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## High Energy Rechargeable Lithium-Metal Cells: Fabrication and Integration

#### JIE XIAO PACIFIC NORTHWEST NATIONAL LABORATORY JUNE 25TH, 2021

#### 2021 DOE VEHICLE TECHNOLOGIES PROGRAM ANNUAL MERIT REVIEW

#### Project ID #: Bat369

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## Overview



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#### Timeline

- Project start date: December 2016
- Project end date: October 2021
- Percent complete: 60%

#### **Budget**

- Total project funding: \$50M (For the entire Consortium
  - DOE share: 100%
- Funding for FY 2020: \$10M (for Consortium)
- Funding for FY 2021: \$10M (for Consortium)

### **Barriers**

- Low energy: Li metal anode will boost cell energy
- Short battery Life: mitigating side reactions will extend the cycling stability

### Partners

- Battery 500 Pls
- 3 national labs
- 10 universities
- GM, Albemarle, Umicore



## **Relevance/Objectives**



Overall Objectives

- Overcome the fundamental issues in building high-energy rechargeable Li metal batteries
- Demonstration of long-term cycling of 500 Wh/kg Li metal cells

#### Objectives of this period

- Identify the cell-level scientific challenges in high-energy rechargeable Li metal batteries: Li-S and Li/NMC
- Demonstrate 350 Wh/kg Li metal pouch cells for 450 stable cycles
- Demonstrate 400 Wh/kg Li metal pouch cells for 100 stable cycles

#### • Impacts

 Accelerate the development of high-energy rechargeable Li metal batteries for future vehicle electrification



## Milestones: Keystone Project 3 for Cell Fabrication, Testing and Diagnosis



Milestones and Go/No-Go Decisions	Date	Status	
Bench mark Li anode architecture with 50 micron Li anode using protocols for 350 Wh/kg cells and achieve over 200 cycles in coins	12/31/2020	Completed	
Optimizing pressure effect for Li-S	3/31/2021	Completed	
Provide new electrolyte formulation for Li-S	6/30/2021	On track	
<ol> <li>Fabricate and test a pouch cell capable of 350 Wh/kg and 450 cycles</li> <li>Fabricate and test a pouch cell capable of 400 Wh/kg and 100 cycles</li> <li>Demonstrate &gt;450 Wh/kg pouch cell</li> <li>Demonstrate good cycling of Li-S pouch cell 300 Wh/kg</li> </ol>	9/30/2021	On track	





