

# Development of High-Capacity Cathode Materials with Integrated Structures

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ES194

# Overview

## Timeline

- Start date: FY12
- End date: On-going
- Percent complete:
  - 50% (This project carries over from FY2012, but was integrated into the 'voltage fade' program)

## Budget

- Total project funding
  - 100% DOE
- Funding in FY12: \$300K
- Funding in FY13: \$300K

## Barriers

- Low energy density
- Cost
- Abuse tolerance limitations

## Partners

- Lead PI: Michael Thackeray
- Collaborators:
  - CSE, Argonne: Brandon Long, Jason Croy, Donghan Kim, Kevin Gallagher, Zonghai Chen (materials design, synthesis and electrochemical characterization)
  - APS, Argonne: Peter Chupas, Karena Chapman, Matthew Suchomel (HR-XRD and PDF analyses )
  - ABR 'Voltage Fade' Team
  - BASF, Toda Kogyo, LG Chem, Envia Systems



# Objectives

- Develop low cost, high-capacity cathode materials with good structural, electrochemical and thermal stability for PHEVs
  - Design and synthesize Li- and Mn-rich oxides with integrated structures, notably ‘layered-spinel’ materials, to *counter the voltage fade phenomenon* observed in ‘layered-layered’ electrode materials
  - Identify and overcome performance degradation issues
  - Exchange information and collaborate closely with others in ABR’s ‘voltage fade’ team
  - Supply promising high-capacity cathode materials for PHEV cell build



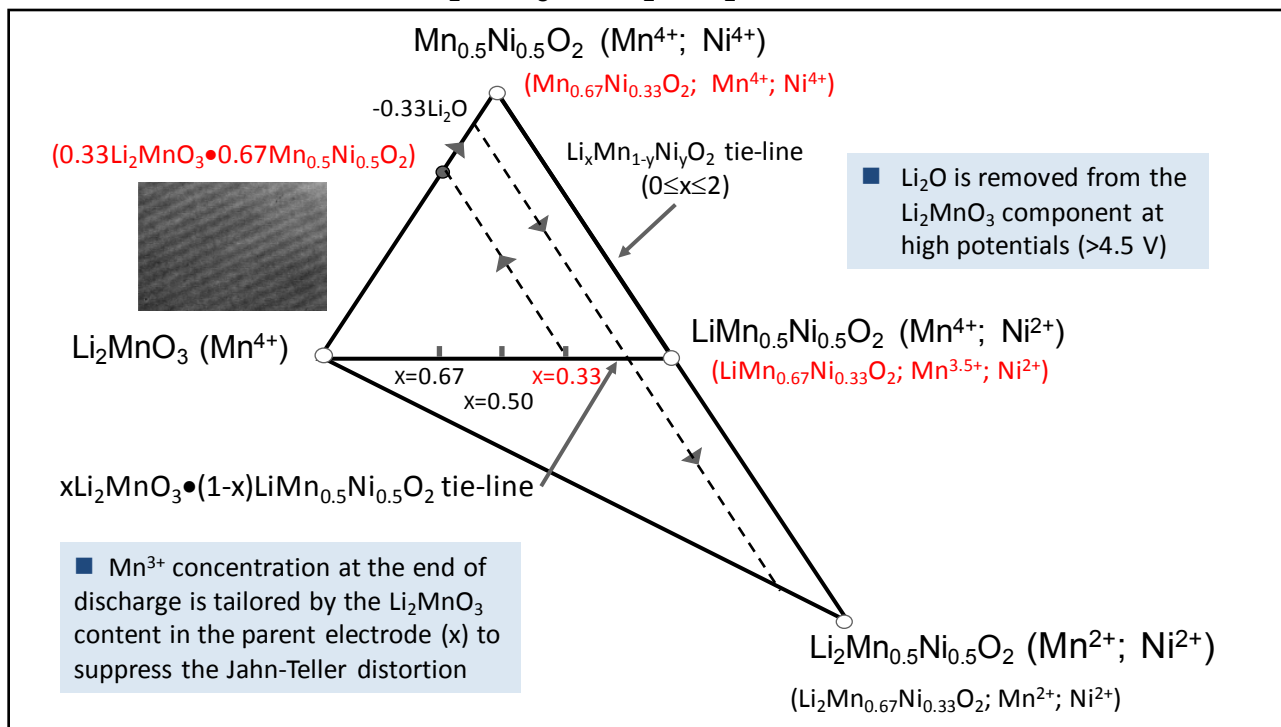
# Milestones FY12

- Explore/optimize the electrode composition using phase diagrams as guide – *on-going*
- Evaluate electrochemical properties of ‘layered-spinel’ electrode materials in lithium half cells and a full Li-ion cell configuration with various anode materials – *on-going*
- Investigate both bulk and surface effects – *on-going*
- Initiate detailed structural analyses of composite electrode structures at the Advanced Photon Source (APS) by X-ray diffraction (XRD), X-ray absorption (XAS) and pair-distribution-function (PDF) analyses – *on-going*

