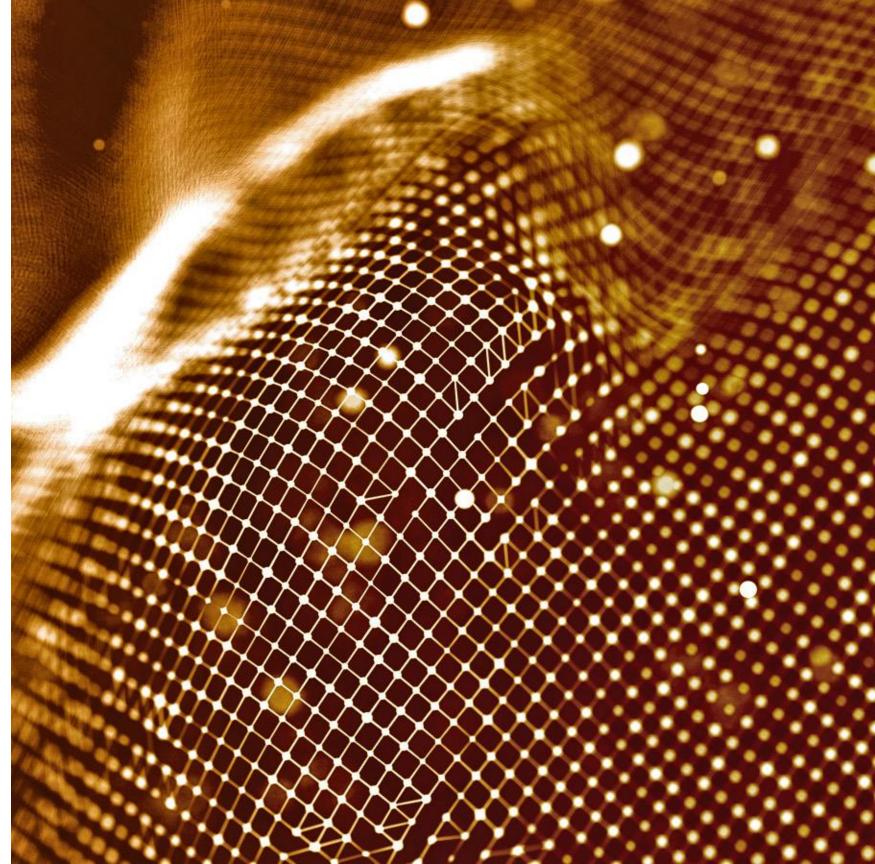


On-line, Real-time Monitoring of Chemical Composition: Process Control and Optimization

May, 15, 2019 Amanda Lines and Samuel A, Bryan



PNNL is operated by Battelle for the U.S. Department of Energy





Contributing team members

PNNL Team:

Sam Bryan Amanda Lines Gregg Lumetta Shirmir Branch Gabe Hall Jarrod Allred

Heather Felmy Adan Schafer Amanda Casella Forrest Heller Susan Adami Brian Riley Susan Asmussen

Industry Partners:

Job Bello/Spectra Solutions

Students/visiting faculty:

Gilbert Nelson Hope Lackey Jen Wilson PoKi Tse



PNNL is operated by Battelle for the U.S. Department of Energy

ORNL Team:

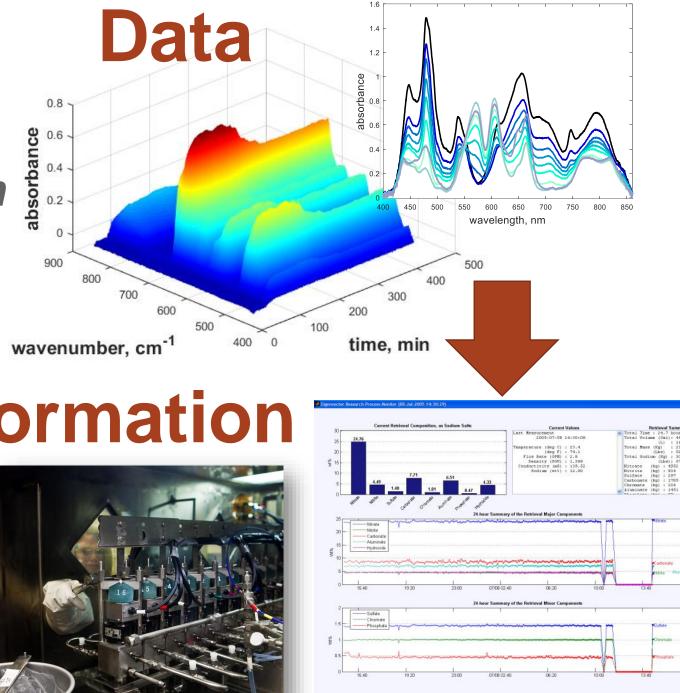
Hunter Andrews Shay Chapel Bill Del Cul **Dianne Ezell** Alex Hackett **David Holcomb** Joanna McFarlane Kristian Myhre



What On-Line Monitoring Provides

Enabling real-time process characterization to supply essential process information

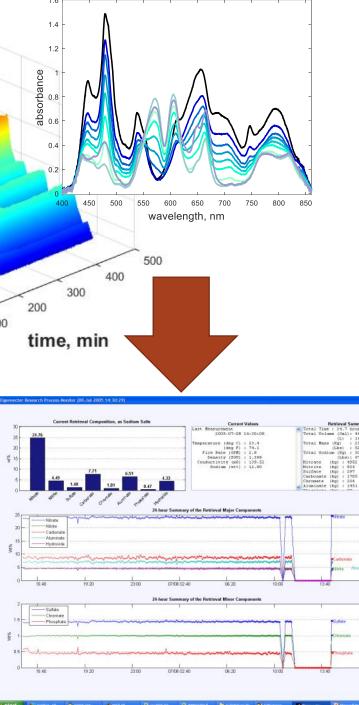
- Design phase
 - Informed and optimized R&D



- Deployment phase
 - Process optimization
 - Process control
 - Real-time characterization

Information





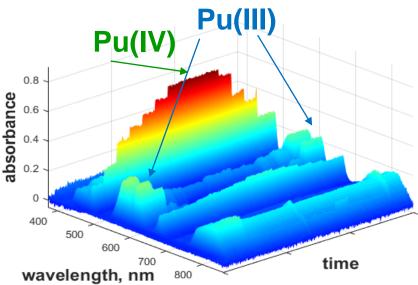
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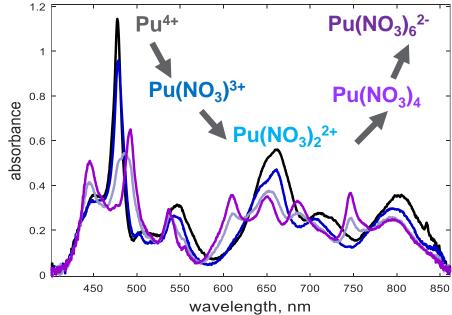


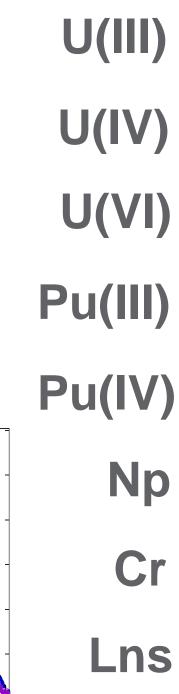
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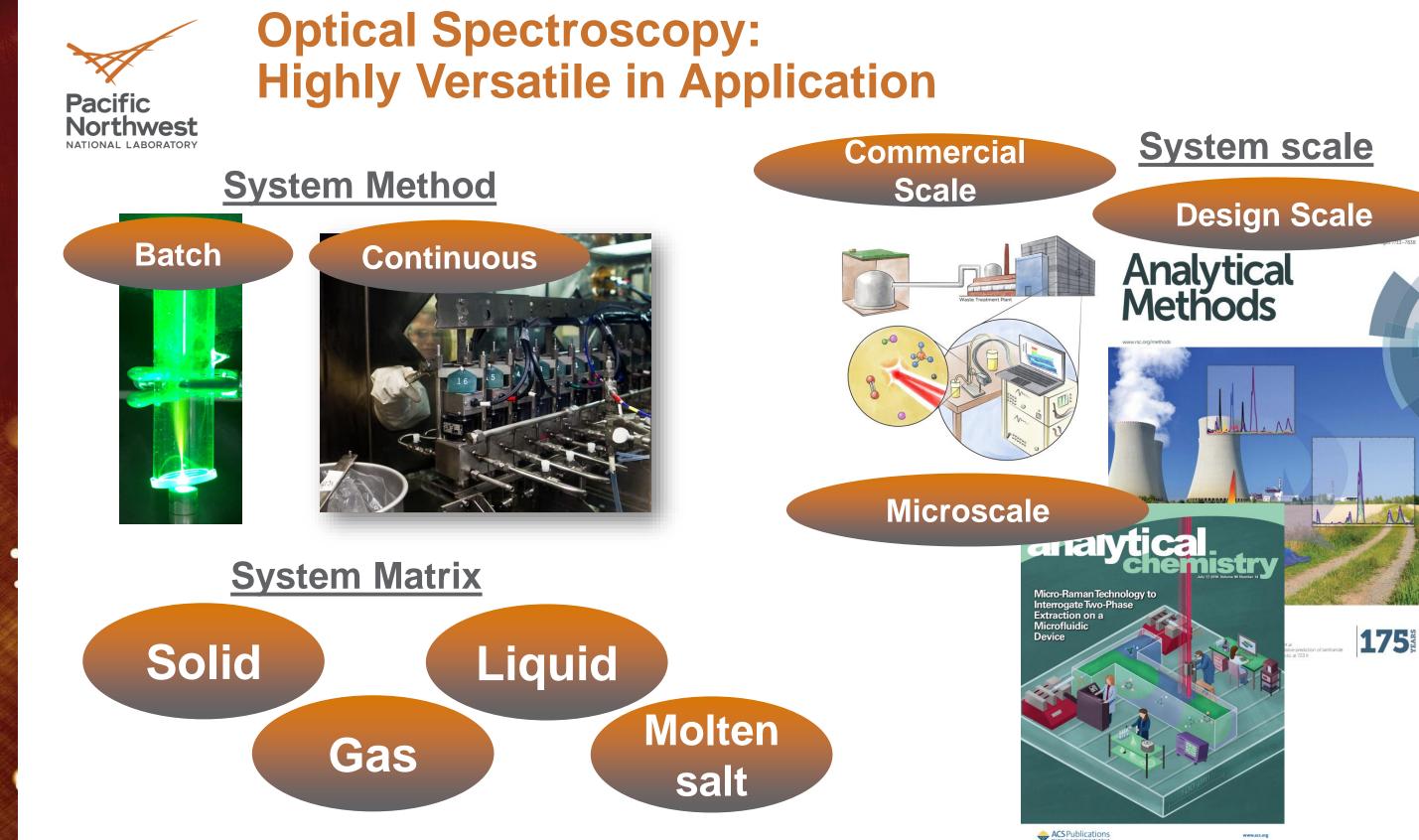
Chemical characterization: Optical spectroscopy