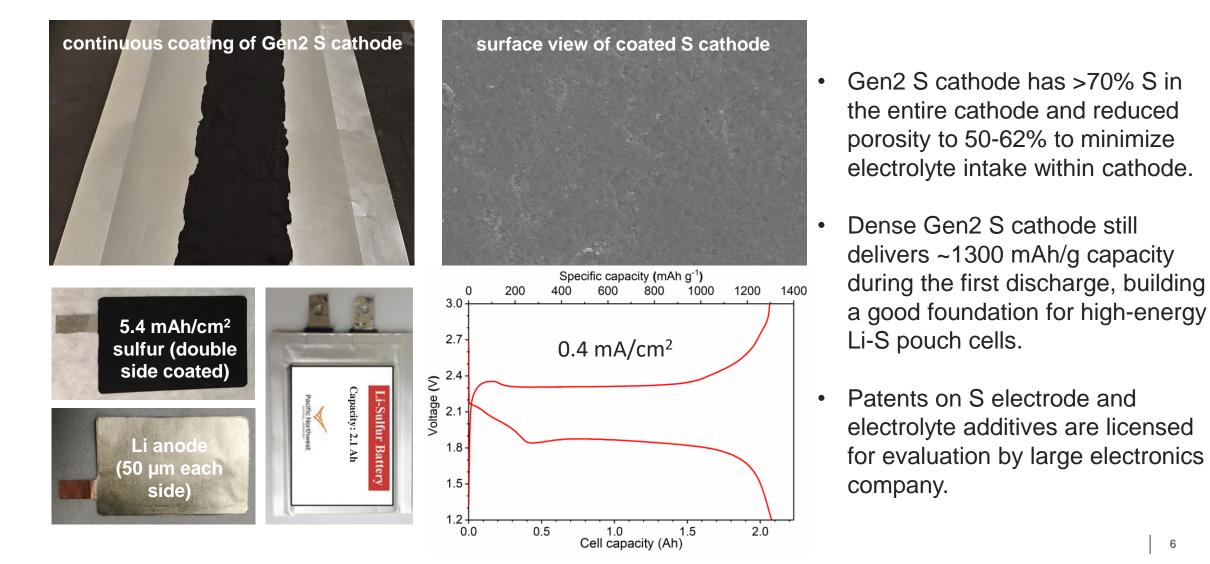


- For Li-S cells: build a dense S/C electrode with minimum parasitic weight and reduced porosity for high-energy Li-S pouch cells
- For Li/NMC cells: balance interfacial reactions to achieve long cycle life
- Develop ML-assisted software to accelerate Li metal pouch cell design and research

