

# Fuel Effects on Emissions Control Technologies

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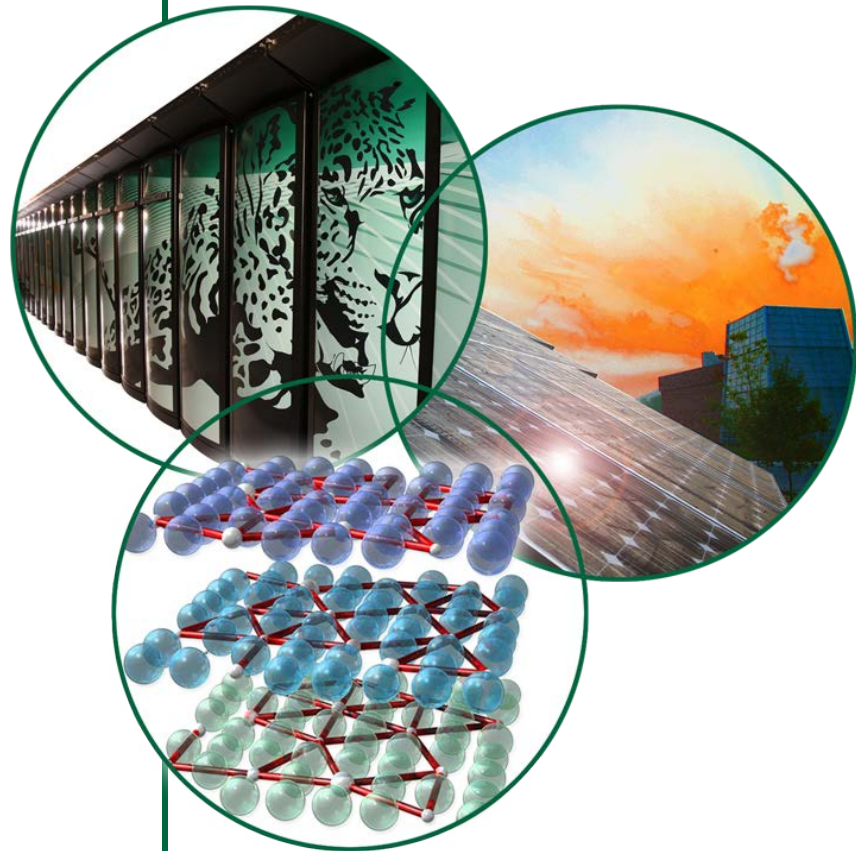
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## 2012 DOE Annual Merit Review

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# Project Overview

## Timeline

- Project is ongoing but re-focused each year to address current DOE and industry needs.
  - FY10 start: Lean-Ethanol LNC
  - FY09 start: Biodiesel-based Na
  - FY08 start: EGR cooler fouling
  - FY07 start: PM kinetics + chemistry

## Budget

- Funding received in
  - FY11: \$725K
  - FY12: \$1,065K (expected)

## Barriers

- Inadequate data and predictive tools for fuel effects on emissions and emission control system impacts. (2.4 D)
- Inadequate data on long-term impact of fuel and lubricants on engines and emissions control systems. (2.4 E)

## Partners

- Collaborators and their roles
  - CLEERS: Evaluation protocols
  - Cummins/Ford/GM/Modine: experimental guidance
  - NREL/Ford/MECA: Biodiesel-aged emissions control devices
  - Research Personnel
    - Penn State University
    - University of Tennessee

# Objectives and Relevance



**Objective:** Provide data in support of predictive tools that can be used to understand fuel-property impacts on combustion and emissions control systems.

**Relevance:** Addresses Fuels Technology barriers D and E: Inadequate data and predictive tools for fuel effects on emissions and emission control system impacts and inadequate data on long-term impacts of fuel and lubricants on engines and emission control systems.

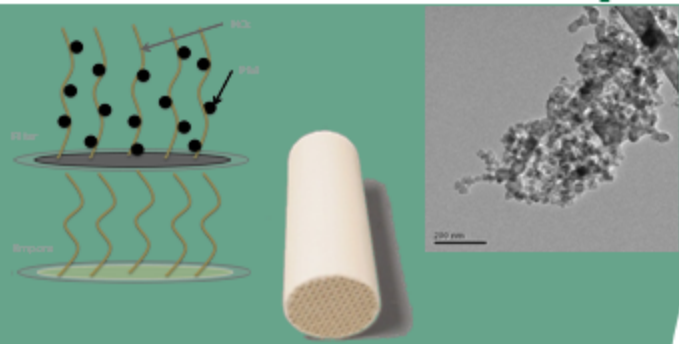
**Changes in fuel properties, whether caused by emerging fuel sources or increasing blend levels, can have unintended and undesirable impacts on increasingly complex vehicles.**