- Provides chemical information
 - Identification and quantification
 - Oxidation State
 - Essential information for control of systems
 - Molecular and elemental species
 - Essential information to understand/control separation efficiency or general system behavior
- Fast
- Robust
- Versatile









www.acs.org



Optical Spectroscopy: Widely Applicable to Analytes

Raman spectroscopy

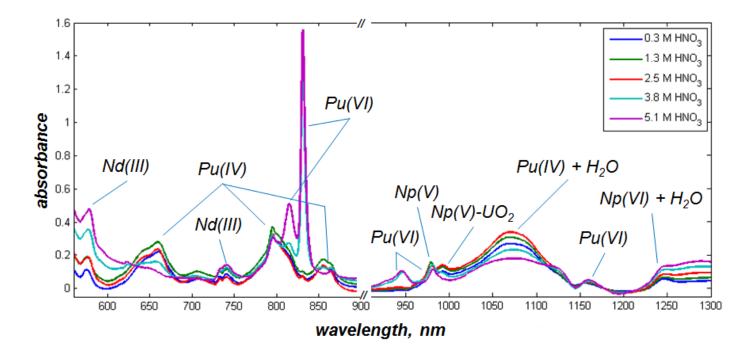
- Actinide oxide ions (UO₂²⁺)
- Organics
 - solvent components and complexants
- Inorganic oxo-anions
 - NO₃⁻, CO₃²⁻, OH⁻, SO₄²⁻
- Water, acid (H⁺), base (OH⁻)
- pH of weak acid buffer systems

Numerous, versatile tools available to capture fingerprints of huge range of fission products/species of interest to the fuel cycle

UV-vis-NIR absorption

Actinides and lanthanides in multiple oxidation states

- Pu (III/IV/VI)
- Np (III/IV/V/VI)
- Various metal-ligand complexes



Several other options

FTIR

- Organic complexants

Light scatter

turbidity

Optical density

Formation of complexes

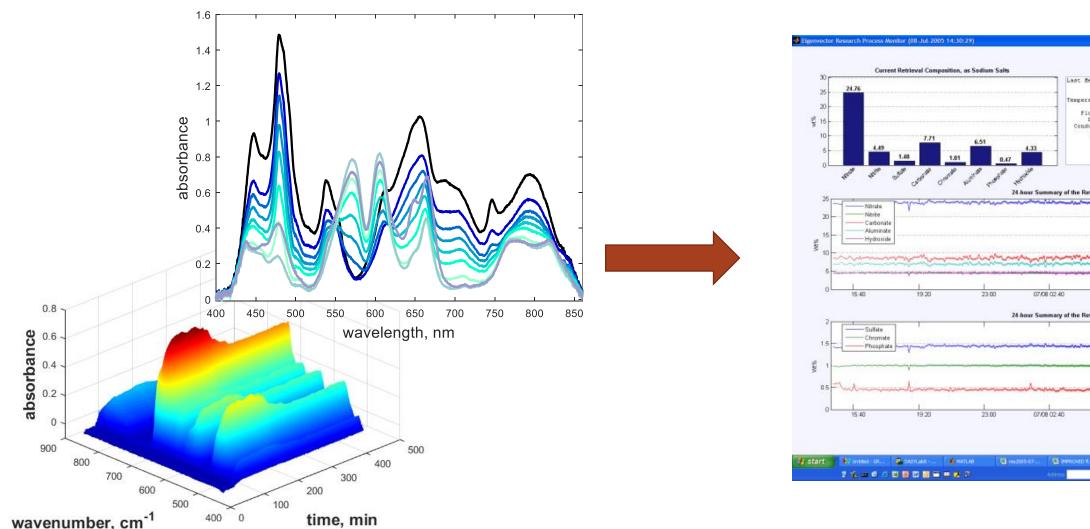


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Primary analytical challenge

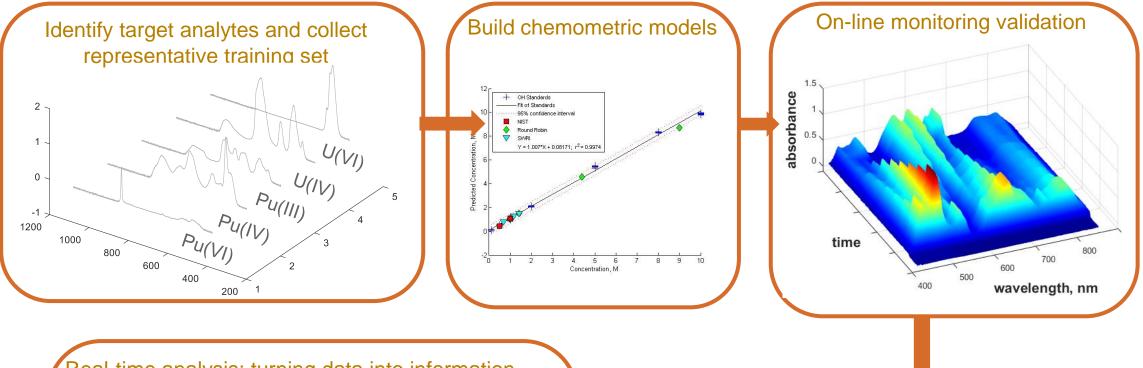
Data

Information



Current Values		Retrieval Summary	
eadurement		e 24.7 hours	~
2005-07-08 14:30:08	Total Volu	me (Gal): 4459 (L) : 16001	
ature (deg C) : 23.4	Total Bans		
(deg F) : 74.1	1000000000	(Lbs) : 52055	
(deg F) : 74.1 ow Rate (OPH) : 2.6	Total Sodi	Luzo (Eg) : 3043	
Density (SGU) : 1.399	Nitrate	(Lbs): 6708	
Sodium (wt%) : 12.90	Nitrite	(Rg) t 4952 (Rg) : 936	
	Sulfate	(kn) : 297	
	Carbonate	(kg) : 1785 (kg) : 206	
	Chromate	(kg) : 206	
	Alusinate	(kg) / 1451	4
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Development of on-line, real-time analysis systems

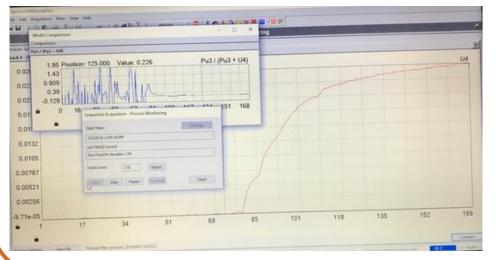


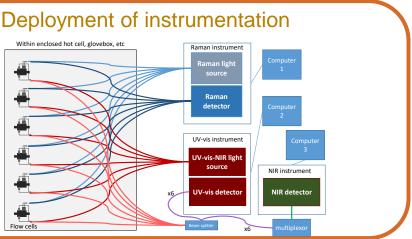


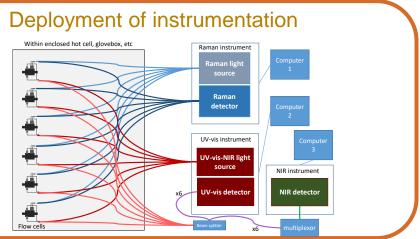
Pacific

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Northwest NATIONAL LABORATORY





