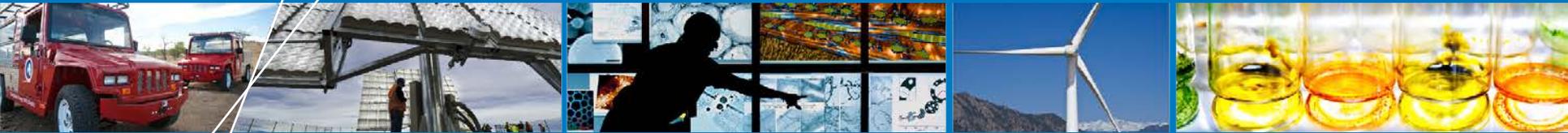


# AMFC Workshop 3 Overview



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(NREL)***

***Sheraton Grand***

***Phoenix, AZ***

***April 1, 2016***

# Agenda

- 8:30 - 8:50 am **AMFC Challenges–Anion Exchange Membrane:** Chulsung Bae (RPI)
- 8:50 - 9:15 am **AMFC Challenges-Electrocatalysis:** Yushan Yan (U. Delaware)
- 9:15 - 9:30 am **BREAK**
- 9:30 - 10:00 am **AMFC Challenges–Membrane Electrode Assembly:** Yu Seung Kim (LANL)
- 10:00 -10:20 am **AMFC Challenges-System/Other Issues (Water/Carbonate):**  
**Miles Page (Elbit Energy)**
- 10:20 -12:00 **BREAKOUT SESSION**  
**Session 1: Research Challenges/R&D Needs**  
AEM – Leader: **Michael Hickner, ORNL (Estrella Main Room)**  
MEA/System – Leader: **Adam Weber, LBNL (Ahwatukee A Room)**  
Catalysts – Leader: **Jacob Spendelow, LANL (Ahwatukee B Room)**
- 12:00 -1:20 pm **LUNCH – ON YOUR OWN**
- 1:20 – 2:20 pm **Joint Session – Out Brief from Breakout Session 1**
- 2:20 – 2:45 pm **AMFC Status:** Dario Dekel (Technion – IIT)
- 2:45 - 3:00 pm **BREAK**
- 3:00 – 4:10 pm **BREAKOUT SESSION**  
**Session 2: Status, Protocols, Milestones**  
AEM – Leader: **Michael Hickner, ORNL (Estrella Main Room)**  
MEA/System – Leader: **Adam Weber, LBNL (Ahwatukee A Room)**  
Catalysts – Leader: **Jacob Spendelow, LANL (Ahwatukee B Room)**
- 4:10 – 5:00 pm **Joint Session – Out Brief from Breakout Session 2**
- 5:00 pm **Concluding Remarks:** David Peterson (DOE)

# Rationale for Workshop

- **AMFCs materials have advanced rather significantly and remain an area of high technical interest**
- **The field has lacked standardization, baselines, or a clear understanding of current status. This impacts setting research priorities.**
- **Workshop goals:**
  - Define *major technical challenges and R&D needs for AMFC technology (Breakout Session 1)*
  - Develop consensus on *Status, Protocols, Milestones for AMFC technology (Breakout Session 2)*

# Overview

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- **While the field is much better understood than it was a few years ago, it is still necessary to provide an overview of AMFCs**
- **Our 4 world leading speakers in the area will provide depth in the most critical areas, this talk looks to provide highlights and fill in some missing areas and then focus on Breakout Sessions and desired outcomes.**

# 2011 AMFC Workshop Findings

**Table 1. Select Highlights of Breakout Sessions**

<b>Breakout Session</b>	<b>Key Highlights</b>
Anion Exchange Membranes – Stability	Membrane stability with Tokuyama membranes has been demonstrated to a level at or near commercial impact. AEM stability remains inferior to proton exchange membranes under conditions relevant to fuel cell operation.
Anion Exchange Membranes – Transport/Conductivity	Conductivity of AEMs is significantly lower than acid membrane analogues. Relatively little transport property data exist for AEMs, but water transport is likely to be an even larger issue in AMFCs than PEMFCs.
Electrocatalysis in High pH Environments	Oxygen reduction under basic conditions using high-performance, durable, non-precious electrocatalysts has been reasonably demonstrated, leaving anode catalysis as the primary concern in stark contrast to acidic systems.
MEA Issues	The most promising AMFC performance and durability reported to date has focused on H <sub>2</sub> as a fuel, and is now commonly achieved without the addition of free electrolyte. Performance of single cells has increased significantly with ~500 mW/cm <sup>2</sup> performance reported and durability in the thousands of hours.
System Issues	System issues will depend on system-specific requirements, but work in this area is necessary to determine how much improvement is needed in each of the other areas to produce viable devices. CO <sub>2</sub> from air or fuel has a major impact on system design and performance.

