
Non-contact Thermophysical Characterization of Solids and Fluids for Gen3 Concentrating Solar Power

Generation 3 Concentrating Solar Power Systems
DE-EE0008379

Topic Area 2B - Gen3 Research and Analysis

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Objectives and Impacts

Problem Statement

- Challenging and time-consuming to measure thermophysical properties of high-temperature heat transfer fluids (HTFs), e.g., particles and molten salts.
- Lack of *in-situ* diagnostic tools to monitor thermophysical properties of HTFs.

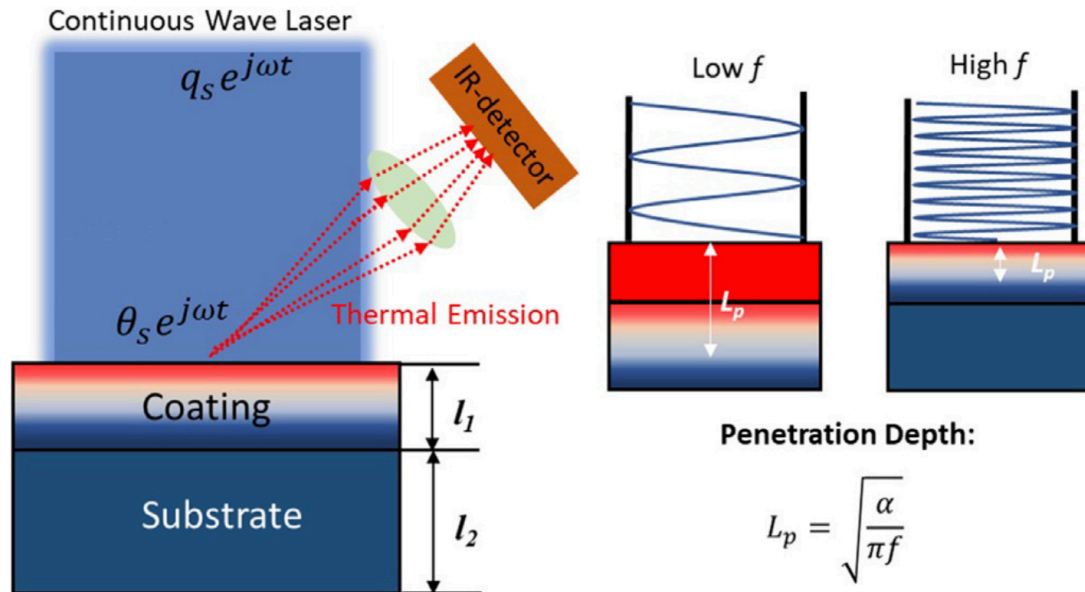
Objectives

- Develop and apply Modulation Photothermal Radiometry (MPR) as an **accurate and non-contact** method to characterize high-temperature thermophysical properties of HTFs.
- Characterize and understand the heat transfer mechanism of HTFs at the flowing state.
- *In-situ* diagnostics of HTFs plants and of their corrosion behaviors.

Impacts to Gen3 CSP

- Provide a convenient tool to measure thermophysical properties of emerging solids (e.g., particles) and fluids (e.g., molten salts) under both **stationary** and **moving** states for Gen3 CSP systems at high temperature
- Transfer the MPR technique a diagnostic tool for other laboratories and *in-situ* tests in other Gen3 projects.

Working Principles of MPR



MPR measurement setup
(a) Schematics of MPR system
(b) Principle of MPR measurement.

At low modulated frequency:

$$\theta_s = q_s \left(\frac{1}{e_m \sqrt{\omega}} + R_{sc} \right)$$

q_s : heat flux

e_m : Thermal effusivity of medium
($e = \sqrt{k\rho c_p}$)

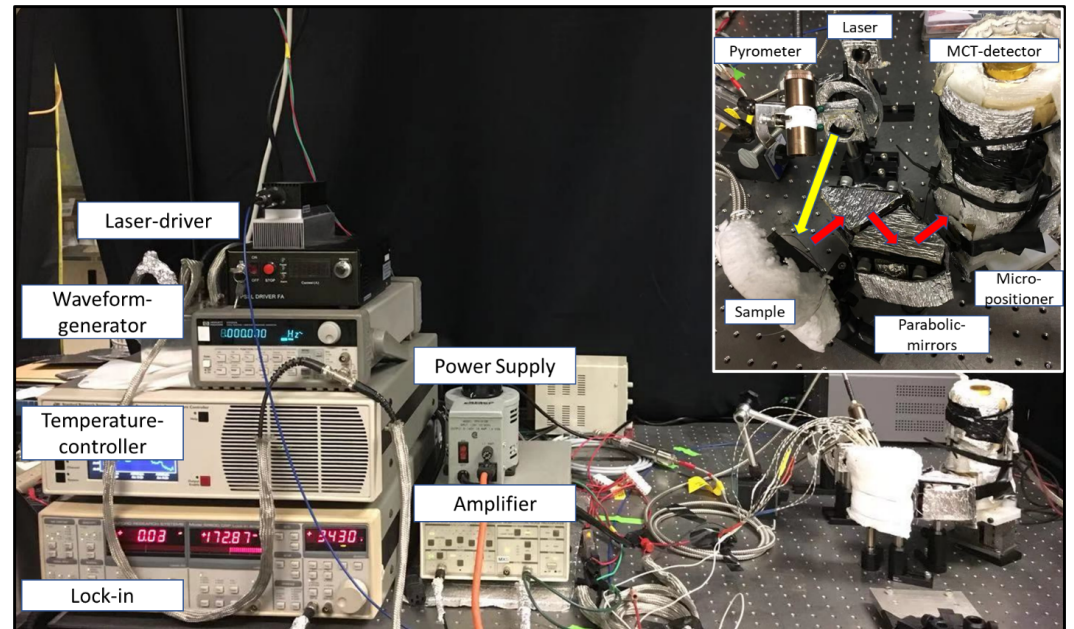
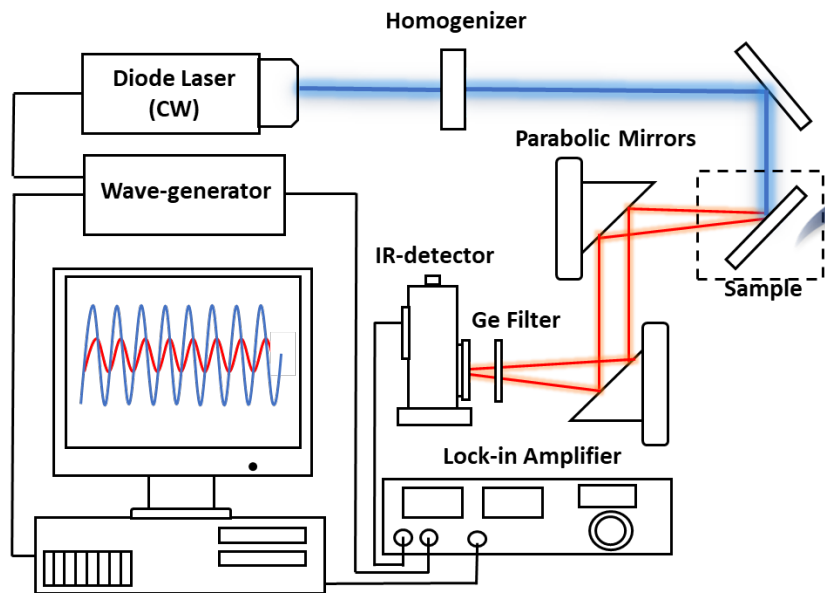
ω : angular frequency

R_{sc} : Thermal resistance of coating and shell.