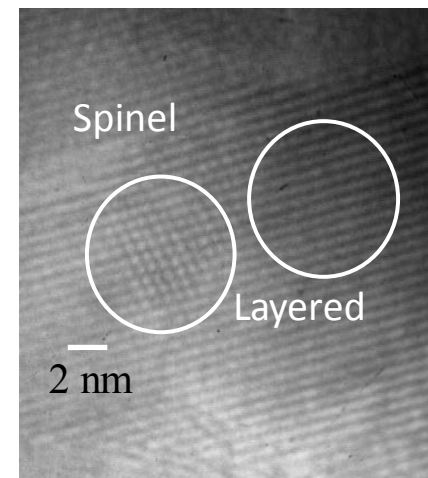


## Background: Integrated Cathode Structures

### 'Layered-Layered' $\text{Li}_2\text{MnO}_3$ - $\text{LiMO}_2$ - $\text{MO}_2$ (M=Mn, Ni) Phase Diagram



'Layered-Spinel'  
 $0.7\text{Li}_2\text{MnO}_3 \bullet 0.3\text{Li}_4\text{Mn}_5\text{O}_{12}$



- Compatibility of ccp planes in layered  $\text{Li}_2\text{MnO}_3$  (C2/m, 001) with those in layered  $\text{LiMO}_2$  (R-3m, 003) and spinel  $\text{LiM}_2\text{O}_4$  (Fd3m, 111) allows structural integration of the two components
- Strategy: Use the 'layered-layered' component to provide high capacity and the spinel component to act as a stabilizer to counter voltage fade

# Strategy / Approach

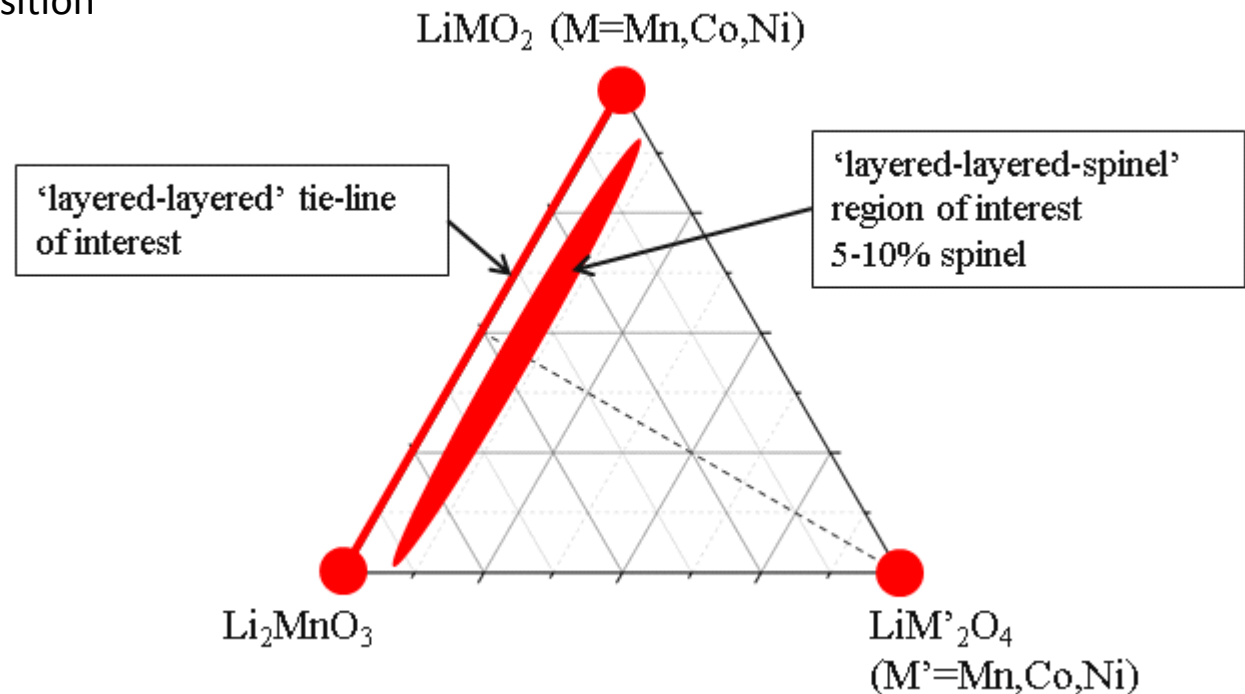
## Part I

- Embed stabilizing spinel component in 'layered-layered' electrode utilizing phase diagram

## Part II

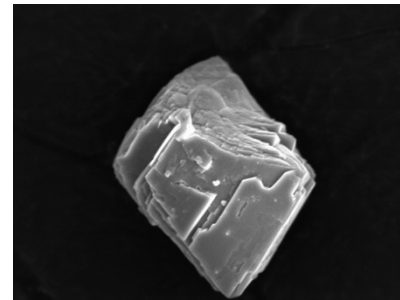
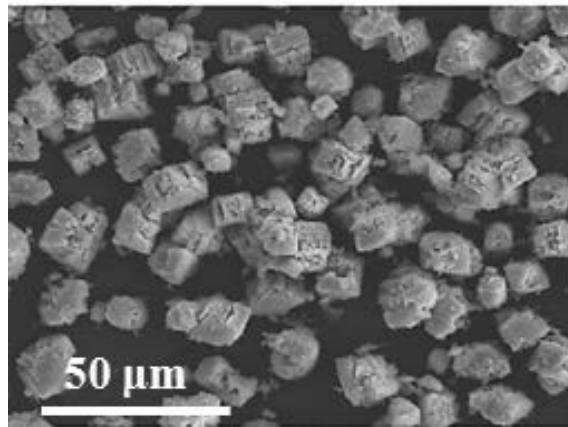
- Structurally and electrochemically characterize 'layered-layered' and spinel end-members of phase diagram to gain understanding of 'layered-layered-spinel' structure and degradation
- Utilize end-member structural knowledge to optimize composite composition