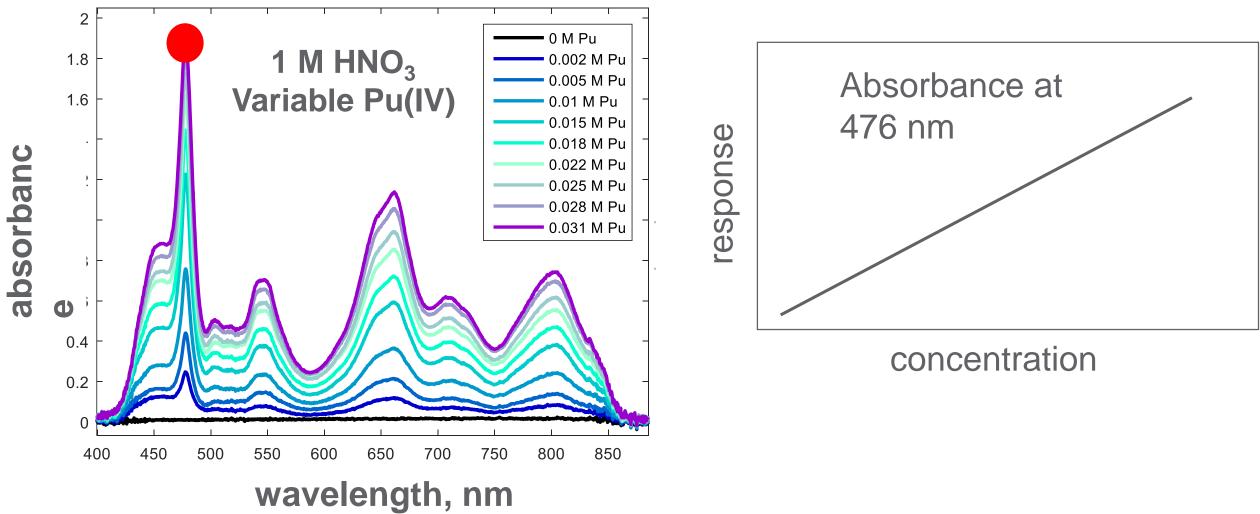




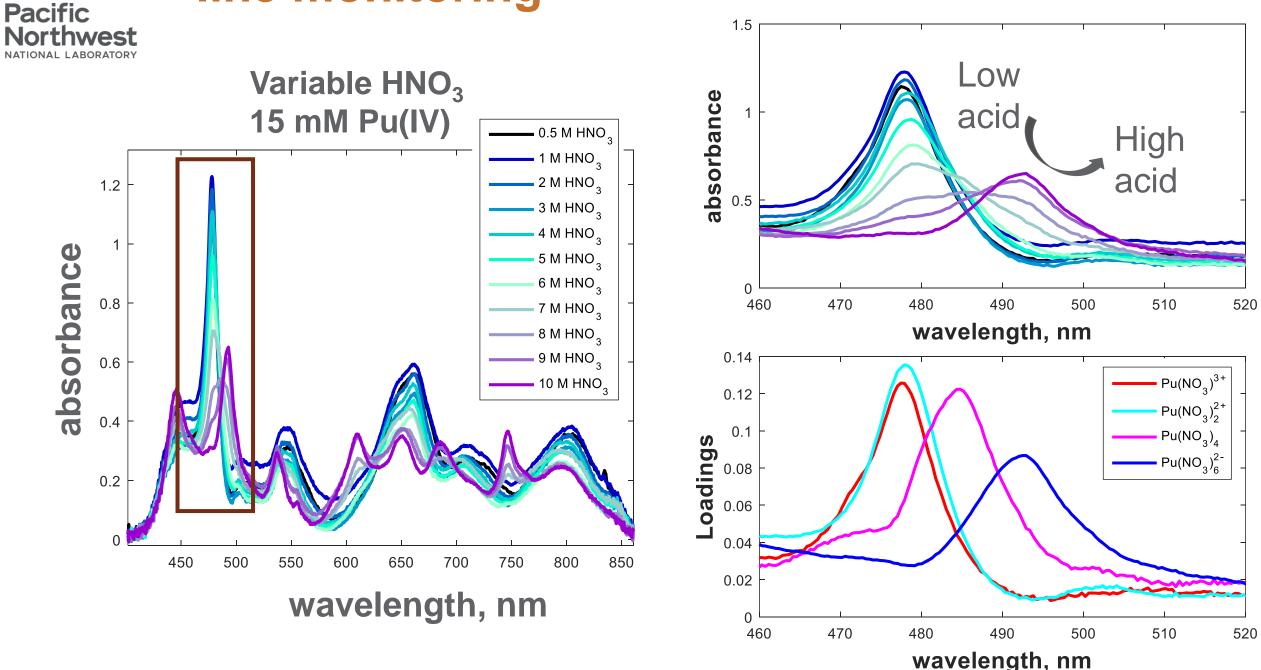


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Developing chemometric models to enable online monitoring

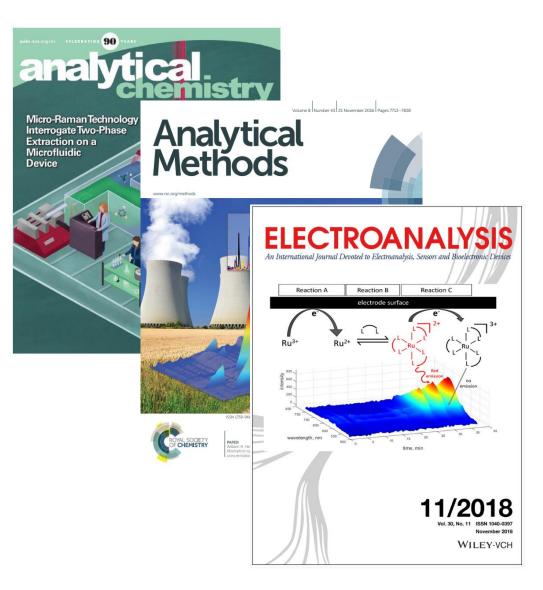


Developing chemometric models to enable online monitoring



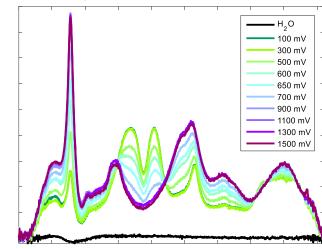
Lines, Adami, Sinkov, Lumetta, Bryan. Multivariate Analysis for Quantification of Plutonium(IV) in Nitric Acid Based on Absorption Spectra. Anal Chem. 2017

Wide Range of Applications Pacific Northwest NATIONAL LABORATOR

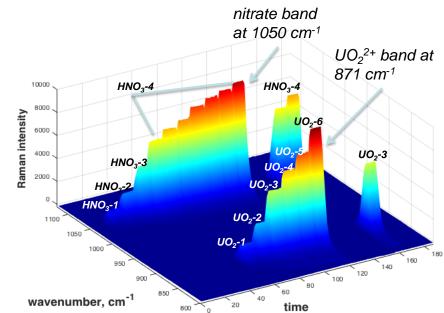


Lines, A.M., S.R. Adami, S.I. Sinkov, G.J. Lumetta, and S.A. Bryan. 2017. "Multivariate analysis for quantification of plutonium (IV) in nitric acid based on absorption spectra." Anal. Chem., 89(14):9354-9359, DOI: 10.1021/acs.analchem.7b02161.

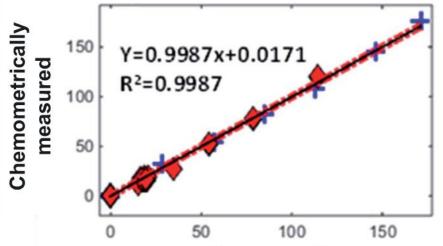
Measurement of Pu in multiple oxidation states



Lines, Adami, Casella, Sinkov, Lumetta, Bryan. Electrochemistry and Spectroelectrochemistry of the Pu(III/IV) and (IV/VI) couples in Nitric Acid Systems. Electroanalysis. 2017



Lines, Bello, Clark, Bryan. Multivariate analysis to quantify species in the presence of direct interferents: micro-Raman analysis of HNO3 in microfluidic environments. Anal Chem. 2018



Schroll, C. A., et al. (2016). "Absorption spectroscopy for the quantitative prediction of lanthania concentrations in the 3LiCI-2CsCl eutectic at 723 K." Analytical Methods 8(43): 7731-7738.

Quantification in presence of direct interferents

Quantification in complex and harsh environments

Known, mM



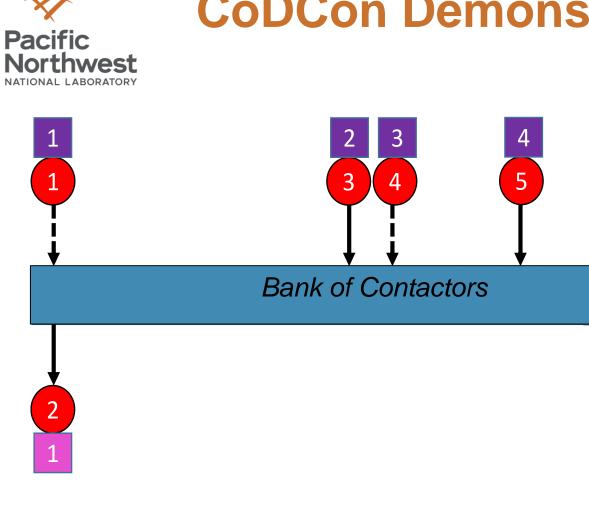
Applications

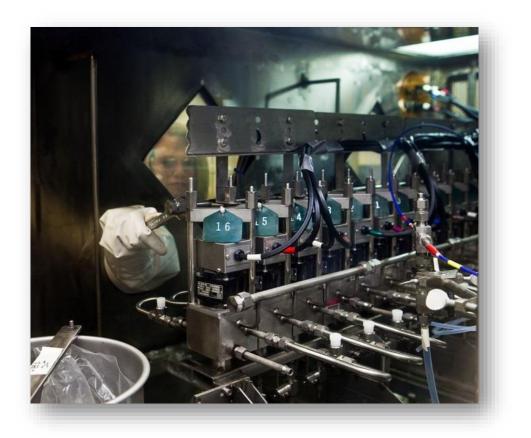
- DOE NE
 - CoDCon (Co-decontamination)
 - Microfluidics, collaboration with CEA, SBIR
 - MSR off-gas treatment
- EM
 - Hanford tank waste processing
 - Off-gas condensate from waste processing
 - In-situ characterization of tank solids
- LDRD
 - Molten salt characterizations
 - Development of fieldable MSR probes