- Thermal effusivity of medium e_m is obtained at low frequency; Thermal conductivity of shell and coating is obtained from the medium and high frequency ranges, respectively.
- Thermal conductivity of medium is obtained with literature or measured values of density ρ and specific heat c_p

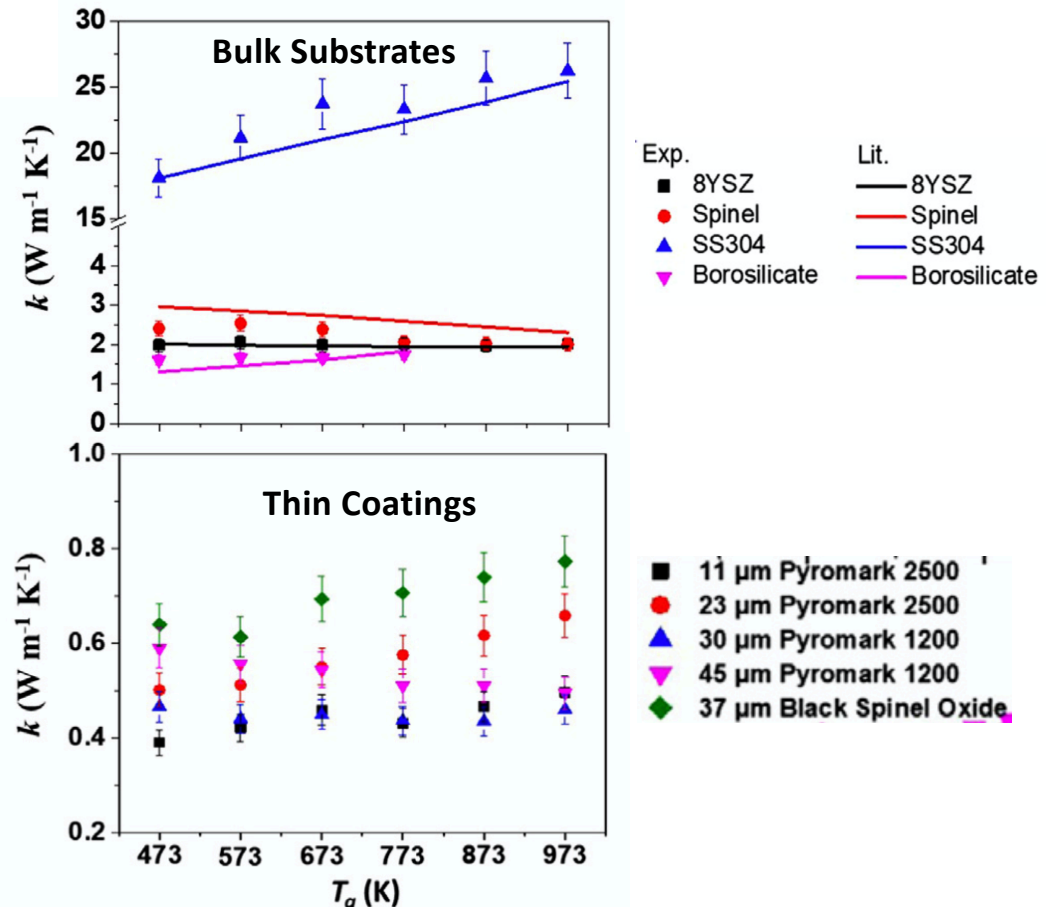
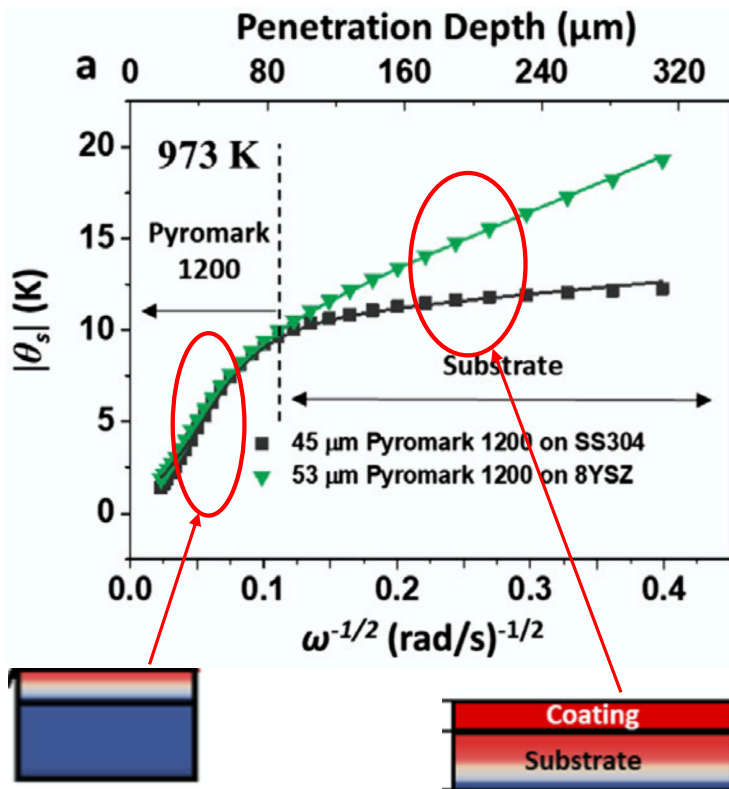
MRP Setup

a



Measurement on Substrates and Coatings

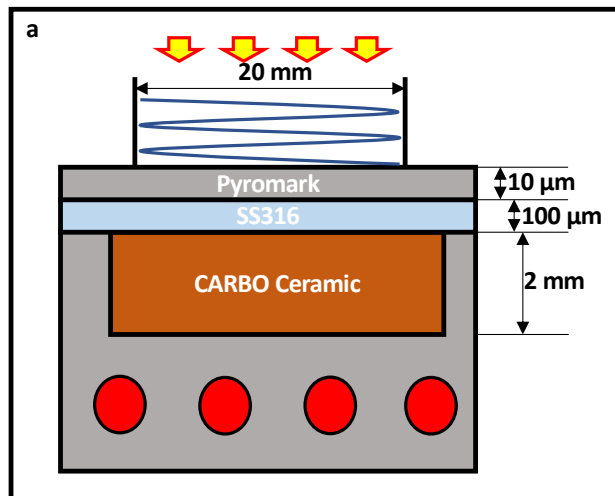
$$\theta_s = q_s \left(\frac{1}{e_m \sqrt{\omega}} + R_{sc} \right)$$