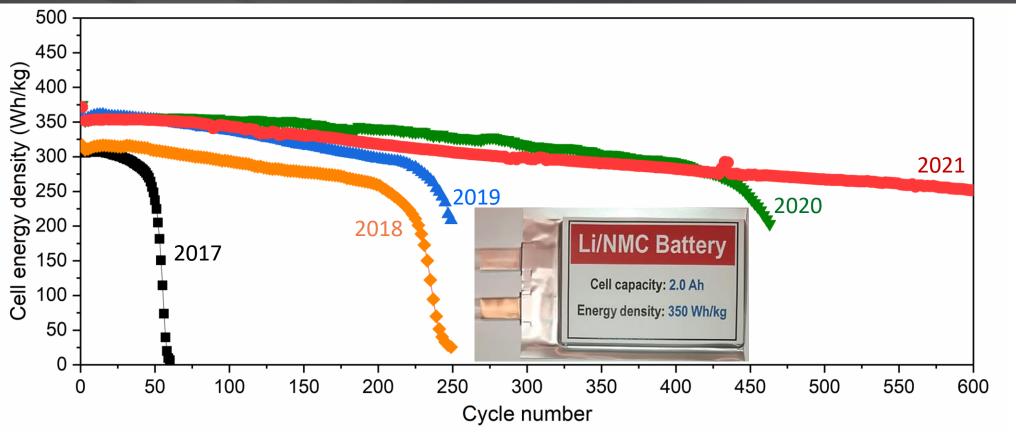


## **Technical Accomplishments:** Gen2 Sulfur Cathode with **Reduced Porosity and High Sulfur Content**





#### **Technical Accomplishments:** 350 Wh/kg Li/NMC622 Pouch Cell Achieved 600 Stable Cycles with 76% Energy Retention

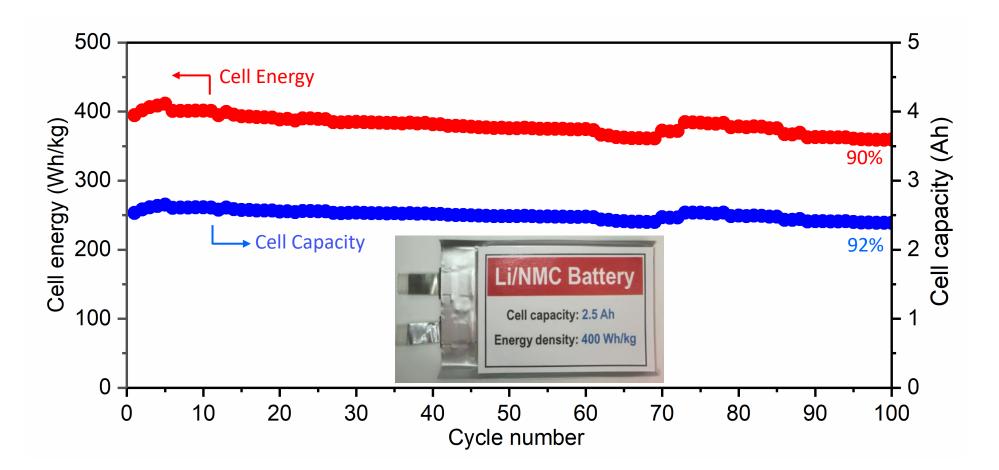


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- In an appropriated designed pouch cell structure, three major factors impact Li metal cycle life:
  - Li thickness
  - electrolyte recipe and amount
  - continuous cell impedance increase due to SEI accumulation: wet SEI vs. dry SEI
- Slowing down the side reactions to retard SEI accumulation: balancing the relative amounts of Li and electrolyte

## Technical Accomplishments: 400 Wh/kg Pouch Cell Based on Li/NMC811





- 400 Wh/kg pouch cell demonstrates stable cycling (still under testing).
- The weights of all cell components are included.



#### **Technical Accomplishments: ML-Assisted Li-Batt App Facilitates Li Metal Battery Research at Industry Relevant** Scales



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LiBatt_Design_App				Export Li Pouch Cell Design Table		
Li-Batt Design Ap	20		BATTERY		Parameters	
Li-batt besign Ap	4		CONSORTIUM	Cathode	Active material	NMC622-4.4
esign Parameters		Design Space			Specific Capacity (mAh/g)	
agnitatanetera		Design opace			Active material %	96%
Run 😯 Reset 🕞 Export		Preference		Electrode width (mm)		
		G			Electrode length (mm)	
Cell		6		The total number of Cathode Electrodes		
Energy density (Wh/kg) 340	Capacity (Ah) 4.34				The total number of Al foil	
Volumetric energy density (Wh/L) 1031	Average discharge voltage (V) 3.76	5			Al foil thickness (um)	
Anode design	Cathode design				Al density (g/cc)	
Li thickness (um) 30.22	Select active material NMC622 - 4.4 V V	E/C (g/Ah)			Active Material Coating weight (mg/cm2)	
Specific capacity (mAh/g) 3860	Specific capacity (mAh/g) 190.67	<b>b</b> <b>y</b> 3			Porosity (%)	33%
CB (N/P) 1.06	Active material % 96 %	С C		Electrolyte	Electrolyte/capacity (g/Ah) (E/C)	
Electrode width (mm) 37.5	Electrode width (mm) 36	2		Anode	Active Material	Li
Electrode length (mm) 55.5	Electrode length (mm) 54				Specific Capacity (mAh/g)	
Cu thickness (um) Mesh: 9 um	Total # of electrodes 20	1			Li thickness (um)	
Electrolyte design	Al foil thickness (um)	2.5	50 Bad		CB (N/P)	
E/C (g/Ah) 4.14	Active material coating wt. (mg/cm2) 30.84	2	40 Color		Electrode width (mm)	
iquid Electrolyte Solid State Electrolyte	Porosity 32.7 %	1	20 jet		Electrode length (mm)	
<u> </u>		CB (N/P) 0.5	10		Cu thickness (um)	
Log			Active material coating weight (mg/cm2)		Cu areal density (mg/cm2)	
prosity changed successfully!			weight (hig/eniz)		Inactive components(Separator, packaging,	
Calculation startedSuccess! It took 10 seconds. Adjust the number of cathode layers and click on the figure to show design parameters.		Li thickness (um) N.A. ▼ Number of cathode layers 20		Tab)		
			2 10 20 30 40 50	Cell	Average discharge voltage (V)	
					Capacity (Ah)	
		4. /4	1		Energy density (Wh/kg)	
nore details here: h	ttps://www.pnnl.gov/n	ews-media/buildi	ng-better-battery-faster		Volumetric energy density (Wh/L)	

• Provide a fast understanding of realistic Li metal pouch cells without having to access pouch cell facility

- Accelerate innovation at relevant scales and help battery prototyping and manufacturing
- Licensed to many institutions and companies

## Responses to Previous Years Reviewers' Comments

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1. "Why the team is focusing on gravimetric energy density when volumetric energy density is more important for vehicles."