- used nuclear fuel simulant (complex signals)
- aqueous and non-aqueous streams
- Micro and lab scale applications
- **Process control** •
- Complex Hanford tank waste chemical • systems
- Process control and validation
- Lab and commercial scale applications
- Design for highly regulated processes \bullet
- High temperature, high corrosion •
- Laying foundation for future advancements

CoDCon Demonstration

6





- → Aqueous stream
- → Organic stream
- Spect

Spectroscopic measurement: U, Pu, and HNO3

Flowrate/density measurement, Coriolis meter

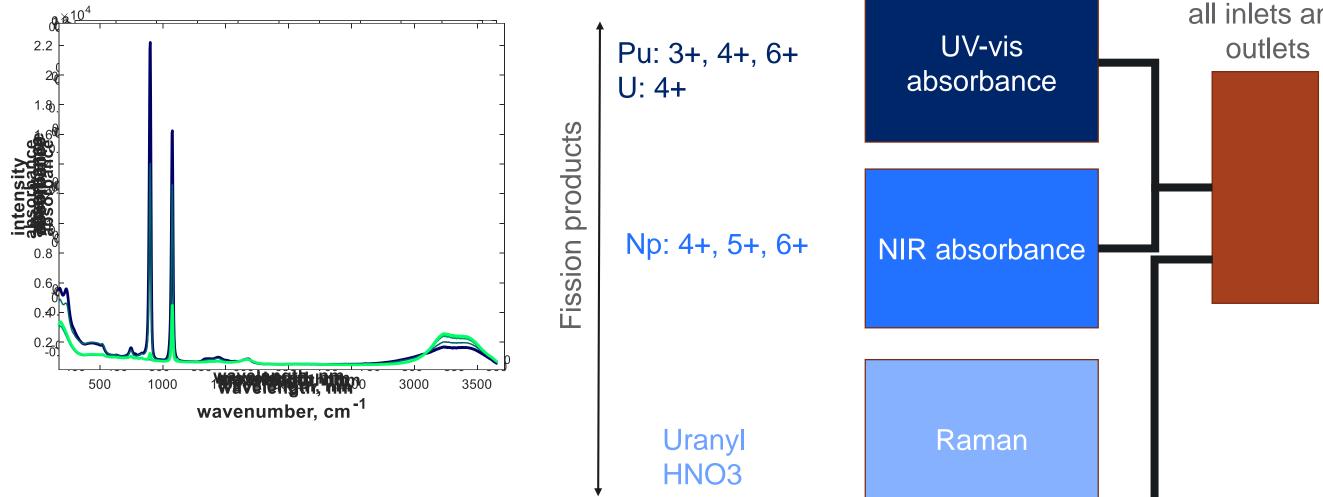
Flowrate/mass measurement, time collection and scales



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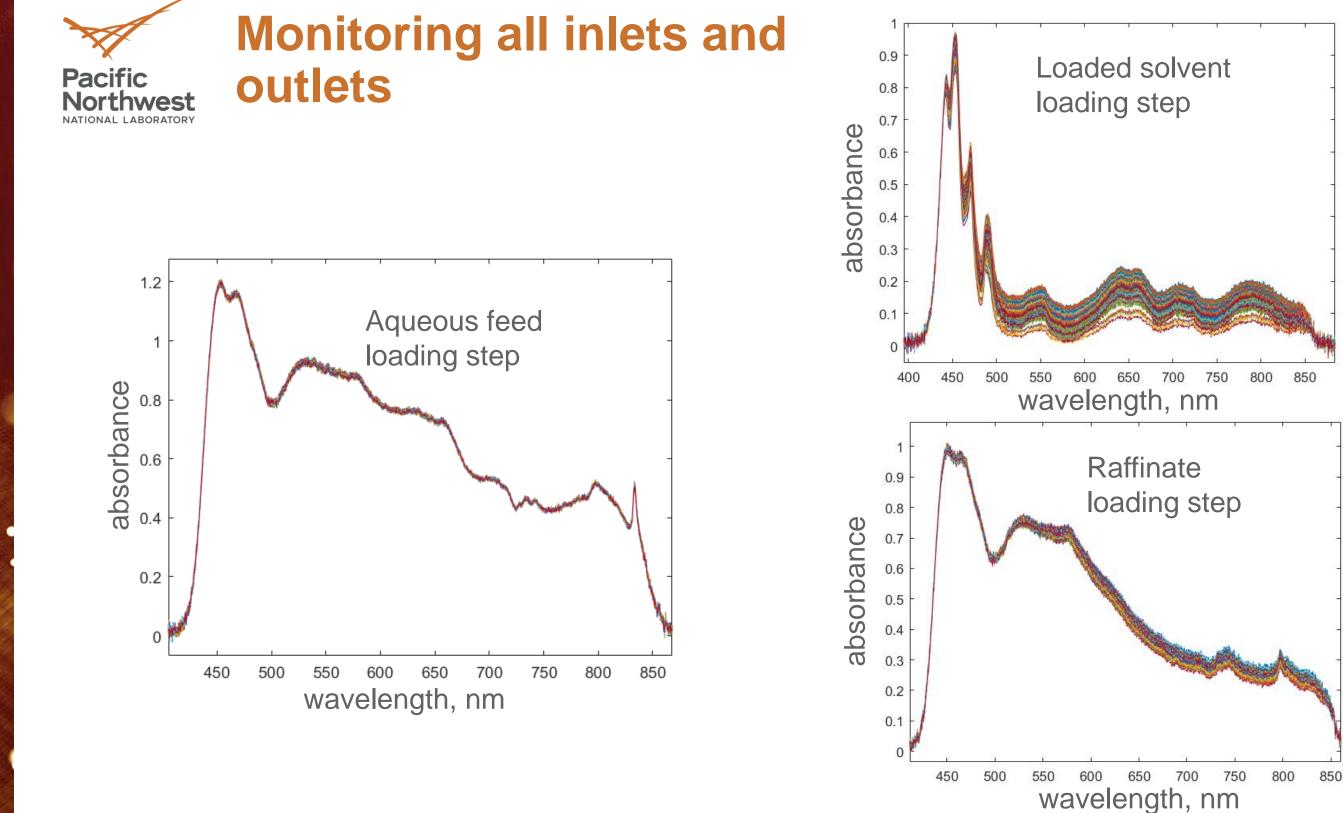
CoDCon on-line monitoring system design

