

Degradation Mechanisms of Urea Selective Catalytic Reduction Technology

Do Heui Kim, Jong Lee, Chuck Peden

Institute for Integrated Catalysis
Pacific Northwest National Laboratory

Chang H. Kim, Steven J. Schmieg,
Michelle H. Wiegenga, and Se H. Oh
General Motors Global R&D Center

David B. Brown
General Motors Powertrain



Program Managers: **Ken Howden and Gurpreet Singh**

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This presentation does not contain any proprietary, confidential, or otherwise restricted information.

ACE027

Timeline

- Start – December 2008
- Finish – November 2011
- 100% Complete

Budget

- Total project funding
 - DOE – \$400K
 - GM – \$327K
- Final year (FY11) funding:
 - \$100K

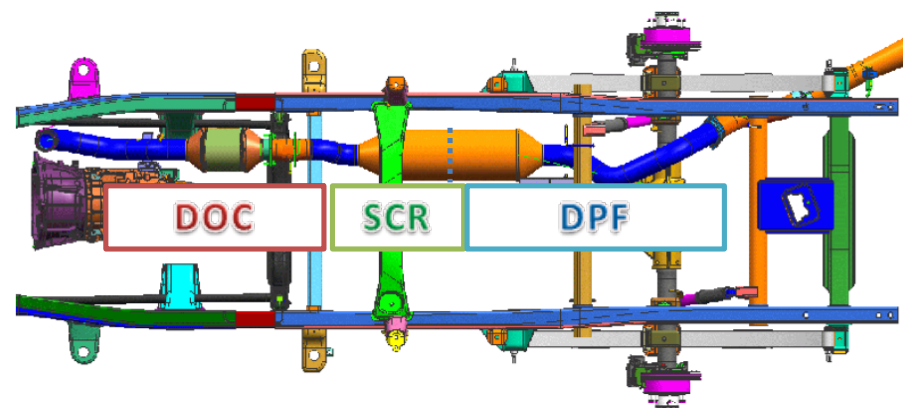
Barriers

- Discussed on next slide

Partners

- Pacific Northwest National Laboratory
- General Motors Global R&D





GM 6.6L Duramax Diesel Engine – 2011 Exhaust System Configuration

- Some of the mechanisms for deactivation of urea SCR and DOC catalysts have been described. However, a detailed understanding of the main factors determining the long-term performance of these catalysts and the interplay between deactivation of the two catalyst systems has yet to be obtained.
- An especially important issue is the relationship between laboratory and vehicle aging. In particular:
 - How well do laboratory aging conditions reproduce sample deactivation in vehicle aged samples at various stages of use?
- Establishing the relevance of rapid laboratory catalyst aging protocols is essential to reducing development cost and time.

- Develop an understanding of the deactivation mechanisms of and interactions between the diesel oxidation catalyst (DOC) and the urea selective catalytic reduction (SCR) catalyst used in light-duty diesel vehicle applications.
- Understand the difference between vehicle aging and aging under laboratory conditions, information essential to provide a rapid assessment tool for emission control technology development.
- Determine the role of the various aging factors impacting long-term performance of these catalyst systems, in order to provide operational information about how catalyst deactivation can be avoided.



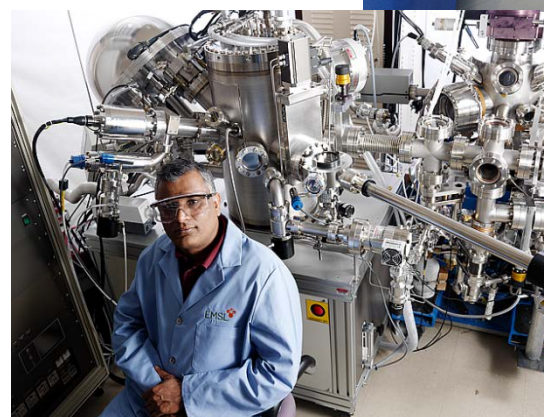
- **Prepare and Process Urea SCR catalyst and DOC catalyst**

- All catalyst samples were provided by GM in monolith form. Both “Model” and “Development” (proprietary) samples were studied in the following forms:

- Fresh and ‘degreened’
 - Lab reactor-aged, oven-aged, and vehicle-aged samples.

- **Utilize catalysis expertise, and state-of-the-art catalyst characterization and testing facilities in PNNL’s IIC to determine deactivation mechanisms and structure/function**

- XRD, XPS, NMR, EPR, TEM/EDX and SEM/EDX
 - NO TPD, H₂ TPR
 - Lab reactor studies





Sample Preparation

- Laboratory Aging
- Vehicle Aging

Measurement

- Catalytic Activity
- Elemental Analysis



DEC-2008 → NOV-2011

Characterization

- *in-situ* XRD
- TEM/EDX
 - XPS
 - NMR
 - EPR
- TPD/TPR

- *Jointly use the new understanding to develop correlations relating the degree of performance deterioration to the catalyst aging parameters.*
- Conference calls were held 5-7 times a year to discuss the results.
- Once a year 'face-to-face' annual reviews are held. The final (close-out) review was held at GM Research in Warren, MI, October 21, 2011.
- Time-line for program was developed and followed (see extra slides below).



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INTEGRATED
CATALYSIS

Thermal Durability of Cu-CHA NH₃-SCR Catalysts for Diesel NO_x Reduction

- Laboratory Aging Test Results
 - Reactor and oven aging matrix
 - 4-step SCR test protocol results (NO oxidation, NO_x conversion, NH₃ oxidation, NH₃ storage)
 - Characterization results (XRD, ²⁷Al-NMR, XPS, TEM/EDX)
- Comparison of Vehicle-Aged and Laboratory-Aged Catalysts



Model Catalyst Formulations		
Component	Washcoat Technology	PGM Loading (g/ft ³)
DOC	PGM w/ zeolite	50 g/ft ³ 10 Pt/1Pd
DOC	PGM w/ zeolite	50 g/ft ³ 5 Pt/1Pd
DOC	PGM w/ zeolite	50 g/ft ³ 2 Pt/1Pd
DOC	PGM w/o zeolite	50 g/ft ³ 5 Pt/1Pd
SCR	Cu-zeolite	
SCR	Fe-zeolite	

