# **Emissions Control for Lean Gasoline Engines**

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# **Project Overview**

## **Timeline**

- New project:
  - Combination of two related efforts
  - Refocused to address current DOE and industry needs
- Lean Gasoline: Started in FY10
- Fundamental Catalysis: Started in FY02

### **Budget**

- Lean Gasoline
  - FY11: \$200k
  - FY12: \$400k
- Fundamental Catalysis
  - FY11: \$200k
  - FY12: \$0k

## **Barriers Addressed**

- Barriers listed in VT Program Multi-Year Program Plan 2011-2015:
  - 2.3.1B: Lack of cost-effective emission control
  - 2.3.1C: Lack of modeling capability for combustion and emission control
  - 2.3.1.D: Durability

### **Collaborators & Partners**

- Umicore
- General Motors
- University of South Carolina
- University of Wisconsin
- Cross-Cut Lean Exhaust Emissions Reduction Simulations (CLEERS)



## **Objectives and Relevance**

# Enabling lean-gasoline vehicles to meet emissions regulations will achieve significant reduction in petroleum use

### Objective:

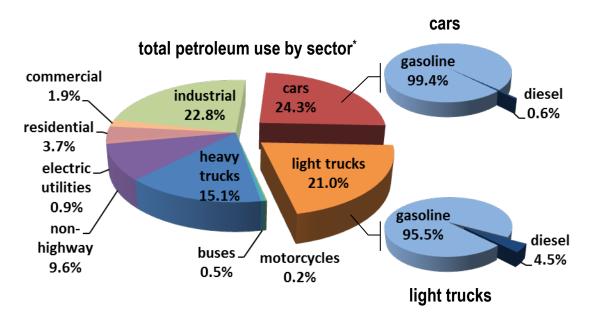
- Address technical challenges to meeting emissions regulations
- Investigate strategies to achieve cost-effective compliance
  - minimize precious metal content while maximizing fuel economy

#### Relevance:

- U.S. passenger car fleet is dominated by gasoline-fueled vehicles.
- Enabling introduction of more efficient lean gasoline engines can provide significant reductions in overall petroleum use
  - thereby lowering dependence on foreign oil and reducing greenhouse gases



## Relevance: small improvements in gasoline fuel economy significantly decreases fuel consumption



Lean gasoline vehicles can decrease **US** gasoline consumption by ~30 million gal/day

- 132,000 million gallons of fuel used by cars and light trucks annually\*\*
- New car and light-truck sales dominated by gasoline engines
- 10% fuel economy benefit † from base case of 22.6/18.1 mpg\*\* has big impact
  - Saves >200,000,000 barrels gasoline annually
  - 5% of overall petroleum used
- **HOWEVER...emissions compliance needed!!!**







### **Current and Future Milestones**

- FY2011: Investigate performance, sulfation and desulfation of Ca-doped LNTs with a Ca-level between 3 and 9% (9/30/2011).
  - Complete; project merged with Lean Gasoline project
- FY2011: Analysis of lean-rich period modification for enabling lower volume LNTs to reduce the higher concentrations of NOx associated with lean gasoline engines (9/30/2011).
  - Complete
- FY2012: Measure the effect of oxygen storage capacity on NH<sub>3</sub> formation by three way catalyst for use in passive SCR emission control strategy. (9/30/2012)
  - On track
- Future years will adopt a quantitative milestone approach to ensure progression towards goal
  of low-cost emissions control solution for fuel efficient lean-burn gasoline vehicles

	FY13	FY14	FY15	FY16	FY17
Fuel economy gain over stoichiometric	7%	10%	10%	12%	15%
Total emissions control devices Pt* (g/L <sub>engine</sub> )	8	7	6	5	4

	Pt-equivalent	
Platinum	\$ 1,504/troy oz.	1.0
Palladium	\$ 463/troy oz.	0.3
Rhodium	\$ 3,582/troy oz.	2.4
Gold	\$ 989/troy oz.	0.7