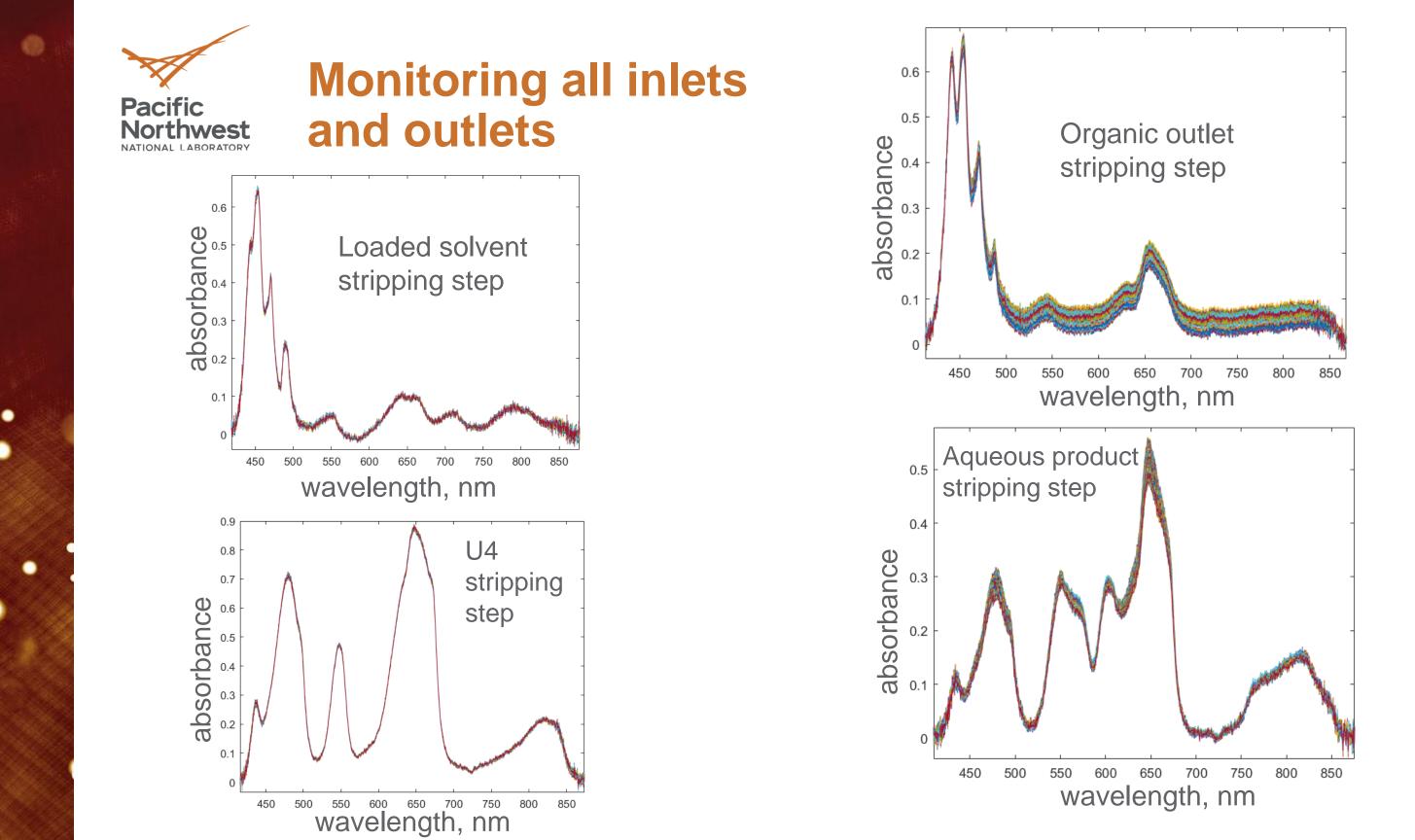






Flow cell installed on all inlets and

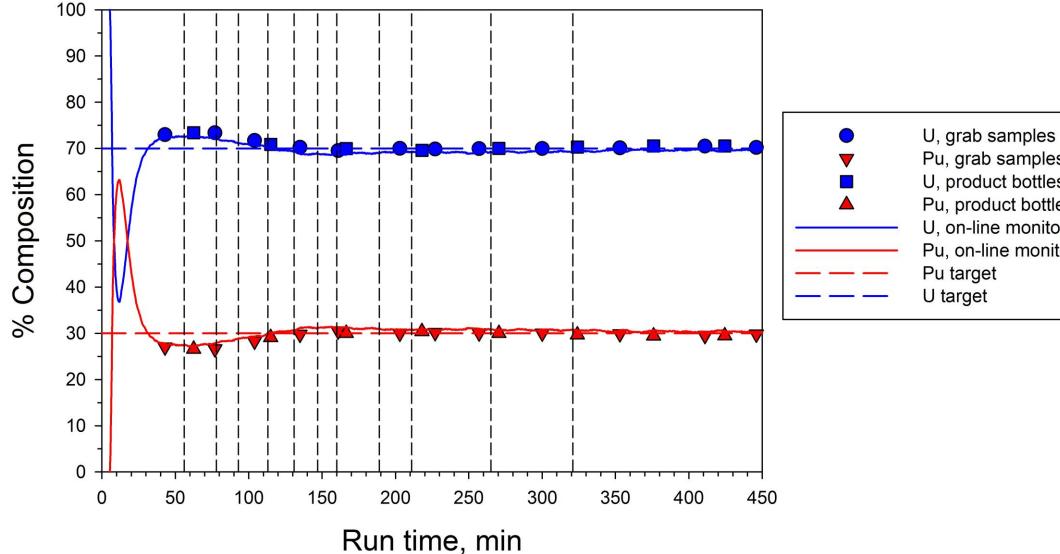




CoDCon demonstration: Enabling Real-Time Process Control Run 3

Pacific

Northwest NATIONAL LABORATORY

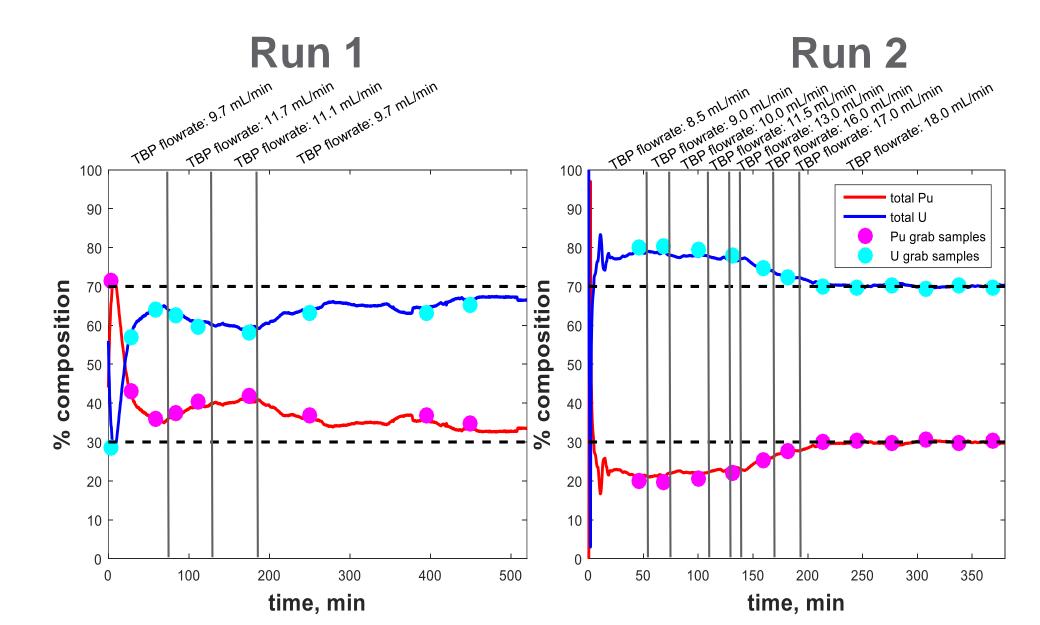


Pu, grab samples U, product bottles Pu, product bottles U, on-line monitor Pu, on-line monitor



Other benefits of on-line monitoring: Identifying Process Equipment Failures

Identification of unexpected separation system behavior

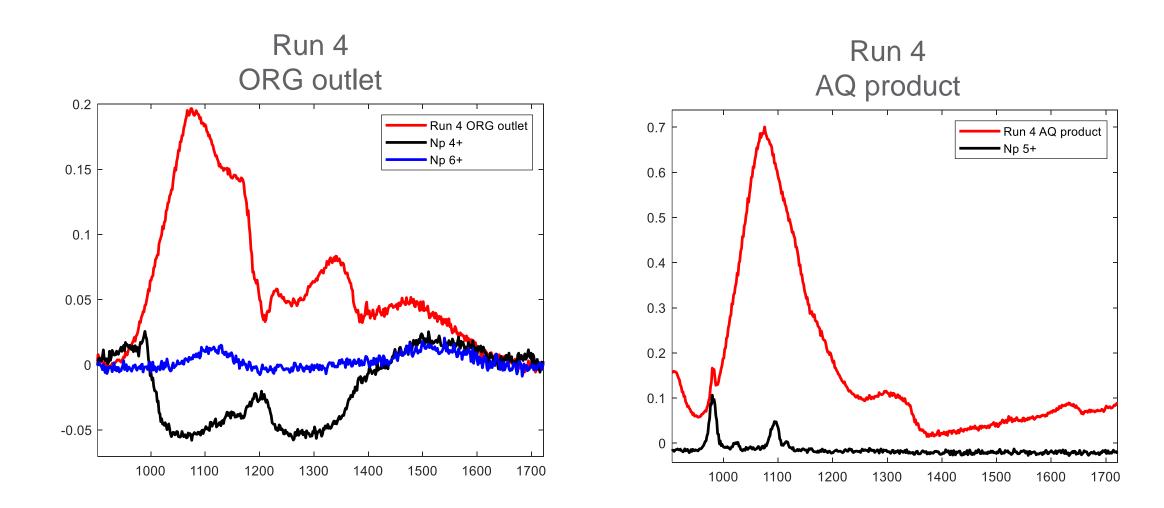






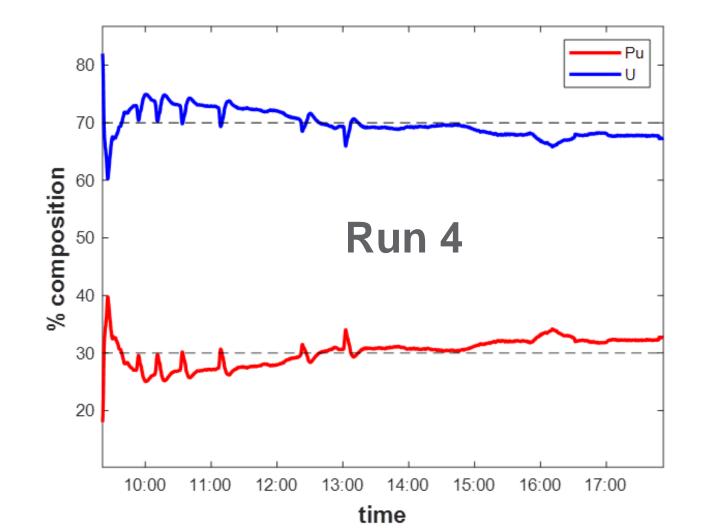
Other benefits of on-line monitoring: Improving Our Fundamental Understanding

• Immediate interrogation of stream allows for capture of short lived species





• Catching system deviations that may not be identified via grab sample collection or gross product characterization

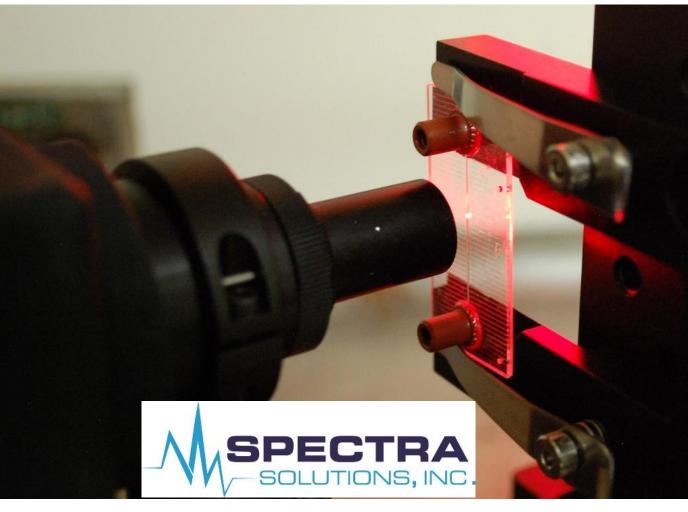


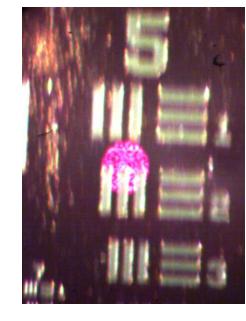


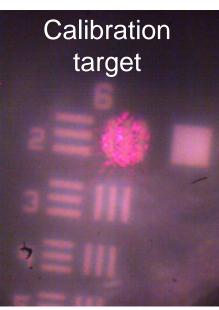
Microfluidics:

Leveraging SBIR and CEA Collaborations

- Fiber optically coupled Raman microscope with integrated video imaging
- High sensitivity Raman system with a focal point capable of measurements inside a microfluidic chip









generation objective focal point diameter ~70 µm

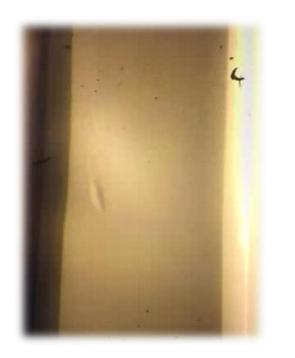
2nd generation objective focal point diameter



Solvent Extraction Monitoring in Microfluidics: simultaneous two-phase measurement

- Two Phase system: 30% TBP-dodecane / nitric acid (HNO₃)
- Simultaneous monitoring of aqueous and organic phases under dynamic flow

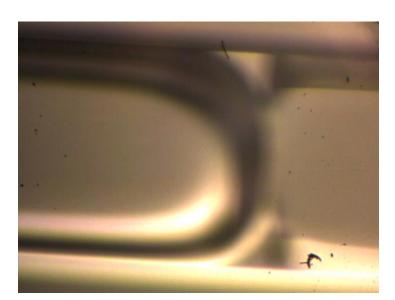


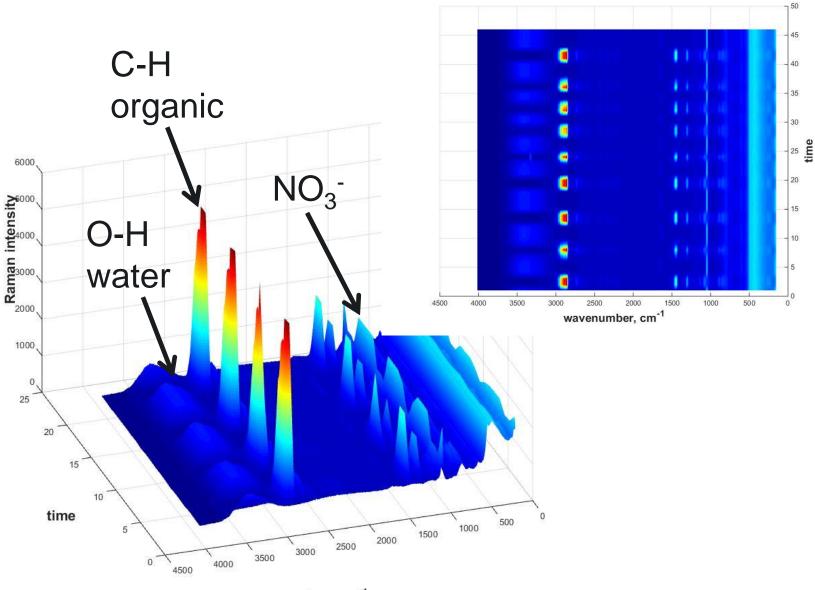


On board camera focused on microfluidic channel



Archetypal binning coupled with chemometric modeling used to simultaneously analyze both phases



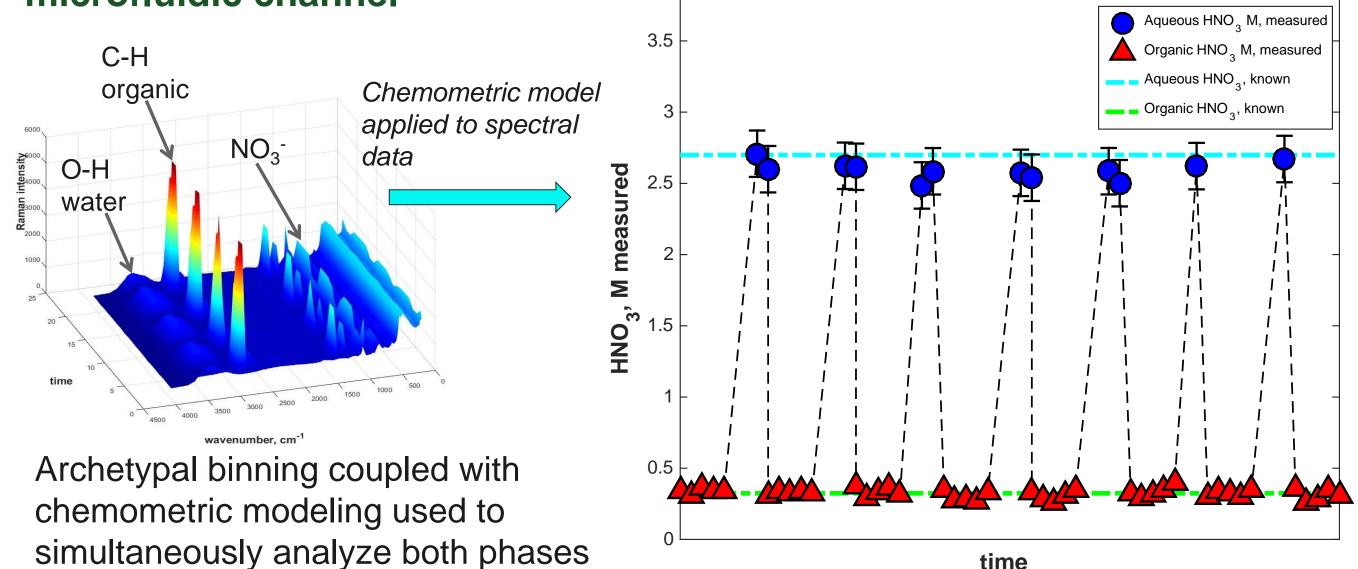


wavenumber, cm⁻¹



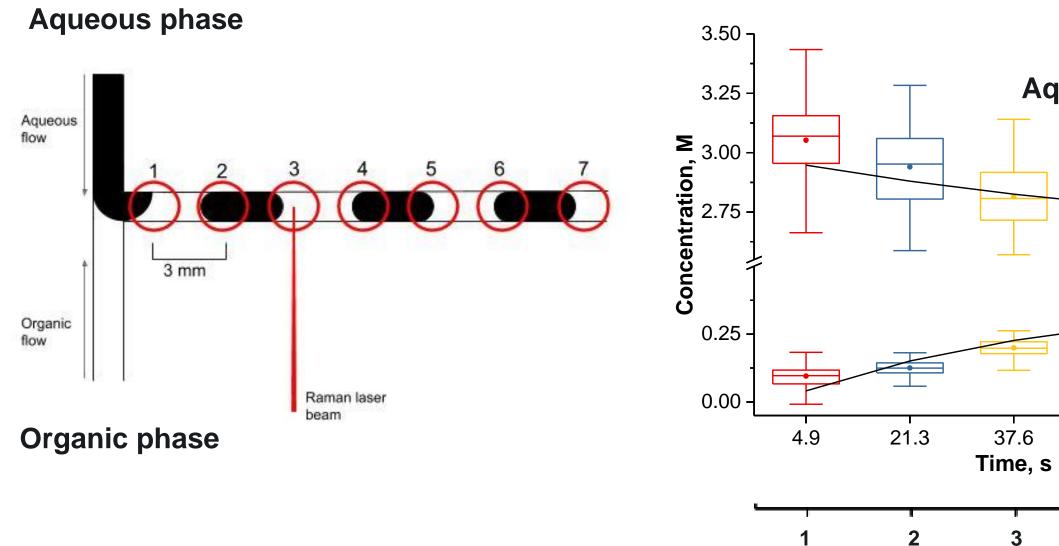
Solvent extraction monitoring in microfluidics: two-phase measurement

real-time measurement of HNO₃ in aqueous and organic phases within microfluidic channel





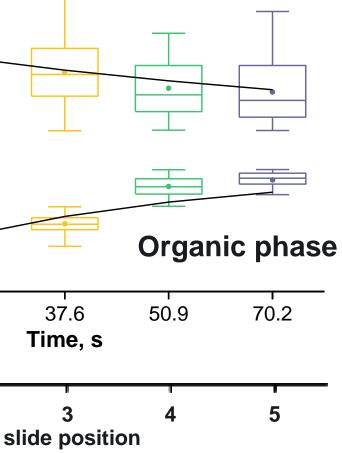
Solvent extraction monitoring in microfluidics: kinetics of two-phase extraction



Anal. Chem., 2018, (90), 8345-8353.



