

Electrochemical Processing of Used Fuel

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Principles followed in establishing the used LWR & FR fuel treatment processes

- Industrially practicable and economic
- Safeguardable system that meets U.S. non-proliferation objectives
- Maximize actinide recovery to maximize resource utilization and enhance repository performance
- Encapsulate fission product waste in engineered waste forms that can be disposed in an environmentally responsible manner; approach also enhances repository performance
- Minimize secondary waste production

Flowsheet consists of five functional areas

- Head-end operations
- Oxide to metal conversion
- Actinide and fission product separations
- Material recycle
- Waste Management



Example Flowsheet for Treatment of Used Fuel

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Types of Process & Facility Monitoring

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Process operational parameters

- Process cell atmosphere monitoring and control
- Temperature monitoring and control
- Leak detection
- Position sensing
- Remote handling
- Remote maintenance
- Commercial, off-the-shelf technology

Process equipment parameters

- High current, low voltage power system
- Electrode potentials vs. reference electrode
- Amount of charge passed
- Salt level and density (or masses) in equipment and in transfer operations
- Amount of salt removed/added
- Composition of molten salt (sampled and realtime)
 - Electroanalytical methods CV, SWV, CA, DP, ASV, etc
 - Spectroscopic methods UV-Vis, LIBS, etc
- Gas flow rates

Process streams/products

- Composition of molten salt (sampled and realtime)
 - Electroanalytical methods CV, SWV, CA, DP, ASV, etc
 - Spectroscopic methods UV-Vis, LIBS, etc
- Composition of U and U/TRU product
- Off-gas composition

NDA analysis

- Composition of initial used fuel feed material
- Composition of material transfers to waste processing (i.e. actinide-free)



Electroanalytical Monitoring of Process Salts

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- Examined several electroanalytical methods for monitoring U and TRU in process salt
 - Square wave voltammetry, chronoamperometry, differential pulse voltammetry, cyclic voltammetry
- For quantitative measurements, electrode area must be fixed or controlled
 - Successfully used standard addition of electrode area by controlled changes in immersion depth
 - Plot of i_{peak} vs Δh gives line with slope corresponding to concentration
 - Improve statistics with multiple measurements at multiple immersion depths to ~1% RSD
- Most promising technique to date is cyclic voltammetry
 - Gives linear response over typical concentration range
 - Peak current is proportional to concentration
 - Results are in good agreement with chemical analysis
 - Semi-differential treatment improves baseline resolution of reduction peaks



Acknowledgments and Contact Information

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