PGM RATIO EFFECT

ZEOLITE EFFECT

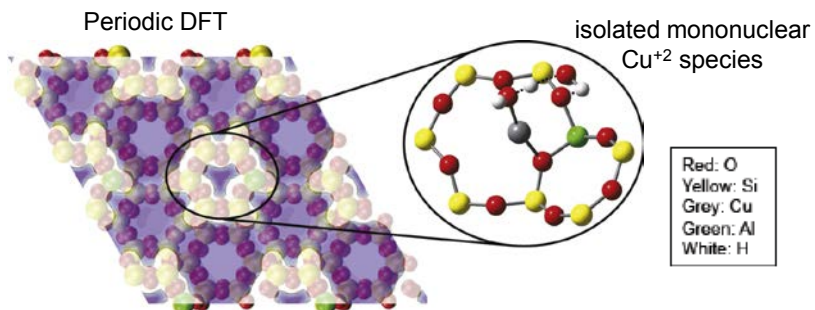
large pore zeolite ~7Å

- HC poisoning issues
- Thermal durability issues



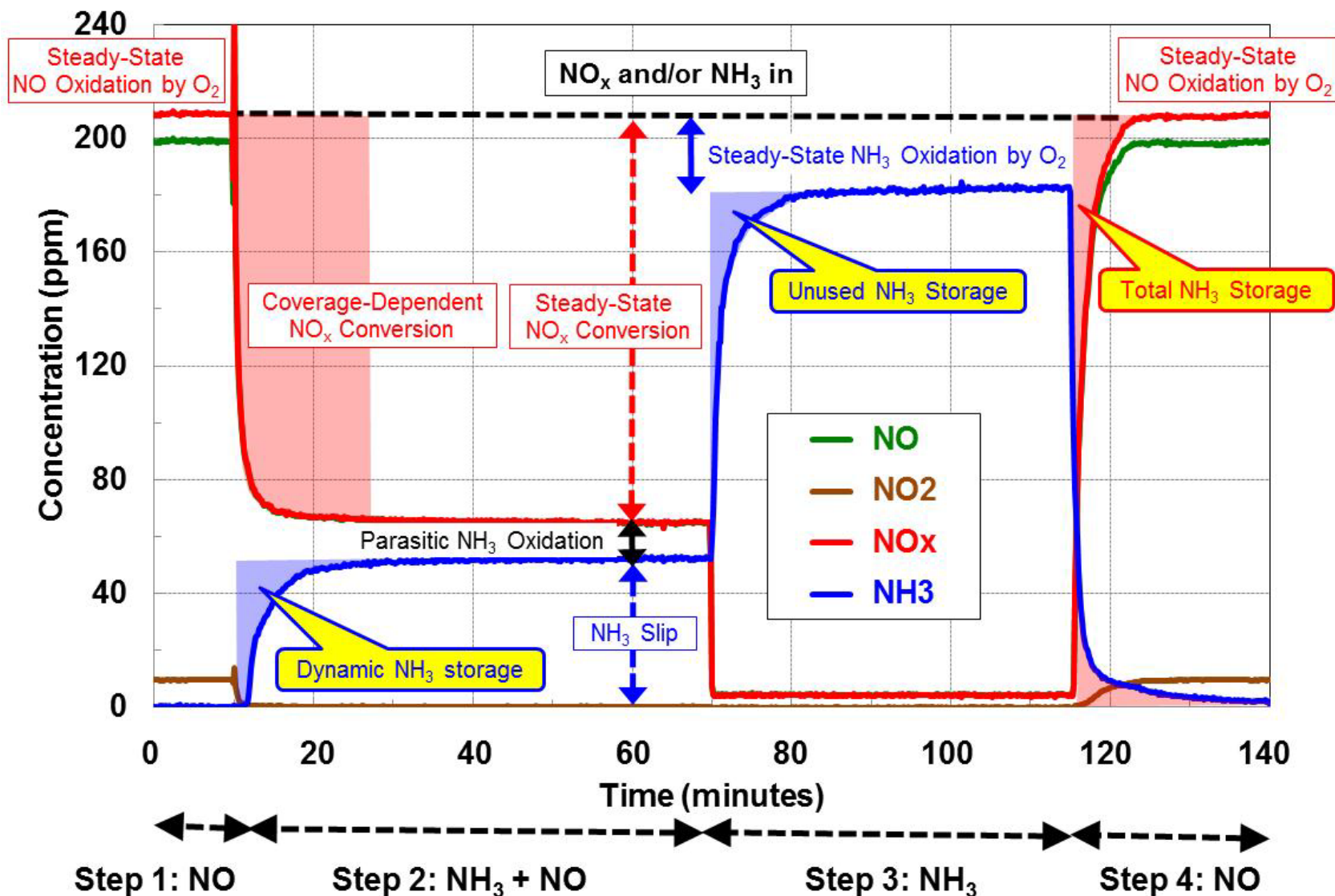
small pore CHA zeolite ~3.8Å

2011 Duramax Catalyst Formulations		
Component	Washcoat Technology	PGM Loading (g/ft ³)
DOC	PGM w/ zeolite	49 g/ft ³ 7 Pt/1Pd
SCR	Cu-zeolite	



Reference: **BASF Corporation**, US Patent 7,610,662 (2009) – 100% ion exchanged chabazite: 2.8% Cu as copper metal, silica to alumina molar ratio ~ 35:1

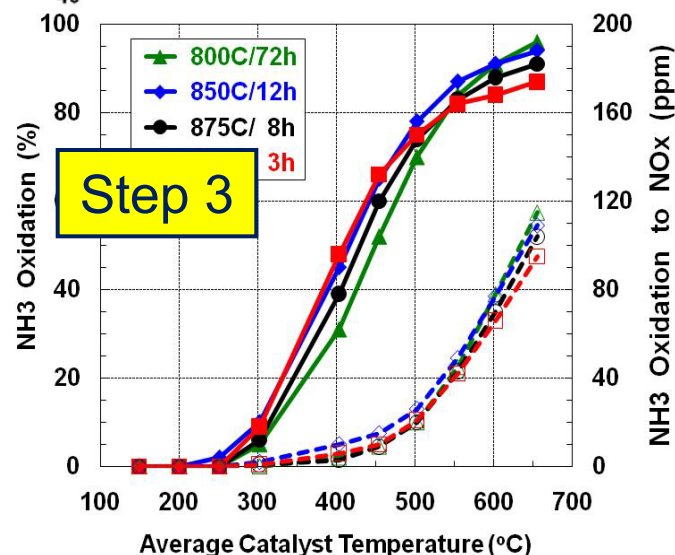
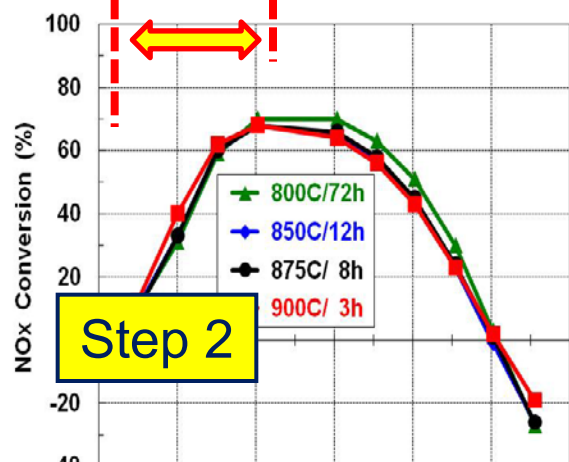
4-Step SCR Test Protocol Used



