

# Synthesis and Characterization of Structured Si-Carbon Nanocomposite Anodes and Functional Polymer Binders

Donghai Wang<sup>1</sup> and Michael Hickner<sup>2</sup>

<sup>1</sup>Department of Mechanical and Nuclear Engineering

<sup>2</sup>Department of Materials Science and Engineering

PENNSSTATE



The Pennsylvania State University

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Project# **ES147**

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# OVERVIEW

## Timeline

***Project Starts:*** Jan. 2011

***Project Ends:*** Oct. 2014

***Percent Completed:*** 20%

## Barriers Addressed

- Power and energy density
- Cycle and calendar life
- Safety

## Budget

- Total funding: \$800K
- FY 2011: \$200K
- FY 2012: \$200K

## Partners

- Ji-Guang Zhang and Jun Liu (PNNL)
- PA Nanomaterials Commercialization Center

# OBJECTIVES

Design **Si-based nanocomposite anodes** with high volume change tolerance, fast kinetics, and low irreversible capacity loss.

- Design new structured Si-carbon nanocomposites to achieve high capacity and good cycling stability.
- Develop new functional polymer binders for Si-based anodes to improve cycling stability.

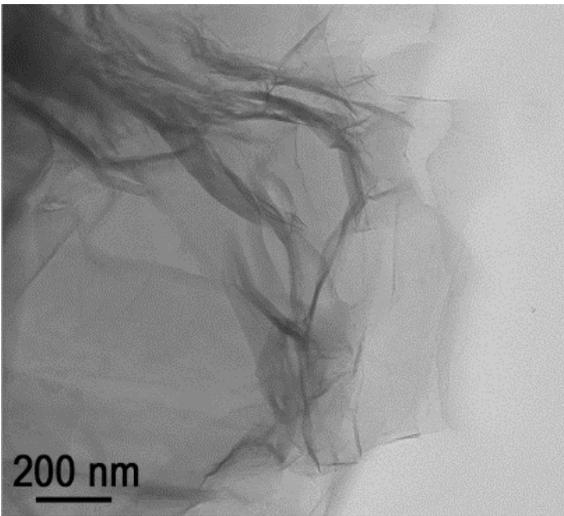
# MILESTONES

- Prepared Si nanoparticles with controlled size and synthesized Si-carbon composites, including Si-graphene and Si@hollow carbon.
- Synthesized new polymer binders for Si-based anodes.
- Characterized electrochemical properties of Si, Si-carbon nanocomposites, and polymer binders in lithium cells up to 200 cycles.
- Achieved improved capacity retention with commercial Si-based anodes.
- Achieved 40% first cycle capacity and 90% coulombic efficiency thereafter.

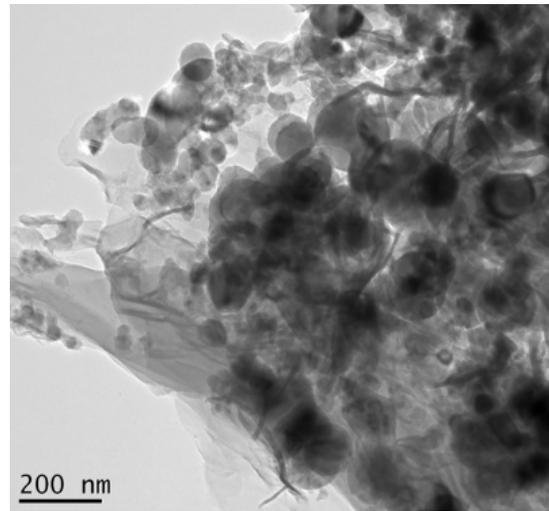
# APPROACH

- **Develop Si-graphene composites** with good electrical contact and electrode conductivity.
- **Develop Si@hollow carbon nanocomposite** to tolerate volume change of Si and improve cycling stability of Si-based anodes.
- **Develop new polymer binders** with controlled elastic properties, ion-conductive moieties, and Si surface binding functionality to stabilize and bridge Si particles.

