

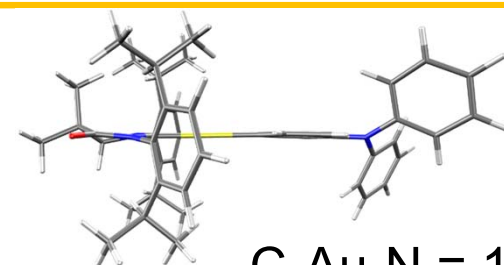
Nφ₂: R=



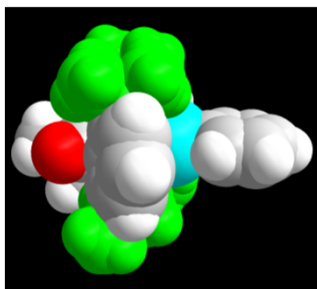
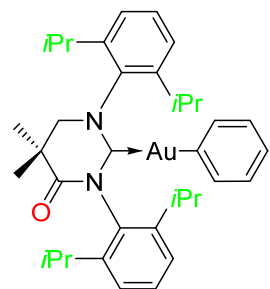
Complex	PS film	
	Excited State	Φ _{PL}
H, Cz	MLCT	< 10 ⁻³
Nφ ₂	ICT	0.38



C-Au-N ~ 120°



C-Au-N = 180°



No steric constraints
on ligand rotation

- Initial thought was that steric interactions were required to maintain linear structure and high F_{PL}.
(WRONG)
- The key is to avoid MLCT excited state and the Renner-Teller distortion it comes with.

Summary

- WOLED: lots of ways being investigated
 - High efficiency and color quality: harvest **ALL** excitons
 - Need platforms with long operational lifetime, which requires long lived blue
 - Operational lifetime of blue OLED is enhanced by short radiative lifetime
- Copper, Silver and Gold may compete well with iridium
 - High efficiency, high radiative rate
 - Easily color tunable
 - Very small S_1/T_1 gap
 - OLED lifetime: coming, we need to rethink host and transport materials