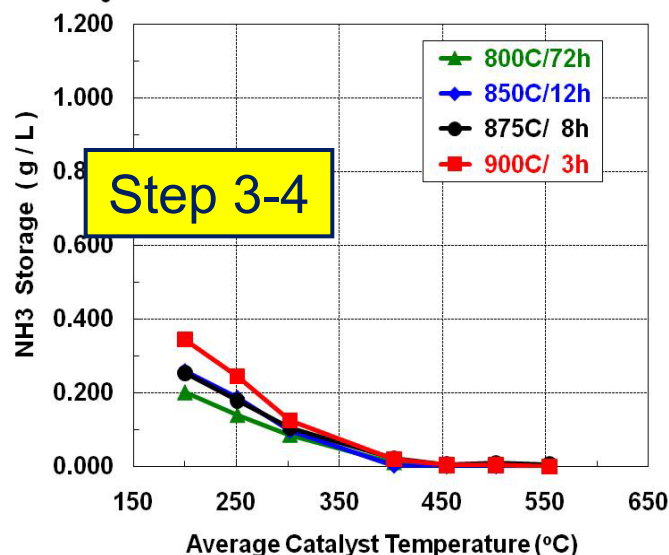
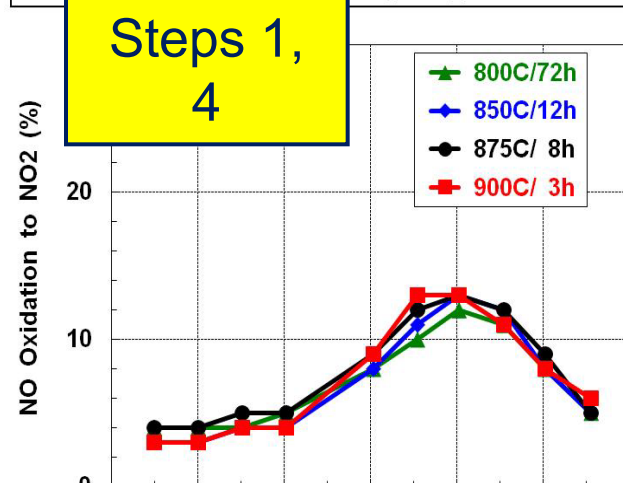
Reference: K. Kamasamudram, N.W. Currier, X. Chen, and A. Yezerets, *Catalysis Today*, 151 (2010) 212-222.



Development Cu-zeolite (Reactor Aged) / SV ~ 30,000 h⁻¹
10% O₂ / 5% H₂O / 8% CO₂ / 200 ppm NO / 200 ppm NH₃ (α = 1.0)



Development Cu-zeolite (Reactor Aged) / SV ~ 30,000 h⁻¹
10% O₂ / 5% H₂O / 8% CO₂ / 200 ppm NO



- Selection criteria of 70% NO_x conversion based on:
 - typical operating T ~ 200 – 400°C
 - 1.0 g/mi engine out NO_x
 - 0.2 g/mi emission standard
 - OBD limit at 0.3 g/mi (1.5x standard)