- We consider both but only mentioned gravimetric energy density in the presentation. For 350Wh/kg reported in this presentation, the volumetric energy density is 780 Wh/L.
- 2. "..there is mention of the final result of its work but not the approach or the details of how the accomplishment was achieved."
- Per the guidelines from export control office, for cells >350Wh/kg energy, we can only report performances without disclosing any detail.
- 3. How many cycles at what C levels could be achieved at 300 Wh/kg?
- At the same C/3 discharge rate, more cycling is expected in 300 Wh/kg cells because of the additional electrolyte.
- 4. The reviewer's response was that there is no theory put forth as to why more electrolyte results in better cycle life but not higher capacity per gram of S.
- It will help both but the intention is to save more electrolyte to extend Li metal cycling in Li-S cells.
  The reviewer observed that the team also wants to reduce the porosity, which seems to go against a higher electrolyte concentration.
- There is no plan to use concentrated electrolyte for Sulfur cathode with high mass loading.

# Collaboration and Coordination with Other Institutions



- Industry:
  - General Motors
  - Albemarle
  - Umicore
- University:
  - SUNY Binghamton: materials selection
  - Univ. Washington: separator coating
  - UC San Diego: testing on PNNL electrolytes
  - Univ. Pittsburg: supplied S/C composite for electrode coating
  - Penn State Univ.: testing of thick NMC and S electrodes made at PNNL
  - Univ. Houston: testing of PNNL new electrolyte
  - Stanford/SLAC: electrodes and electrolyte testing
  - UT Austin: Supplied high-Ni NMC to PNNL for evaluation
  - Univ. Maryland/Army research Lab: electrolyte development
- National Laboratory
  - Idaho National Lab: independent testing of PNNL-made pouch cell; co-develop safety protocol
  - Brookhaven Nation Lab: characterization of PNNL fabricated electrodes/electrolytes
  - SLAC: new electrolyte characterization



## **Remaining Challenges and Barriers**



- Push the cell energy towards 500 Wh/kg with stable cycling by integrating innovation and engineering
  - Baseline performance will accelerate the innovation
- Balance of high energy and cycle life of Li metal cells
- Dendrite-induced cell shorting
  - C/10 charging is used for now to decouple cell shorting and cell failure caused by Li/electrolyte depletion.



## **Proposed Future Work**



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- Li-S pouch cell design to balance cell energy and cycling
- Move towards 500 Wh/kg Li metal pouch cells
- Improve the utilization of both lithium metal and electrolyte
- Complete the pressure study to further extend the cell cycling



Any proposed future work is subject to change based on funding levels.





- Improved S% and reduced electrode porosity to minimize electrolyte intake within cathode side without sacrificing S utilization.
- Demonstrated 350 Wh/kg and 400 Wh/kg Li pouch cell with stable cycling
- Developed Li metal pouch cell design software to accelerate research innovation and support industry for Li metal cell prototyping and manufacturing







- DOE/EERE/VTO: Battery500
- Key contributors: C. Niu, D. Liu, D. Lv, L.Shi, Joshua Lochala, Cassidy Anderson, B.Wu, W. Xu, J. Zhang, J. Liu
- Battery500 PIs and their teams