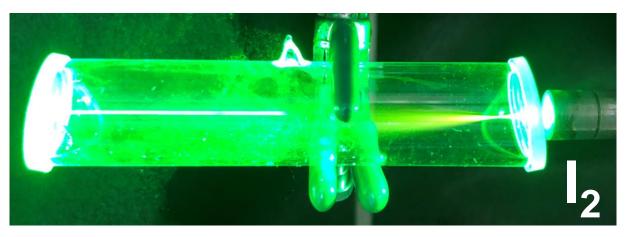
Aqueous phase

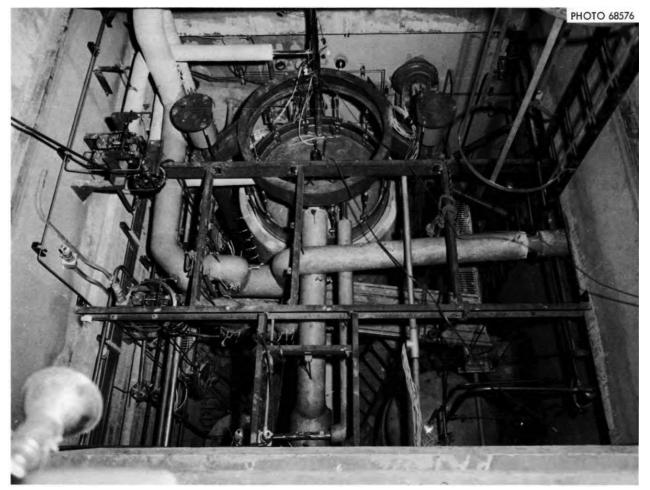




Molten Salt Reactor Off-gas Treatment On-line Monitoring to quantify Iodine Species

- Collaboration with ORNL
- Building off-gas treatment systems and integrating on-line monitoring
- Duel methods
 - LIBS at ORNL
 - Raman at PNNL



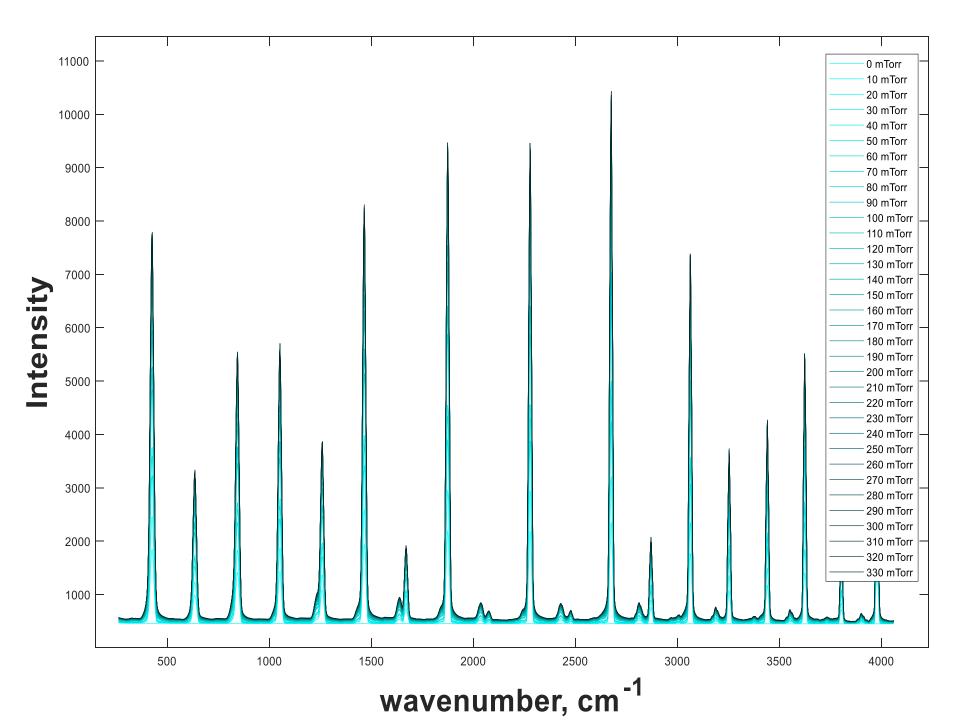


MSRE fuel processing hot cell. R.B. Lindauer, ORNL-TM-2578 (1969)

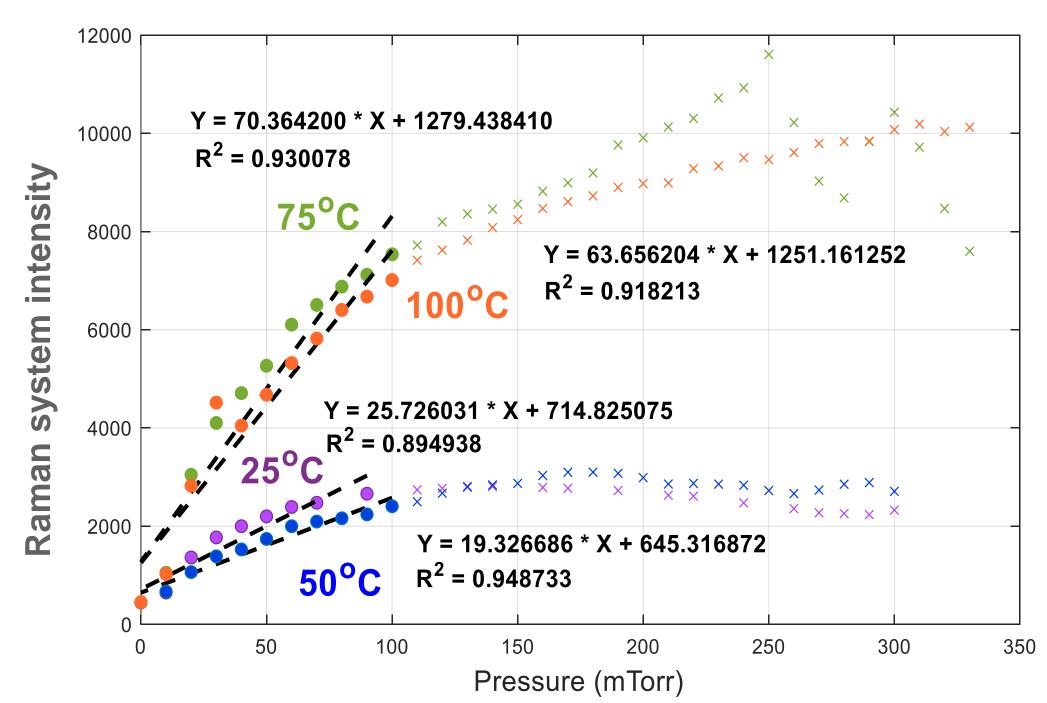


Analysis of I₂ fingerprints: variable pressure

- I₂ exhibits strong and unique fingerprint
- Data shows excellent correlation between signal intensity and I₂ partial pressure



Analysis of I₂ fingerprints: As a function of temperature and pressure



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Pacific

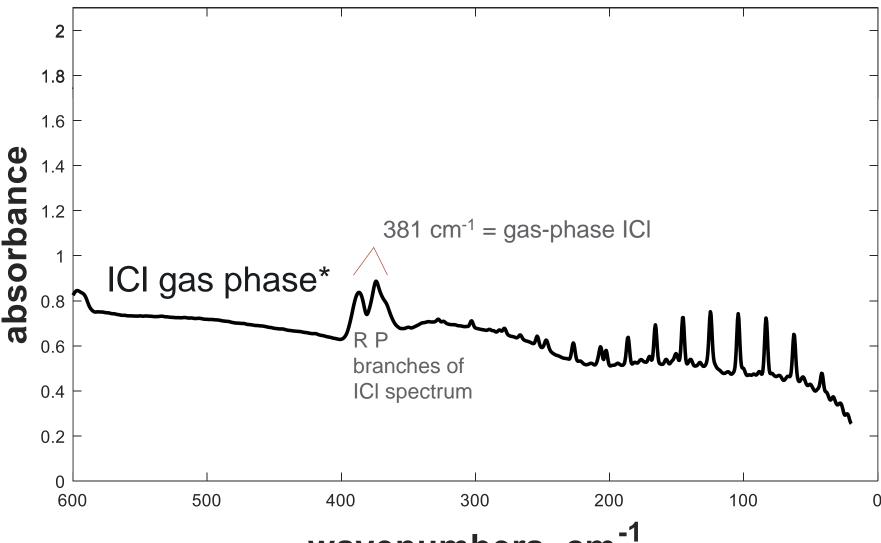




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FTIR analysis of ICI fingerprints

- ICI can also be identified via FTIR
- Sensor fusion of Raman and FTIR additional benefits
 - Multiple methods of detecting HCI, HF, etc
 - Potential to measure temperature of gas based on spectral fingerprint



wavenumbers, cm⁻¹

*Band assignment: **J. Chem. Phys.** 52, 399 (1970).

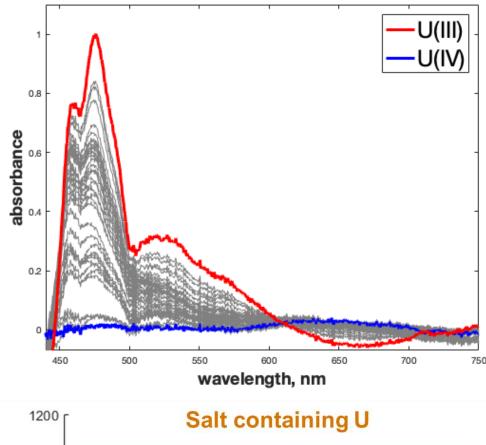


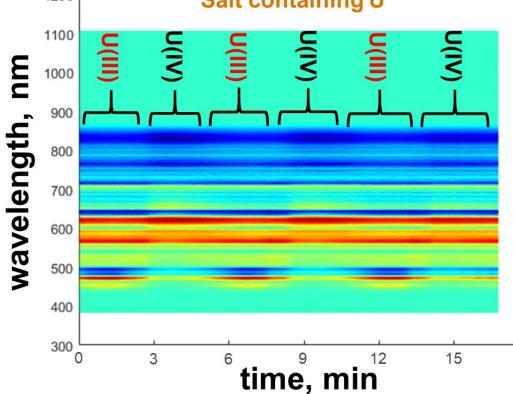


- On-line monitoring as well as fundamental analysis of salt melts
- Applied to CI salts, with plans to test NaOH and F salts