- Spinel stability
- $\text{Li}_2\text{MnO}_3$  activation and Li reserve
- Layered capacity



# Methods

- Electrode materials made from solid state reaction of mixed metal oxalates (formed via co-precipitation) and  $\text{Li}_2\text{CO}_3$  precursors
- Materials heated to 550°C for 12 hours in air at 2°C/min
- Final heating typically 850°C for 12 hours in air
- Electrochemistry performed in half-cell configuration
- SEM micrographs of typical final product shown below



Cleaved crystallites, believed to be result of  $\text{H}_2\text{O}/\text{CO}_2$  loss from oxalate precursor, contribute to porosity of secondary particles

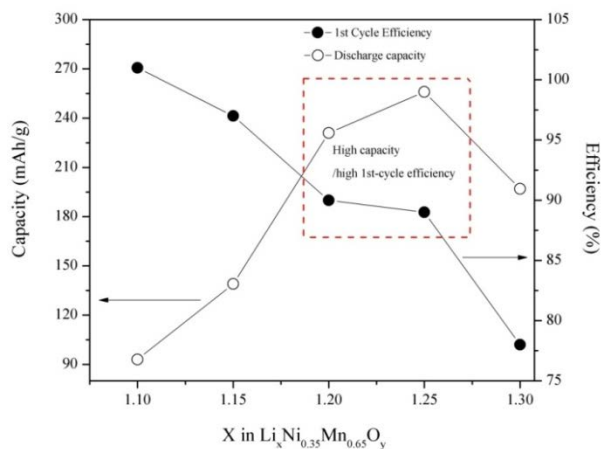
# Li<sub>x</sub>Mn<sub>0.65</sub>Ni<sub>0.35</sub>O<sub>y</sub> System

RECAP: Strategy: Embed spinel component in Mn/Ni system

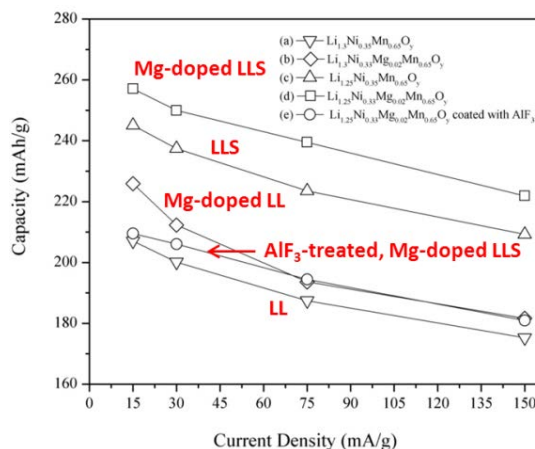
■ The Li<sub>x</sub>Mn<sub>0.65</sub>Ni<sub>0.35</sub>O<sub>y</sub> system has end-members of 0.3Li<sub>2</sub>MnO<sub>3</sub>•0.7LiMn<sub>0.5</sub>Ni<sub>0.5</sub>O<sub>2</sub> (x=1.3; y=2.3), and LiMn<sub>1.3</sub>Ni<sub>0.7</sub>O<sub>4</sub> (x=0.5; y=2)

■ Lowering the Li : transition metal ratio in xLi<sub>2</sub>MnO<sub>3</sub>•(1-x)LiMO<sub>2</sub> embeds a spinel component in the structure