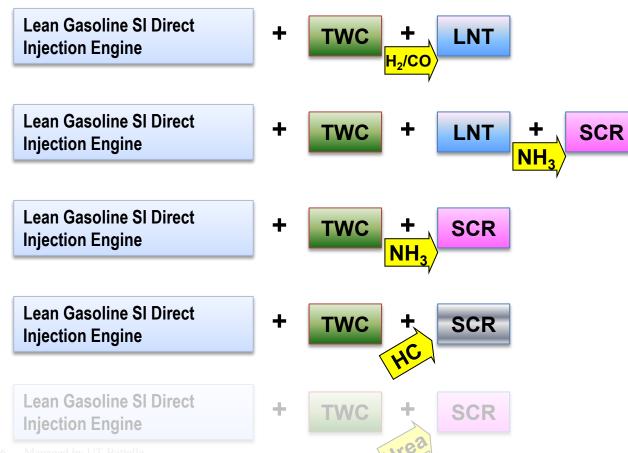
<sup>\* -</sup> will use Pt equivalent to account for different costs of Pt, Pd and Rh; 5-year average value fixed at beginning of project



# Approach: Technology Options and Critical Issues Related to Cost and Performance

- Goal: Enable Tier II Bin 2 Emission Compliance for Lean Gasoline Engine
- Focus on NOx, CO, HC (PM may be issue for DI engines, but outside of project scope; new project starting)
- Technologies: TWC = Three-Way Catalyst
  LNT = Lean NOx Trap

**SCR = Selective Catalytic Reduction** 



### Specific Key Issues:

Cost, Durability, Fuel Penalty, Operating Temp., etc...

LNT Capacity and Cost
HC Slip Control

LNT NH<sub>3</sub> Optimization HC Slip Control

TWC NH<sub>3</sub> Production HC Slip Control

Temperature Performance HC Supply and Slip Control

Urea Tank/Injector Cost Customer Acceptance

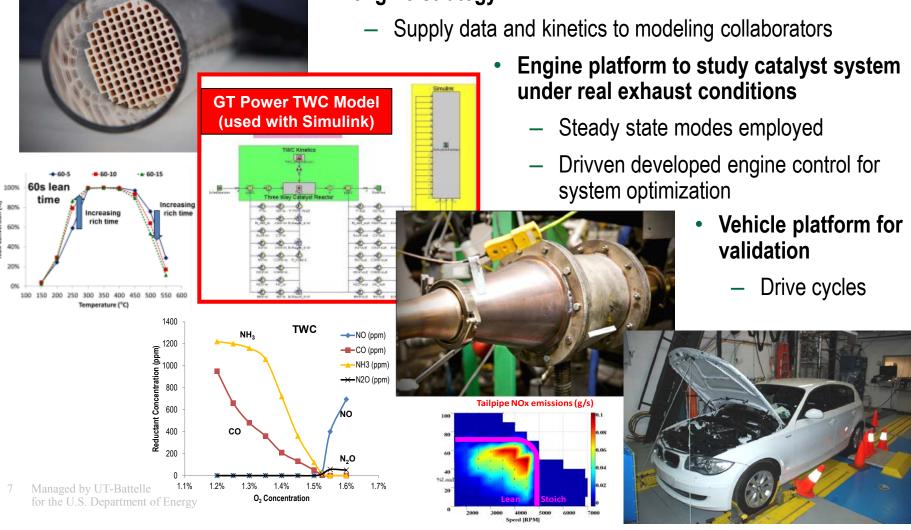
Not in Project Scope

Managed by UT-Battelle for the U.S. Department of Energy

## Overall approach to meeting long term milestones

- Improve catalyst understanding and optimize emissions control
  - Catalyst technology focus → Efficiency gains from lean vs. stoichiometric operation

Understanding from bench reactor catalyst data guides engine strategy



# **Collaborations and partners**

- Currently all collaborations are based on mutual interest
  - No subcontracts between partners
- General Motors
  - Monthly teleconferences
- Umicore
  - Catalyst supplier for the commercial LNT and TWCs
  - Facilitating range of catalysts with varying PGM and functionality
  - Monthly teleconferences
- University of South Carolina
  - Visiting graduate student operating and analyzing bench reactor data
- University of Wisconsin
  - Modeling components; based on global kinetics
- CLEERS
  - share results and identify research needs