# Milestones for FY 2012

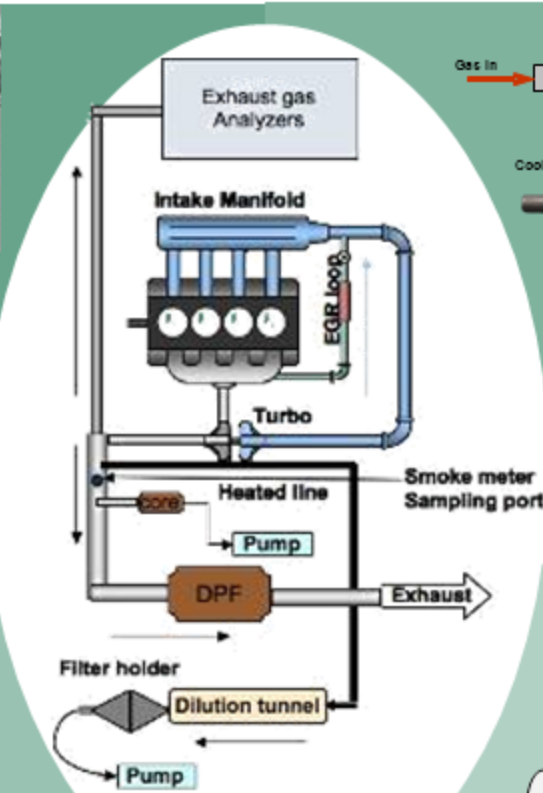
- **Establish EGR cooler performance and degradation metrics. (\$300K)**
  - Investigate low-temperature deposit formation characteristics.
  - Determine whether velocity-induced shear forces can remove deposits.
- **Determine key reactions for NO<sub>x</sub> reduction over silver-alumina catalyst. (\$200K)**
- **Determine sodium, potassium, calcium impact on DOC-DPF-SCR system. (\$300K)**
- **Investigate soot reactivity relationships with engine operational parameters. (\$75K)**
- **Assess Properties, Emissions, and Compatibility of Emerging Fuels. (\$190K)**
  - Implement a pyrolysis-GC/MS capability to enable improved study of fuel and lubricant impacts on soot characteristics and catalyst poisoning.

# Approach

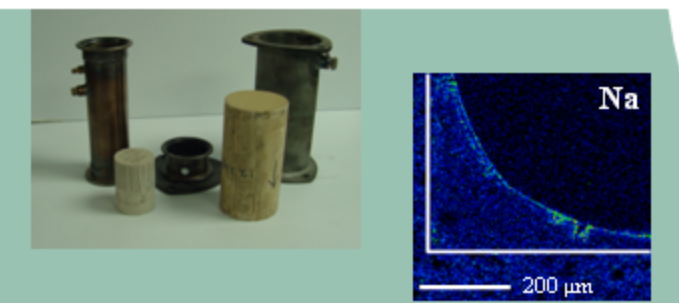
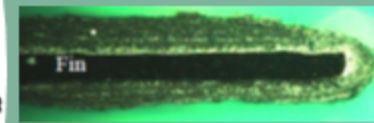
Bring together targeted, engine-based, micro-reactor, and bench reactor studies with in-depth characterization of PM, HCs, and emissions control devices to better understand fuel interactions with emissions control components.



Study fuel- and engine-specific PM to support reduced fuel penalty for emissions control.

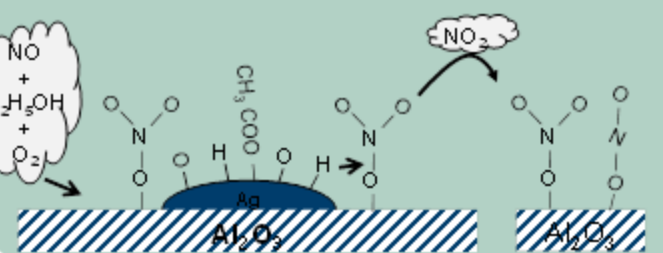


Enable examination of EGR cooler deposits to support models and component development.



Identify specific Na-related impact for each emissions control device.

Identify mechanisms to utilize advantageous fuel characteristics.



New characterization techniques and engagement with industry are key enablers.

# Summary of Technical Accomplishments

- Confirmed that HC-significant deposits that may lead to plugging can result from low-T, low-Re operation, particularly for low-volatility fuels.
- Identified pathway for advantageous use of ethanol in gasoline to enable non-precious metal lean  $\text{NO}_x$  aftertreatment for high-efficiency lean-burn engines.
- Determined Na affects SCR oxidation sites; launched expanded collaborative effort with Ford, NREL, MECA, and NBB to rapidly assess full-size systems.
- Determined that soot oxidation kinetics depend on fuel properties but also on engine operating conditions.
- Identified fuel-range HCs as the dominant species responsible for de-activation of SCR catalysts through direct analysis of the monolith by pyrolysis GC/MS.

### **Does use of biodiesel blends and other non-traditional fuel formulations worsen cooler fouling compared with ULSD?**

**Benefit:** Mitigating concerns and potential issues with EGR cooler fouling associated with expanded biodiesel use can enable broader utilization to displace petroleum consumption.

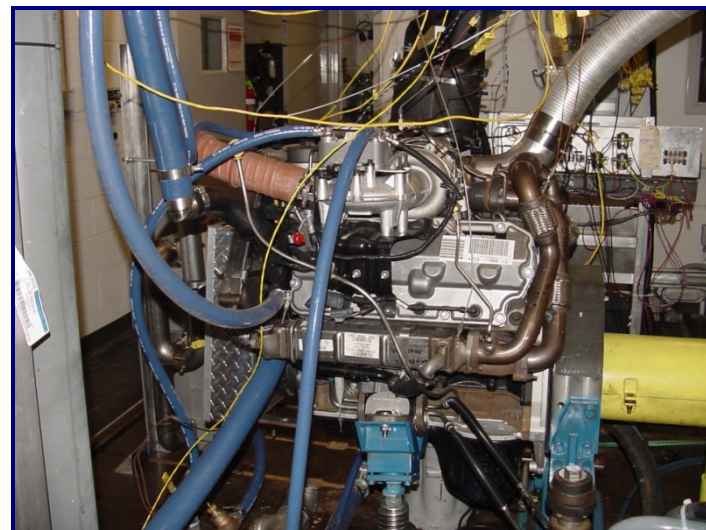
**Accomplishments:** Demonstrated that HC-significant layers can build up extensively without effectiveness loss, potentially leading to plugging.

- Showed that extended operation at low-Re, low-T conditions can build significant deposit layers high in fuel-HCs with very little effectiveness loss.
- Found that removal by flow-induced shear is suppressed in HC-rich deposits.
- Determined that spall material is not re-entrained diesel particulate, but rather has different characteristics than typical diesel particles.