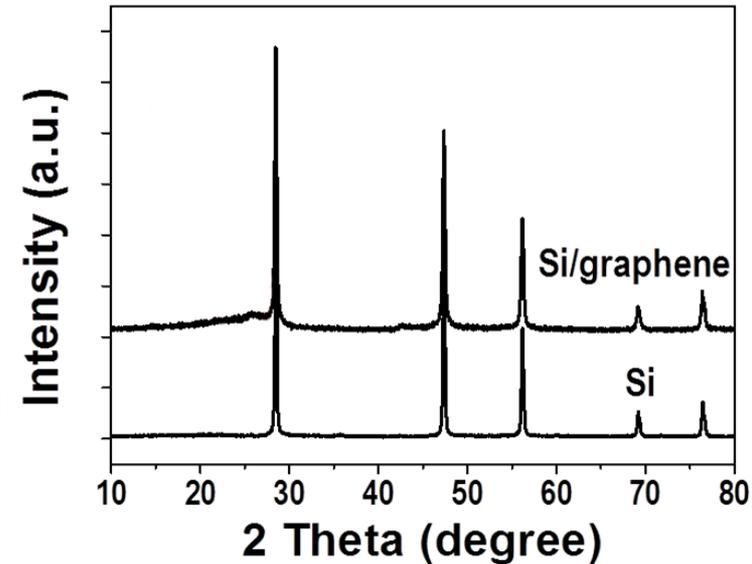
# SI-GRAPHENE NANOCOMPOSITES



Graphene Sheets



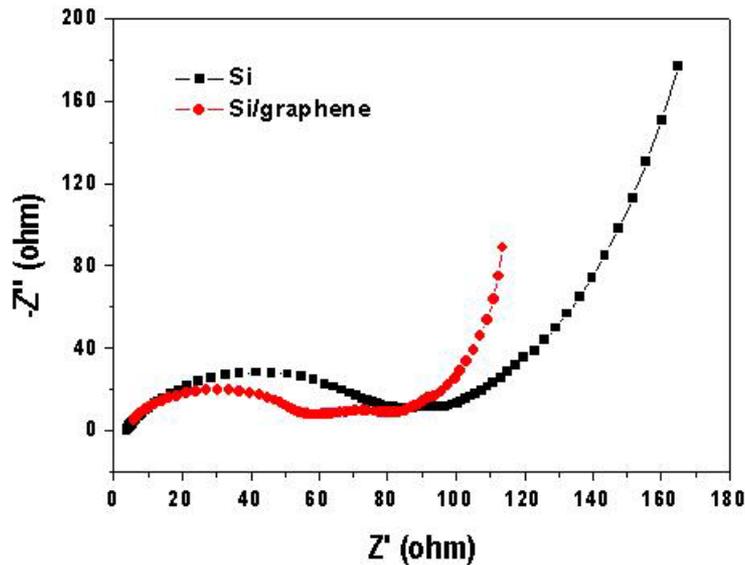
Si-Graphene



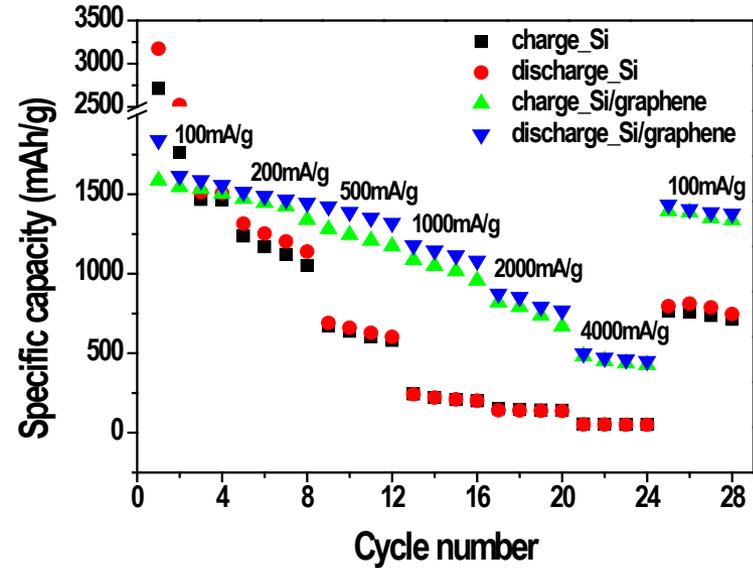
XRD of Si and Si-  
Graphene

- Si particles (<200nm) synthesized by magnesiothermic reduction.
- Si particles coated onto graphene sheets.

# GRAPHENE IMPROVES CONDUCTIVITY AND CYCLING STABILITY



Impedance Analysis of Electrodes



Rate Performance Comparison

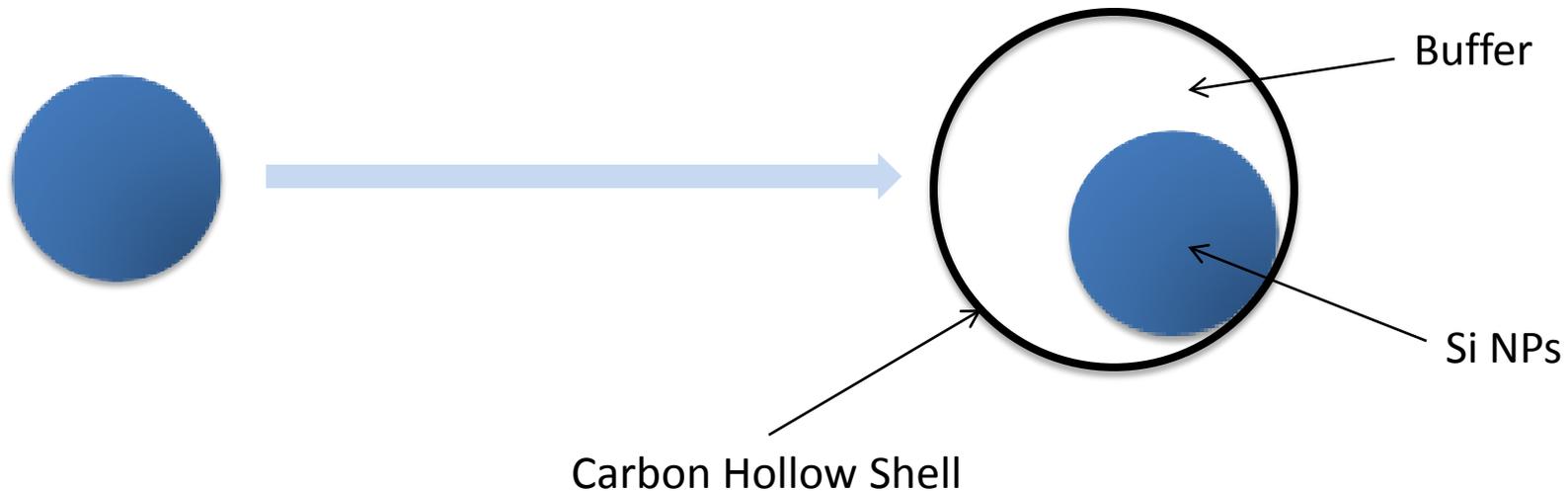
- The Si-graphene nanocomposite shows increased electrode conductivity.
- Incorporating graphene improves the cycling stability of Si anodes, possibly due to the good electric contact between Si and graphene sheets.