# **Summary of Technical Accomplishments**

- Progressed towards lean gasoline engine dynamometer platform
  - Obtained two lean gasoline vehicles (BMW 120i)
    - Engine to be removed from one for full control in test cell
    - Other vehicle to be kept intact for chassis-studies
  - Subcontract to Drivven to develop full pass controller in progress
  - Transient drive cycle data uploaded to CLEERS website
- Measured impact of lean/rich duration on NOx reduction and product yields
  - Maximized NOx conversion with constant fuel penalty with:
    - short, concentrated rich dose for high temperatures T>450°C
    - long, low concentration rich dose for T<300°C</li>
  - Identified peak NH<sub>3</sub> and N<sub>2</sub>O formation at 250°C
  - Established desulfation temperature: Peak desulfation at 550-580°C (sulfated to 3.4 g/L)
- Identified TWC NH<sub>3</sub> production conditions for passive SCR approach
  - Two TWC technologies studied
    - High PGM, Pd-based technology with <u>no</u> OSC (oxygen storage capacity)
    - Low PGM, Pd/Rh-based technology with OSC
  - Demonstrated controlled NH<sub>3</sub> production while rich with >98% CO/HC conversion



## Engine and vehicle platform is the lean burn gasoline version of the BMW120i - 2.0L L4

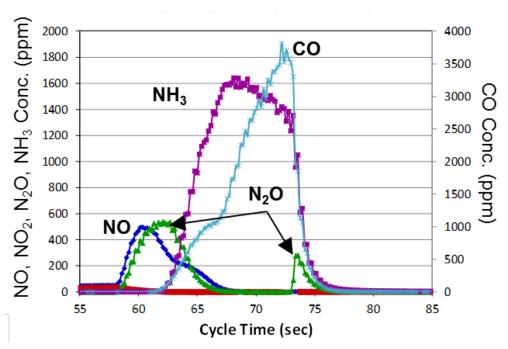


- Vehicle based on industry suggestion
  - BMW lean-burn gasoline vehicle
  - Close-coupled Three-way Catalyst (TWC)
  - Underfloor Lean NOx Trap (LNT)
    - New CLEERS reference catalyst
  - Total PGM = 7.5 g/L-engine
  - As calibrated, does not meet current US emissions standards
- Bench reactor studies focused on catalyst technologies relevant to this application
  - Umicore recommended TWC
    - Pd-only with no oxygen storage component (OSC) in front zone
    - Pd/Rh with OSC in rear zone
  - Identical LNT to BMW



# LNT: Lean/rich durations varied to establish impact on NOx reduction and NH<sub>3</sub> formation

- Investigate 3 lean times and 3 rich times for a total of nine conditions
  - Hold rich dose/fuel penalty constant for all cases
  - Stoichiometric reductant concentrations for nitrate reduction to carbonate + OSC reduction
- Develop understanding on tradeoffs of long and short durations
  - Is it better to have a short rich time with high reductant concentrations?
  - What is the impact on conversion and  $NH_3$  and  $N_2O$  formation?



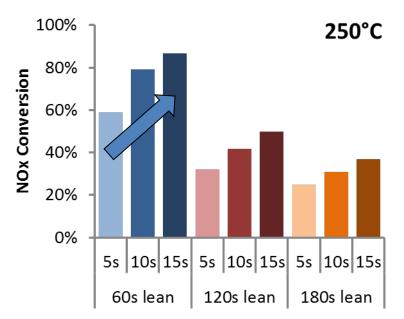
Lean / Rich times	CO	$H_2$	$C_3H_6$
60 s lean/5 s rich	2.01%	0.67%	0.112%
60 s lean/10 s rich	1.01%	0.34%	0.056%
60 s lean/15 s rich	0.67%	0.22%	0.037%
120 s lean/5 s rich	2.83%	0.94%	0.157%
120 s lean/10 s rich	1.42%	0.47%	0.079%
120 s lean/15 s rich	0.94%	0.31%	0.052%
180 s lean/5 s rich	3.65%	1.22%	0.203%
180 s lean/10 s rich	1.82%	0.61%	0.101%
180 s lean/15 s rich	1.22%	0.41%	0.068%