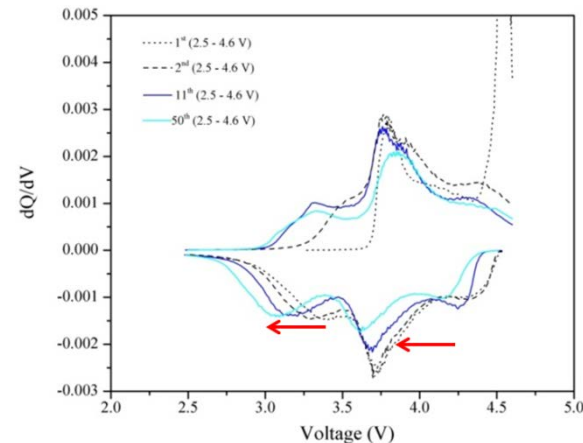
1<sup>st</sup> Cycle Efficiency



Doping / Rate Capability



Electrochemical Performance



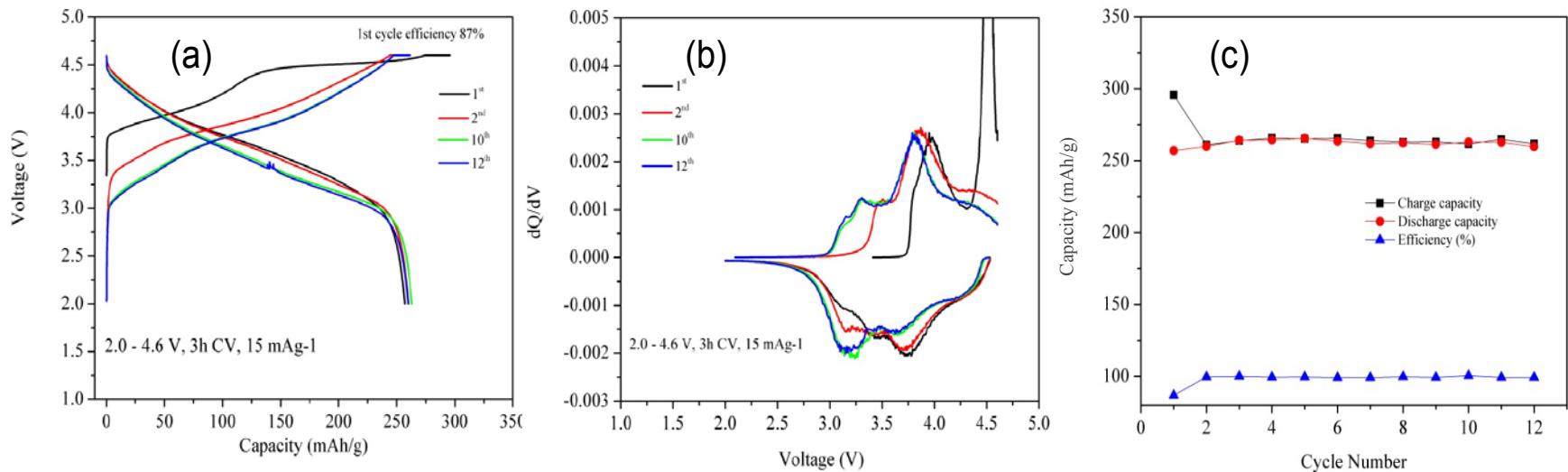
- 5-10% spinel provides best performance
- 1<sup>st</sup> cycle efficiency of 99% achieved for x=1.25 when cycled at 4.6-2.0 V
- ‘Layered-layered-spinel’ electrodes exhibit higher capacities than ‘layered-layered’ electrodes
- Voltage fade phenomenon still present



# Co Containing Mn-rich 'layered-layered-spinel'

Strategy: Add Co to 'layered-layered-spinel' (5% spinel)

- Addition of Co content to 'layered' component of Mn-Ni rich composite
- $\text{Li}_{1.4}\text{Ni}_{0.22}\text{Co}_{0.12}\text{Mn}_{0.66}\text{O}_{2.42}$  alternatively  
 $0.95[0.44\text{Li}_2\text{MnO}_3 \bullet 0.56\text{LiNi}_{0.39}\text{Co}_{0.22}\text{Mn}_{0.39}\text{O}_2] \bullet 0.05\text{Li}_{0.5}\text{Mn}_{0.75}\text{Ni}_{0.25}\text{O}_2$
- Co assumed to be in 'layered-layered' component although variations in cation distribution likely



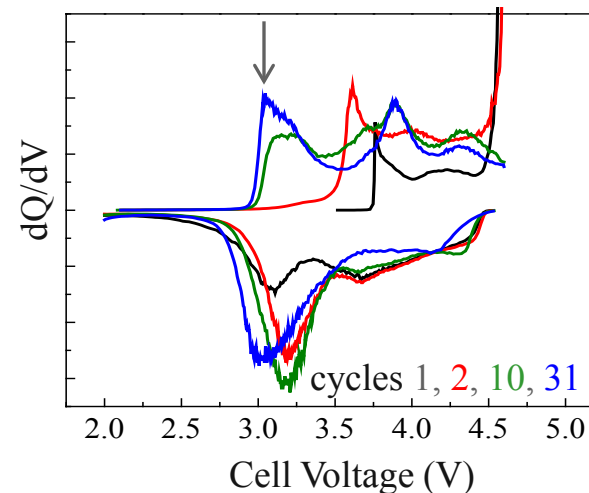
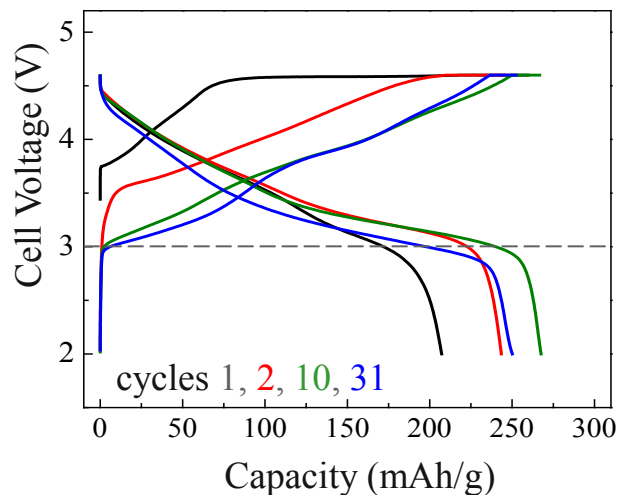
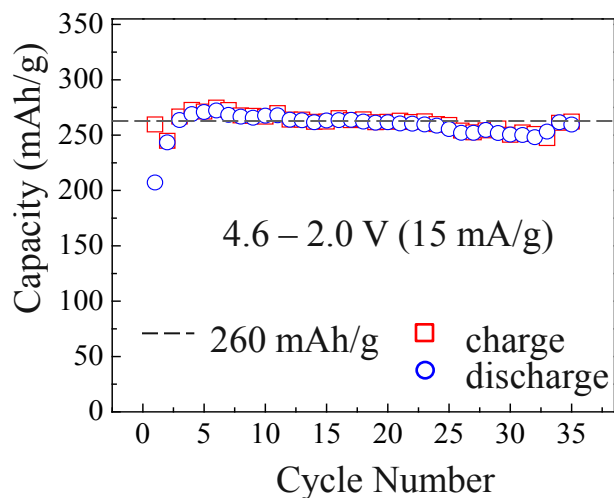
- High capacity and cycling stability achieved
- Spinel-like regions formed during cycling improve stability, but lowers voltage