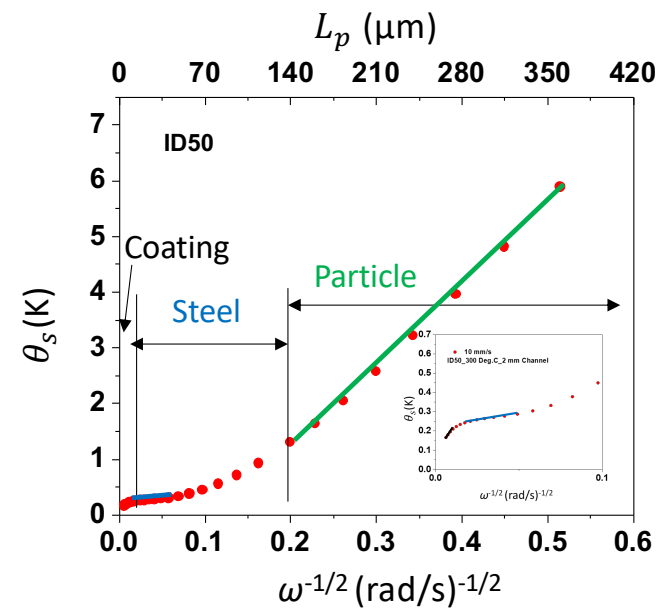
Zeng *et al.* "Measurement of High-temperature Thermophysical Properties of Bulk and Coatings Using Modulated Photothermal Radiometry." *Intl. J Heat Mass Transf.* 170 (2021): 120989.

MPR Measurements on Stationary Particles

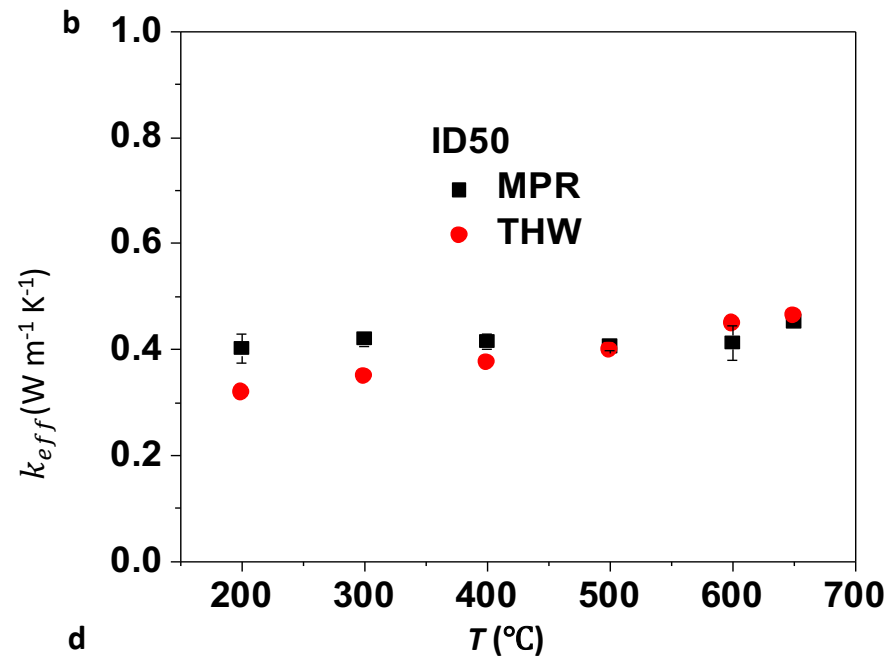
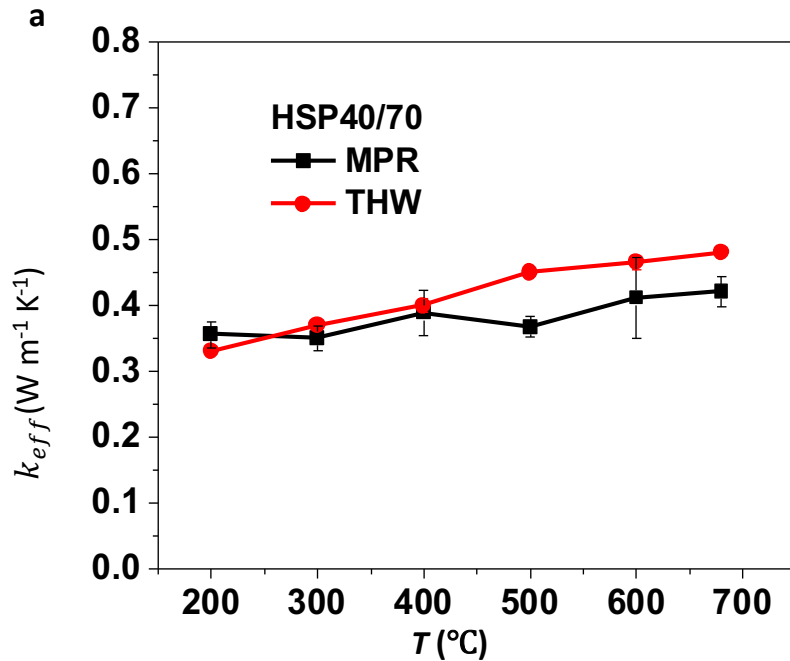
Configuration for *stationary* particle measurement



Typical thermal response of a three-layer system (coating-steel-particle)



MPR Measurements on Stationary Particles

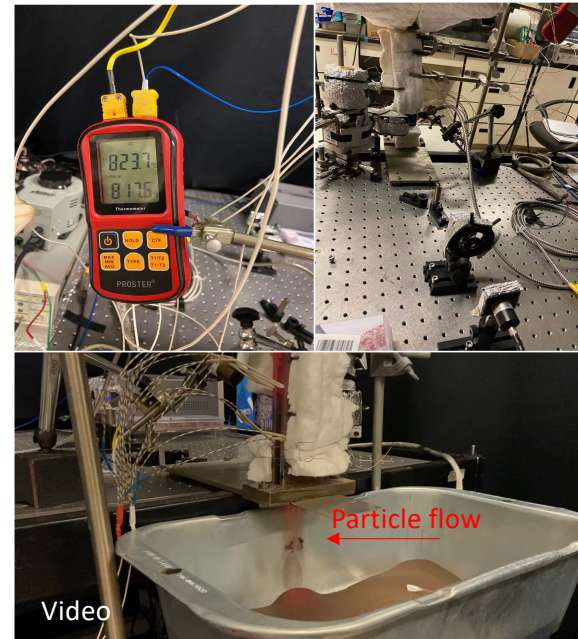
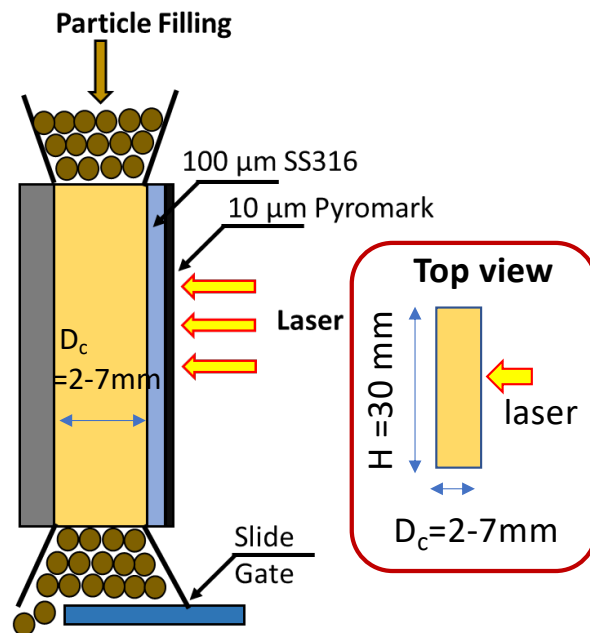