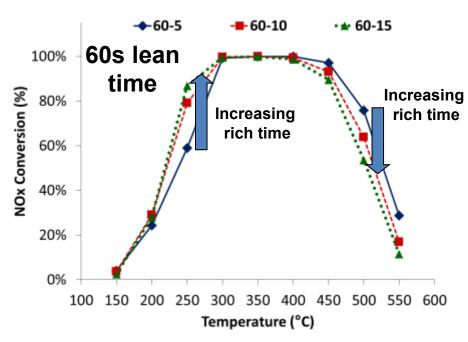
Example products during rich portion of 60s lean and 15s rich at 250°C

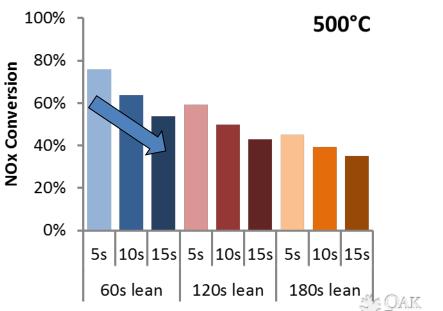


# Practical lean/rich timing variation impacts NOx conversion performance

- Short lean times and long, low concentration rich dose favored at T<350°C</li>
  - At 250°C, NOx conversion varies from 25% to 85%...fuel penalty is constant
  - Slow NOx release and reduction kinetics
- Need fast, high concentration rich dose to convert stored NOx at T>400°C
  - Fast NOx release and reduction kinetics



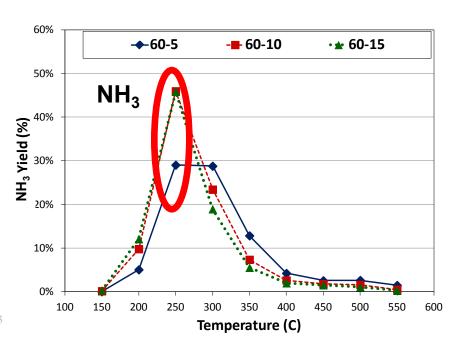


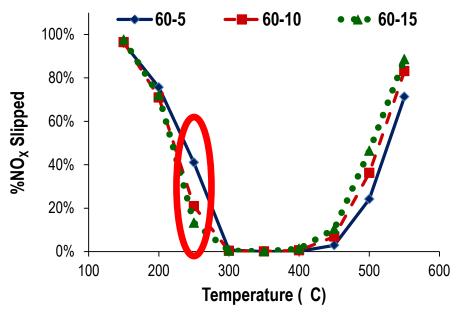


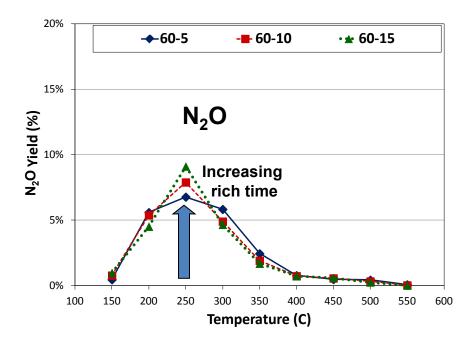
## NH<sub>3</sub> generation occurs between 200 and 350°C; potential of LNT+SCR

$$Yield_{NH_3} = \left(\frac{NH_3(produced)}{NO_X(fed)}\right)$$

- Up to 45% NO fed goes to NH<sub>3</sub>
- Comparing NOx and NH<sub>3</sub> slip at 250°C illustrates potential LNT+SCR system
- Unfortunately, N<sub>2</sub>O yield also peaks at these conditions