<http://energy.gov/eere/fuelcells/alkaline-membrane-fuel-cell-workshop>

# 2011 AMFC Workshop Findings

Table 1. Select Highlights of Breakout Sessions

Breakout Session	
Combined at this Workshop	Anion Exchange Membranes – Stability
	Anion Exchange Membranes – Transport/Conductivity
	Electrocatalysis in High pH Environments
Combined at this Workshop	MEA Issues
	System Issues

3 Break Out Sessions

Membranes  
Electrocatalysts  
MEAs

<http://energy.gov/eere/fuelcells/alkaline-membrane-fuel-cell-workshop>

# Breakout Session 1

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*Major technical challenges and R&D needs for AMFC technology*

# Recent Advances

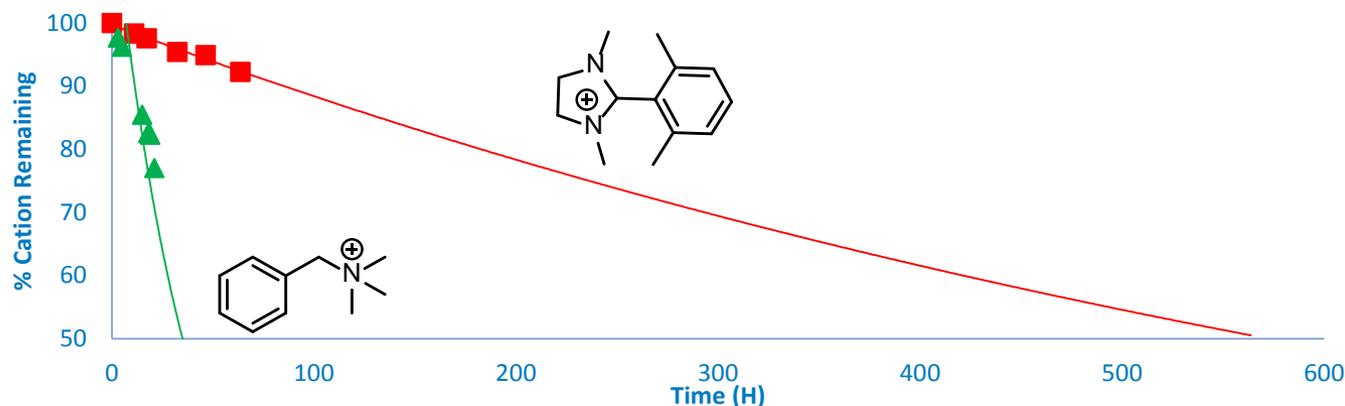
*Perhaps the most exciting advance in the AEM area in the past decade has come from the high hydroxide stabilities of covalently tetherable cations recently reported.*

benzyltrimethyl ammonium (BTMA<sup>+</sup>) stability at 80°C of less than 10% degradation after 5,000 hours.

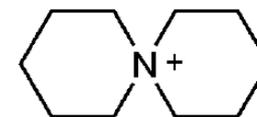
M Sturgeon, C Macomber, C Engrakul, H Long, and B Pivovar, J. Electrochem. Soc., 162 (4) F366-F372 (2015).

Our results suggest benzyl imidazole is ~100x more stable than BTMA.

0.1M cation in 2M KOH(aq) @ 140 °C



6-azonia-spiro[5.5]undecane  
20x improvement.