Similar activity loss occurs at shorter times as aging temperature increases

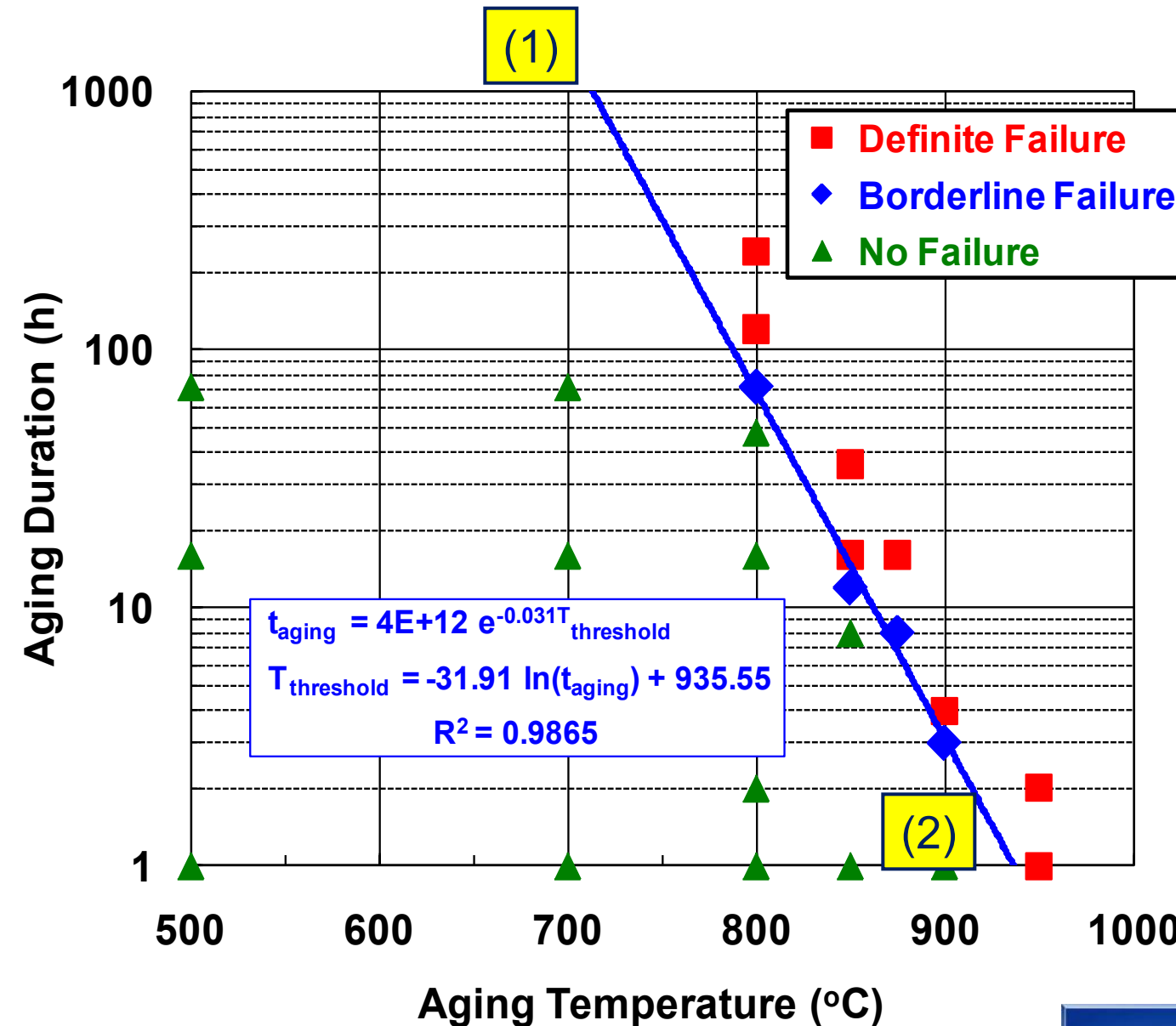


SCR Reactor Aging Matrix

Aging Temperature (°C)	Aging Duration (h) [Reactor Aged: 6 L/minute AIR + 12.5% H ₂ O]											
	1	2	3	4	8	12	16	36	48	72	120	240
500	✓						✓			✓		✓
700	✓						✓			✓		✓
800	✓	✓			✓		✓	✓	✓	✓	✓	✓
850	✓	✓		✓	✓	✓	✓	✓				
875				✓	✓		✓					
900	✓	✓	✓	✓								
950	✓	✓										

→ 33 catalysts aged and tested





Implications for operation of SCR catalyst:

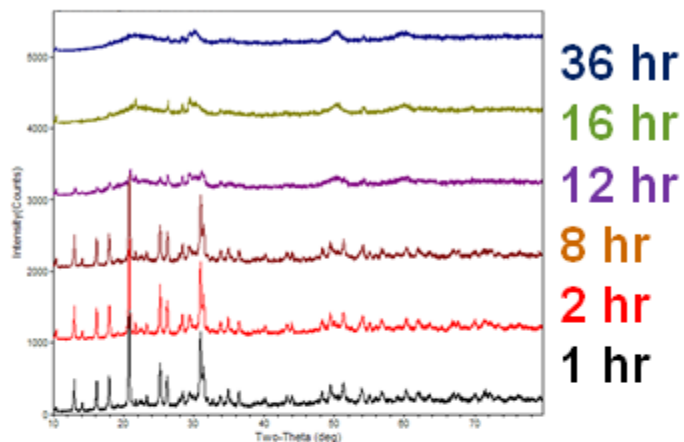
(1) Catalyst would still provide sufficient NO_x conversion efficiency up to 1000 h at 700°C