# METHODS TO FURTHER IMPROVE STABILITY OF SI-CARBON COMPOSITES

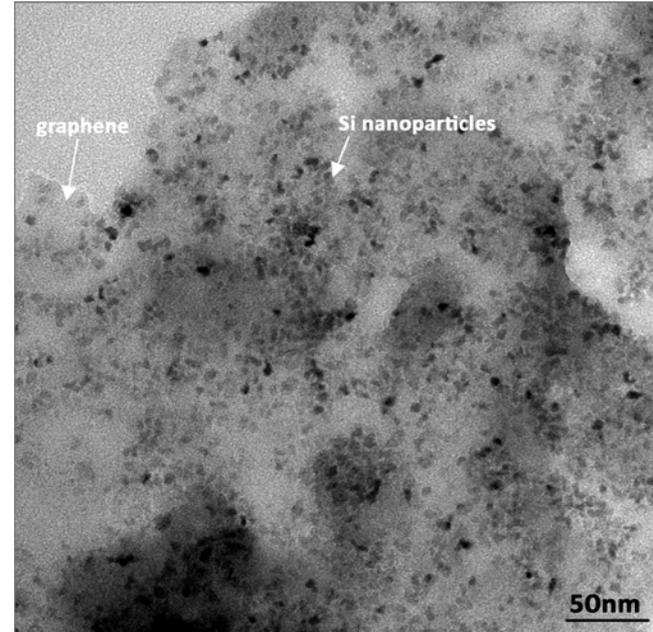
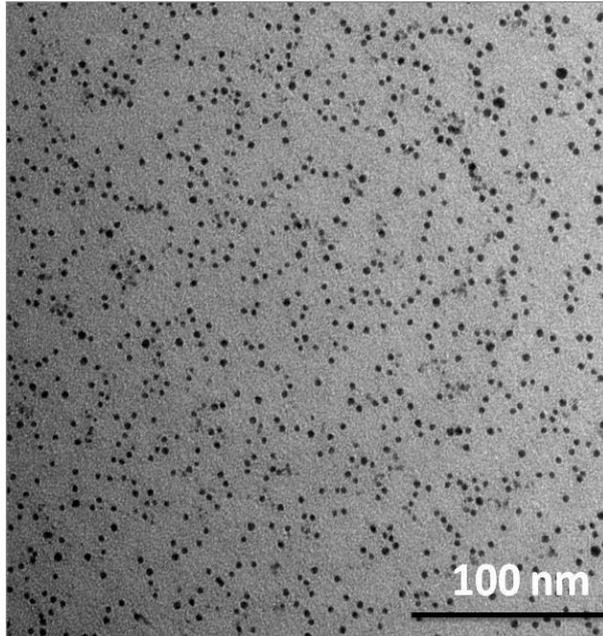
1. Decrease Si nanoparticle size



