### Innovative Manufacturing and Materials for Low-Cost Lithium-Ion Batteries

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## **Overview**

#### Timeline

- Start date October 2011
- End date September 2014
- Percent complete 30%

#### Budget

- Total Funding: \$2,999,127
  - DOE Share \$2,249,127
  - Contractor Share \$750,000
- Funding Received:
  - FY12: \$382,875 (10/1-9/30)
  - FY13: \$190,820 (10/1-current)

#### **Barriers/Targets**

- Cost reduced to \$300/kWh by 2014
- Energy density to 300 Wh/kg by 2015

#### **Partners**

#### Interactions/Collaborations

- Madico, Inc. Electrode Stack Mfg.
- Dow Kokam, LLC Battery Mfg. and Testing
- University of Rhode Island Electrolyte
- Ashland -Solvents & Polymers
- **Project Lead** *Optodot Corporation*

Thank you to the DOE Vehicle Technologies Program for their support and funding of this project!

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# **Project Objectives**

#### **Project Long-term Objective**

 Reduce the cost, weight, and volume of the cell's inactive components by at least 20%, and preferably by at least 40%, while maintaining cell performance

#### **Project Immediate Objectives (Mar-12 to Mar-13)**

 Complete initial prototype with much thinner separator and current collector layers



# **Project Milestones for FY12**

Month/Year	Milestone	Status
Jan-12	Initial ceramic separator design	Complete
Jan-12	Initial anode coated stack design	Complete
April-12	Initial cathode coated stack design	Complete
April-12	Initial low cost electrolyte design for lithium salt	Complete
June-12	Initial current collection and termination design	Complete
June-12	Initial low cost electrolyte design for solvents complete	Complete



# **Project Milestones for FY13**

Month/Year	Milestone	Status
Dec-12	Initial coated stack cell development and testing	To be completed May-13
Dec-12	Initial electrolyte for energy cell design	To be completed May-13
Mar-13	Anode coated stack optimization and scale up for energy cells	To be completed June-13
June-13	Optimized low cost and/or safer electrolyte for energy cells	On schedule
June-13	Cathode coated stack optimization and scale up for energy cells	On schedule
Sep-13	Initial anode coated stack design for power cells	On schedule
Sep-13	Current collection and termination design optimization and scale up	On schedule

## Approach: Battery Stack Manufacturing Process

#### **COATING**



- Utilizes a roll-to-roll process
  - Lower cost
  - Higher efficiencies
  - Wider Widths
- Electrode coatings can be in lanes to provide electrodefree current collector zones for termination
- Release substrate is removable, enabling interleaving of anode and cathode coated stacks



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## **Approach: Battery Stack Manufacturing Process**



- Lower cost and lighter inactive components
  - Nanopore nature and compressive strength of ceramic separator enables overcoating with anode and cathode layers followed by calendering
  - Thinner separator reduces inactive components and cell level cost significantly
  - Thinner & lighter current collector layers



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 Demonstrated first example of an anode coating onto a separator layer followed by calendering and showed good cycling results for this coated anode stack in half cells



NOTE: Release substrate was removed prior to cross-sectional SEM

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 Demonstrated first example of a cathode coating onto a separator layer and showed good cycling results for this coated cathode stack in half cells



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• Demonstrated first example of depositing copper and aluminum current collector layers onto coated anode and cathode stacks, respectively, and successfully cycled these stacks in coin cells



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Cycling of initial prototype separator/electrode stack full coin cells with 2 um Cu and 6 um Al depositions on the anode and cathode respectively

*NOTE:* Electrodes were not calendered and not vacuum dried in these prototypes

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- Developed new versions of 8-10 micron thick ceramic separator layers with an all-nanoporous design and very narrow pore size distributions ranging from 25-75 nm average pore diameter.
- Using the Argonne National Lab battery cost model, preliminary cost savings from the 8 micron ceramic separator component including the reduced usage of electrolyte were estimated to be greater than 20% cost savings for the inactive components of the cell. This estimated savings meets the cost reduction objectives of this project from the thinner ceramic separator alone.



### **Collaborations/Subcontractors**

- Madico (industry) on manufacturing processes of mixing & coating of the ceramic separator
- **Dow Kokam** (industry) on electrode coatings
- URI (academic) on cell cycling testing on various ceramic separator and coated stack designs & on new electrolytes
- Ashland (industry) on polymer selection for battery coatings

# **Future Work**

### **Remainder of FY13**

- Optimization and scale-up of anode stack, cathode stack, and current collector/termination
- Develop initial designs for anode stack and electrolyte for high rate/power cells

### FY14

 Optimization and scale-up of coated stack design with new electrolyte for both energy and power cells

### **FY14 Final Deliverables**

- Cost assessment of Li-ion cell manufactured using the current vs. improved designs
- Deliver 18 cells of baseline design and of new coated stack design with cell test plan and report on performance and abuse tolerance of these cells



# **Summary**

- Meeting the at least 20% improved cost, volume and weight, as well as the performance requirements, for the key inactive components of Li-ion cells and developing a low cost next generation manufacturing process will help meet the DOE goals of cost reduction to \$270/kWh by 2017 for PHEVs and to \$150/kWh by 2020 for EVs.
- Demonstrated the first examples of electrode/separator coated stacks by leveraging off of the all-nanoporous ceramic separator, the key enabling technology of this project
- The use of the 8 micron thick ceramic separator alone meets the project's > 20% cost improvement objective for the inactive components
- Our 4 partners and subcontractors, Madico, Dow Kokam, URI, and Ashland, are providing coating and converting expertise and equipment, battery assembly and testing capability, electrolyte expertise, and polymer and solvent expertise.