(2) The “not-to-exceed” operating temperature of this catalyst would be 935°C

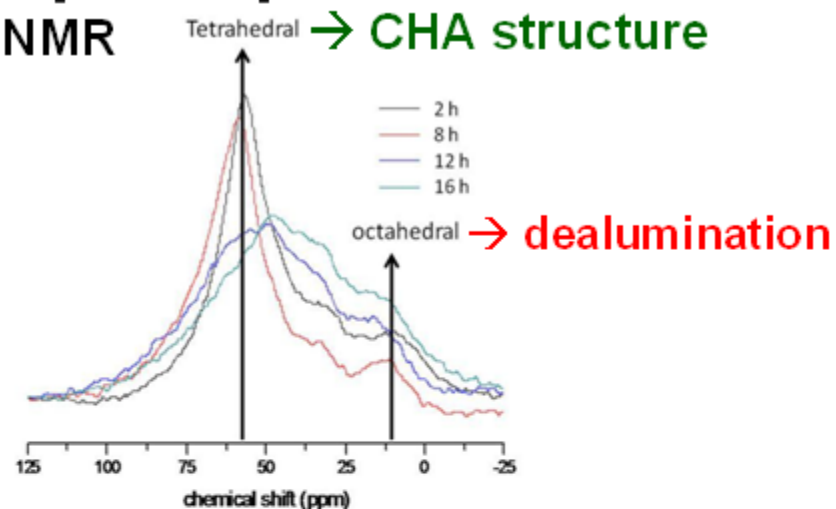


1. Changes in zeolite structure → abrupt collapse

XRD

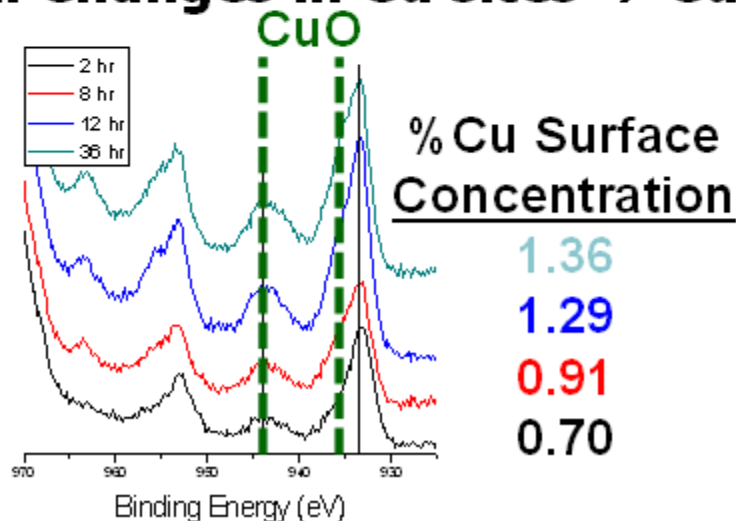


^{27}Al NMR

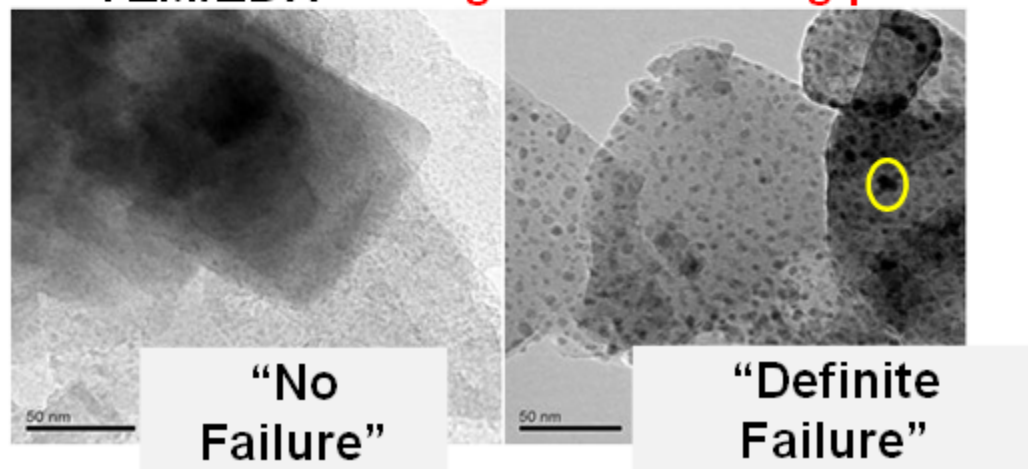


2. Changes in Cu sites → CuO agglomeration

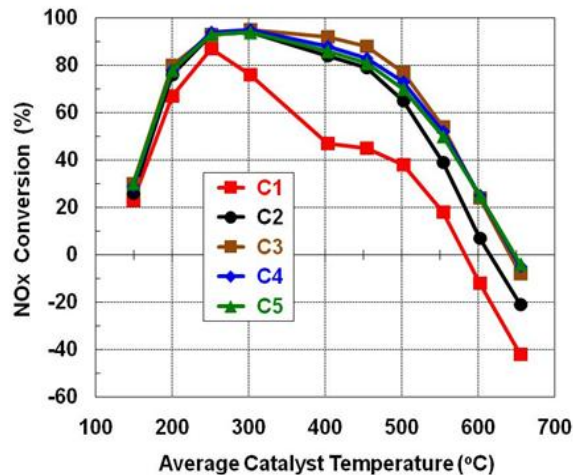
XPS



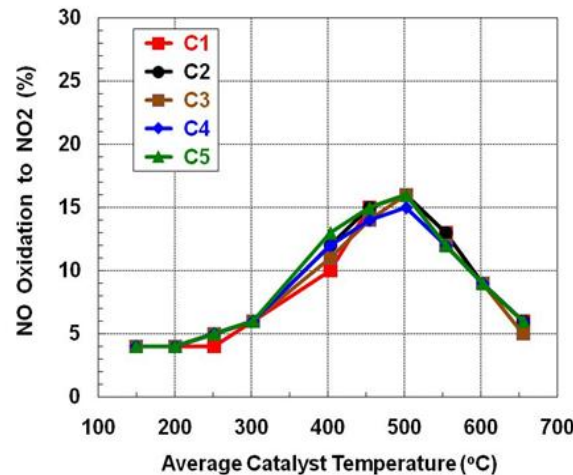
TEM/EDX → Large Cu-containing particles



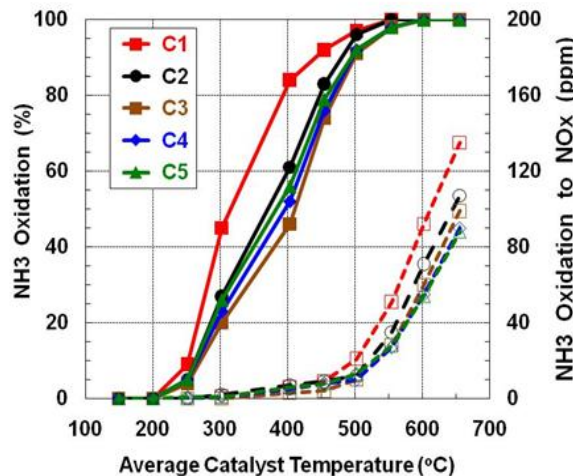
Development Cu-zeolite (Engine Aged 135K) / SV ~ 30,000 h⁻¹
10% O₂ / 5% H₂O / 8% CO₂ / 200 ppm NO / 200 ppm NH₃ ($\alpha = 1.0$)



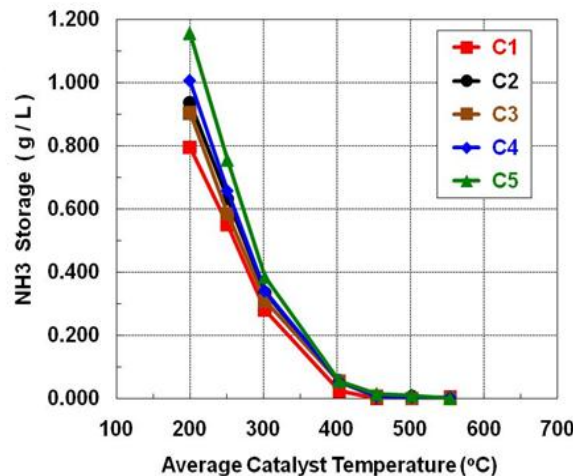
Development Cu-zeolite (Engine Aged 135K) / SV ~ 30,000 h⁻¹
10% O₂ / 5% H₂O / 8% CO₂ / 200 ppm NO



Development Cu-zeolite (Engine Aged 135K) / SV ~ 30,000 h⁻¹
10% O₂ / 5% H₂O / 8% CO₂ / 200 ppm NH₃



Development Cu-zeolite (Engine Aged 135K) / SV ~ 30,000 h⁻¹
10% O₂ / 5% H₂O / 8% CO₂ / 200 ppm NO / 200 ppm NH₃ ($\alpha = 1.0$)



As function of axial position:

- NO_x conversion decreases with first inch (C1) much worse at high T
- No difference in NO oxidation
- NH₃ oxidation and NO_x formation worse for C1
- NH₃ storage decreases C5-C1
- XRD results indicate C1 slightly more thermally damaged

→ **C4** selected as representative of vehicle aging

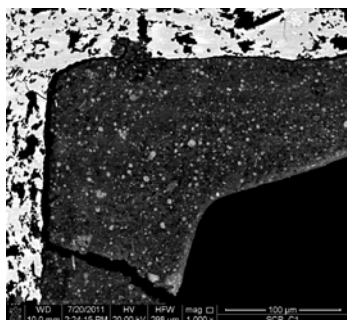


- 1) Primary deactivation modes are the destruction of the zeolite structure and the agglomeration of the active metal, well correlated with DeNO_x activity.
- 2) Damage equation for laboratory aging with respect to time and temperature was obtained.
- 3) Determined reactor aging conditions and oven aging conditions required to reproduce vehicle aged and threshold aged (OBD) catalyst performance:

Reactor Aged: 135,000 Mile = 16h / 800°C , OBD Threshold = 12h / 850°C

Oven Aged: 135,000 Mile = 72h / 700°C , OBD Threshold = 54h / 850°C

- 4) Detailed XRD, XPS, ICP and SEM measurements of the vehicle-aged catalysts (not shown) showed high concentrations of Pt and engine oil components on at the front end of the monolith.



Analyzed for C, O, Al, Si, P, S, Cl, Ca, and Zn as a function of depth for monolith cores from the front to the back of the brick.