2. Provide built-in buffer volume for Si expansion

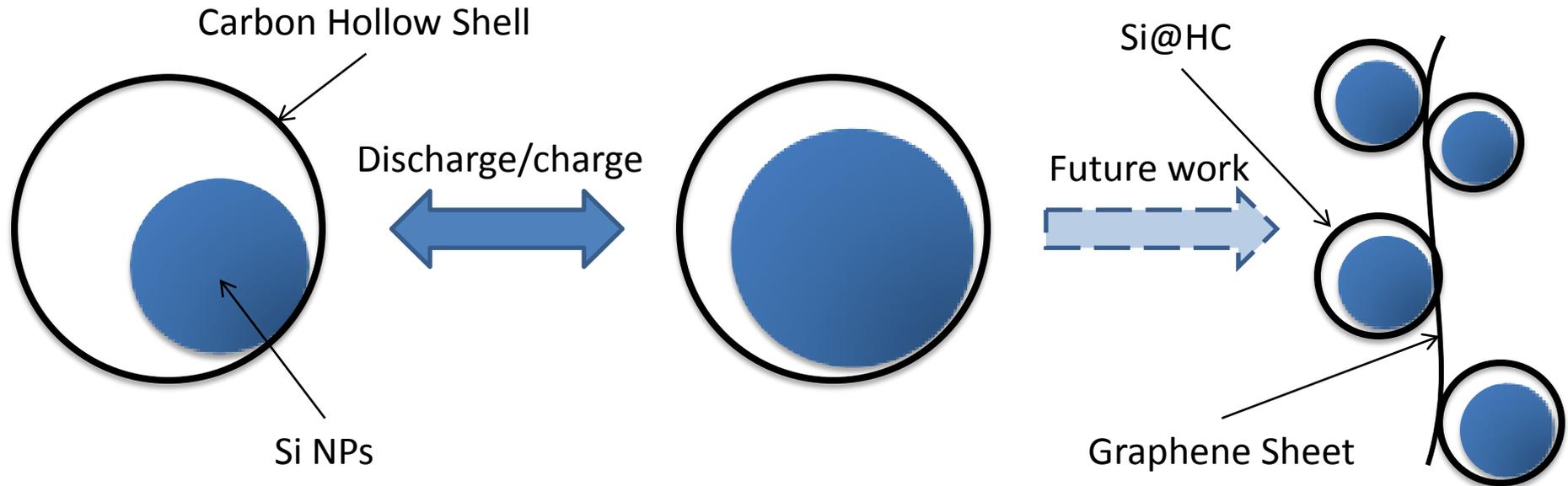


# SI NANOPARTICLES AND SI-GRAPHENE HYBRID



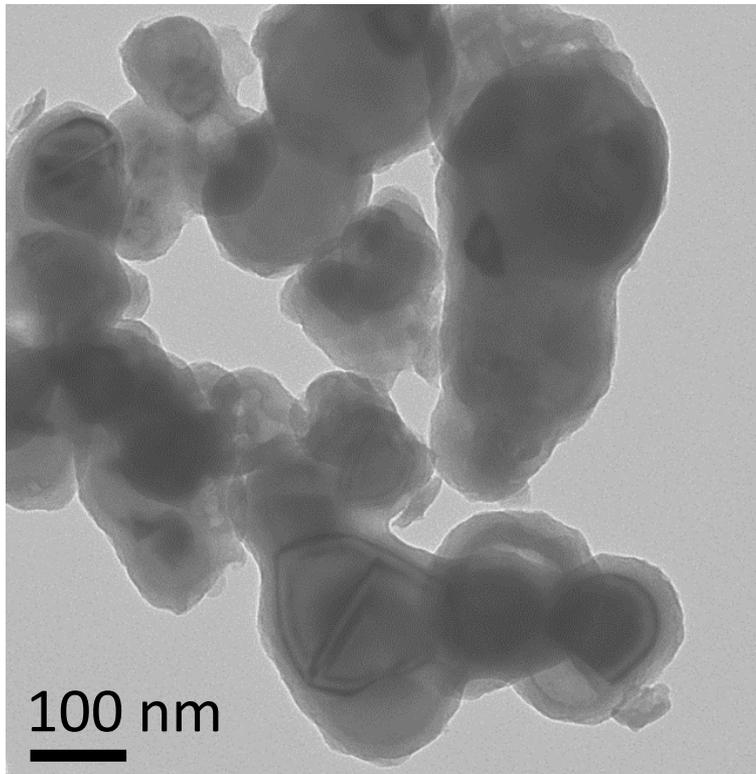
- Synthesized Si nanoparticles with diameter of 5-10 nm.
- Synthesized Si-graphene nanocomposites with well-dispersed Si nanoparticles, providing Si with buffer space for superior volume change tolerance.
- Performance testing in battery cells underway.

# Si@HOLLOW CARBON WITH CONTROLLED BUFFER VOLUME

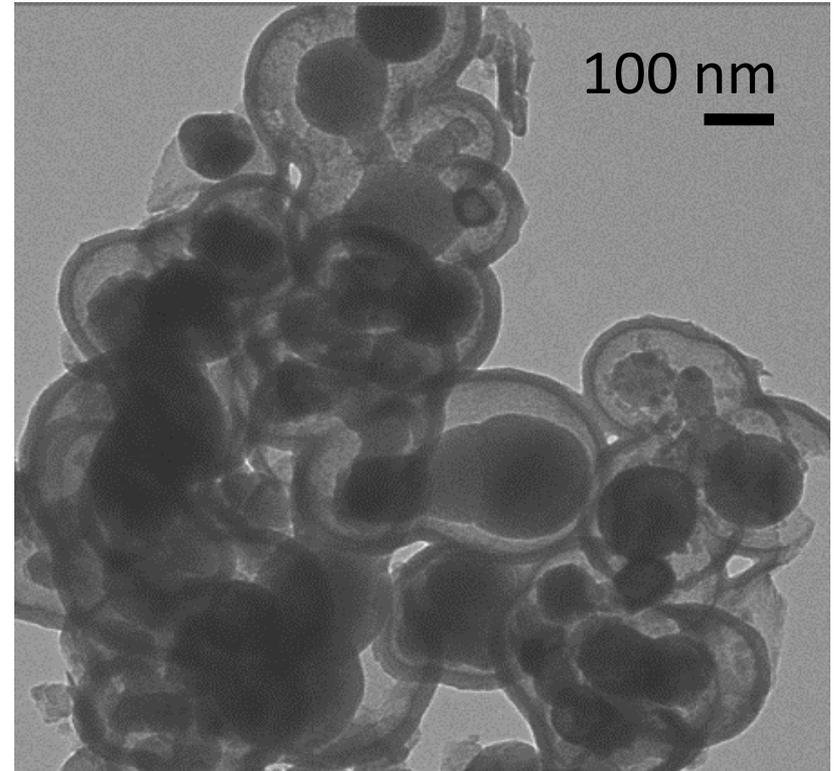


- Synthesized Si@hollow carbon nanocomposites – commercial Si nanoparticles inside hollow carbon spheres.
- Structure allows Si to expand/contract freely inside carbon support structure.

# Si@CARBON AND Si@HOLLOW CARBON NANOCOMPOSITES



**Si@carbon nanocomposite**



**Si@hollow carbon nanocomposite**

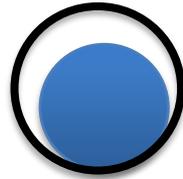
- Used commercial Si nanoparticles.
- Synthesized Si@C and Si@HC nanocomposites.

