# Hot Pressing of Reinforced Lithium Nickel Manganese Cobalt Oxide (Li-NMC) All-Solid-State Batteries with Sulfide Glass Electrolyte

Project ID: BAT482

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GENERAL MOTORS

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### **Overview**

#### Timeline

Project Start: October 1, 2019
 Project End: June 30, 2023
 Percent Complete: 33%

### Budget

□ Total project funding

- DOE: \$1,000,000
- Industrial cost share: \$250,000

FY2020 funding: \$388k
 FY2021 funding: \$299k

#### **Partners**

□ Project lead: General Motors LLC

### **Barriers**

- ❑ All-solid-state batteries should achieve a useable specific energy @ C/3 of ≥ 350 Wh/kg with similar operational specifications as those of conventional Li-ion batteries.
- All-solid-state batteries (ASSBs) are stuck in a paradigm of low energy density or limited functionality:
   Li-ion configurations typically have energy density < 170 Wh/kg.</li>
  - □ Li metal configurations may require slow charging at elevated temperature and high stack pressure.
- □ The performance of ASSBs with sulfide solid-state electrolyte (SSE) is limited by the fact that they are essentially green tapes with 10 20 % porosity.
  - Porosity provides a conduit for Li deposits in the separator.
  - Porosity limits the energy density of the composite cathode.



### Relevance

#### **Objectives**

- Eliminate porosity in both the separator and the cathode by assembling ASSB using a hot-pressing process and appropriately formulated sulfide glass solid-state electrolytes.
- Demonstrate a hot-pressed cathode with a reversible capacity of > 120 mAh/g (June 2021 Go/No-Go).
  - Eliminate porosity in the composite cathode to increase cell energy density.
  - Demonstrate the feasibility of cathodesupported hot-pressed glass separator.

#### Impact

- Elimination of porosity may allow ASSB to escape the paradigm of low energy or limited functionality to help achieve program goals of:
  - □ ≥ 350 Wh/kg useable specific energy
     @ C/3
  - □ 15 year calendar life
  - 1000 cycle life (C/3 deep discharge with <20% energy fade)</p>
  - $\Box \leq 100/kWh \cos t$



### **Milestones**

Month/Year	Milestone Type	Description of Milestone or Go/No-Go Decision
March 2020 (FY20)	Milestone	Baseline cathode performance metric: Develop a suitable baseline system with a reversible capacity of about 120 mAh/g.
June 2020 (FY20)	Milestone	Establish protective coating on cathode: Select the best coating method and coating chemistry.
March 2021 (FY21)	Milestone	<i>Interfacial characterization:</i> Determine the parameters required to prepare cathode samples via FIB/SEM lift-out and to analyze samples via HRTEM. ( <b>completed</b> )
June 2021 (FY21)	Milestone	Catholyte candidate selected: Performing catholytes will be identified. (completed)
June 2021 (FY21)	Go/No-Go Decision Point	Design capable of meeting performance requirements: Demonstrate a hot pressed cathode with a reversible capacity of > 120 mAh/g. Analysis indicates technical approach capable of achieving performance targets. (Go decision)



## Approach

- 1) Apply GM's hot-pressed, reinforced glass separator technology to ASSB.<sup>1</sup> Cathode support provides pathway towards thinner separator for increased energy density.
- 2) Elucidate the mechanisms that promote NCM-catholyte thermal stability to enable hot pressing of cathode composites.

### Phenomena to assess during hot pressing:

Consolidation of softened glass SSE by viscoplastic flow
 Devitrification of glass SSE
 NCM/SSE interfected reaction

□ NCM/SSE interfacial reaction

□ NCM microcracking (especially important for high Ni-content CAM).

[1] T. Yersak et al., *Int'l J. of Applied Glass Science*, 12, 1, (2021): 124 – 134.
[2] Y. J. Nam et al., *Nano Letters*, 15, (2015): 3317 – 3323.



**Milestone (March FY20)** – *Establish protective coating on cathode:* Select the best coating method and coating chemistry.