Effective thermal conductivity (k_{eff}) of CARBO **HSP 40/70** (dia. $\sim 430 \mu m$) and **ID50** ($\sim 275 \mu m$) agree well with that measured by the transient hot-wire (THW) method ^[1]

[1] Chung et al., "Measurement and Analysis of Thermal Conductivity of Ceramic Particle Beds for Solar Thermal Energy Storage", *Sol. Energy Mater Sol. Cells.* 230, 111271 (2021),

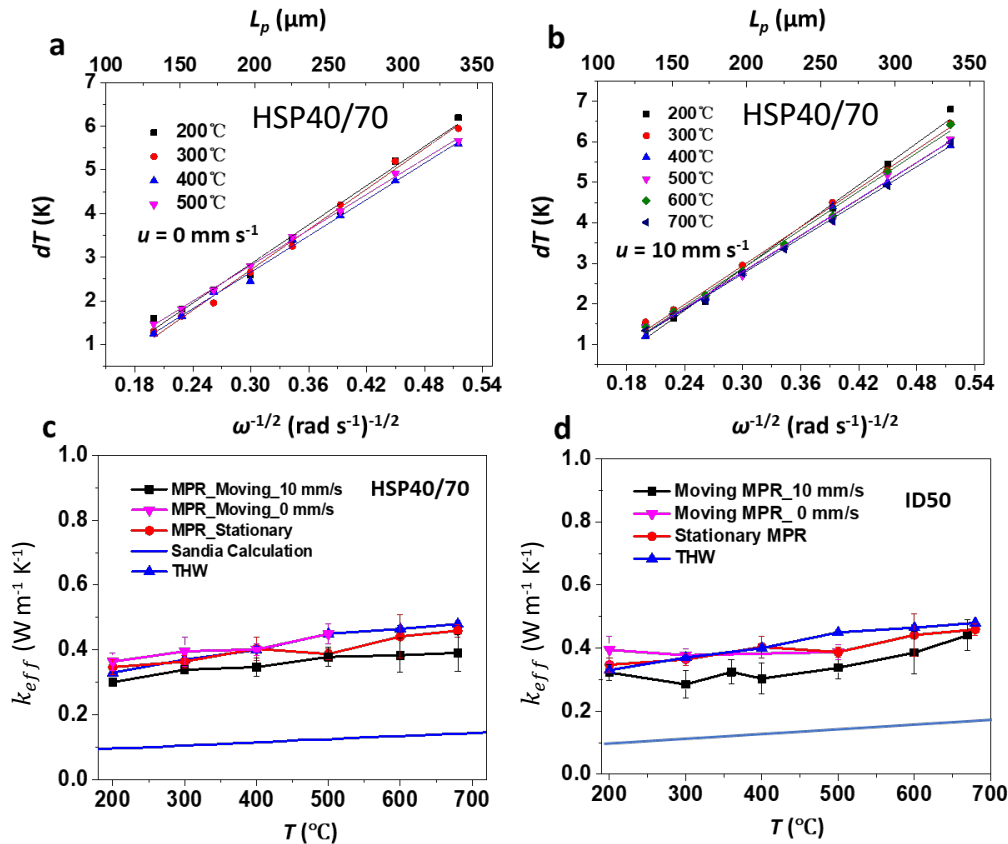
MPR Measurements on Moving Particles

MPR setup for high-temperature Moving Particle Measurement



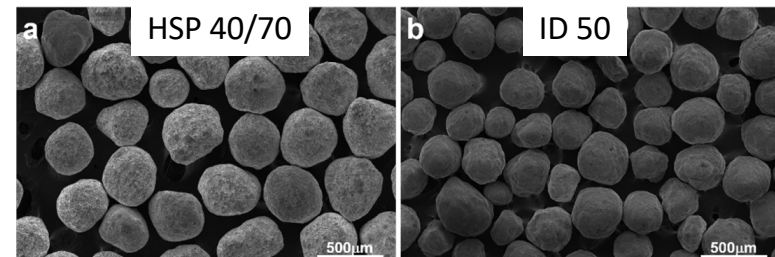
- Multiple High-temperature heaters are used to achieve the target temperature ($\sim 700^\circ\text{C}$).
- High-capacity reservoir (> 30 Gallon) is used for high flowing velocity tests ($\sim 30\text{ mm s}^{-1}$ for 4 mm channel and 60 mm s^{-1} for 2 mm channel).

k_{eff} of Moving Carbo Particles



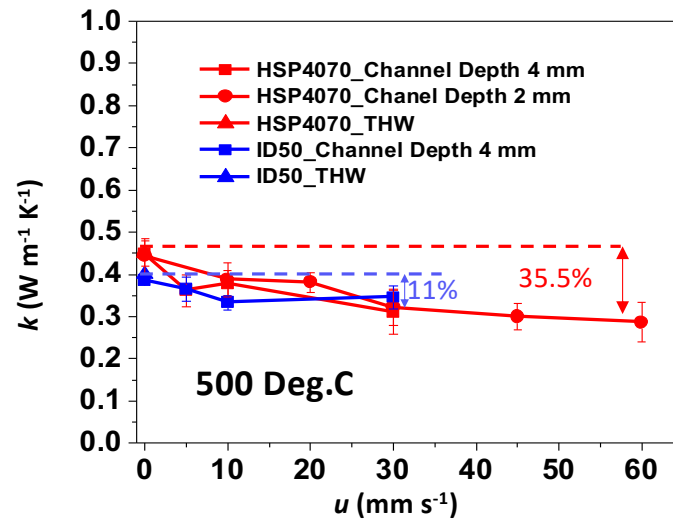
Thermal response of HSP 40/70 in the *moving* particle at
(a) 0 mm s^{-1} and
(b) 10 mm s^{-1} ;
 k_{eff} of moving particles for
(c) HSP4070 and
(d) ID50 (channel depth: 4 mm)

- k_{eff} at 0 mm s^{-1} is close to that measured by THW and stationary MPR setup
- k_{eff} at 10 mm s^{-1} is **15-20% lower**