Thermal Durability of Diesel Oxidation Catalysts

- Laboratory Aging Test Results
 - Deactivation as a function of PGM Ratio
 - Effects of the additional zeolite
 - Characterization results (XRD, XPS, TEM/EDX)
- Deactivation of Vehicle-Aged Catalysts
- Comparison of Vehicle-Aged and Laboratory-Aged Catalysts
 - Accelerated aging protocol



Model Formulations		
Component	Washcoat Technology	PGM Loading (g/ft ³)
DOC	PGM w/ zeolite P10Z-F	50 g/ft ³ 10 Pt/1Pd
DOC	PGM w/ zeolite P5Z-F	50 g/ft ³ 5 Pt/1Pd
DOC	PGM w/ zeolite P2Z-F	50 g/ft ³ 2 Pt/1Pd
DOC	PGM w/o zeolite	50 g/ft ³ 5 Pt/1Pd
SCR	Cu-zeolite	N/A
SCR	Fe-zeolite	N/A

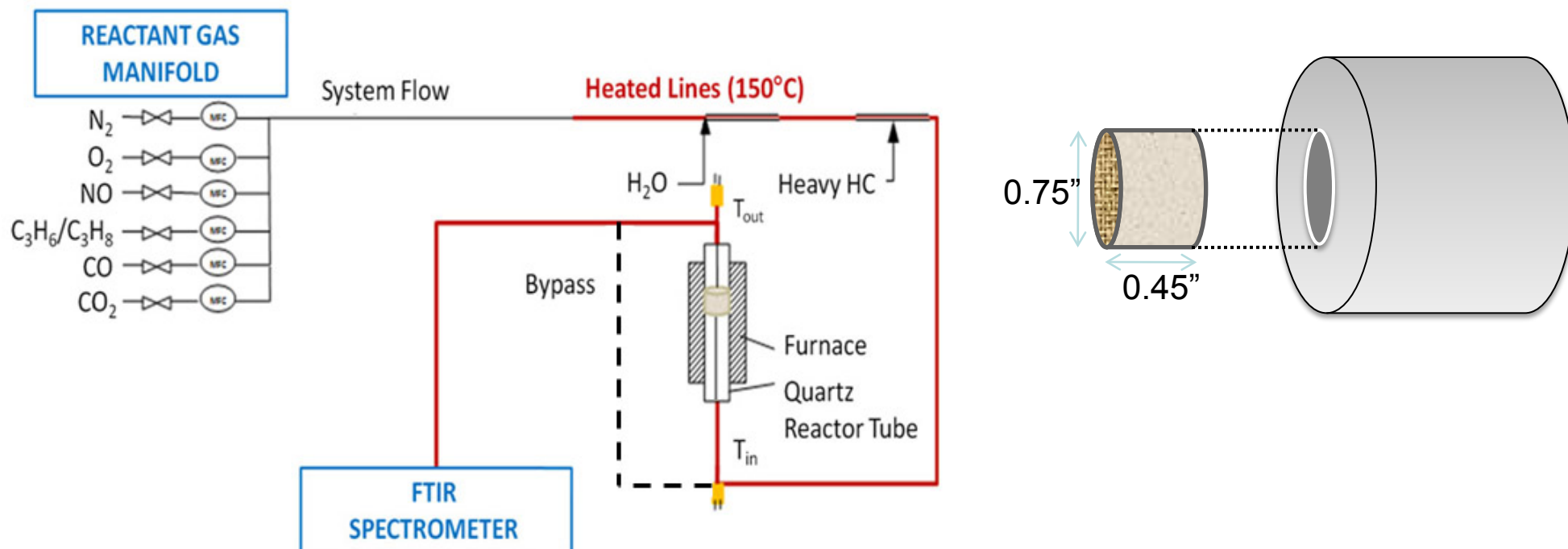
PGM RATIO EFFECT

ZEOLITE EFFECT

2011 Duramax Catalyst Formulations		
Component	Washcoat Technology	PGM Loading (g/ft ³)
DOC	PGM w/ zeolite	49 g/ft ³ 7 Pt/1Pd
SCR	Cu-zeolite	N/A