- ❑ LiNbO<sub>3</sub> is the dominant cathode active material (CAM) coating material for sulfide SSE-based ASSB, but it may not be the optimal choice to promote thermal stability.
- Three coatings investigated: LiNbO<sub>3</sub>,<sup>3</sup> Li<sub>3</sub>PO<sub>4</sub>,<sup>4</sup> and Li<sub>2</sub>O-ZrO<sub>2</sub> (LZO)<sup>5</sup>
   Coatings characterized by differential scanning calorimetry (DSC),
- Coatings characterized by differential scanning calorimetry (DSC), X-ray photoelectron spectroscopy (XPS), and/or high-resolution transmission electron microscopy (HR-TEM)



6

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[3] N. Ohta et al., *Electrochemistry Communications*, 9, (2007): 1486 – 1490.
[4] Y. -G Lee et al., *Nature Energy*, 5, (2020): 299 – 308.
[5] S. Deng et al., *Energy Storage Mater.*, 27, (2020): 117-123.

**Milestone (June FY20)** – *Baseline cathode performance metric:* Develop a suitable baseline system with a reversible capacity of about 120 mAh/g.

- □ Hot-pressing technology should be evaluated using commercially relevant cathode.
- Three considerations: NCM coating, composite preparation method (HM = hand mixing vs. BM = ball milling), and electrode loading.
- □ Cell description:
  - Cathode: 7:3 (w/w) NCM85105@LiNbO<sub>3</sub>:β-Li<sub>3</sub>PS<sub>4</sub>
  - **Ω** Separator: 3wt.% Kevlar fiber pulp in  $\beta$ -Li<sub>3</sub>PS<sub>4</sub>
  - Anode: Indium



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**Milestone (March FY21)** – *Interfacial characterization:* Determine the parameters required to prepare cathode/catholyte samples via FIB/SEM lift-out and to analyze samples via HRTEM.

HR-TEM samples successfully prepared by focused ion beam (FIB) lift-out technique; however, samples were beam damaged despite cyrogenic temperature (-170 °C) and utilization of beam dose control.



#### Alternative methods to study the buried NCM/SSE interface: Electrochemical evaluation XPS DSC – reaction kinetics Cyclic voltammetry (CV)



**Milestone (June FY21)** – *Catholyte candidate selected:* Best performing catholytes will be identified. **Go/No-Go (June FY21)** – *Design capable of meeting performance requirements:* Demonstrate a hot-pressed cathode with a reversible capacity of > 120 mAh/g.

□ Three catholytes evaluated in trials:

- **D** Baseline:  $β-Li_3PS_4$ **D** Candidate A:  $Li_3PS_4 + \frac{1}{2}Lil$

**Candidate B:**  $Li_3PS_4 + \frac{1}{2}LiBr$  (may also be expressed as "75Li<sub>2</sub>S-25P<sub>2</sub>S<sub>5</sub> + 20 mol% LiBr," or "60Li<sub>2</sub>S-20P<sub>2</sub>S<sub>5</sub>·20LiBr")

Material System	Cold Press density (g/cc)	Hot Press density (g/cc)	True density (g/cc)	Cold Press porosity (%)	Hot Press porosity (%)
7:3 (w/w) NCM85105@LiNbO <sub>3</sub> :β-Li <sub>3</sub> PS <sub>4</sub>	2.36	2.47	3.36	29.7	26.7
7:3 (w/w) NCM85105@LiNbO <sub>3</sub> :Candidate A	2.63	2.95	3.69	28.8	20
7:3 (w/w) NCM85105@LiNbO3:Candidate B	2.63	2.95	3.55	26.1	16.9



- Stability of the buried NCM/SSE interface evaluated using DSC, CV, and XPS.
   Energy of activation for β-Li<sub>3</sub>PS<sub>4</sub>/NCM exothermic reaction calculated to be 259 kJ/mol whereas for Candidate B/NCM exothermic reaction it's closer to 264 kJ/mol. Used Kissinger method.
- □ XPS data show interfacial thermal SSE decomposition is similar to that of interfacial anodic SSE decomposition.<sup>6</sup>



10

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NCM microcracking quantified using digital image analysis.
 High Ni content CAM are susceptible to microcracking<sup>7</sup> under the compressive stress applied during ASSB fabrication (370 MPa). In some cases, hot-pressing exacerbates micro-cracking.