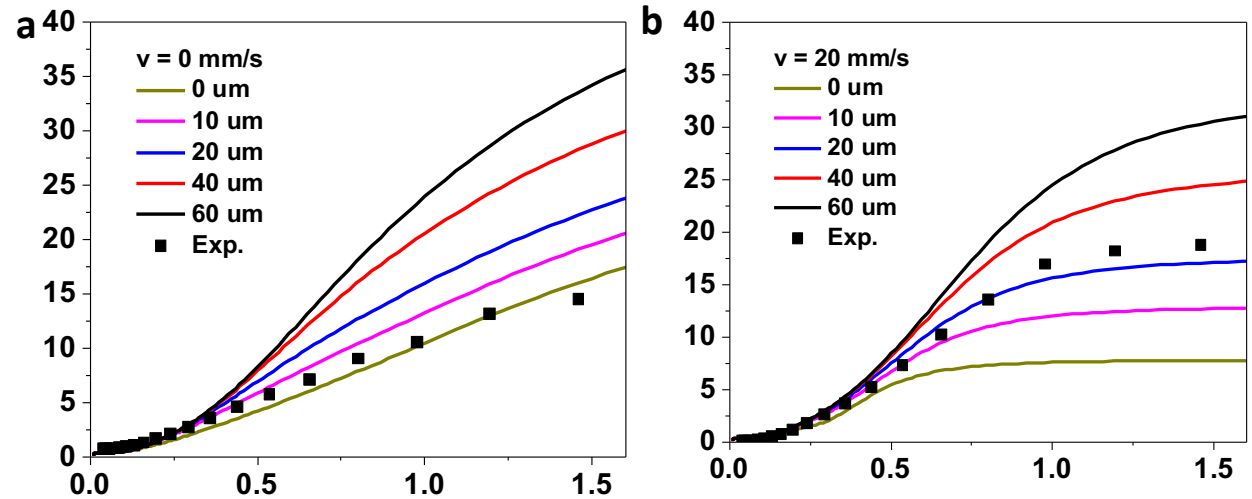
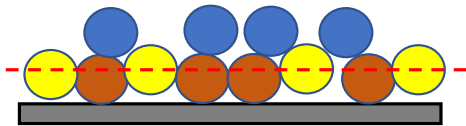
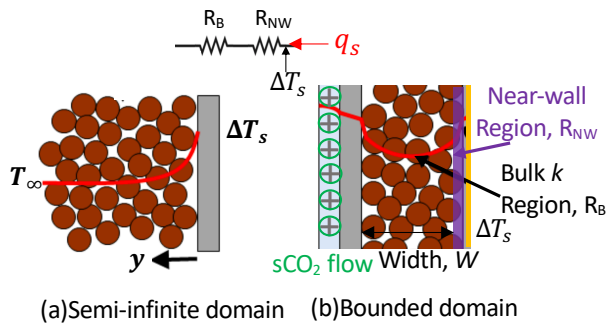
k_{eff} vs. velocity of Moving Carbo Particles

Comparison between ID 50 ($\sim 275 \mu\text{m}$) and HSP 40/70 ($\sim 430 \mu\text{m}$)



- Thermal conductivity decreases more for the HSP 40/70, possibly due to the large particle size.
- The effect of channel size is not significant in the tested velocity range for HSP 40/70.

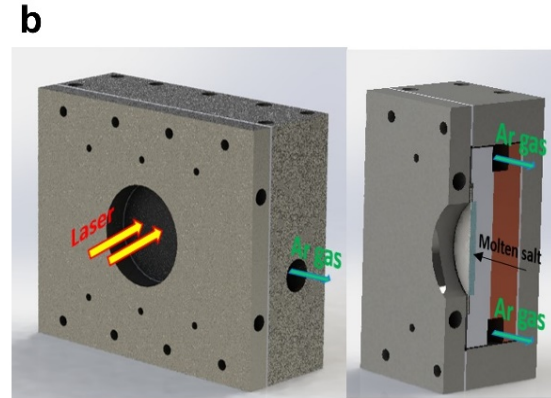
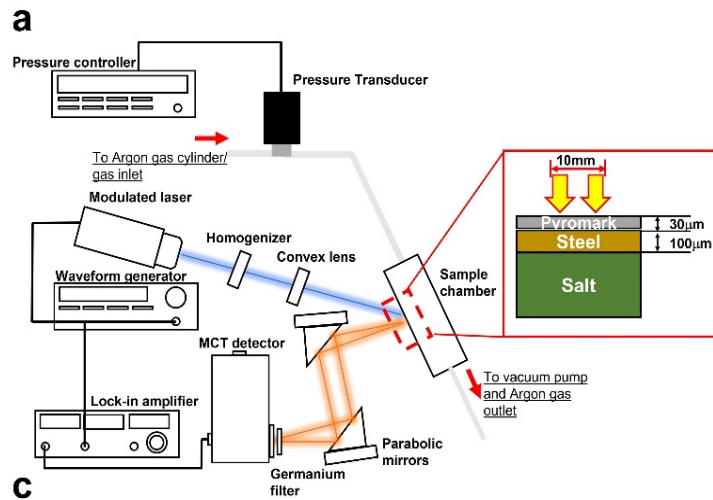
Near-wall and Bulk Resistance of Moving Particles



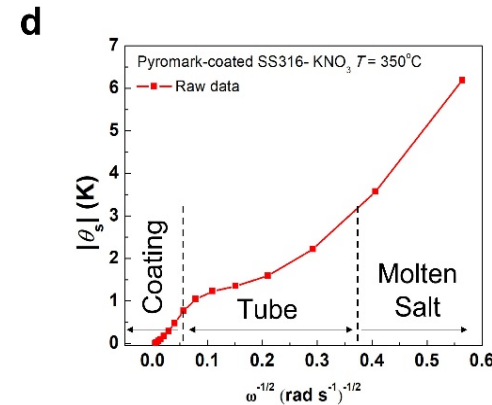
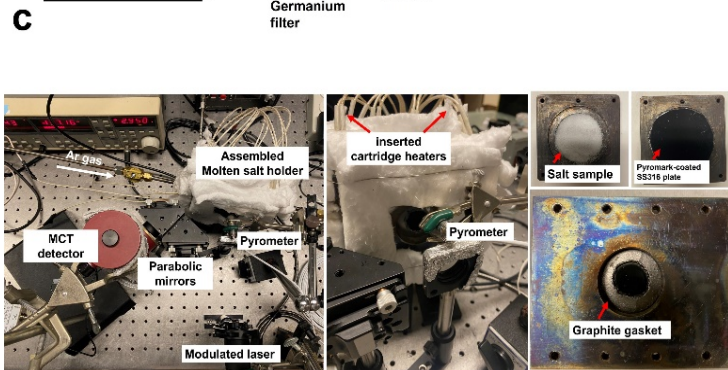
Fitting of Moving ID-50 with different airgap thickness at
(a) 0 mm s^{-1} and (b) 20 mm s^{-1} flowing velocity

- The finite-element model (COMSOL model, solid lines) captures the trend of the thermal response well
- To fit the experimental results, both the velocity and the airgap thickness effects should be included. Fitting results show $D_{air} = 20\text{-}40 \mu\text{m}$ for ID50 ($D_p \sim 275 \mu\text{m}$) at 20 mm s^{-1} .
- More measurements and analysis are being conducted to quantify both the near-wall air-gap resistance and bulk thermal conductivity of particles under various operation conditions.