# Si@HOLLOW CARBON NANOCOMPOSITES WITH CONTROLLED BUFFER VOLUME

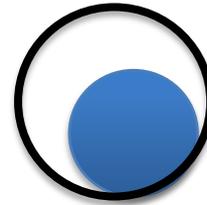
Si@C



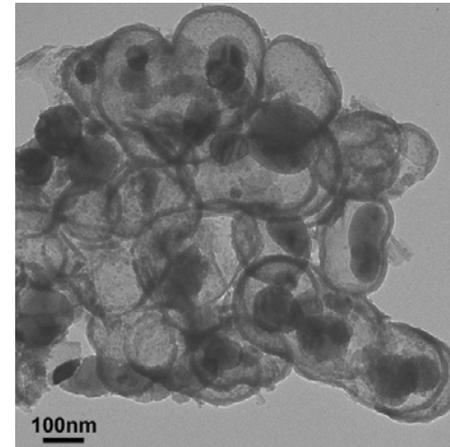
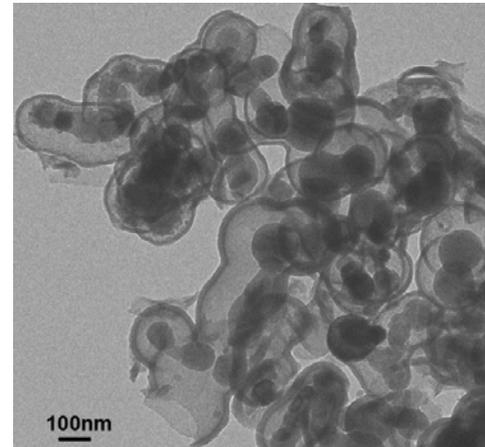
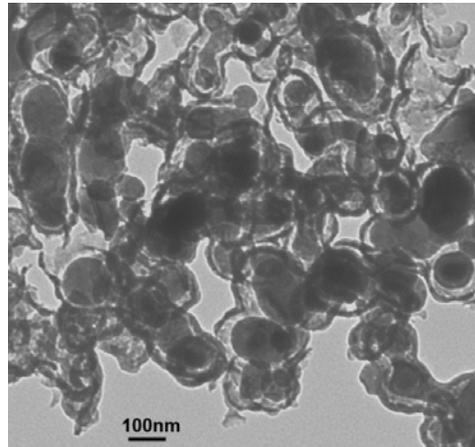
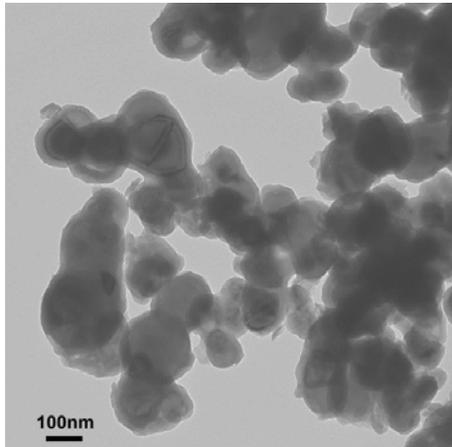
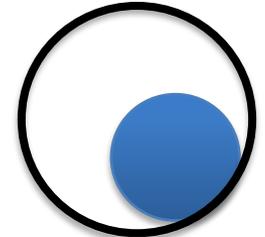
Si@HC-1.5



Si@HC-3



Si@HC-6

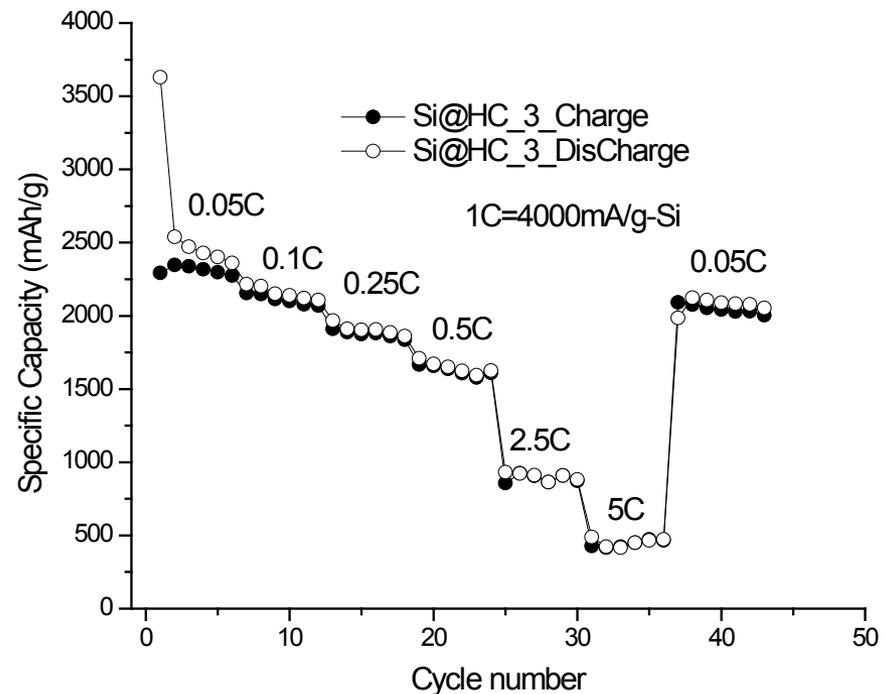
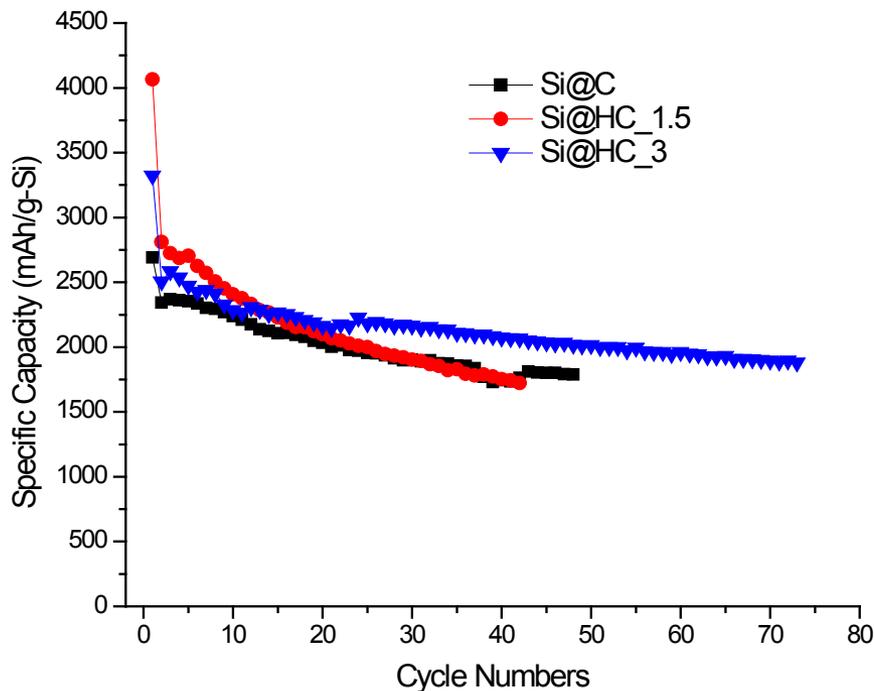


Buffer to Si Volume Ratio = 0, 1.5, 3 and 6.