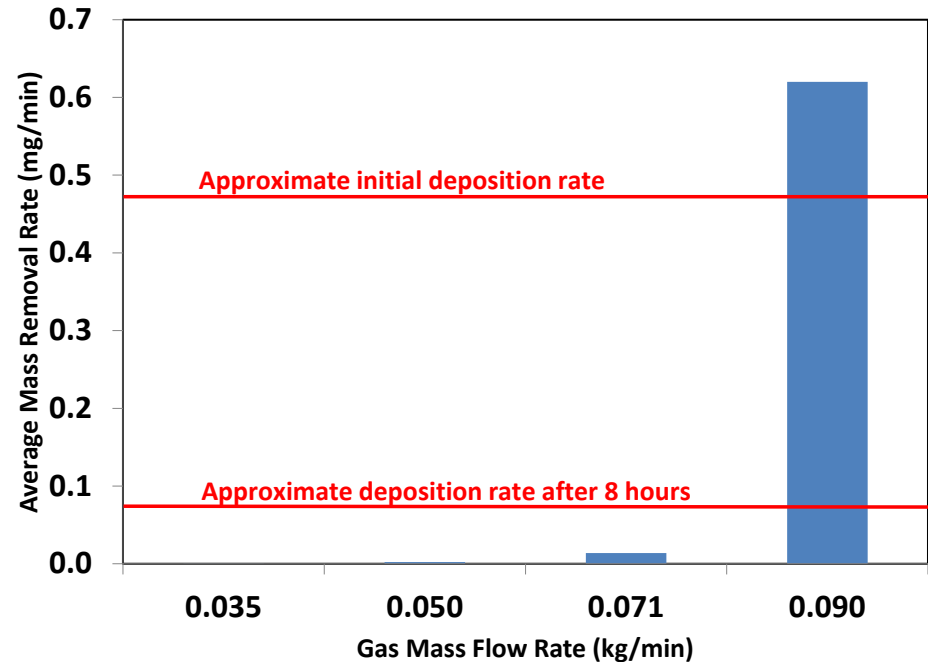
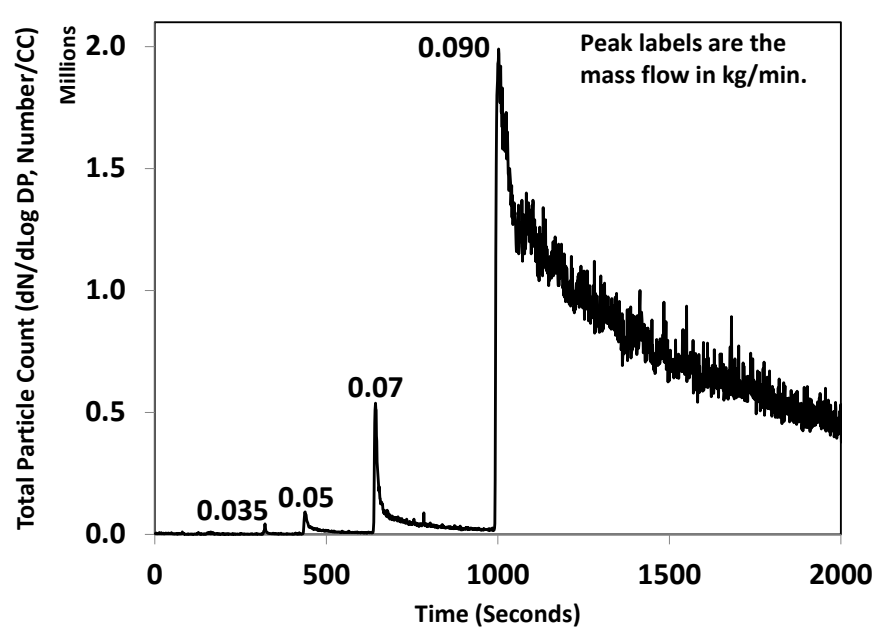
# Surrogate EGR cooler tubes are employed to enable multiple analyses of deposits.

- Ford 6.4-L V-8 used as exhaust generator.
- Surrogate tubes provide more accessible samples for study than full-size coolers.
- Exhaust passed through surrogate EGR cooler tubes at constant flow rate and coolant temperature.
  - Tubes were  $\frac{1}{4}$  inch square cross-section stainless tubes.
  - Thermal effectiveness of tubes is assessed during exposure.
- Subsequent analyses of tube deposits:
  - Total mass of deposits
  - Volatile / non-volatile deposit mass
  - GC/MS characterization of the deposit HCs
  - Deposit layer thermal and physical properties





**One key question in cooler fouling is whether deposits are removed by shear forces; results show this is possible but may not be practical.**



## Spalling of Particles

- Significant beginning at 0.071 kg/min/tube or 45 m/s gas velocity.
- Insignificant at flows comparable to or lower than the deposition flow rate. (<15 m/s)
- Removal velocity causes high pressure drop (~1 PSIG); likely not practical.
- Spalls are not simply re-entrained diesel particulate (see supplemental slides).

Practical removal of deposits likely requires processes or geometries that impose an uplift force on the deposit to augment flow-induced shear forces.