#### PFIB cross section image of NCM85105/β-LI<sub>3</sub>PS<sub>4</sub> composite



#### **Processed image**



White = SSE particle Black = pore Blue = intact NCM particle Red = damage NCM particle

		% of damaged particles	% of intact particles
NCM85105/β-	CP	20.50%	79.50%
Li <sub>3</sub> PS <sub>4</sub>	<mark>HP</mark>	<mark>31.25%</mark>	68.75%
NCM85105/	СР	20.00%	80.00%
Candidate B	HP	<mark>15.40%</mark>	84.60%

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11

Addition of 2 wt.% carbon black mitigates the effects of NCM 85-10-5 microcracking.
 Carbon black does increase the initial side reaction.

**Cold Pressed** 

Hot Pressed (200 °C)

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During hot pressing the glassy catholyte may devitrify. In some cases, ceramic precipitates may have higher ionic conductivity than the mother glass.

The newly reported superionically conductive δ-Li<sub>3</sub>PS<sub>4</sub> phase,<sup>8</sup> precipitated from the LiBr-doped Li<sub>3</sub>PS<sub>4</sub> glass, may compensate for thermal degradation of NCM/SSE interface.



δ-Li<sub>3</sub>PS<sub>4</sub> crystal structure<sup>8</sup>



SSE system	Phase	lonic conductivity (mS/cm)
Li <sub>3</sub> PS <sub>4</sub>	glass	0.25
	$\beta$ -Li $_3$ PS $_4$	0.29
Li <sub>3</sub> PS <sub>4</sub> + ½Lil	glass	1.2
	Li <sub>7</sub> P <sub>2</sub> S <sub>8</sub> I [ref 9]	0.63
Li <sub>3</sub> PS <sub>4</sub> + ½LiBr	glass	0.54
	δ-Li <sub>3</sub> PS <sub>4</sub>	1.77

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13

[8] A. D. Bui et al., ACS Applied Energy Materials, 4, (2021): 1-8.
[9] E. Rangasamy et al., J. of the American Chemical Society, 137, (2015), 1384 – 1387.

Milestone (June FY21) – Catholyte candidate selected: Performing catholytes will be identified. Go/No-Go (June FY21) – Design capable of meeting performance requirements: Demonstrate a hot-pressed cathode with a reversible capacity of > 120 mAh/g.

□Electrochemical performance of NCM622 cells meets performance criteria.

Cathode composite	% damaged particles	% intact particles
HP NCM622/β- Li <sub>3</sub> PS <sub>4</sub>	8.42%	91.58%
HP NCM622/ Candidate B	14.50%	85.50%



14

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□ Thermal stability is predicted by a glass' short range structure and not by presence of halide dopants.

□ Li<sub>3</sub>PS<sub>4</sub>, otherwise known as 75Li<sub>2</sub>S·25P<sub>2</sub>S<sub>5</sub>, is a fully de-networked glass composed predominantly of stable  $PS_4^{3-}$  tetrahedra. □ 70Li<sub>2</sub>S·30P<sub>2</sub>S<sub>5</sub> consolidates most effectively by hot pressing, however, it has poor thermal stability versus NCM CAM since it contains labile  $P_2S_7^{4-}$  and  $P_2S_6^{4-}$  structural units.

Material System	Cold Press density (g/cc)	Hot Press density (g/cc)	True density (g/cc)	Cold Press porosity (%)	Hot Press porosity (%)
NCM622 + g-70Li <sub>2</sub> S·30P <sub>2</sub> S <sub>5</sub>	2.863	3.499	3.483	17.8	0
$NCM622 + g\operatorname{-75Li}_2S\operatorname{-25P}_2S_5$	2.863	3.149	3.455	17.1	8.86



### **Responses to Previous Year Reviewers' Comments**

□ This is the first year that the project has been reviewed.



### **Collaboration and Coordination**

Michigan Technological University – Prof. Erik Herbert
 Agreement in place for MTU to measure mechanical properties of sulfide separator glasses with nano-indentation.

The Iowa State University – Prof. Steven W. Martin
 Material transfer agreement to be executed to evaluate processability of Iowa State's glass electrolyte formulations.