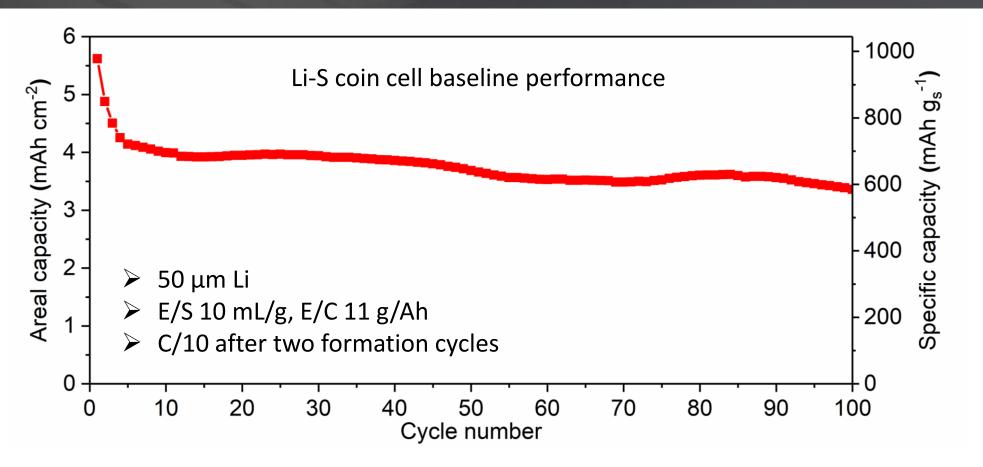


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## **Technical Backup Slides**

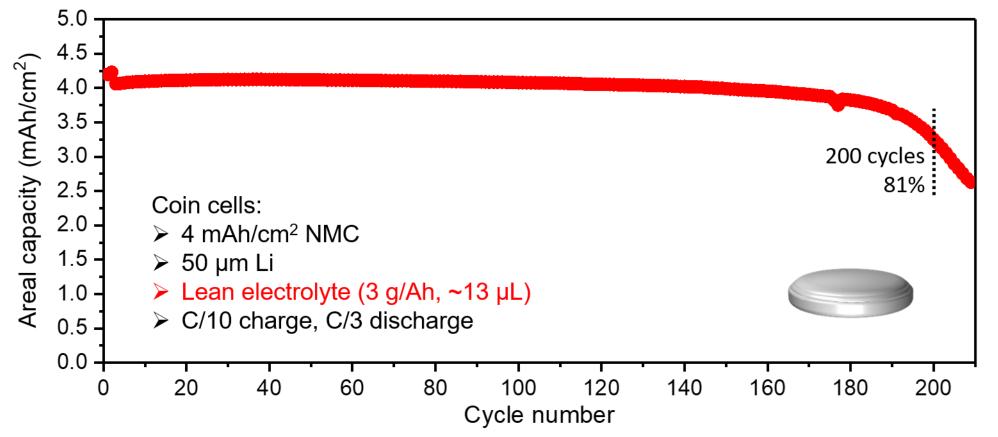


## Technical Accomplishments: Established Li-S Coin Cell Testing Protocol and Baseline Performance for Benchmarking Pacific Northwest National Laboratory Pacific Northwest Proudy Obstated by Battelle Since 1965



- Li-S coin cell protocol has been updated to ensure testing are conducted at similar conditions and results can be reproduced across different teams in the consortium.
- Effective new concepts need to surpass the baseline performance of Li-S coin cells.

## <u>Technical Backup:</u> Li/NMC Coin Cell Baseline Performance

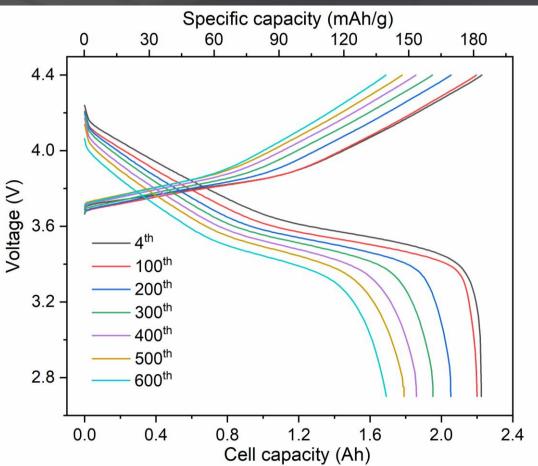


- FY21 Li/NMC coin cell protocol has been updated
- Baseline performances of Li/NMC tested using the protocol has been provided to the consortium for benchmarking.



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#### Technical Backup: Pacific Northwes NATIONAL LABORAT Pouch Cell Polarization Increases after Extensive Proudly Operated by Batterie



Upon extensive cycling, cell polarization gradually increases in Li pouch cells due to the SEI buildup and impedance increase.