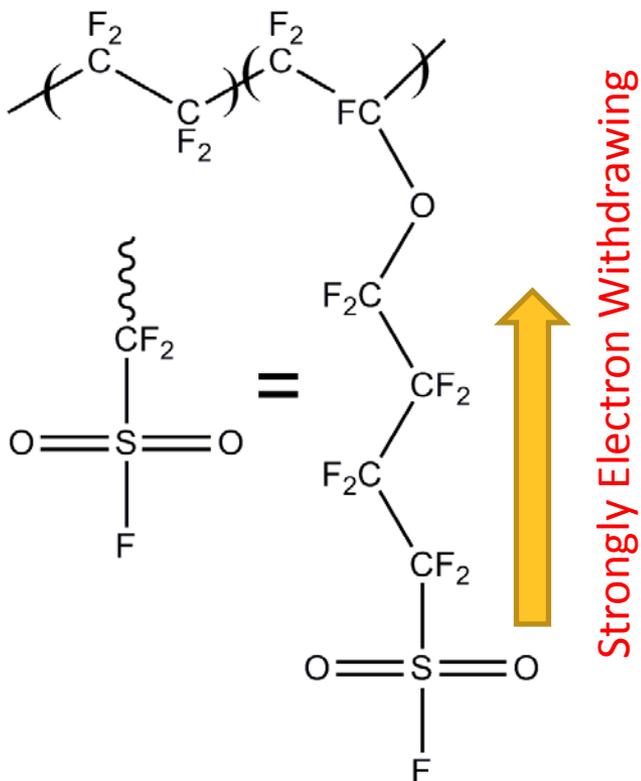


6-azonia-spiro[5.5]undecane

Marino, M.G. and K.D. Kreuer, Chemsuschem, 2015. 8(3): p. 513-523

# Research Needs

## AEMs



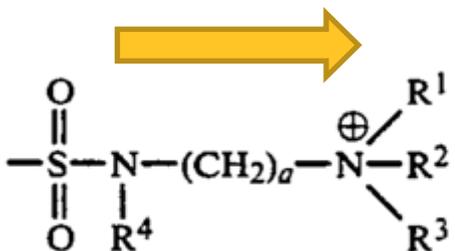
Strongly Electron Withdrawing

Backbone and tether stability remain concerns.

Can PF linking strategies be developed that have stability

Can straight hydrocarbon polymers withstand operating conditions and provide desired properties (conductivity, mechanical properties)

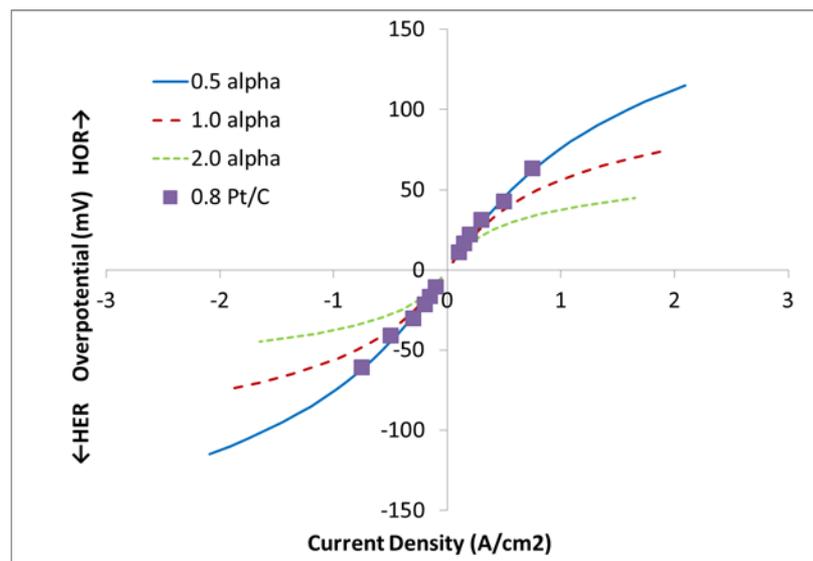
Electron Deficient



# Research Needs

## Electrocatalysis

**Electrocatalysis** - Hydrogen reactions are the concern rather than oxygen reaction (PEM).



Reference	Catalyst	$i_{O,HOR}$ (mA/cm <sup>2</sup> )
Strmcnik et al., 2013	Pt/C	~1.0
“	Pt(0.1)Ru(0.9)/C	>6.5
Durst et al., 2014	Pt/C	1.0
Wang et al., 2014	Pt/C	0.3
“	Pt(0.6)Ru(0.4)/C	0.7
St. John et al., 2015	Pt/C	0.49
“	Pt(0.8)Ru(0.2)/C	1.42