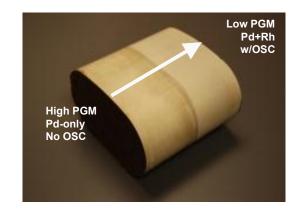


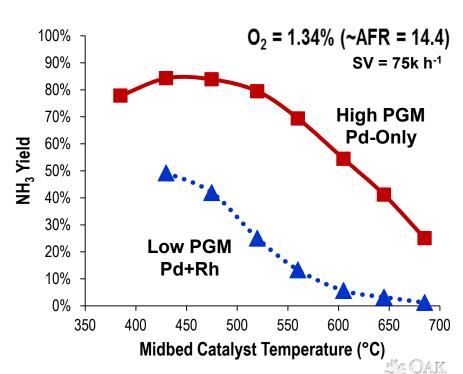
# Understanding NH<sub>3</sub> generation over TWC may enable TWC+SCR system without Urea

- Measure performance of TWCs as a function of temperature and components
  - Overall NO, CO, C<sub>3</sub>H<sub>6</sub> conversion
  - N<sub>2</sub>O and NH<sub>3</sub> formation



- High PGM Pd-only with no oxygen storage component (OSC)
  - 0/4.4/0 g Pt/Pd/Rh
- Low PGM Pd+Rh with OSC in rear zone
  - 0/0.8/0.21g Pt/Pd/Rh
- Including LNT into this study will give the following catalyst types:
  - PGM-only
  - PGM + OSC
  - PGM + OSC + NOx storage

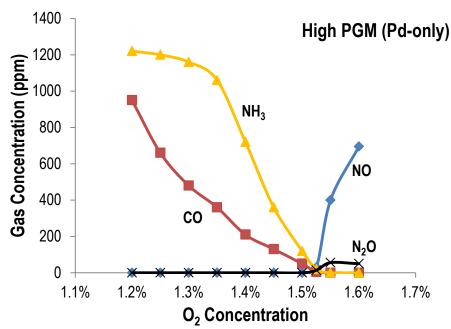


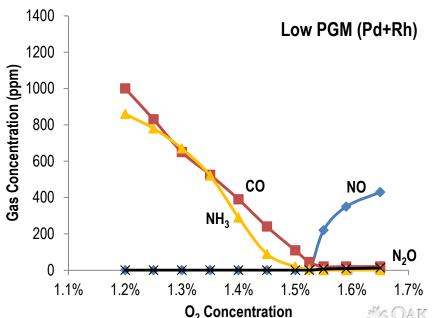


# TWC is effective and tunable NH<sub>3</sub> generator

- NH<sub>3</sub> readily generated; varies with PGM
  - For high PGM (Pd-only) TWC:
    - All NO fed converted to NH<sub>3</sub> when very rich
  - For low PGM (Pd+Rh) TWC:
    - NH<sub>3</sub> production is still significant but reduced
- At all conditions, >95% CO conversion
  - C<sub>3</sub>H<sub>6</sub> not observed in effluent
  - >98% CO conversion for O<sub>2</sub>>1.35% (AFR ~14.4)
- N<sub>2</sub>O formation observed under lean conditions and varies with PGM content
  - Up to 56 ppm with high PGM (Pd-only) TWC
  - Less than 10 ppm with low PGM (Pd+Rh) TWC

~AFR	O <sub>2</sub>	NO	СО	H <sub>2</sub>	$C_3H_6$
14.6	1.59%	0.12%	1.80%	0.60%	0.10%
14.4	1.34%	0.12%	1.80%	0.60%	0.10%
14.2	1.06%	0.12%	1.80%	0.60%	0.10%

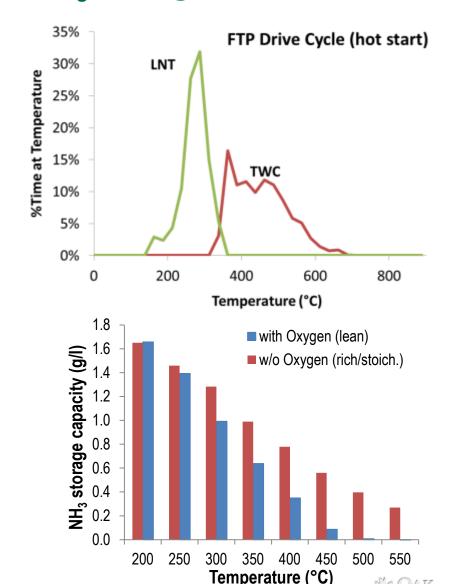




Midbed Temperatures: 460-500°C

### NH<sub>3</sub> production over LNT and TWC occurs at temperatures relevant to vehicle operation and NH<sub>3</sub> storage on SCR