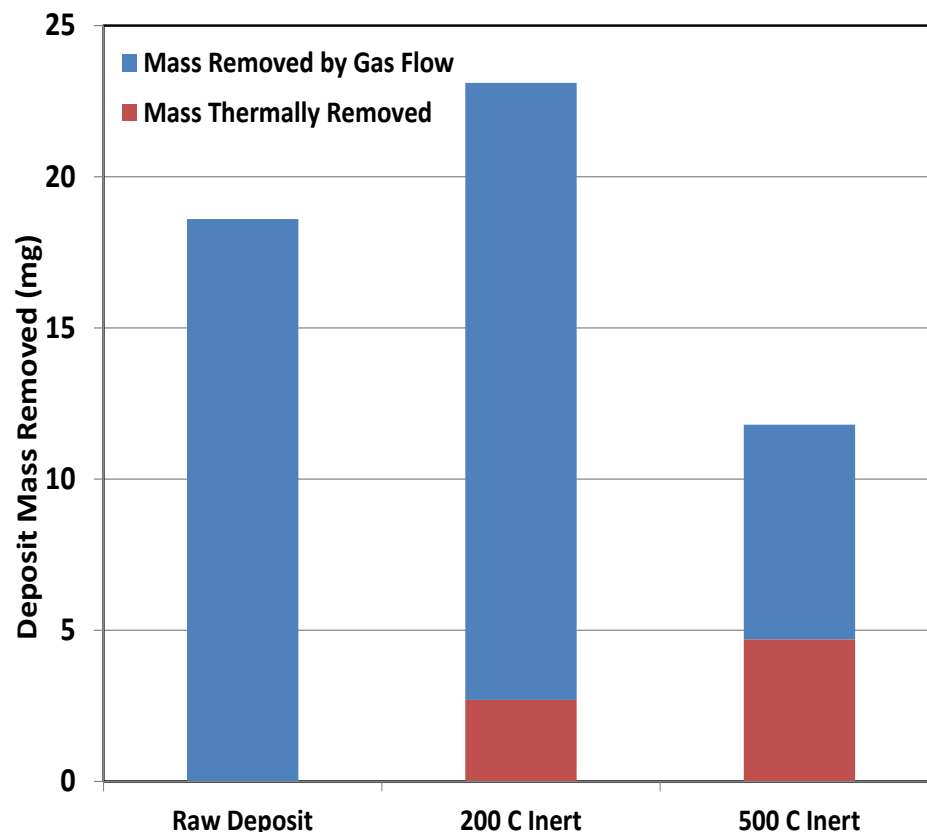
## Differing removal rates have been observed.

### HC-Significant Deposits:

- Shear-induced removal is suppressed for HC-dominated deposits.
- Density (thickness), tensile strength, or both?

### Low-HC Deposits:

- Raw deposit and 200C case have similar mass removal.
- 500C case showed much less mass removal.
  - Sintering of the deposit.
  - Higher resistance to shear.



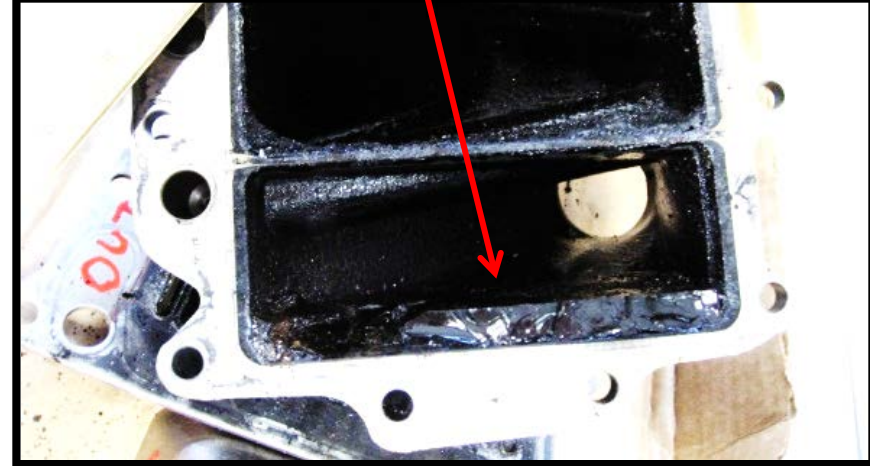
Total deposit mass before removal ~ 60mg.

Deposit thermochemical history (including HC content and chemistry) can significantly influence deposit removal through spallation.

## **Fuel-related HCs can contribute to fouling through the formation of a relatively dense deposit under low-temperature conditions.**

- Operation at low-T, low-Re conditions develops HC-rich deposits.
- Experiments have shown high deposit mass with very little effectiveness loss.
  - Density must increase significantly.
- Deposit HC content >50%.
- Can continue unabated for long periods.

Image of a glass-like deposit in an OEM cooler diffuser; ~ 1cm thick and 80% volatile content.



HC deposition is strongly a function of the wall temperature and volatility of the HCs; expanded use of higher-boiling HCs (such as biodiesel) in fuel is expected to increase the risk of plugging.

## **Fuel Effects – Exploiting Alcohols in Gasoline to Enable Lean NO<sub>x</sub> Reduction**

**Can broad inclusion of bio-derived alcohols in fuel may improve the viability of fuel-efficient lean-burn vehicles by enabling a non-precious metal route to NO<sub>x</sub> reduction?**

**Benefit: Use of lean-burn technology can reduce the tank-mileage penalty associated with high ethanol blend levels and encourage broader utilization. Taking advantage of the properties of ethanol will enable the use of non-precious metal catalyst systems for lean-NO<sub>x</sub> reduction.**