# Research Needs

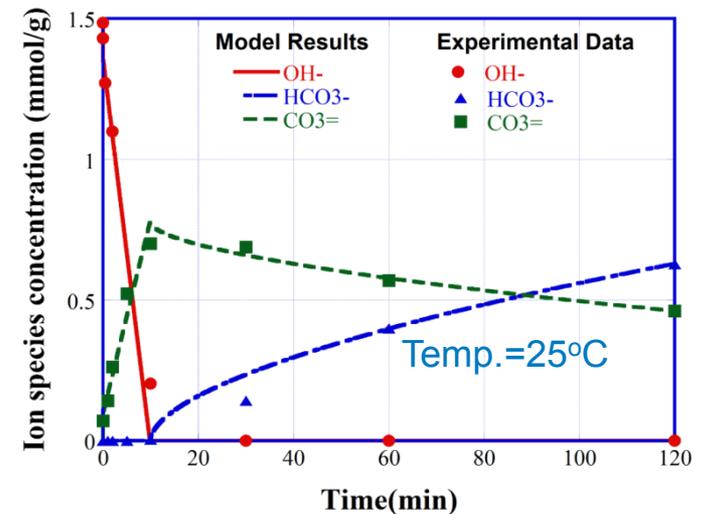
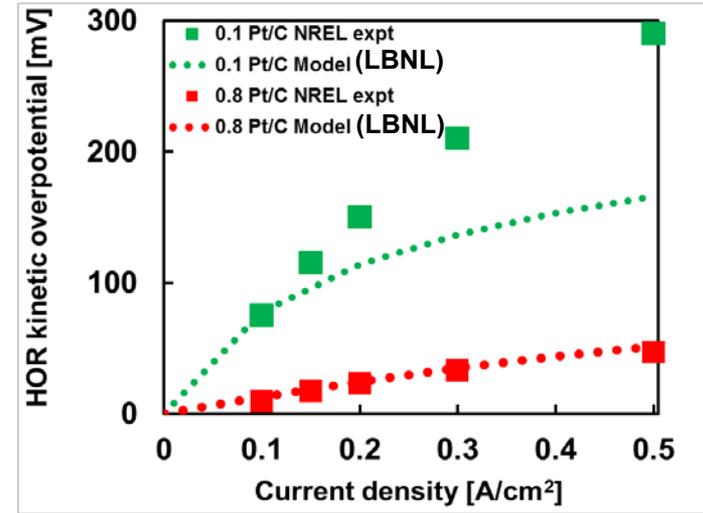
## MEAs/Systems

**Electrode Fabrication** – Optimized performance, ionomer solutions.

**Cell Performance/Durability** - Optimization, diagnostics and quantification of losses.

**Water Management** – A more significant issue in AMFC than PEMFCs. Impacts operating temperatures.

**Carbonate** –  $\text{CO}_2$  a major concern for performance and operating conditions.



Experimental data from Yanagi, H. and K. Fukuta, ECS Transactions, 2008. **16**(2): p. 257-262.

# Breakout Session 2

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## *Status, Protocols, Milestones for AMFC technology*

# Target/Status Tables for AEMs and AMFCs

Table 3.4.12 Technical Targets: Electrocatalysts for Transportation Applications

Characteristic	Units	2005 Status <sup>a</sup>		Stack Targets	
		Cell	Stack	2010	2015
Platinum group metal (pgm) total loading <sup>b</sup>	mg PGM / cm <sup>2</sup> electrode area	0.45	0.8	0.3	0.2
Cost	\$ / kW	9	55 <sup>c</sup>	5 <sup>d</sup>	3 <sup>d</sup>
Durability with cycling					
Operating temp ≤80°C	hours	>2,000	~2,000 <sup>e</sup>	5,000 <sup>f</sup>	5,000 <sup>f</sup>
Operating temp >80°C	hours	N/A <sup>g</sup>	N/A <sup>g</sup>	2,000	5,000 <sup>f</sup>
Electrochemical area loss <sup>h</sup>	%	90	90	<40	<40
Mass activity <sup>i</sup>	A / mg Pt @ 900 mV <sub>iR-free</sub>	0.28	0.11	0.44	0.44
Specific activity <sup>j</sup>	μA / cm <sup>2</sup> @ 900 mV <sub>iR-free</sub>	550	180	720	720

The information in Tables like these is helpful for both quantifying the status of a technology and setting research goals.

Protocol's similarly beneficial.