Participation in 2020 Oak Ridge National Lab SSE virtual workshop (5/15/20)
 Led session: "[Solid-state battery] Architecture – Design for Strength"
 Workshop findings reported in recent ACS Applied Energy Materials Paper (see Publications and Presentations)



### **Remaining Challenges and Barriers**

□ Scale up challenge: Translation of technology to a high throughput roll-to-roll process

□ Process specifications (T, t, P) require optimization

□ Reduction of pressure (370 MPa)

□ Reduction of time (10 minutes)

□ Binder, reinforcement, and solvent system design requires explication

□ Tolerance of hot-pressing process to dry room conditions currently unknown

□ Integration of cathode support with separator to be addressed in the following budget period.

Any proposed future work is subject to change based on funding levels.

18



### **Proposed Future Research**

- Design of a supported separator with process specifications matching that of the cathode support developed in this first budget period.
- □ Verifying separator performance (*e.g.*, critical current density, capacity, and cycle stability)
- □ Ensuring that separator performance is maintained at target thickness of < 40 µm. Management of defects by interlayer technology is crucial.</p>

Month/Year	Milestone Type	Description of Milestone or Go/No-Go Decision
Sept. 2021 (FY21)	Milestone	Test cell implemented: Implementation of a 3-electrode test cell. (in progress)
Dec. 2021 (FY22)	Milestone	<i>Multifunctional reinforcement:</i> Decision regarding use of multifunctional reinforcement. ( <b>not started</b> )
March 2022 (FY22)	Milestone	<i>Moisture stability demonstrated:</i> Demonstrate that $H_2S$ generation of target separator glasses and catholytes can be cut by 50% in a -40 °C dewpoint dry room. ( <b>not started</b> )
June 2022 (FY22)	Milestone	Separator glass candidates selected: Selection of separator glass candidates. (not started)
June 2022 (FY22)	Go/No-Go Decision Point	Design capable of meeting performance requirements: Establish a critical current density of 1 mA/cm <sup>2</sup> and a capacity of 4 mAh/g for > 100 cycles for a separator. Analysis indicates technical approach capable of achieving performance targets. ( <b>not started</b> )

Any proposed future work is subject to change based on funding levels.

19

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### Summary

□ Our work verifies the possibility of utilizing cathode support for GM's hot-pressed, reinforced sulfide glass separators.

□ The thermal stability of the buried NCM/SSE interface was studied as a function of catholyte SSE composition:  $(1-y)[xLi_2S \cdot (1-x)P_2S_5] \cdot yLiM$  (M = Br, I).

□ Thermal degradation of the buried NCM/SSE interface is analogous to anodic degradation of the same interface during extended cycling

**L**iBr halide dopant promotes the formation of a new metastable  $\delta Li_3PS_4$  phase with superionic conductivity (1.77 mS/cm at 25 °C). The formation of this new phase compensates for thermal degradation of the buried NCM/SSE interface.

Prevention of CAM microcracking during cell manufacture should be a universal consideration for ASSB.





#### Additional DSC data

• Why did we choose not to use LZO coating? Because it consistently showed lowest exothermic onset temperature.



 While DSC shows higher onset for Candidate A catholyte, it does have a higher peak response.





#### Additional XRD data

β-Li<sub>3</sub>PS<sub>4</sub> (HT 2 hours, 240°C)  Primary ceramic phases present after hot pressing are δ-Li<sub>3</sub>PS<sub>4</sub>, Li<sub>7</sub>P<sub>2</sub>S<sub>8</sub>I, and β-Li<sub>3</sub>PS<sub>4</sub> for LPSB, LPSI, and LPS mother glasses, respectively.



23 GM RESEARCH

#### **DSC reaction kinetics study**



DSC response for LPS/NCM, LPSI/NCM, and LPSB/NCM composites when heated at 5K/min.  Fractional reaction completion as a function of time for cathode composites. Legend same as for (a). Kissinger plot of cathode composites and the energy of reaction computed at 33% of reaction completion. Legend same as for (a).

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24

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**Additional XPS data** 

S2P SPECTRA, ALL OTHERS





**BINDING ENERGY (EV)** 

S2P SPECTRA, HOT PRESSED