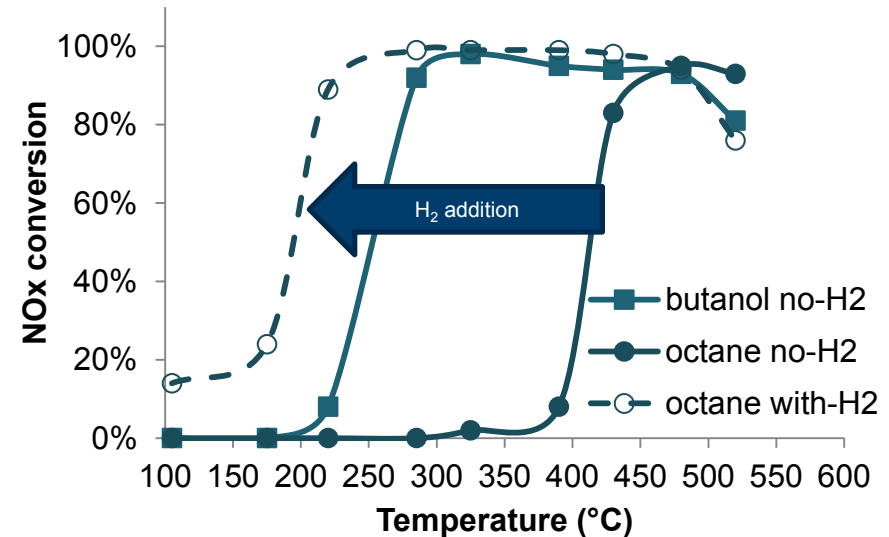
### **Accomplishments:**

- 1. Demonstrated high NO<sub>x</sub> reduction reactivity with high selectivity to N<sub>2</sub> using ethanol as a reductant over silver-alumina catalyst.**
- 2. Illustrated role of ethanol is similar to HC-SCR activation observed with H<sub>2</sub>.**

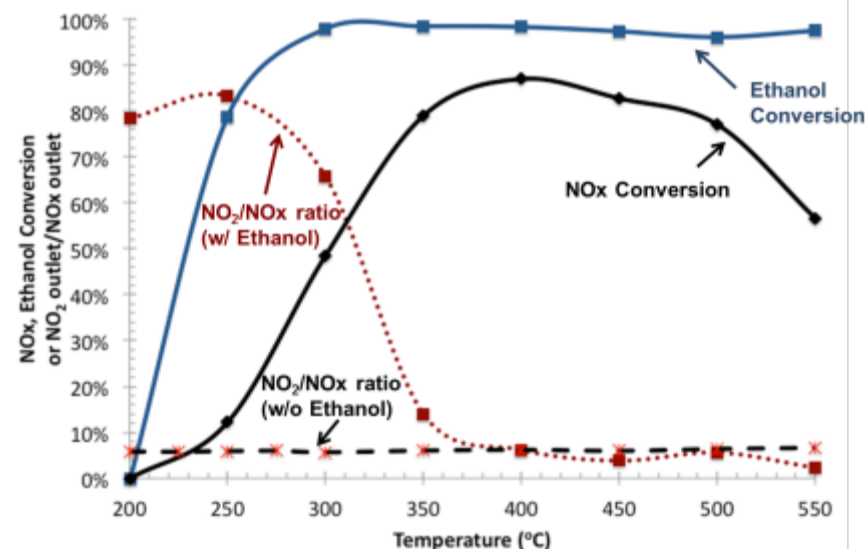
# Inclusion of bio-derived alcohols in gasoline creates unique opportunities for PGM-free $\text{NO}_x$ reduction catalysts to support lean-burn engines.

- Ethanol/butanol inclusion in gasoline may promote  $\text{NO}_x$  reduction using octane/higher-HCs.
- $\text{NO}_x$  reduction activity using octane is enabled by  $\text{H}_2$ ; similar to ethanol mechanism.
- Ethanol and butanol show low temperature  $\text{NO}_x$  conversion on silver catalysts.
  - Similar  $\text{NO}_x$  reactivity to  $\text{H}_2$ +octane.
  - Suggests similar mechanism.

EISA mandates have resulted in virtually all gasoline containing ethanol; this presents an opportunity to capitalize on the beneficial properties of ethanol/butanol.



From R. Burch, *Cat. Rev. Sci. Eng.* 46 (2004) 271.



From Toops et al., in press, *Catal. Today* (2012)

# Recent fundamental study determined the beneficial role of ethanol in NO<sub>x</sub> reduction with silver/alumina catalysts.

- **Ethanol + O<sub>2</sub> readily forms acetates**

- de-hydrogenation is first step
- Leads to adsorbed H (H—\*)

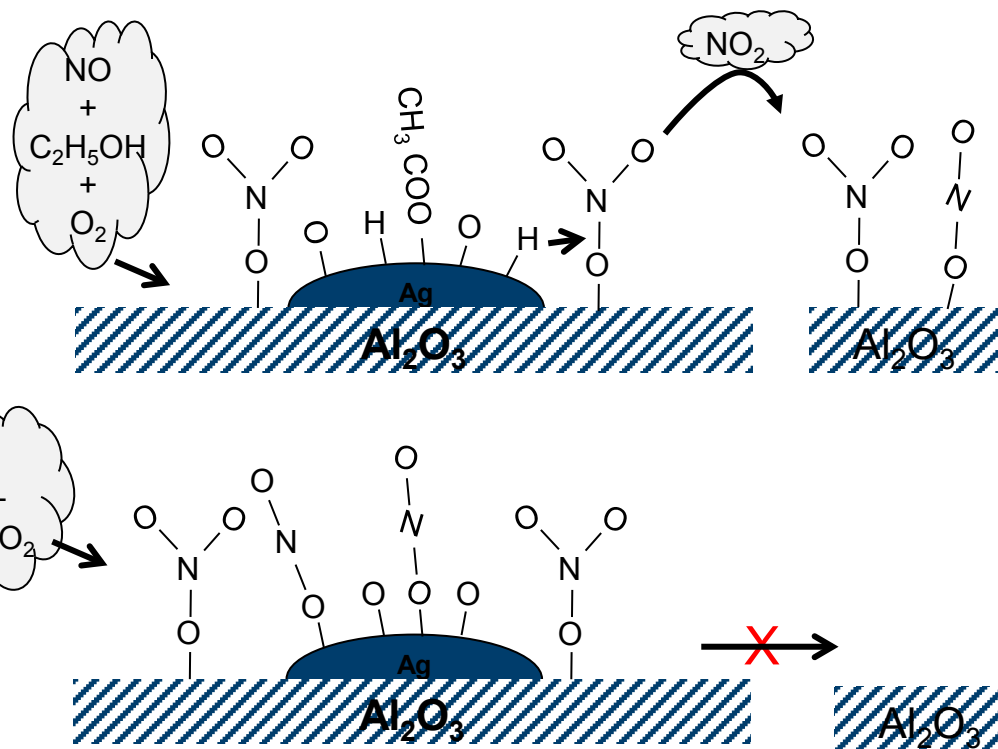
- **H—\* releases nitrates as NO<sub>2</sub>**