MPR Measurement of Stagnant Fluids



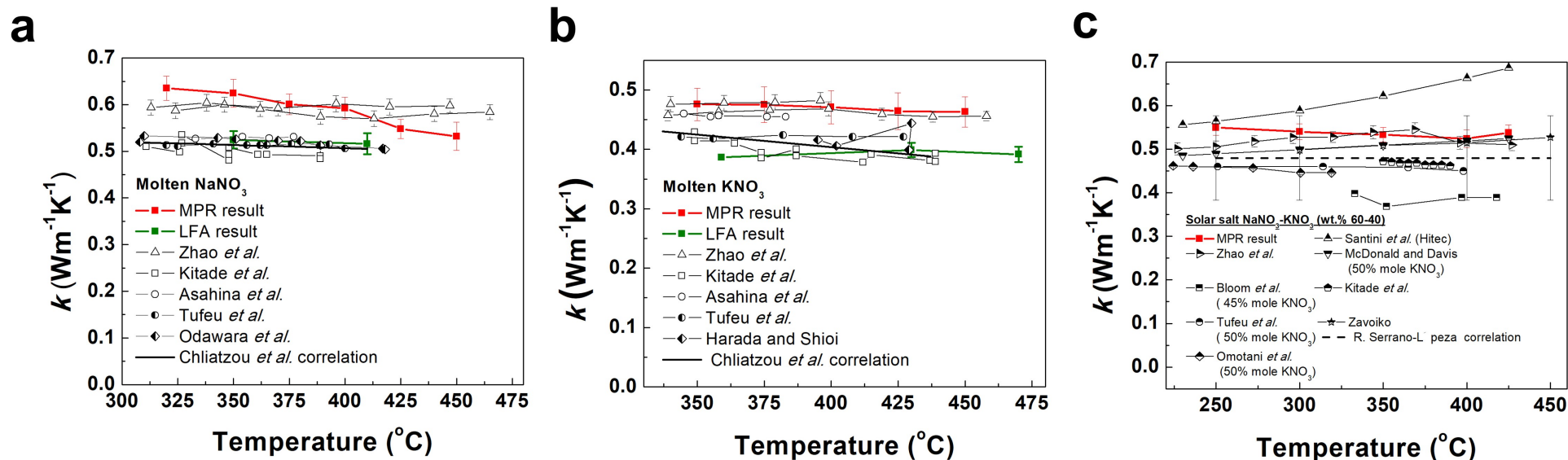
$$\theta_s = q_s \left(\frac{1}{e_m \sqrt{\omega}} + R_{sc} \right)$$



- (a) Schematic of the MPR setup;
- (b) Schematic of the MPR molten salt holder
- (c) Photographs of the MPR setup;
- (d) Plot of $|\theta_s|$ versus $\omega^{-1/2}$ of a typical MPR measurement of stagnant molten salts

MPR Measurement of Stagnant Fluid

Molten Nitrate Salts

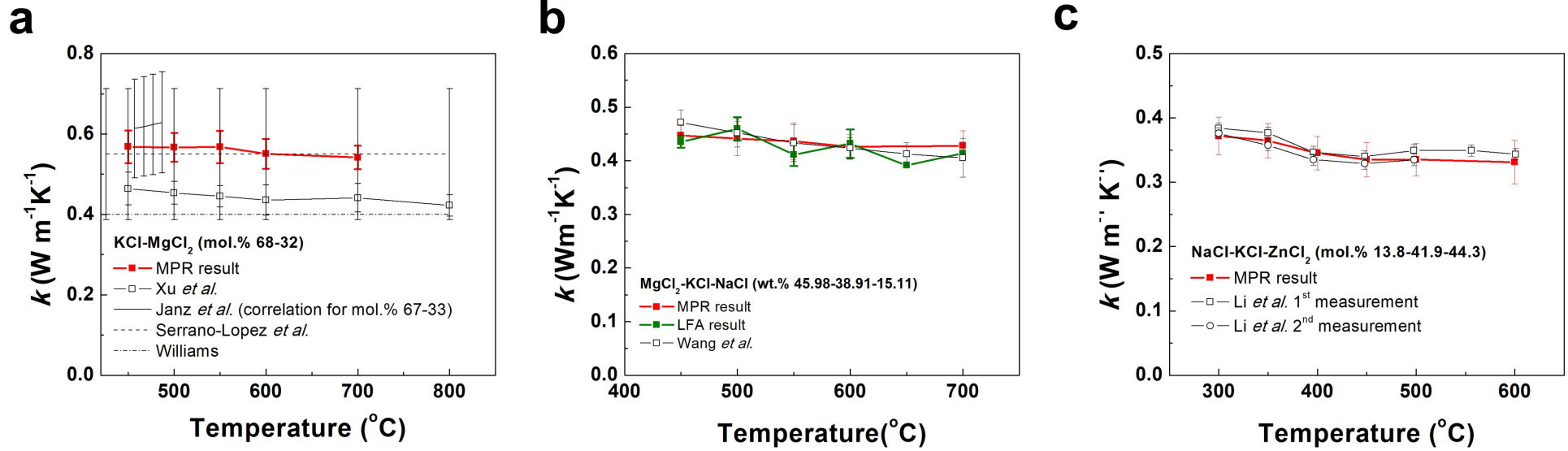


The measured k of the molten NaNO₃, KNO₃ and solar salt mixture, NaNO₃-KNO₃ (wt.% 60-40) are higher than the LFA results and most of the literature values but agree well with the results from a frequency-domain THW method by Zhao *et al.* (UCSD).

- [1] Zhao, Andrew Z., Matthew C. Wingert, and Javier E. Garay. "Frequency-Domain Hot-Wire Measurements of Molten Nitrate Salt Thermal Conductivity." *Journal of Chemical & Engineering Data* (2020).
- [2] Chliatzou, Ch D., et al. "Reference correlations for the thermal conductivity of 13 inorganic molten salts." *Journal of physical and chemical reference data* 47.3 (2018): 033104.
- [3] Tufeu, R., et al. "Experimental determination of the thermal conductivity of molten pure salts and salt mixtures." *International journal of thermophysics* 6.4 (1985): 315-330.
- [4] Odawara, Osamu, Isao Okada, and Kazutaka Kawamura. "Measurement of the thermal diffusivity of HTS (a mixture of molten sodium nitrate-potassium nitrate-sodium nitrite; 7-44-49 mole%) by optical interferometry." *Journal of Chemical and Engineering Data* 22.2 (1977): 222-225.
- [5] Bloom, Ho, A. Doroszowski, and S. B. Tricklebank. "Molten salt mixtures. IX. The thermal conductivities of molten nitrate systems." *Australian Journal of Chemistry* 18.8 (1965): 1171-1176.
- [6] Tufeu, R., et al. "Experimental determination of the thermal conductivity of molten pure salts and salt mixtures." *International journal of thermophysics* 6.4 (1985): 315-330.
- [7] Omotani, T., Yuji Nagasaka, and A. Nagashima. "Measurement of the thermal conductivity of KNO₃-NaNO₃ mixtures using a transient hot-wire method with a liquid metal in a capillary probe." *International Journal of Thermophysics* 3.1 (1982): 17-26.
- [8] Santini, R., et al. "Measurement of thermal conductivity of molten salts in the range 100-500/sup 0/C." *Int. J. Heat Mass Transfer;(United Kingdom)* 27.4 (1984).
- [9] McDonald, John, and Howard Ted Davis. "Thermal conductivity of binary mixtures of alkali nitrates." *The Journal of Physical Chemistry* 74.4 (1970): 725-730.
- [10] Pflieger, Nicole, et al. "Thermal energy storage—overview and specific insight into nitrate salts for sensible and latent heat storage." *Beilstein journal of nanotechnology* 6.1 (2015): 1487-1497.
- [11] Serrano-López, Roberto, Jordi Fradera, and Santiago Cuesta-López. "Molten salts database for energy applications." *Chemical Engineering and Processing: Process Intensification* 73 (2013): 87-102.