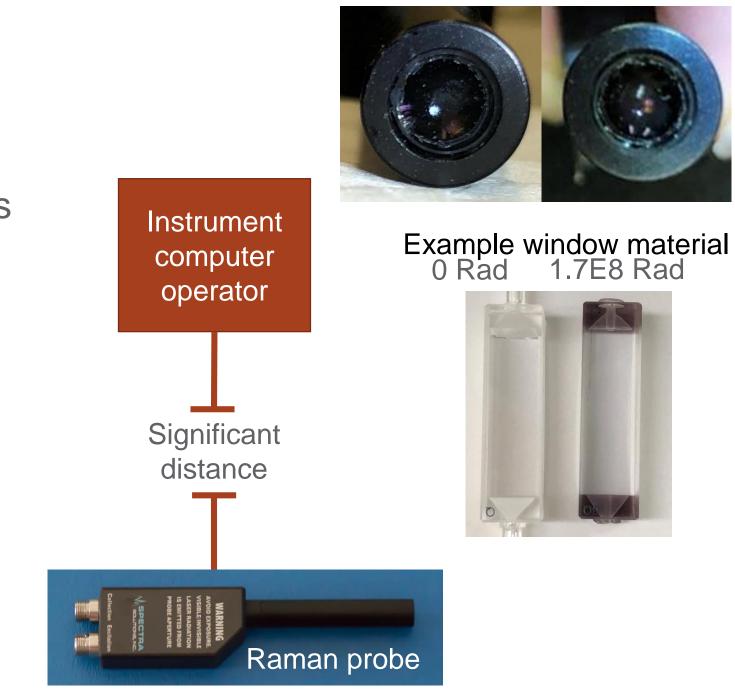






Application Within Harsh Environments for face

- High temperatures (>600°C)
- Highly corrosive liquids and gases
- High radiation dose

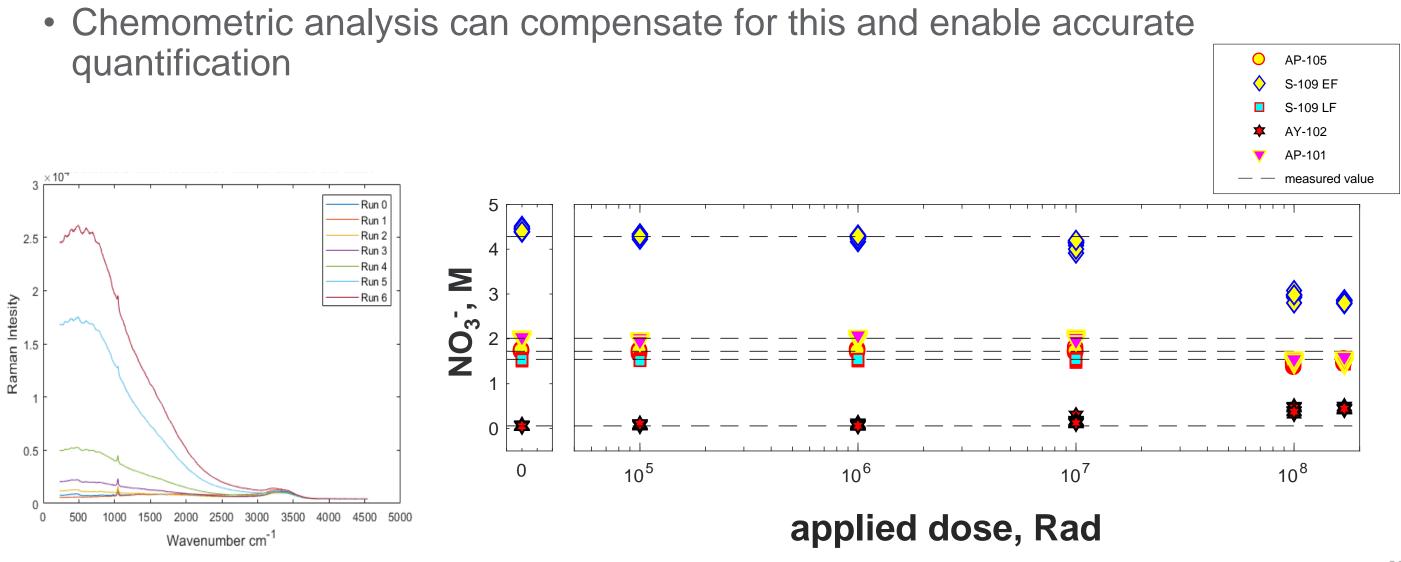






Modeling of data during irradiation dose tests

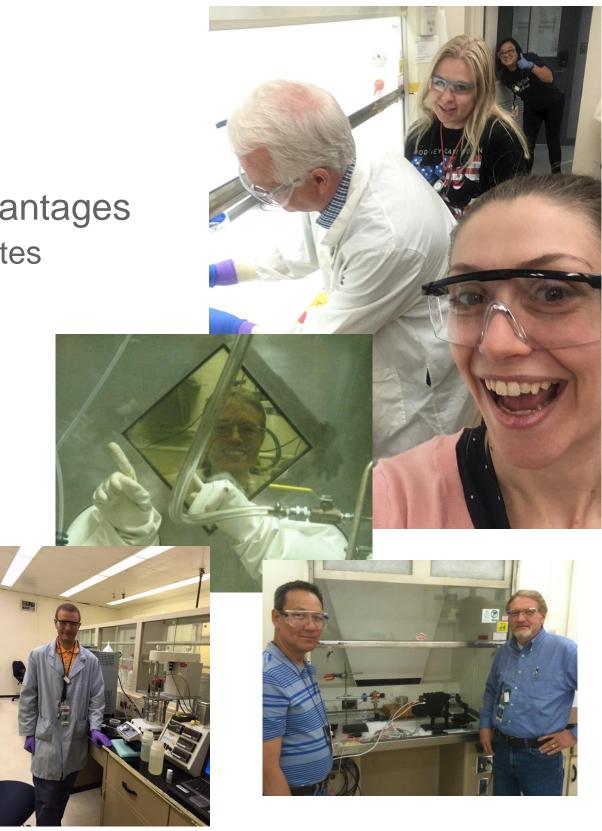
- Spectral background changes with radiation dose
- quantification





Conclusions

- On-line monitoring can offer significant advantages
 - Identification and quantification of target analytes
 - Fast, versatile application
- Real-time characterization
 - Safeguards
 - Process control/optimization





Acknowledgements

- U.S. Department of Energy, Office of Nuclear Energy
- Small Business Innovative Research (SBIR) Grant, Office of Science (SC); collaboration with Spectra Solutions Inc.
- Visiting Faculty Program and Next Generation Safeguards Internship program
- PNNL LDRD programs

Team:

Sam Bryan Amanda Lines Gregg Lumetta Shirmir Branch Gabe Hall Jarrod Allred

Job Bello Amanda Casella Forrest Heller Susan Adami Brian Riley Susan Asmussen

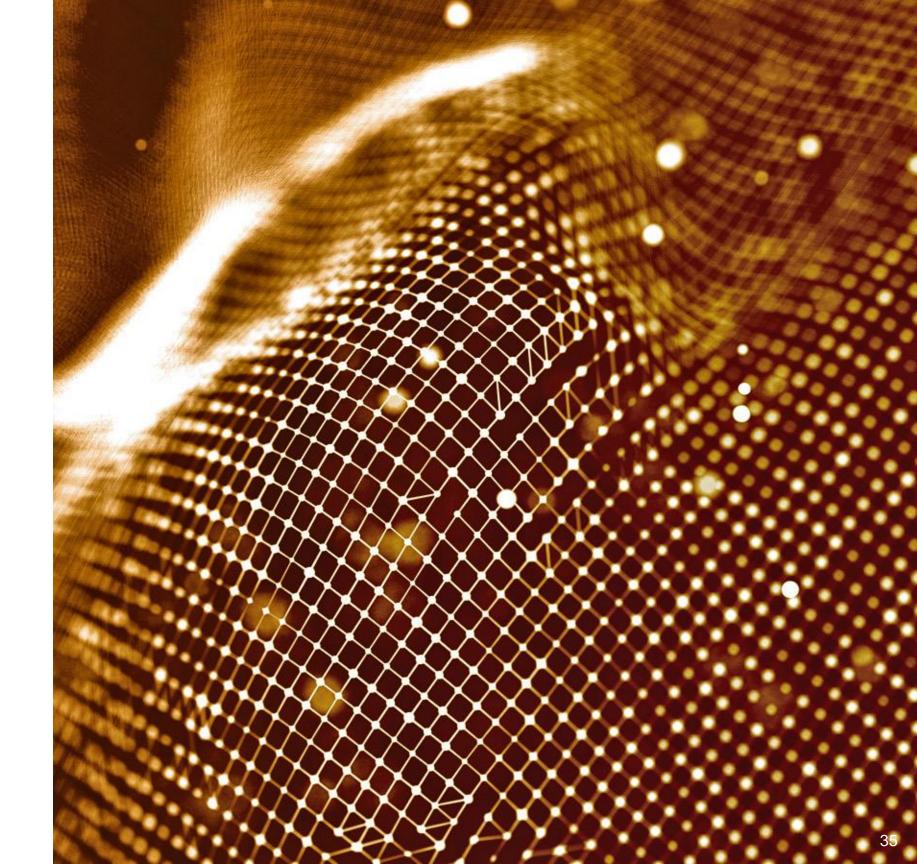
Students/visiting faculty:

Gilbert Nelson Hope Lackey Jen Wilson PoKi Tse



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Thank you





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