- **Histogram of catalyst temperatures** during drive cycle with BMW 120i
  - FTP (hot start)
  - 200-350°C for LNT
  - 350-600°C for TWC
- LNT: max NH<sub>3</sub> yield at 250-300°C
- TWC: tunable NH<sub>3</sub> production 350-600°C
- NH<sub>3</sub> production temperatures mesh well with NH<sub>3</sub> storage temperatures on SCR
  - More NH<sub>3</sub> storage occurs under rich/stoichiometric conditions
  - However switching from rich to lean will result in NH3 release if over-saturated



# Lean Gasoline Engine Research Platform Development In Progress

- Two BMW 120i lean gasoline engine vehicles were procured from Europe and are being mapped for engine controller development
  - One vehicle's engine will be pulled for installation on engine dynamometer
- Subcontract to Drivven for controller development in progress
  - End controller will have full control capability for research specific engine operation and OEM map operation for general catalyst and reference studies

#### <u>Tasks in Controller Development:</u>

- 1. Procure Vehicles (complete)
- 2. Verify lean operation (complete)
- 3. Drivven visit for electronic signal probing (complete)
- 4. Shipment of one vehicle to Drivven (complete)
- 5. Calibration of sensors, injectors, etc. (in progress)
- 6. Harness wiring and mapping preparation (in progress)
- 7. Full mapping exercise on vehicle at ORNL (May-June 2012)
- 8. Control software development (June 2012)
- 9. Engine installation at ORNL (July 2012)
- 10. Controller installation at ORNL (July 2012)
- 11. Controller operation and verification (August 2012)



## Additional future work

- Study effects of catalyst composition by varying:
  - PGM content, OSC content
  - Adding NOx storage capacity to TWC
  - Low temperature CO/HC oxidation
- Impact of lean/rich cycling on TWC
  - Investigating strategies for minimizing N<sub>2</sub>O formation
- Age catalysts and evaluate end of life performance
  - Define durability requirements, time and temperature as related to sulfur exposure and desulfation requirements
- Improve modeling capability to better guide research and meet milestones
- Finalize milestones and achieve year one target

	FY13	FY14	FY15	FY16	FY17
Fuel economy gain over stoichiometric	7%	10%	10%	12%	15%
Total emissions control devices Pt* (g/L <sub>engine</sub> )	8	7	6	5	4



## **Summary**

- Relevance: Enabling lean gasoline vehicles will impact significantly on US petroleum use
- Approach: Cost-effective emissions control facilitates introduction of lean-gasoline engines
  - Employ multi-platform approach to optimize catalysts and lean engine operation
  - Employ annual cost and efficiency milestones to push technology
- Collaborations: Industrial (GM and Umicore) and Academic (U-So. Carolina and U-Wisconsin)
- <u>Technical Accomplishments</u>:
  - Progressed towards lean gasoline engine dynamometer platform
  - Measured impact of LNT lean/rich duration on NOx reduction and product yields
  - Identified TWC NH<sub>3</sub> production conditions for passive SCR approach

#### • Future Work:

- Complete and install DRIVVEN-developed engine controller for engine/chassis operation
- Vary OSC and NOx storage component on LNT and TWCs under cycling conditions
- Develop and employ aging routine to rapidly age catalysts to end of life
- Meet first cost and efficiency milestone of the multi-year program

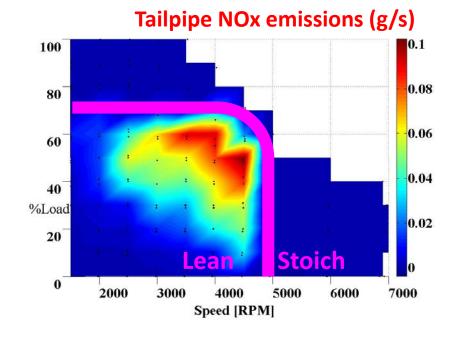


# **Technical back-up slides**



# 4-15% Fuel Economy Benefit <u>but</u> Challenge of Emissions Exceeding U.S. Regulation Levels

- Vehicle designed to meet emissions levels required by European regulations
- NOx emission levels exceed U.S. Tier II Bin 5, 0.05 g/mile at 50k miles
  - Bin 2 → 0.02 g/mile
- NOx emissions during lean operation are problematic
  - Data shown is from engine vehicle evaluation at ORNL