2011 DURAMAX FORMULATION



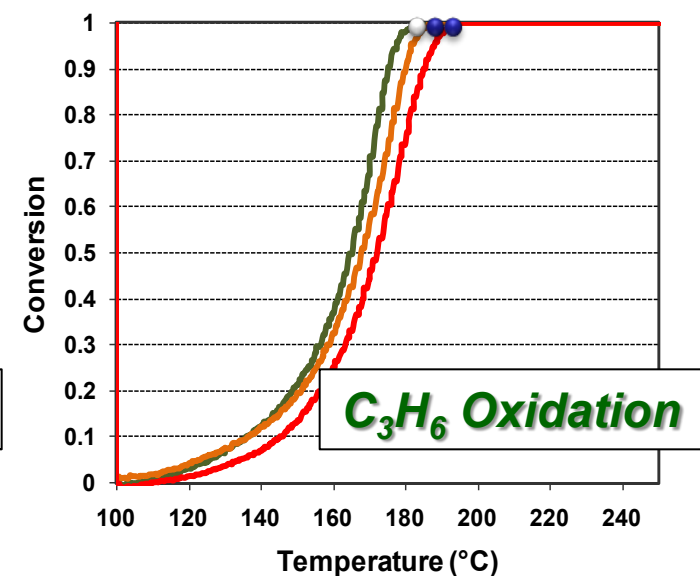
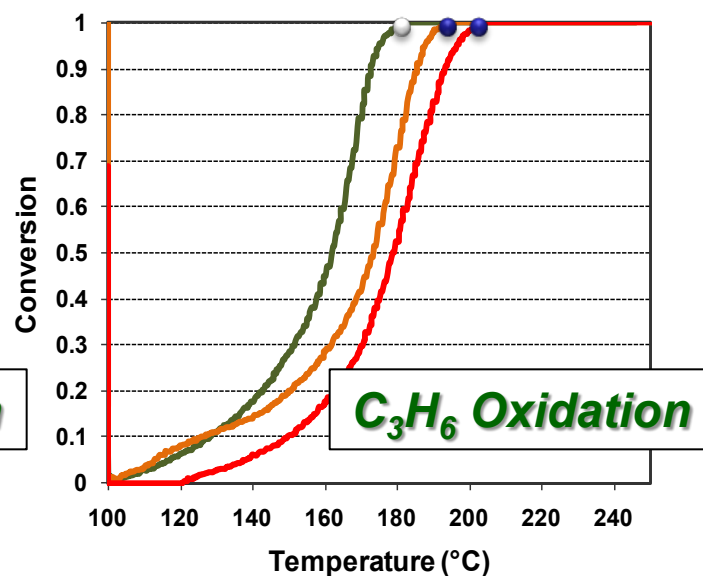
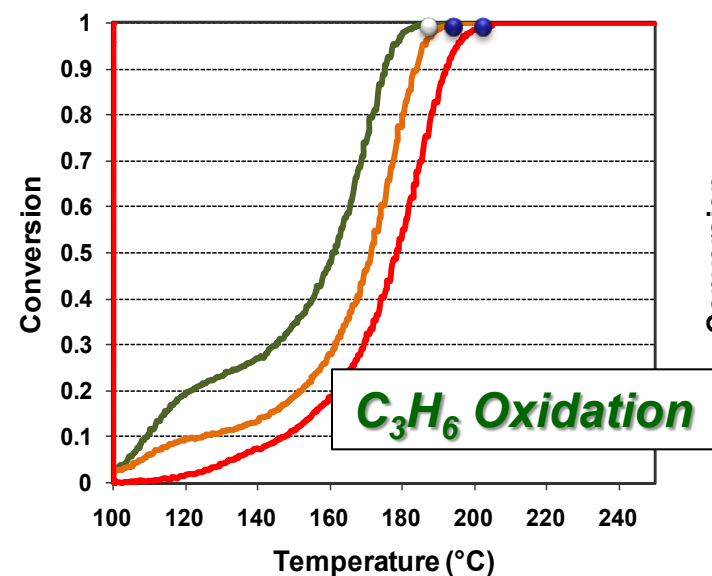
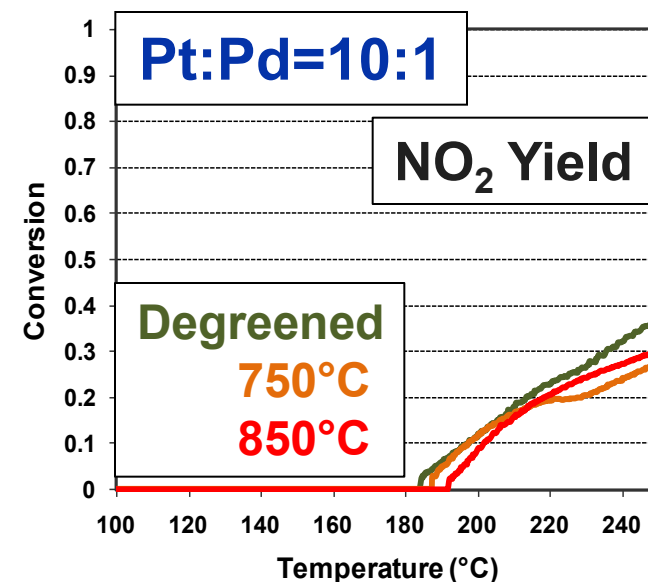
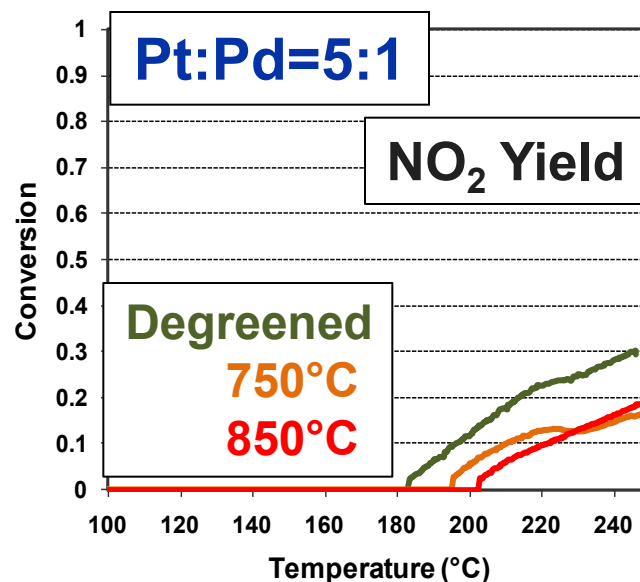
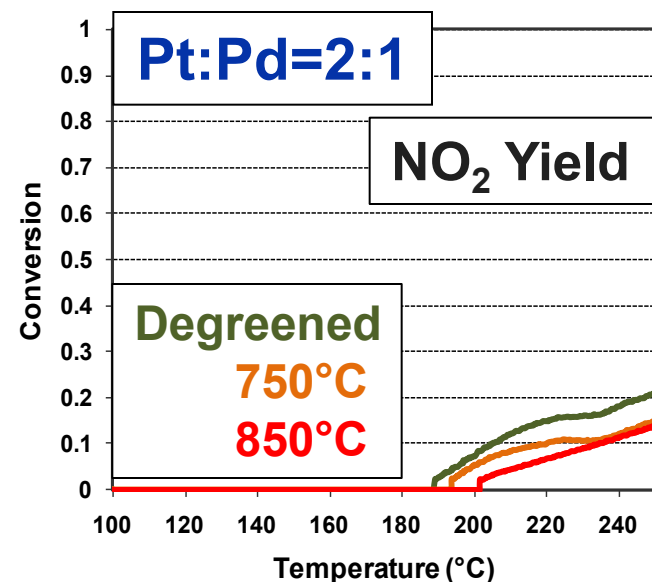