# Baseline Mn-rich ‘layered-layered’ Member

## Part II Strategy: Evaluate performance of Mn-rich ‘layered-layered’ member

- The effect of  $\text{Li}_2\text{MnO}_3$  content in  $x\text{Li}_2\text{MnO}_3 \cdot (1-x)\text{LiMn}_{0.5}\text{Ni}_{0.5}\text{O}_2$
- Comparison of  $x = 0.7, 0.8$ , and  $0.95$  with various annealing temperatures
- $x = 0.7$  ( $850^\circ\text{C}$ ) showed best performance (data below)

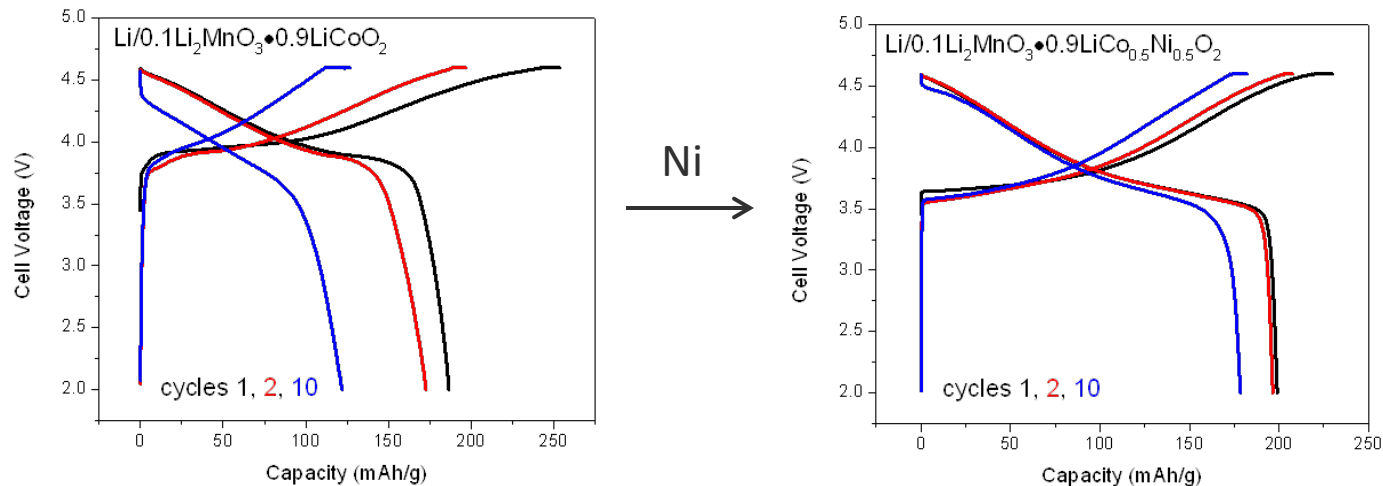


- $\text{Li}_2\text{MnO}_3$  component activated during first 2 cycles before reaching steady capacity of  $\approx 260 \text{ mAhg}^{-1}$
- Voltage profile shows the cell delivered  $\approx 240 \text{ mAhg}^{-1}$  above 3.0 V for  $\approx 10$  cycles

# Increasing Ni Content in $0.1\text{Li}_2\text{MnO}_3 \cdot 0.9\text{LiCo}_y\text{Ni}_{1-y}\text{O}_2$

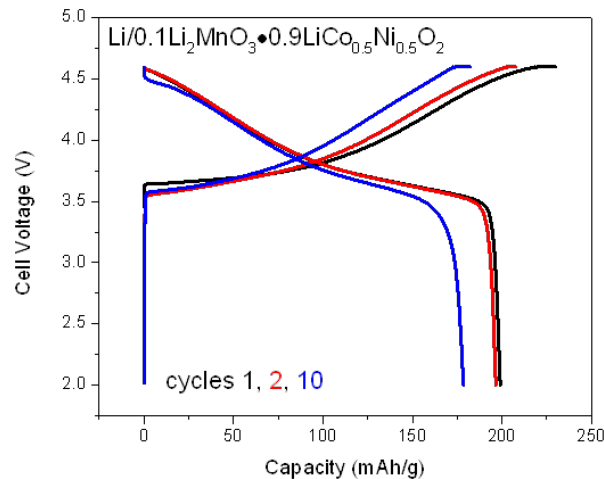
Strategy: Incorporate stabilizing component to Co/Ni-rich electrodes

- Evaluating the addition of a stabilizing  $\text{Li}_2\text{MnO}_3$  component, which can act as a reserve for Li, and Ni substitution to conventional  $\text{LiCoO}_2$  cathode materials for improved performance