Table 3.4.13 Technical Targets: MEAs

Characteristic	Units	2005 Status <sup>a</sup>	2010	2015
Operating temperature	°C	<80	<120	<120
Inlet water vapor partial pressure	kPa	50	<1.5	<1.5
Cost <sup>b</sup>	\$ / kW	60 <sup>c</sup>	10	5
Durability with cycling				
At operating temp of ≤80°C	hours	~2,000 <sup>d</sup>	5,000 <sup>e</sup>	5,000 <sup>e</sup>
At operating temp of >80°C	hours	N/A <sup>f</sup>	2,000	5,000 <sup>e</sup>
Unassisted start from low temperature	°C	-20	-40	-40
Performance @ ¼ power (0.8V)	mA / cm <sup>2</sup> mW / cm <sup>2</sup>	200 160	300 250	300 250
Performance @ rated power	mW / cm <sup>2</sup>	600	1,000	1,000
Extent of performance (power density) degradation over lifetime <sup>g</sup>	%	5 <sup>h</sup>	10	5
Thermal cyclability in presence of condensed water		Yes	Yes	Yes

# Status, Protocols, Milestones

## AEMs

Cycle	Cycle from 0% RH (2 min) to 90°C dewpoint (2 min), single cell 25-50 cm <sup>2</sup>	
Total time	Until crossover >2 mA/cm <sup>2</sup> or 20,000 cycles	
Temperature	80°C	
Relative Humidity	Cycle from 0% RH (2 min) to 90°C dewpoint (2 min)	
Fuel/Oxidant	Air/Air at 2 SLPM on both sides	
Pressure	Ambient or no back-pressure	
Metric	Frequency	Target
Crossover*	Every 24 h	≤2 mA/cm <sup>2</sup>
Shorting resistance**	Every 24 h	>1,000 ohm cm <sup>2</sup>

\* Crossover current per USFCC “Single Cell Test Protocol” Section A3-2, electrochemical hydrogen crossover method.

\*\* Measured at 0.5 V applied potential, 80°C and 100% RH N<sub>2</sub>/N<sub>2</sub>. Compression to 20% strain on the GDL.

**AEMs** – PEM developed protocols may be inappropriate

What are most appropriate metrics? (conductivity, mechanical properties)

What are most appropriate test conditions? (OH<sup>-</sup> form, vs Cl<sup>-</sup> form, T, RH)

What about durability? (Specific durability tests, anything beyond chemical?)

# Status, Protocols, Milestones

## Electrocatalysis

**Table E1 Electrocatalyst Cycle and Metrics**

Cycle	Triangle sweep cycle: 50 mV/s between 0.6 V and 1.0 V. Single cell 25-50 cm <sup>2</sup>	
Number	30,000 cycles	
Cycle time	16 s	
Temperature	80°C	
Relative Humidity	Anode/Cathode 100/100%	
Fuel/Oxidant	Hydrogen/N <sub>2</sub> (H <sub>2</sub> at 200 sccm and N <sub>2</sub> at 75 sccm for a 50 cm <sup>2</sup> cell)	
Pressure	Atmospheric pressure	
Metric	Frequency	Target
Catalytic Mass Activity <sup>a</sup>	At Beginning and End of Test minimum	≤40% loss of initial catalytic activity
Polarization curve from 0 to ≥1.5 A/cm <sup>2b</sup>	After 0, 1k, 5k, 10k, and 30k cycles	≤30 mV loss at 0.8 A/cm <sup>2</sup>
ECSA/Cyclic Voltammetry <sup>c</sup>	After 10, 100, 1k, 3k, 10k, 20k and 30k cycles	≤40% loss of initial area

- Mass activity in A/mg @ 150 kPa abs backpressure at 857 mV iR-corrected on 6% H<sub>2</sub> (balance N<sub>2</sub>)/O<sub>2</sub> {or equivalent thermodynamic potential}, 100% RH, 80°C normalized to initial mass of catalyst and measured before and after test.
- Polarization curve per protocol in Table E3.
- Sweep from 0.05 to 0.6 V at 20 mV/s, 80°C, 100% RH.