Increasing buffer volume 

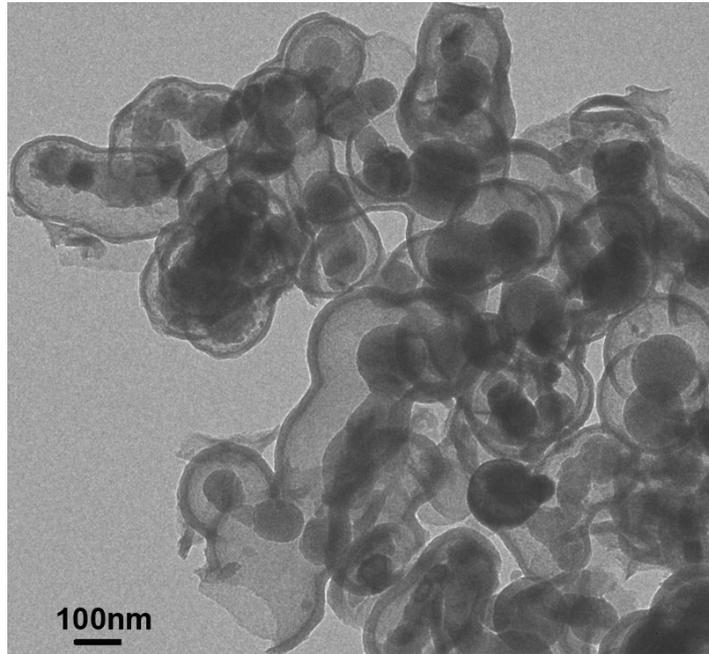
- Si@carbon and Si@hollow carbon nanocomposites with controlled buffer volume were developed.
- Si content is currently 48-62 wt %.

# CYCLING PERFORMANCE OF Si@C AND Si@HC NANOCOMPOSITES

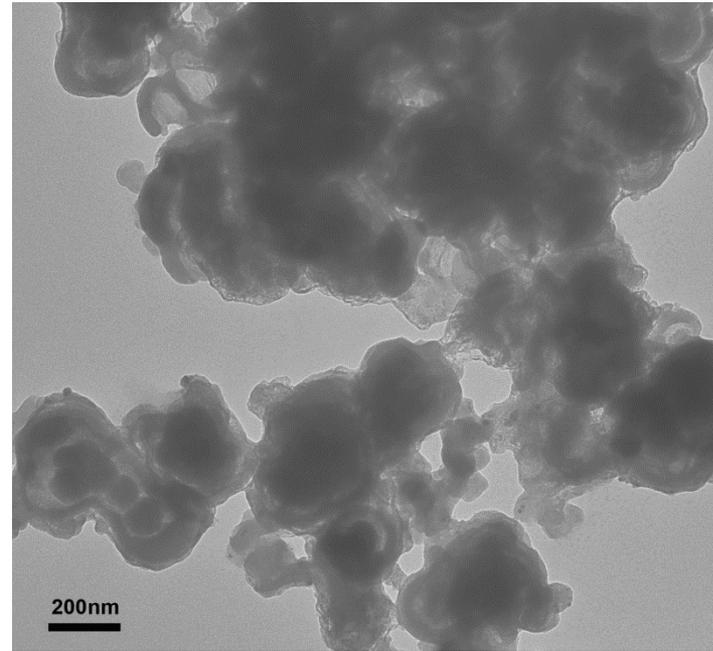


- Si@HC with buffer/Si volume ratio of 3 has the best capacity retention among the Si@C and Si@HC nanocomposites.
- 1<sup>st</sup> cycle capacity: 2250 mAh/g of Si at 400 mA/g.
- Capacity above 1000 mAh/g of Si@HC after 50 cycles.