MPR Measurement of Stagnant Fluid

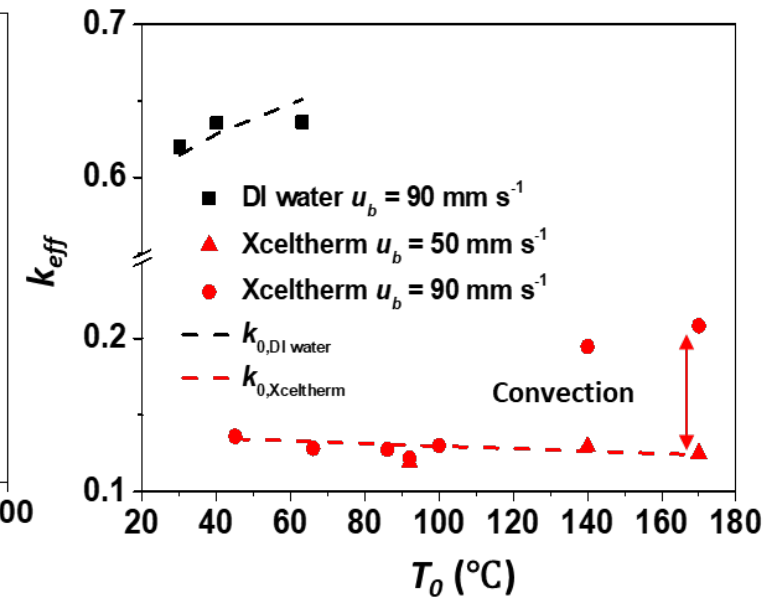
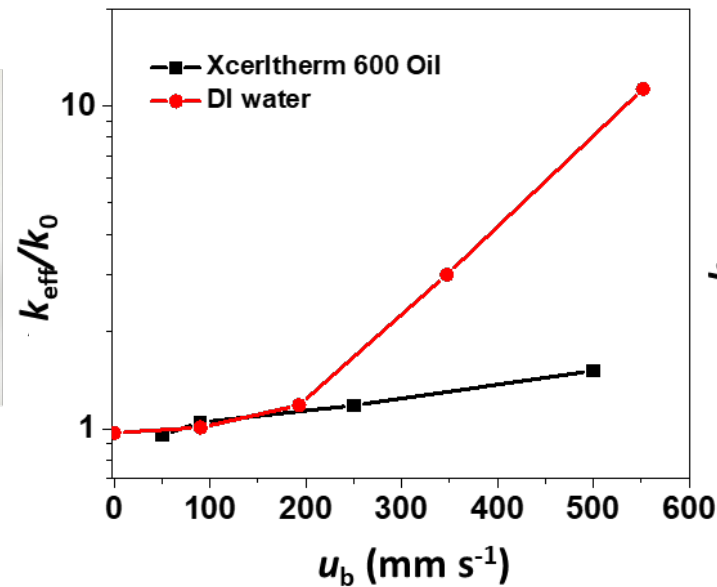
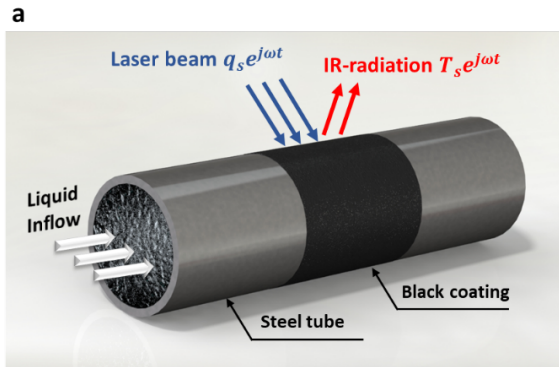
Molten Chloride Salts



The measured k of the molten KCl-MgCl₂, MgCl₂-KCl-NaCl, and NaCl₂-KCl-ZnCl₂ are consistent with the reported values and the reported correlations

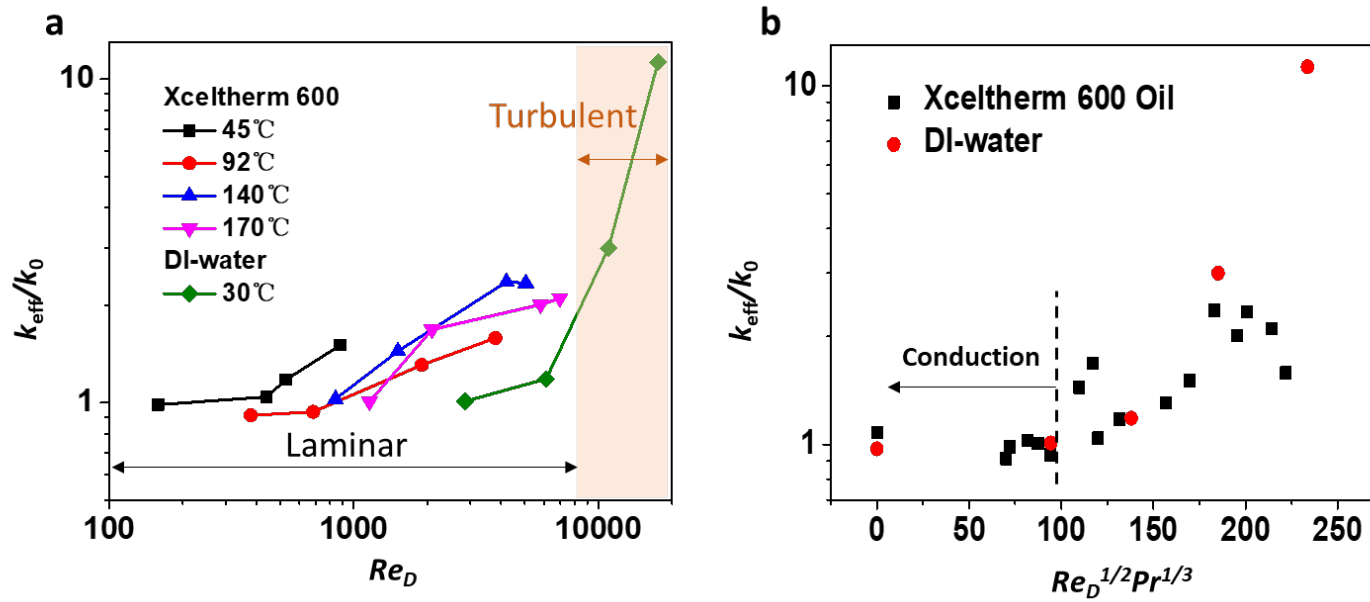
[1] X. Xu, X. Wang, P. Li, Y. Li, Q. Hao, B. Xiao, H. Elsenriechy, D. Gervasio, Experimental test of properties of KCl-MgCl₂ eutectic molten salt for heat transfer and thermal storage fluid in concentrated solar power systems, *Journal of Solar Energy Engineering*, 140(5) (2018).
 [2] M.S. Sohal, M.A. Ebner, P. Sabharwall, P. Sharpe, Engineering database of liquid salt thermophysical and thermochemical properties, Idaho National Laboratory (INL), 2010.
 [3] R. Serrano-López, J. Fradera, S. Cuesta-López, Molten salts database for energy applications, *Chemical Engineering and Processing: Process Intensification*, 73 (2013) 87-102.
 [4] D. Williams, Assessment of candidate molten salt coolants for the NGNP/NHI heat-transfer loop, Oak Ridge National Lab.(ORNL), Oak Ridge, TN (United States), 2006.
 [5] X. Wang, J.D. Rincon, P. Li, Y. Zhao, J. Vidal, Thermophysical Properties Experimentally Tested for NaCl-KCl-MgCl₂ Eutectic Molten Salt as a Next-Generation High-Temperature Heat Transfer Fluids in Concentrated Solar Power Systems, *Journal of Solar Energy Engineering*, 143(4) (2021) 041005.
 [6] P. Li, E. Molina, K. Wang, X. Xu, G. Dehghani, A. Kohli, Q. Hao, M.H. Kassaei, S.M. Jeter, A.S. Teja, Thermal and transport properties of NaCl-KCl-ZnCl₂ eutectic salts for new generation high-temperature heat-transfer fluids, *Journal of Solar Energy Engineering*, 138(5) (2016).