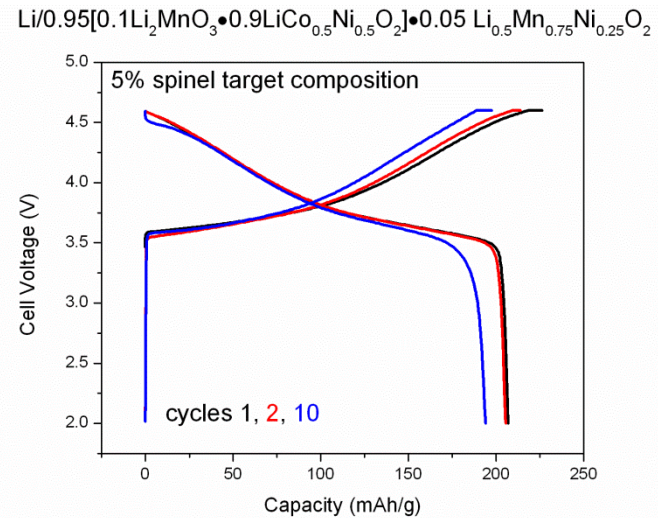


- Increasing Ni content in  $0.1\text{Li}_2\text{MnO}_3 \cdot 0.9\text{LiCo}_y\text{Ni}_{1-y}\text{O}_2$  system improves 1<sup>st</sup> cycle efficiency from 74% ( $y=1$ ) to 87% ( $y=0.5$ )
- Cycling stability improved at cycle 10 from  $120 \text{ mAhg}^{-1}$  ( $y=1$ ) to  $178 \text{ mAhg}^{-1}$  ( $y=0.5$ )
- $190 \text{ mAhg}^{-1}$  of capacity delivered above 3.5 V during first cycles

# Embedding a Spinel Component in $0.1\text{Li}_2\text{MnO}_3 \bullet 0.9\text{LiCo}_{0.5}\text{Ni}_{0.5}\text{O}_2$



Embed  
5% spinel  
→

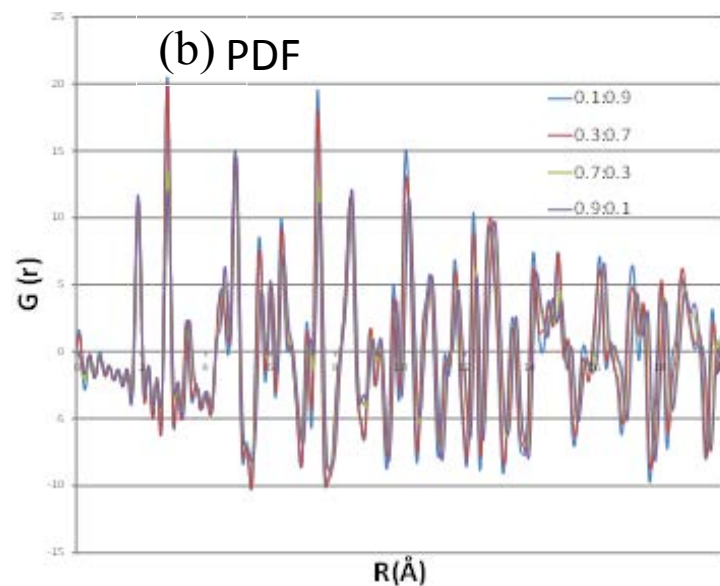
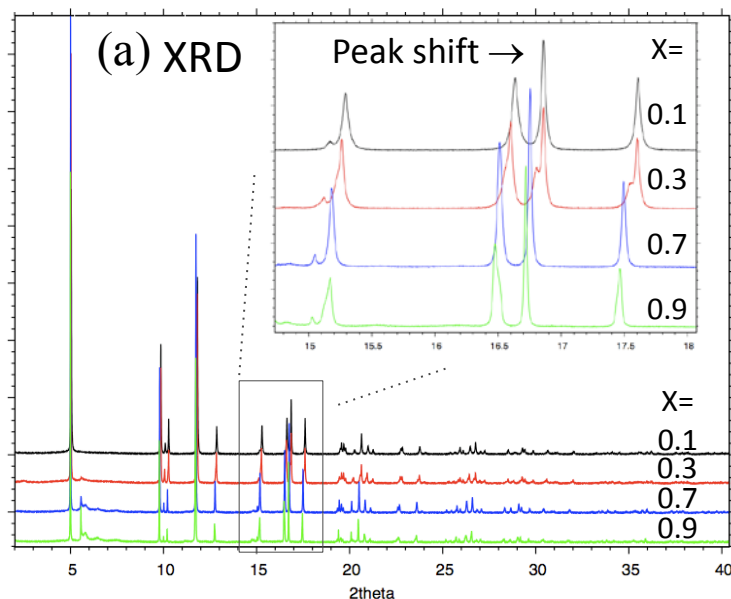


- 5% spinel phase improves:
  - First cycle efficiency (86% to 91%)
  - Cycling stability improved at cycle 10 from 178 mAhg<sup>-1</sup> to 188 mAhg<sup>-1</sup>
- Can cycling be improved by smaller voltage window?
- What is spinel structure in Co/Ni-rich electrodes?  
What is behavior of Mn, Co, and Ni in spinel?
  - Utilize X-ray characterization to understand structure
  - Study spinel components individually

# X-ray Structural Analyses of $x\text{Li}_2\text{MnO}_3 \cdot (1-x)\text{LiCoO}_2$

- High-resolution X-ray diffraction (XRD) and pair-distribution function measurements (PDF) have been initiated at the Advanced Photon Source (APS)

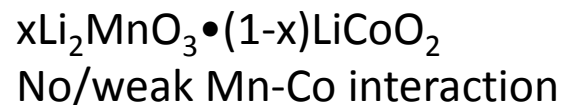
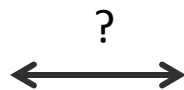
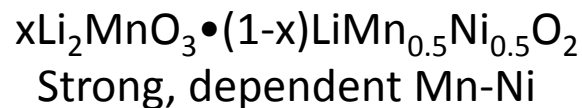
## Recap:



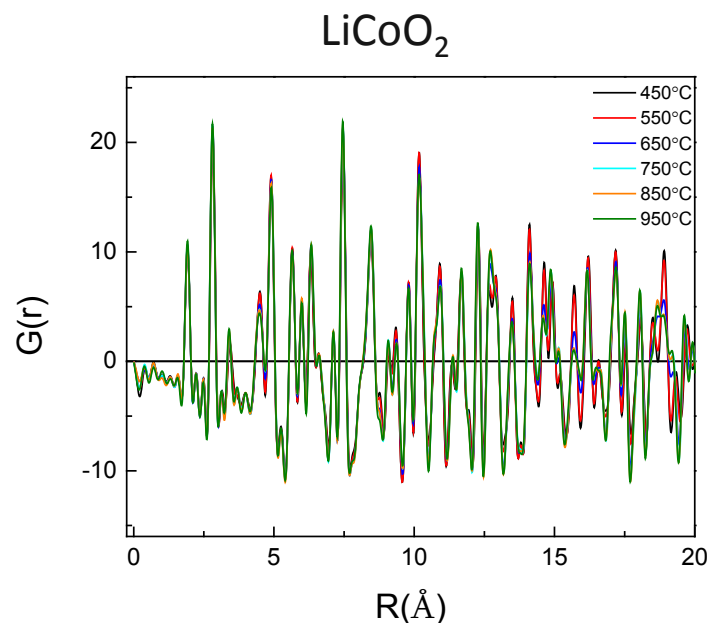
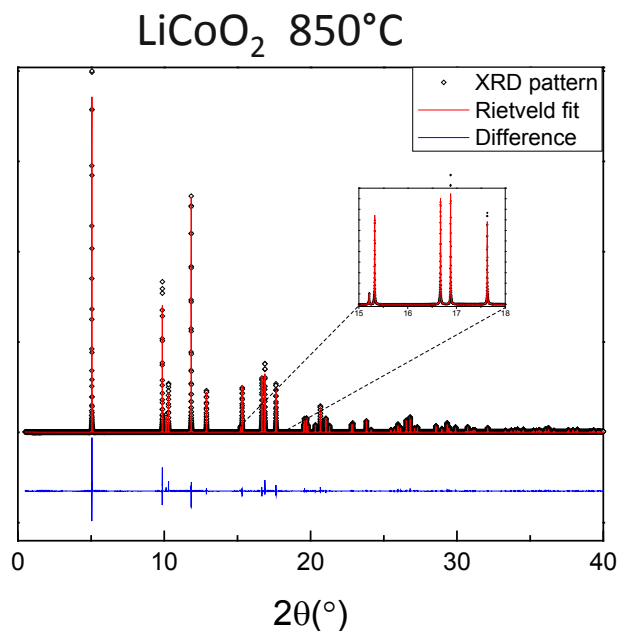
- XRD data suggests complex structure behavior as a function of x; “solid solution” of  $\text{Li}_2\text{MnO}_3$  and  $\text{LiCoO}_2$  domains in a common ccp oxygen array
- PDF data show short range environment of Mn, Co remains constant as a function of x (to  $R \approx 15\text{Å}$ ) (M-M correlations: CN = 3 for Mn in  $\text{Li}_2\text{MnO}_3$  layers; CN = 6 for Co in  $\text{LiCoO}_2$  layers)
- How do end-members behave? What is composition/atomic arrangement?

# X-ray Structural Characterization

Strategy: Stabilize ‘layered-layered’ Mn/Ni system with atoms that have strong layered character → Co



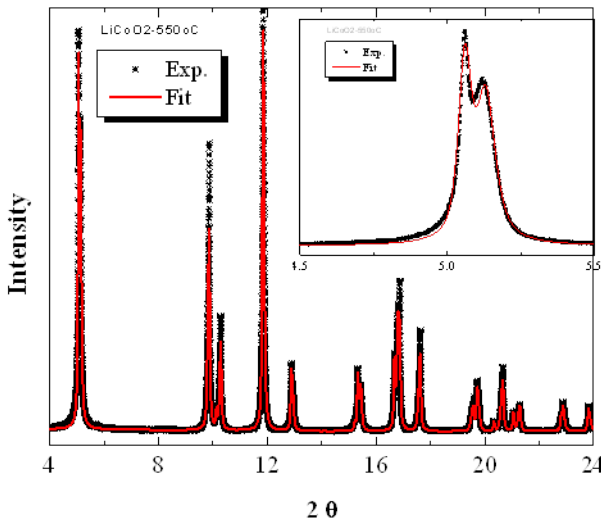
- What happens to domains and individual components of a “domain structure”, i.e.  $x\text{Li}_2\text{MnO}_3 \bullet (1-x)\text{LiCoO}_2$ , with the addition of Ni?
- Utilize high-resolution synchrotron XRD and PDF for structural characterization of end-members



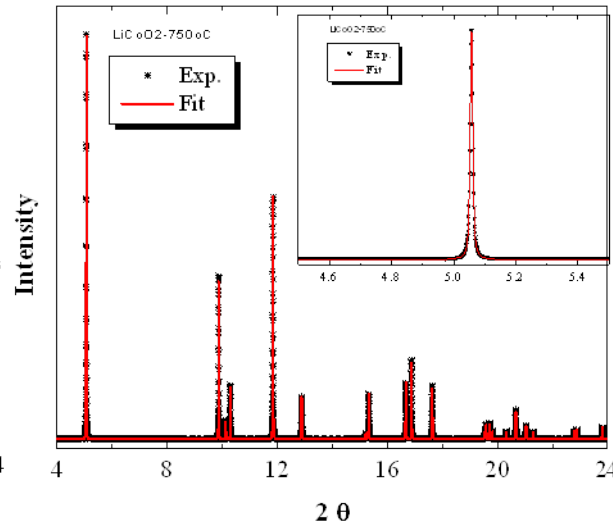
# Baseline XRD of $\text{Li}_2\text{MnO}_3$ and $\text{LiCoO}_2$ End-members