Improved Lean
NOx catalysis
required for
deployment of
lean gasoline
vehicles

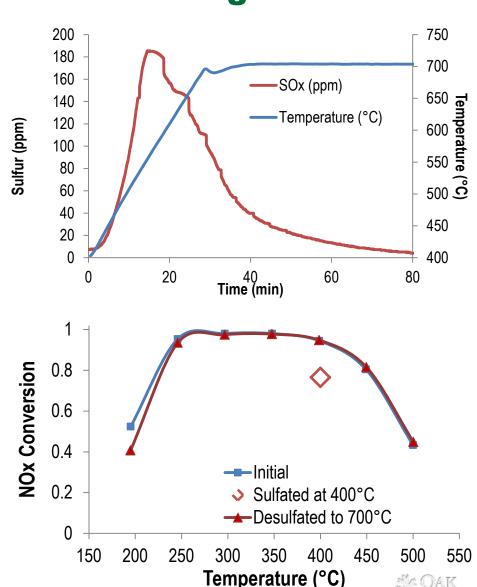
Drive Cycle	Fuel Economy Improvement*	NOx Emissions (g/mile)
FTP	10.0%	0.11
HFET	14.6%	0.11
US06	4.4%	0.35





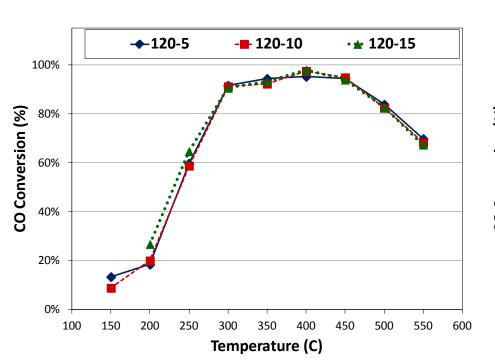
# Peak LNT desulfation at 550-580°C and full performance recovery after desulfating to 700°C

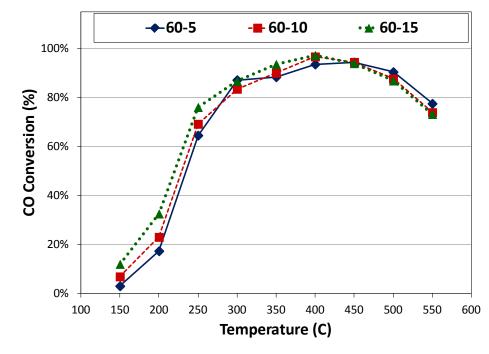
- Sulfation to 3.4 g/L<sub>cat</sub> at 400°C
  - 45 ppm SO<sub>2</sub> accelerated
  - NOx conversion drops from 95% to 77%
- LNT shows low desulfation temperature
  - Sulfur removal occurs readily above 400°C
  - Peaks at 550-580°C
  - Slow release up to 700°C
- Desulfation to 700°C results in complete performance recovery
  - Suggests targeting 700-750°C for aging conditions

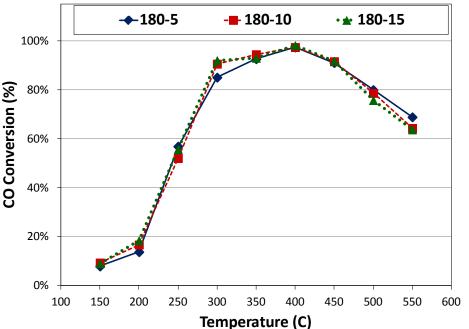


# Cycle averaged CO conversions

- Essentially independent of lean rich timing
  - at 300-500C, >80% conversion all conditions
- CO utilization decreases at higher temperatures
  - Less NOx stored and fast release
  - Mismatch in reductant feed and NOx release