# SI NANOPARTICLES REMAIN IN HOLLOW CARBON STRUCTURES



Fresh Si@HC

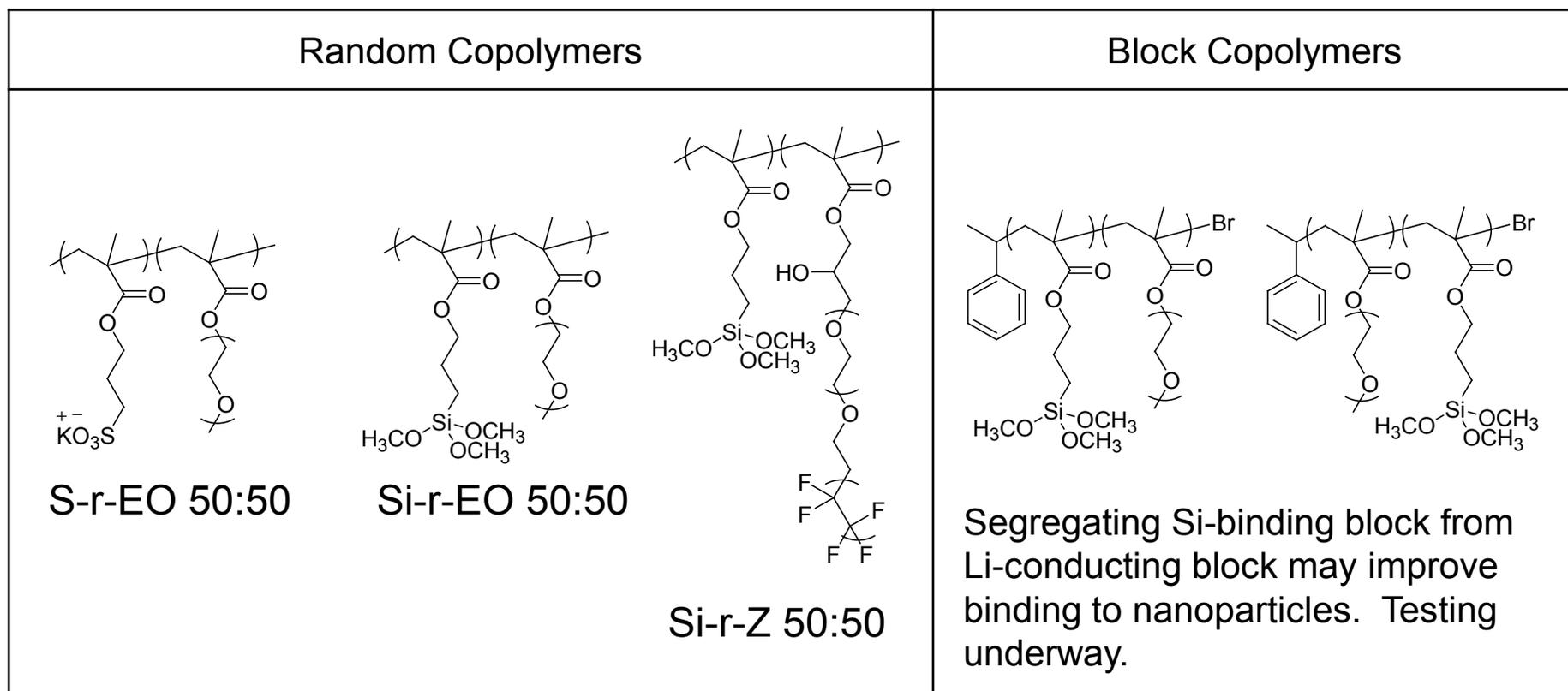


Si@HC after 20 cycles of discharge/charge

- TEM investigation shows Si nanoparticles confined within the hollow carbon shells after repeated discharge/charge.
- SEI coatings were observed on Si@HC particles.

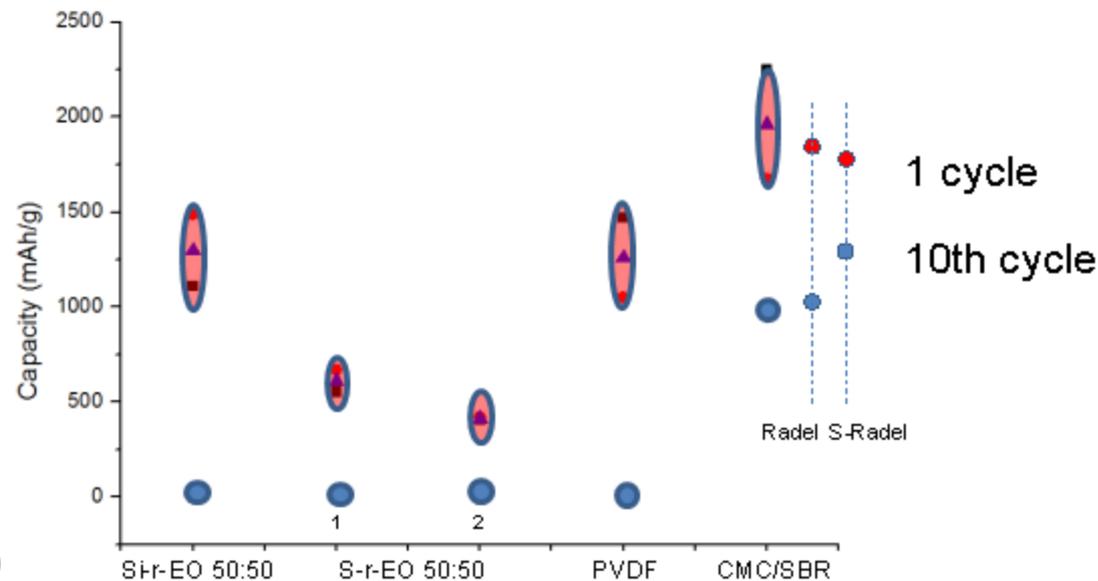
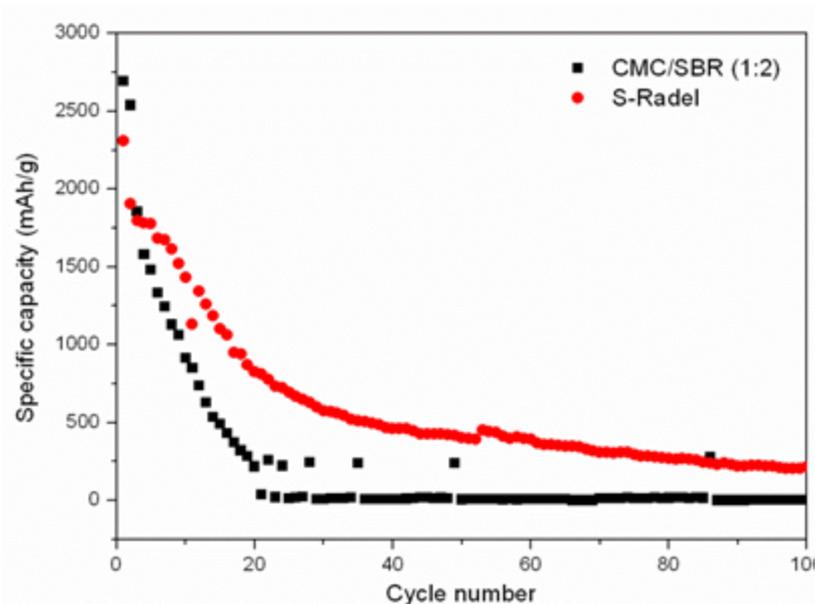
# LI-CONDUCTING BLOCK COPOLYMER BINDERS

Random and block copolymers were designed to have Li-conducting EO segments and Si NP surface binding groups. Modular synthesis was used to substitute various functional groups easily without drastically modifying synthetic methodology.