**ORR – 0.6-0.9V cycling may be fine to just take PEM as this replicates operating conditions, but is this an issue for AMFCs?**

**What about support corrosion?**

**What about HOR? What should be catalyst activity target or assessment condition for HOR?**

# Status, Protocols, Milestones

## MEAs/Systems

Table E3 Polarization Protocol

Test Point #	Current Density [A/cm <sup>2</sup> ]	Anode Inlet H2% (balance N2) inlet/dry	Anode H2 Stoich [-]	Anode Dewpoint Temp [°C]	Anode Inlet Temp [°C]	Anode Pressure outlet [kPaabs]	Cathode Inlet O2% inlet/dry	Cathode Inlet N2% inlet/dry	Cathode O2 Stoich [-]	Cathode Dewpoint Temp [°C]	Cathode Inlet Temp [°C]	Cathode Pressure Outlet [kPaabs]	Cell/Stack control Temp [°C]	Test pt. Run Time min	Set Point Transit time s
Break-in															
B1	0.6	100%	1.5	59	80	150	21%	79%	1.8	56	80	150	80	20	0
Reduction															
R1	0	100%	1.5	59	80	150	21%	79%	1.8	59	80	150	80	1 Until V> 0.1V	0
R2	0	100%	1.5	59	80	150	0%	100%	1.8	59	80	150	80		0
Polarization curve															
P1	0.2	100%	1.5	59	80	150	21%	79%	1.8	59	80	150	80	3	0
P2	0.4	100%	1.5	59	80	150	21%	79%	1.8	59	80	150	80	3	0
P3	0.6	100%	1.5	59	80	150	21%	79%	1.8	59	80	150	80	3	0
P4	0.8	100%	1.5	59	80	150	21%	79%	1.8	59	80	150	80	3	0
P5	1	100%	1.5	59	80	150	21%	79%	1.8	59	80	150	80	3	0
P6	1.2	100%	1.5	59	80	150	21%	79%	1.8	59	80	150	80	3	0
P7	1.4	100%	1.5	59	80	150	21%	79%	1.8	59	80	150	80	3	0
P7	1.6	100%	1.5	59	80	150	21%	79%	1.8	59	80	150	80	3	0
P8	1.8	100%	1.5	59	80	150	21%	79%	1.8	59	80	150	80	3	0
P9	2	100%	1.5	59	80	150	21%	79%	1.8	59	80	150	80	3	0
P10	1.8	100%	1.5	59	80	150	21%	79%	1.8	59	80	150	80	3	0
P11	1.6	100%	1.5	59	80	150	21%	79%	1.8	59	80	150	80	3	0
P12	1.4	100%	1.5	59	80	150	21%	79%	1.8	59	80	150	80	3	0
P13	1.2	100%	1.5	59	80	150	21%	79%	1.8	59	80	150	80	3	0
P14	1	100%	1.5	59	80	150	21%	79%	1.8	59	80	150	80	3	0
P15	0.8	100%	1.5	59	80	150	21%	79%	1.8	59	80	150	80	3	0
P16	0.6	100%	1.5	59	80	150	21%	79%	1.8	59	80	150	80	3	0
P17	0.4	100%	1.5	59	80	150	21%	79%	1.8	59	80	150	80	3	0
P18	0.2	100%	1.5	59	80	150	21%	79%	1.8	59	80	150	80	3	0
P19	0.1	100%	1.5	59	80	150	21%	79%	1.8	59	80	150	80	3	0
P20	0.05	100%	1.5	59	80	150	21%	79%	1.8	59	80	150	80	3	0
P21	0.02	100%	1.5	59	80	150	21%	79%	1.8	59	80	150	80	3	0
P22	0.05	100%	1.5	59	80	150	21%	79%	1.8	59	80	150	80	3	0
P23	0.1	100%	1.5	59	80	150	21%	79%	1.8	59	80	150	80	3	0
P24	0.2	100%	1.5	59	80	150	21%	79%	1.8	59	80	150	80	3	0

Stoichs for points below 0.2A/cm<sup>2</sup> at 0.2A/cm<sup>2</sup> equivalent flow

Specific cell protocols exist for PEM performance and durability testing.

For AMFC, what reactants should be used (O<sub>2</sub> or CO<sub>2</sub> free air), what stoich, temperature, humidification, break-in procedure, etc.

How can comparisons be made easily between groups.

# Workshop Deliverables

- **Report to DOE**
  - Define *major technical challenges and R&D needs for AMFC technology (Breakout Session 1)*
  - Develop consensus on *Status, Protocols, Milestones for AMFC technology (Breakout Session 2)*

3 Break Out  
Sessions

Membranes  
Electrocatalysts  
MEAs

- 
- **Thank you all again for your participation**
  - **We look forward to an interesting and productive Workshop.**