MPR Measurement of Flowing Fluid



The intrinsic k_0 of the flowing fluids can be obtained at a **low flow velocity**, where the forced convection effect is small.

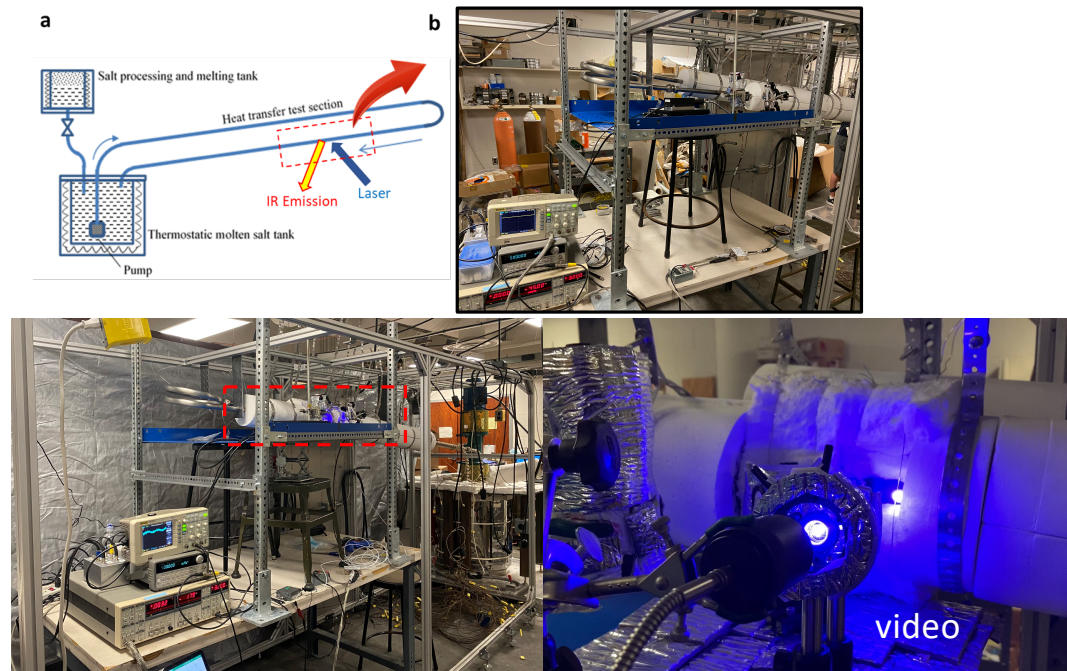
Critical Conditions to Obtain k_o of Flowing Fluids



To obtain the intrinsic thermal conductivity (k_o) of flowing fluid, $Re_D^{1/2} Pr^{1/3}$ should be less than ~ 100 , which agrees with the analysis (thermal penetration depth shorter than both the momentum and thermal boundary layers).

MPR on Flowing Molten Salt (in progress)

We have integrated the MPR section with the thin wall-thickness (100 μm) and set up the MPR tools for the molten salt loop at U. Arizona . The MPR system is set up and the loop is being prepared for flowing molten salt test soon



(a) Schematics of molten salt loop with MPR setup;
(b), (c) and (d) Photo and video of molten salt loop in test.

Summary of the MPR Measurement

For Particles:

- Measured k_{eff} of stationary particles, which agrees well with that by THW.
- Measured k_{eff} of moving particles with temperature up to 700°C and velocity up to 60 mm s⁻¹. k_{eff} decreases by 10-40% depending on the particle morphology and velocity.
- MPR shows the potential to differentiate the near-wall thermal resistance and bulk thermal conductivity of moving particles in a particle heat exchanger.

For Molten salts:

- Measured k of a variety of nitrate and chloride salts using MPR up to 700°C.
- Systematic measurement and analysis of flowing fluid using MPR and identify the critical condition to obtain intrinsic thermal conductivity of flowing fluid.
- MPR tool integration with molten salt loop in U. Arizona and continuing work on the *In-operando* flowing molten salts measurement.

Discussions on Plans and Challenges for Future *In-operando* Measurements



Plan

For Moving Particles:

- Systematic measurement on CARBO ceramic particles and silica sands with a wide range operation conditions, i.e., particle sizes, channel sizes and velocity at high temperature
- Analysis of near-wall thermal resistance and bulk thermal conductivity of moving particles.

For Molten salts:

- Understand the difference between conventional thermal conductivity measurement method (i.e., steady-state method, transient hot wire, and LFA), compared with the MPR technique.
- Measure the thermal conductivity flowing molten salts and study the corrosion effect.

Challenges

For Moving Particles:

- Thermal noise and 2D heat spreading issue at the low frequency measurement (down to 0.1 Hz for the sensitivity on R_{NR}).
- Mechanical strength of the thin MPR section when integrated with particle heat exchangers.

For Molten salts:

- Corrosion and thermal stress on the thin MPR section in the flowing molten salt loop.