Strategy: Study end-members of 'layered-layered' system as baseline to composite structure

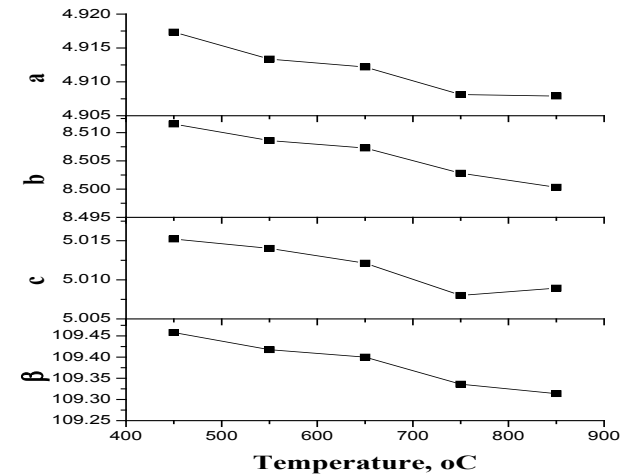
XRD fit of  $\text{LiCoO}_2$  at 550°C



XRD fit of  $\text{LiCoO}_2$  at 750°C



X-ray PDF fit of  $\text{Li}_2\text{MnO}_3$



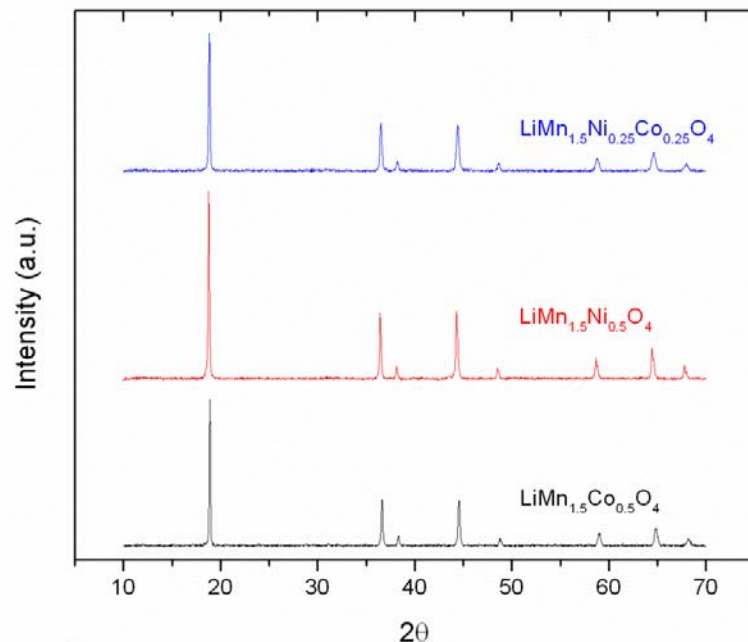
- $\text{LiCoO}_2$  data shows spinel phase decreasing with increasing synthesis temperature
- $\text{Li}_2\text{MnO}_3$  data shows decreasing lattice parameters with increasing synthesis temperatures
- Knowledge gained from structural analysis will be directed towards composite 'layered-layered' material

# Spinel End-members

Strategy: Investigate spinel end-members for structural and electrochemical insight into composite electrodes

- Evaluation of spinel materials for optimal starting composition and insight into Mn, Ni, and Co mixed composites coupled with X-ray structural analysis
- Structures synthesized and electrochemical testing in progress

XRD of materials heated at 750°C for 12 hours



Electrochemical  
testing in progress



# Future work

## ■ Composition optimization

- Continue to screen 'layered-layered-spinel' electrodes and 'layered-layered' & 'spinel' end-members to determine optimized composition and the spinel content to circumvent voltage fade;
- Select two most promising compositions/chemistries for exhaustive electrochemical evaluation and characterization of their chemical, physical and thermal properties;
- Evaluate electrodes in a full lithium-ion cell configuration.

## ■ Collaboration

- Collaborate with other ABR participants, academic and industrial partners to understand and combat voltage fade phenomena and the cause thereof.

For example, continue to explore detailed structure-electrochemical relationships of relatively simple 'layered-layered' & end-member baseline materials (e.g.,  $x\text{Li}_2\text{MnO}_3 \bullet (1-x)\text{LiCoO}_2$ ) and 'layered-layered-spinel' derivatives by XRD, XAS and PDF analyses with collaborators at the Advanced Photon Source.



# Summary

- 'Layered-layered' composite electrode structures, stabilized by a spinel component, hold promise for countering voltage fade
- 'Layered-layered-spinel' electrodes counter the first-cycle irreversible capacity loss of 'layered-layered' electrodes
- Determination of optimal high  $\text{Li}_2\text{MnO}_3$  content to enhance capacity and stability of 'layered-layered' composite
- Increasing Ni content in high Co content electrodes provides stability
- Many questions are still to be answered: Further work is required 1) to obtain a detailed understanding of the reasons for voltage fade phenomena, 2) identify optimized 'layered-layered-spinel' compositions to best counter these effects (a basis for collaborative studies with academia and industry)



# Acknowledgment

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