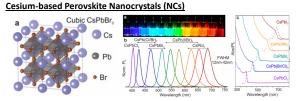


Stable Perovskite Core-Shell Nanocrystals as Down-Converting **Phosphors for Solid State Lighting**

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Introduction

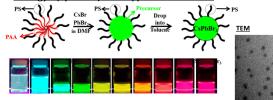


- · Cesium-based all inorganic perovskite nanocrystals have bandgap energies through the entire visible spectrum (410-700nm), narrow FWHM (12-42nm), excellent quantum yield (QY, 50-90%), and short radiative lifetimes (1-29 ns).1
- The bandgap is easily tunable by either changing the halide composition between chlorine. bromine, and iodine or by controlling the size of the nanocrystals or quantum dots (QDs).1
- Potential applications are in LEDs, X-ray Detectors, Lasers, etc.
- HOWEVER, long term stability is currently the major problem.
- Compositional Mixing Humidity Deterioration Phase Instability
- Colloidal Aggregation
- Thermal Degradation

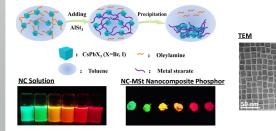
Motivation & Objectives

- Improve stability of colloidal NCs by synthesizing NCs inside a unimolecular star-like block copolymer nanoreactor or impregnating pre-synthesized NCs in metal stearate.
- Utilize permanently tethered PS chains on the CsPbBr₃ surface to improve colloidal stability (prevent aggregation) in any solvent that dissolves PS.
- Form a protective shell layer around the CsPbBr₃ surface that can block penetration of water or other solvents that will break the perovskite crystal structure.
- Embed NCs in a matrix to improve water, composition, and phase stability

Strategy 1 (Synthesis) – PS-capped Perovskite NCs

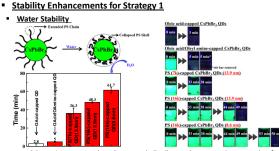


Strategy 2 (Synthesis) – Perovskite-MSt Nanocomposites

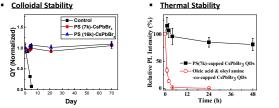


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Results & Discussion

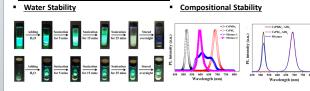


- Schematic represents the polystyrene shell collapse when exposed to water, thus forming a dense protective shell around the perovskite NCs.
- Star-like PAA-b-PS nanoreactors with same inner block PAA length but different outer block PS length (7k & 16k) was used to investigate the effectiveness of the PS shell in protecting the NC. PS-CsPbBr₃ NCs were photoluminescent for up to 20 times longer compared to conventional
- CsPbBr₂ NCs with linear ligands.



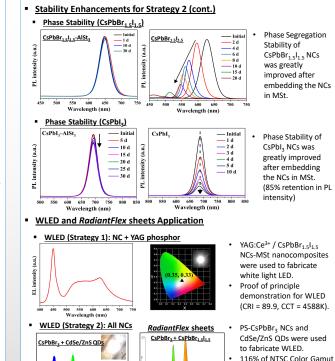
- Conventional CsPbBr₃ NCs with linear ligands showed drastic decay under ambient conditions due to colloidal aggregation. (>90% PL Decay in 5 days)
- PS-CsPhBr, NCs were stable under ambient conditions for more than 2 months (No PL Decay in 70 days; No PL Shift in 70 days; No Change in FWHM in 70 days) PS-CsPbBr₃ NCs showed 85% retention in PL at RT after storage at 80°C for 48 h while conventional perovskite NCs completely lost PL during this time period.

Stability Enhancements for Strategy 2



- CsPbBr₃-MSt nanocomposites were photoluminescent even after directly exposing to water and sonicating for 25 mins while the CsPbBr₃ NCs alone completely lost PL.
- CsPbBra-MSt nanocomposites did not undergo any compositional mixing and maintained their distinct PL wavelengths even after 24 h of mixing while the NCs not embedded in MSt matrix quickly underwent compositional mixing in just 5 minutes.

References 1. Protesescu et al. Nano Letters 2015, 15, 3692.



Results & Discussion

& 163% of sRGB Color LER=332 lm/W CCT=4218k Gamut. RadiantFlex sheets prepared with perovskite NC-MSt nanocomposites Wavelength (nm)

Conclusions

- Perovskite NCs with PLQY up to 81% and green emission bandwidth as narrow as 17.7 nm, RadiantFlex sheets with LER up to 332 lm/W prepared
- PS-CsPbBr₂ NCs & CsPbX₂-MSt nanocomposites were both much more stable compared to conventional CsPbBr₃ NCs when exposed to water.
- PS-CsPbBr₃ NCs displayed greatly enhanced thermal stability in solution as well as in a polymer matrix compared to conventional CsPbBr₂ NCs.
- CsPbX₂-MSt nanocomposites showed great enhancements in compositional as well phase stability compared to conventional CsPbX₃ NCs.

Acknowledgement

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