- enables adsorption away from Ag;
- Increases N population on catalyst surface.
- new reactive/adsorption site becomes available

- **Similar to H<sub>2</sub> enhanced HC-SCR**

- **NO+O<sub>2</sub> yields minimal nitrite/nitrate formation**

- low mobility w/o ethanol
- Illustrates NO inhibition
- DRIFTS supported finding



## References Consulted:

A.B. Mhadeshawar et al., Appl. Catal. B 89 (2009) 229; S. Chansei et al., J. Catal. 276 (2010) 49; Y.H. Yeom et al., J. Catal. 238 (2006) 100; Shimizu et al., Appl. Catal. B 30 (2001) 151. R. Burch, Cat. Rev. Sci. Eng. 46 (2004) 271.

Hydrogen from dissociative adsorption of ethanol facilitates NO<sub>x</sub> reduction over Ag/Al<sub>2</sub>O<sub>3</sub>; may enable utilization of higher HCs by silver catalysts.



**Is the <5 ppm Na specification for biodiesel sufficient to ensure performance of emissions control devices over the vehicle lifetime?**

**Benefit:** Insuring that the biodiesel blend levels above 5% in the marketplace does not adversely impact emissions control devices will enable broader utilization of blends such as B20.

### **Accomplishments:**

- 1. Identified that oxidation functionality is the most impacted characteristic of SCR-performance when exposed to Na equivalent of 450,000 miles of B20.**
- 2. Began large collaboration between ORNL, NREL, NBB, MECA, and Ford to perform full-size catalyst study with current commercial emissions control system.**

# **Current 5 ppm Na limit is a concern for emissions control devices, especially for heavy duty applications.**

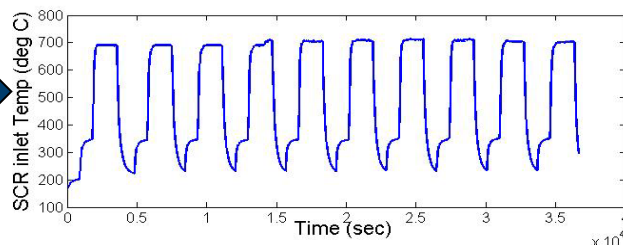
- **NaOH or KOH typically used in biodiesel production**
  - Liquid-phase catalyst
- **Current ASTM specification**
  - 5 PPM for total alkali content (Na+K)
  - 5 PPM Calcium + Magnesium
- **Primary concerns are its impact on**
  - ash accumulation in DPF,
  - mechanical strength of substrates
  - zeolite-based SCR deactivation
- **Collaboration with NREL, NBB, MECA, and Ford to perform full-size catalyst study with current commercial emissions control system**

# Based on initial findings in studies at ORNL/NREL/Ford large collaborative effort undertaken to establish thresholds

- Collaboration with NREL, NBB, MECA, and Ford focusing on light duty applications
  - full production exhaust systems from 2011 Ford F250; DOC→SCR→DPF



Thermal cycling



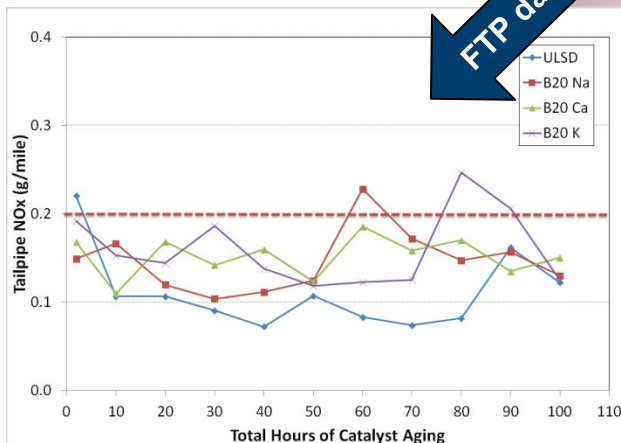
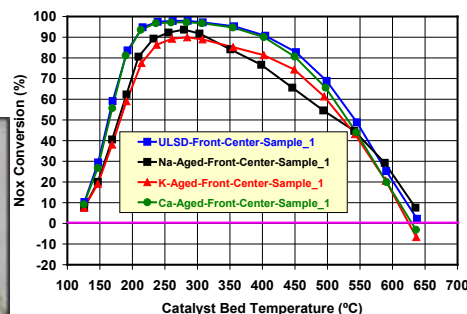
**Ongoing study expected to conclude this FY**

## Ford leading catalyst functionality study

- Sectioned study of DOC and SCR
- Determine what level of Na, Ca, and K leads to decrease in performance

## NREL leading engine operation

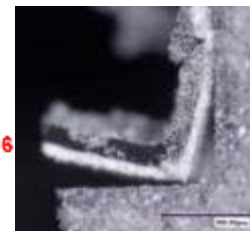
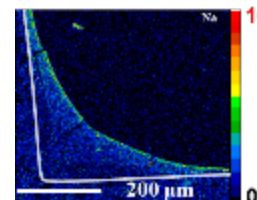
- Na, K, and Ca plus thermal exposure; ULSD-only too
- Coordinating with MECA+NBB



FTP data

## ORNL leading materials characterization

- DPF materials characterization
  - Co-funded by Propulsion Materials
- Identify level of Na, Ca, and K in each section of the emissions control devices