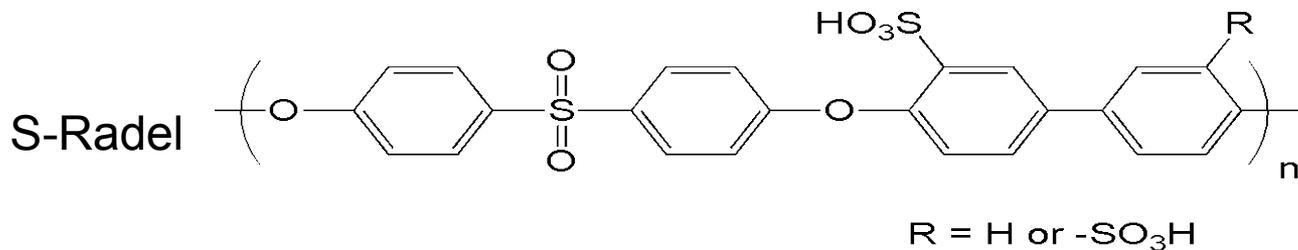
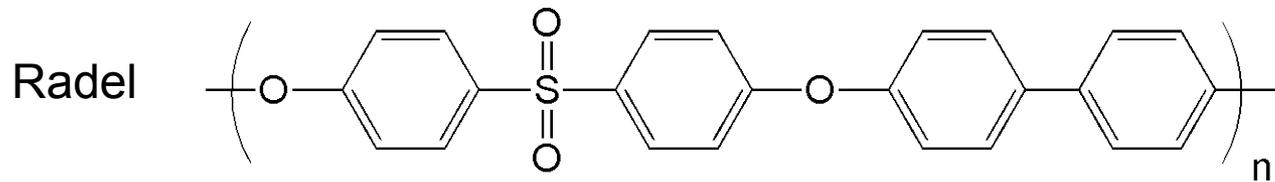
# INITIAL BINDER CYCLING TESTS

- Symbols show 1<sup>st</sup> cycle and 10<sup>th</sup> cycle capacity of Si NP-based anodes with various binders.
- EO random polymers had poor cycling performance. Investigating blocks now.
- Radel<sup>®</sup> samples have similar initial capacity retention to CMC/SBR.



# STIFF-BACKBONED POLYMER BINDERS

- Radel poly(sulfone)-based polymers not as prone to swelling in liquid Li-ion battery electrolytes compared to EO samples.
- Investigate how mechanical properties influence Si-binder anode performance.



## Moduli

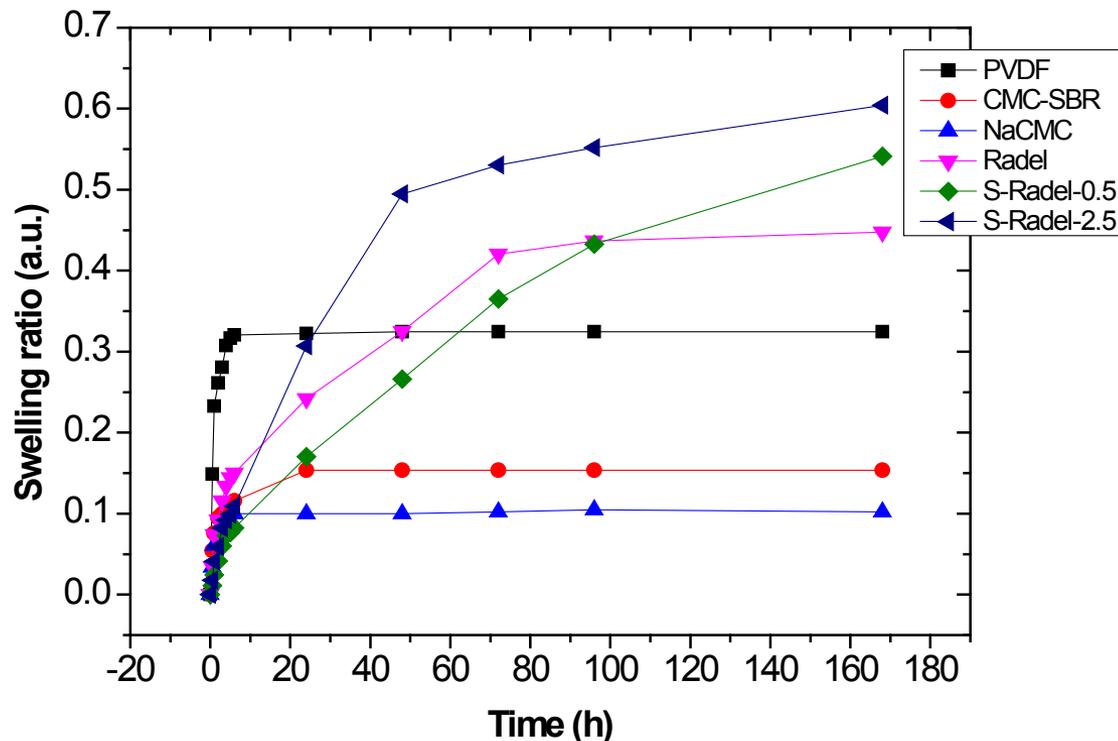
Radel: 2 GPa

PVDF: 200 MPa

PEO: 20 MPa

# POLYMER BINDER SWELLING STUDY

Swelling experiments performed in electrolyte mixtures to determine rate and extent of binder dissolution



*Key information* relating the swelling profile to the cell performance. Next iteration of polymers will incorporate crosslinkable groups to limit the swelling in electrolyte.

# FUTURE WORK

- Study of SEI formation on Si-graphene and Si@HC nanocomposites.
- Synthesis, characterization, and testing of amorphous Si-carbon nanocomposite.
- Continued investigation of binder failure mechanisms.
- Design of low-swelling binders.

# SUMMARY

- Synthesized Si-graphene, Si NP-graphene, Si@C, and Si@HC nanocomposites.
- Characterized and tested the electrochemical performance of our new composites. Found performance improvements over conventional Si-based materials.
- Developed new binders for Si-based anodes.
- Tested several promising binders and elucidated potential failure mechanisms to direct further work.

## **Acknowledgment**

Support for this work from DOE-EERE, Office of Vehicle Technologies is gratefully acknowledged – Tien Duong and David Howell