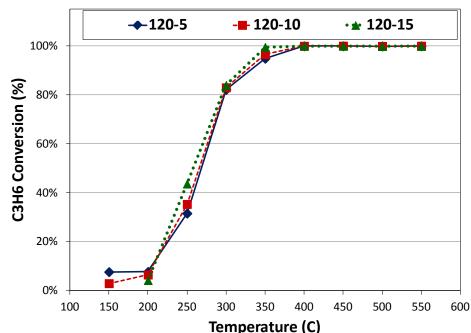


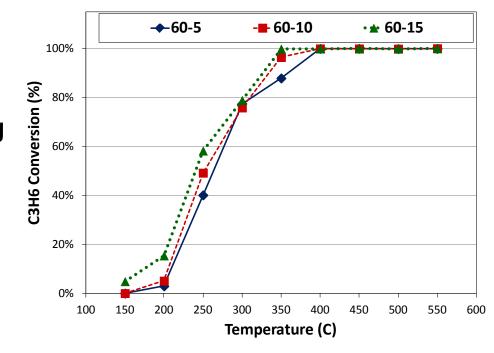


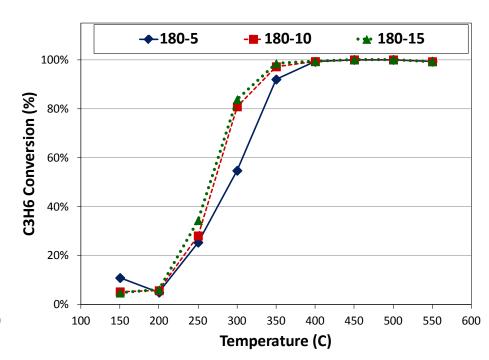


# Cycle averaged C<sub>3</sub>H<sub>6</sub> conversions

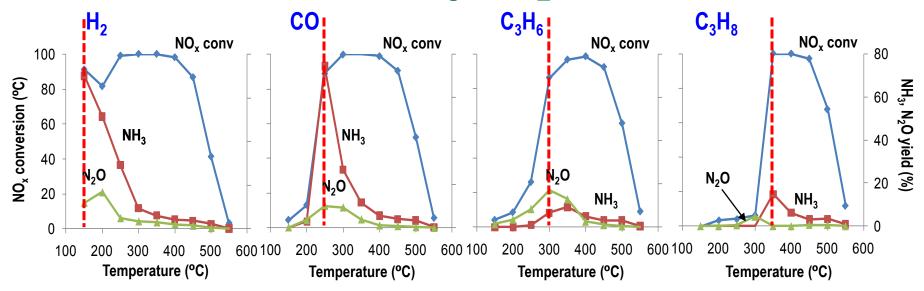
- Similarly independent of lean/rich timing
  - at T>300, >80% conversion all conditions
    - Except 180s lean / 5s rich
- Looks like a typical light off curve
- Represents propylene disappearance rather than complete oxidation







# Reductant type affects spatial distribution of NH<sub>3</sub> chemistry controlling NH<sub>3</sub> & N<sub>2</sub>O yields



- Light-off temperature is highly dependent on reductant type: H<sub>2</sub> < CO < C<sub>3</sub>H<sub>6</sub> < C<sub>3</sub>H<sub>8</sub>
  - At light-off: near max conv. reached using whole LNT
  - Above light-off: max conv. using partial length of LNT
- NH<sub>3</sub> & N<sub>2</sub>O yields peak at light-off T
  - Significant slip possible: NH<sub>3</sub> ~70%; N<sub>2</sub>O ~20%
  - At light-off: reactions forming NH<sub>3</sub> (reductant + stored NO<sub>x</sub>) & N<sub>2</sub>O (NH<sub>3</sub> + stored NO<sub>x</sub>) maximized
  - Above light-off: NH<sub>3</sub>-consuming reactions without generating N<sub>2</sub>O increase (NH<sub>3</sub> + CeO<sub>2</sub>; decomposition)