# Collaborators and Partners

- **Fuel Effects on EGR Cooler Fouling**
  - Ford/GM/Modine: experimental guidance
  - ORNL Advanced Manufacturing Center (DOE-AMO)
- **Lean NO<sub>x</sub> catalysis for ethanol fueled vehicles**
  - Galen Fisher will provide commercially relevant catalysts and serve as a technical advisor
    - Formerly of Delphi, now Adjunct Professor at Univ. Michigan
- **Fuel Effects on Emissions Control Devices**
  - NREL/Ford/MECA: Biodiesel-aged collaborative effort
  - University of Tennessee: graduate research
  - CLEERS: evaluation protocols
- **Fuel Effects on PM Oxidation Kinetics**
  - Penn State University: microscopy and PM structural analysis, generation of biodiesel PM with medium-duty engine.



# Future Directions (Beyond FY12)

- **Fuel Effects on EGR Cooler Fouling:**
  - Future EGR efforts will focus on low-temperature conditions (such as drop-to-idle) where fuel-related HCs can play a significant role in cooler performance and degradation.
- **Exploiting Alcohols in Gasoline to Enable Lean-NO<sub>x</sub> Reduction:**
  - Investigate ethanol and butanol ability to improve HC-SCR NO<sub>x</sub> reduction reactivity with Gasoline blends.
  - Study formation of NH<sub>3</sub> under lean operation (potential for hybrid SCR system)
- **Biodiesel-based Na Impacts on Emissions Control Devices:**
  - Study concluding; will present assessment of 5 PPM limit impact on light-duty engineering margins.
  - Assessing the need to perform similar study with heavy-duty and lubricant communities.
- **Assessment of Properties, Emissions, and Compatibility of Emerging Fuels:**
  - Utilize pyrolysis GC/MS capability to investigate fuel and lubricant contributions to PM formation in GDI and other advanced engines.
- **Soot oxidation kinetics work is concluding in FY12. A follow-on activity investigating GDI PM production and characteristics is being initiated.**

As always, we welcome specific concerns from industry, whether in these areas or other topics, for future studies.

# Summary

- **Relevance:** This project is targeted towards providing data and predictive tools to address gaps in information needed to enable increased use of NPBFs. (DOE Technical Barrier)
- **Approach:** The approach being pursued is to bring together targeted, engine-based studies using NPBFs with in-depth characterization of PM and HCs to better understand behavior for specific technologies. (Currently emissions control devices and EGR Systems)
- **Collaborations:** Collaborations with several industry stakeholders and universities are being used to maximize the impact of this work.
- **Technical Accomplishments:**
  - Low-Re, Low-T conditions produce HC-significant deposits that may lead to plugging; removal of deposits by flow shear requires very high velocities.
  - Identified pathway for advantageous use of ethanol in gasoline to enable non-precious metal lean NO<sub>x</sub> aftertreatment for high-efficiency lean-burn engines.
  - Determined Na affects SCR oxidation sites; launched expanded collaborative effort with Ford, NREL, MECA, and NBB to rapidly assess full-size systems.
  - Determined that soot oxidation kinetics depend on fuel properties but also on engine operating conditions.
- **Future Work:** Plans are in place; industry input towards those plans or other fuel-emissions control effect concerns is needed and welcomed.



# **TECHNICAL BACKUP SLIDES**

## Fuel Effects on EGR Cooler Fouling

**Spalls from the EGR cooler deposit are not simply re-entrained diesel particulate.**

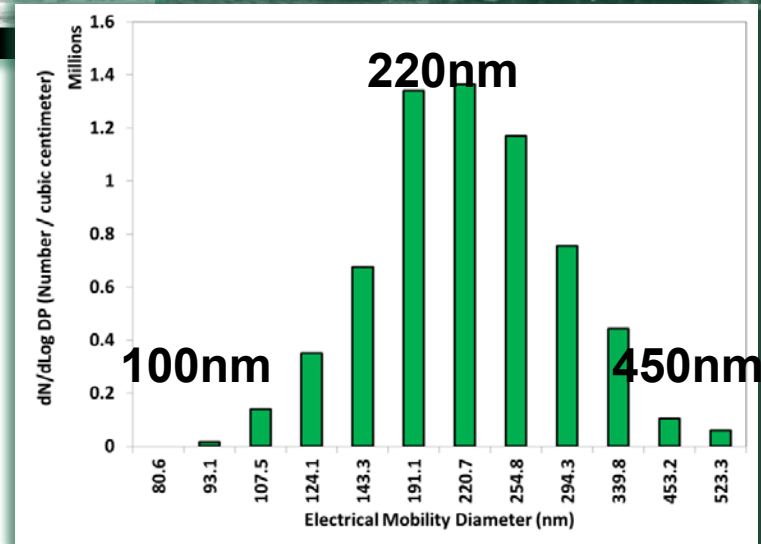
Image of a spall

1  $\mu\text{m}$

Deposit before spallation

Deposit after spallation

- Spalls are much larger than typical diesel particulates.
- Areas experiencing spallation become rougher than the parent deposit.
- Observations point to the formation of a cohesive deposit even during short time periods.

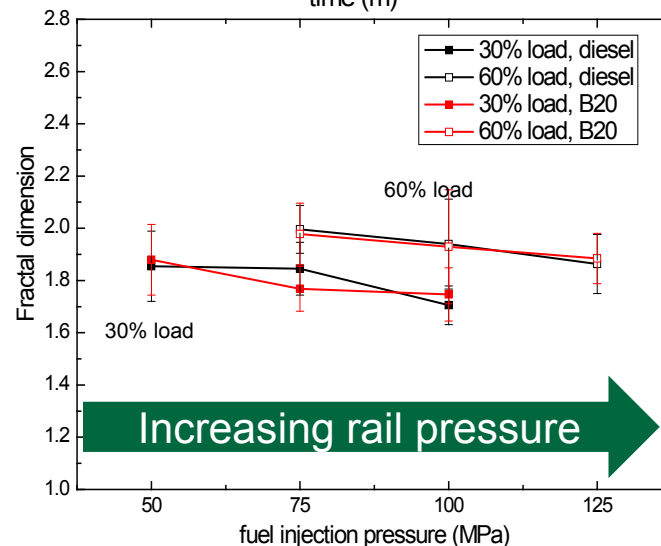
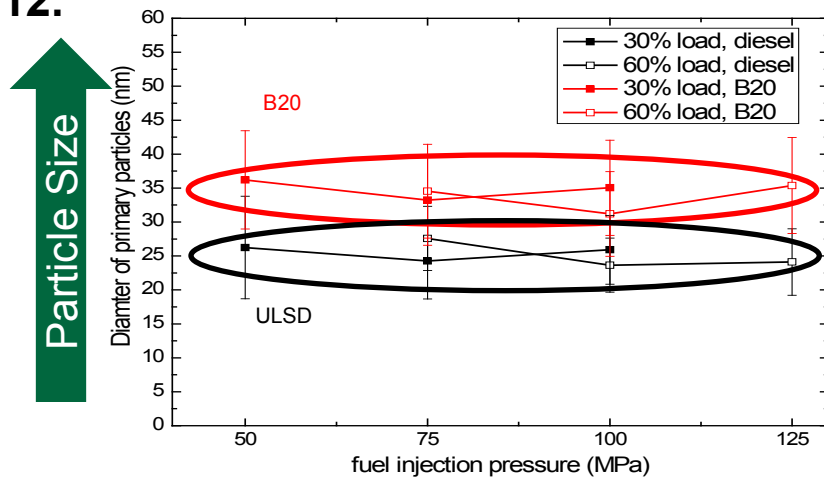
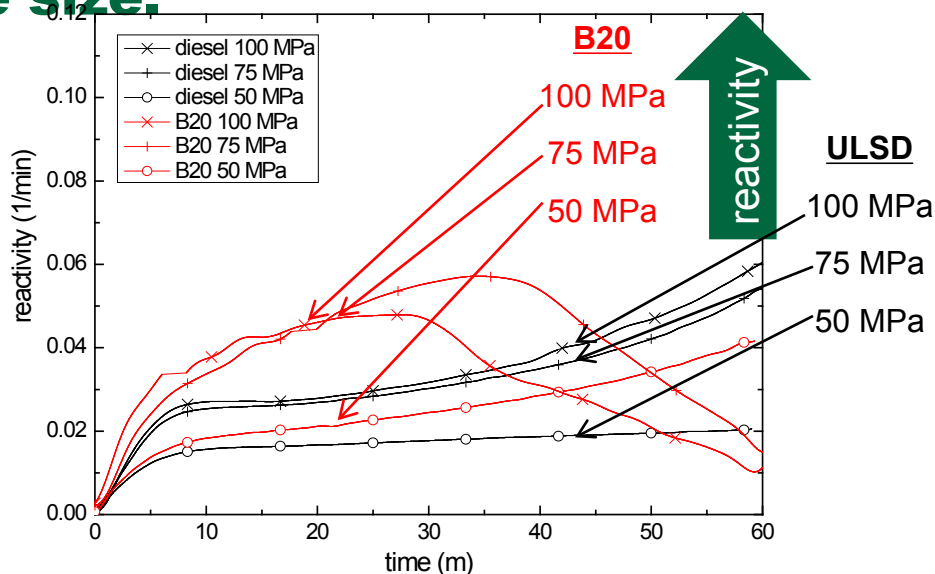


Formation and removal of deposit spalls is complex; critical-velocity models may not adequately capture the underlying physics.

## Fuel Effects on Soot Oxidation Kinetics

### Fuel-type and engine operating conditions affect soot reactivity and primary particle size.

- B20 soot more reactive than ULSD.
- Higher injection pressure increases reactivity.
- B20 soot has higher average primary particle size
- Soot disorder (fractal dimension) increases with higher load.
- Study being brought to completion in FY12.



Engine operating conditions do influence subsequent soot oxidation kinetics.

# **Summary of key biodiesel PM oxidation findings**

- **Rate of O<sub>2</sub>-based oxidation of the fixed carbon component of diesel PM is directly dependent on available surface area.**
  - Surface area evolves differently with burnout based on fuel type.
  - A single Arrhenius rate expression with  $E_A = 113 \pm 6$  kJ/mol can be used to fit surface normalized rates for all fuels studied.
- **O<sub>2</sub>-based oxidation rates of nascent PM can be modeled as the sum of a volatiles evolution term and an Arrhenius FC consumption term.**
- **NO<sub>2</sub>-based oxidation of diesel PM by appears to be very different from oxidation by O<sub>2</sub> and shows little fuel dependence.**
- **DPF models need to account for biodiesel blending, prior oxidation history and O<sub>2</sub>/NO<sub>2</sub> concentrations for predicting regeneration rate.**
- **In addition to biodiesel level, increasing injection pressure increases soot reactivity.**

# **Recommendations for Future Work in PM Oxidation Kinetics**

- Continue resolving appropriate kinetics for  $\text{NO}_2$  PM oxidation and role of surface area.
- Identify how the combined kinetics for  $\text{O}_2$  and  $\text{NO}_2$  should be modeled.
- Resolve the appropriate functional form to use for the PM devolatilization rate in the presence of oxidants.
- Expand the range of biofuels and engines evaluated.
- Incorporate the combined  $\text{O}_2/\text{NO}_2$ /devolatilization kinetics models into an integrated DPF device model suitable for vehicle simulations.