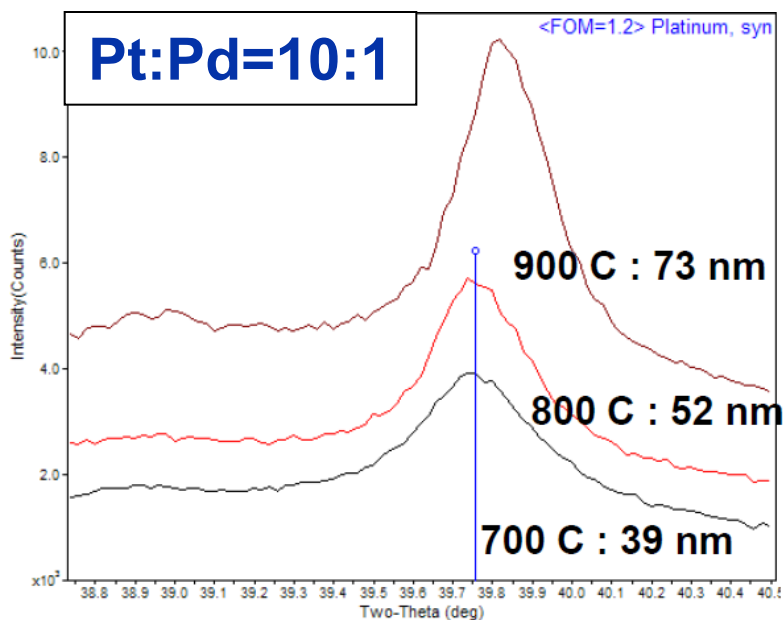
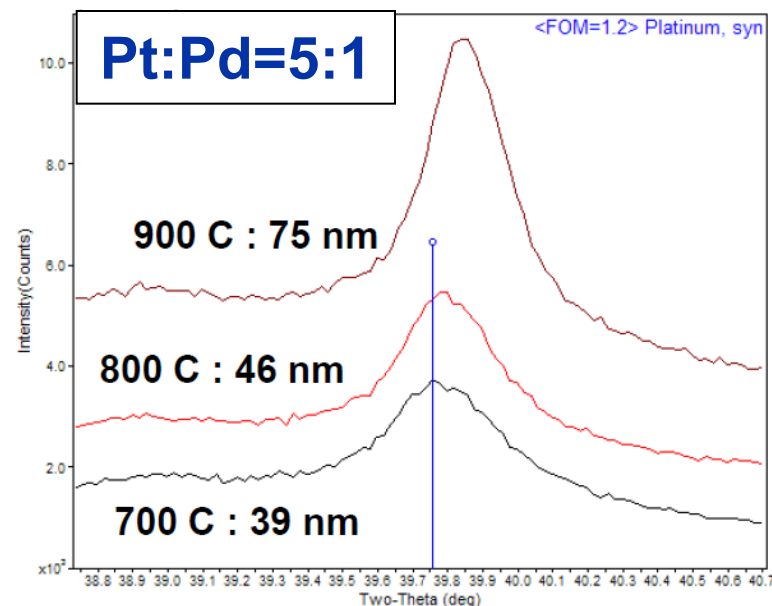
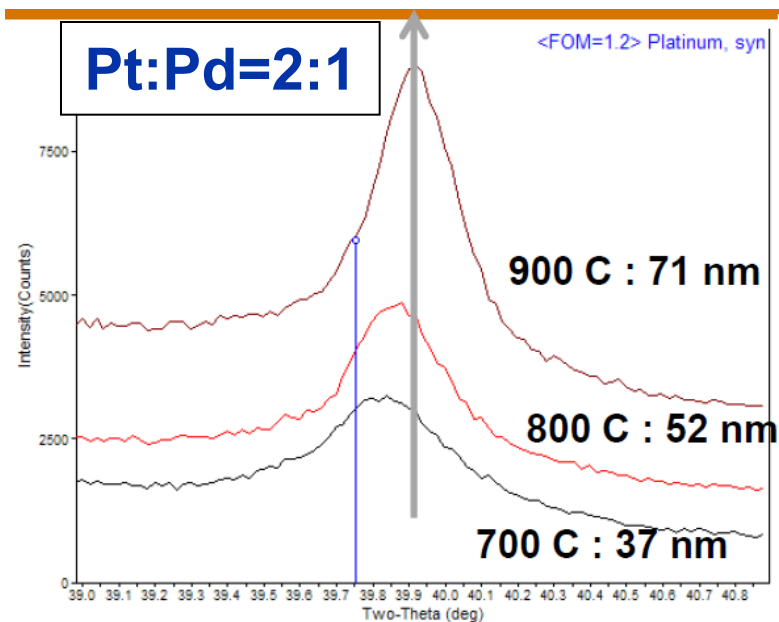
- GHSV: 30,000 h⁻¹

NO	HC(C ₁)	C ₃ H ₆ :C ₃ H ₈	O ₂	H ₂ O	N ₂
200 ppm	1050 ppm	2.9 : 1	8 %	8 %	Balanced

- Measured Components: NO/NO₂/HC/CO/CO₂/H₂O



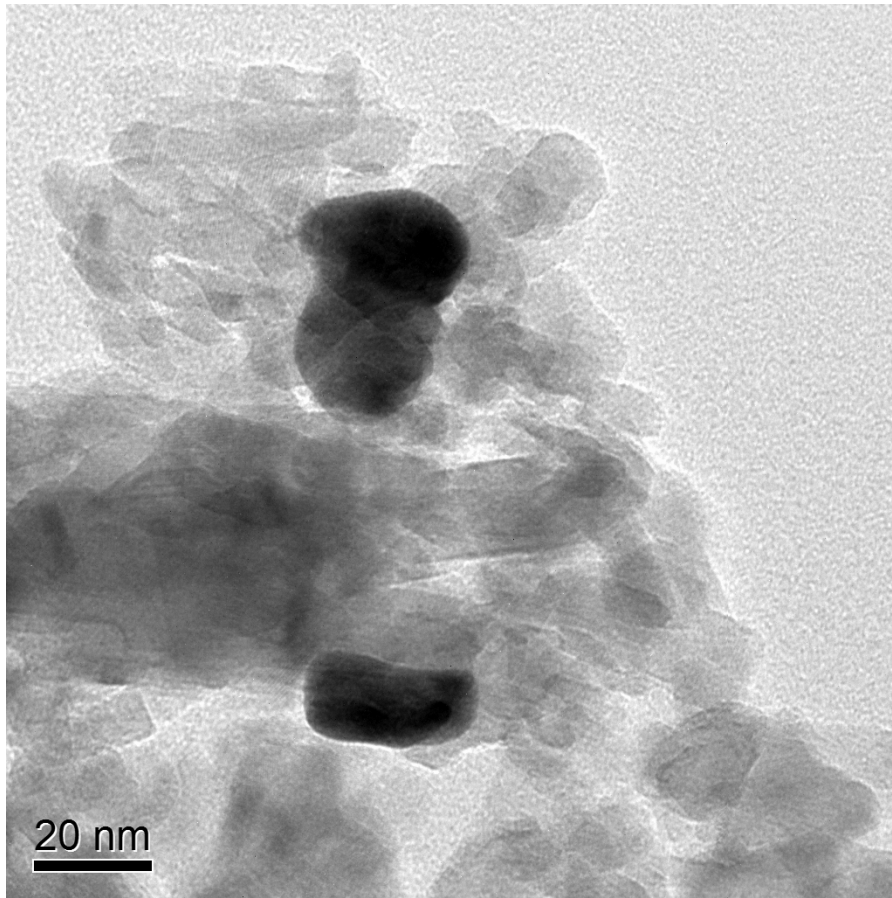




- XRD peak height, shape and location were used to probe deactivation.
- Both particle growth and an increase in Pd concentration occurred at temperatures above 800 °C.

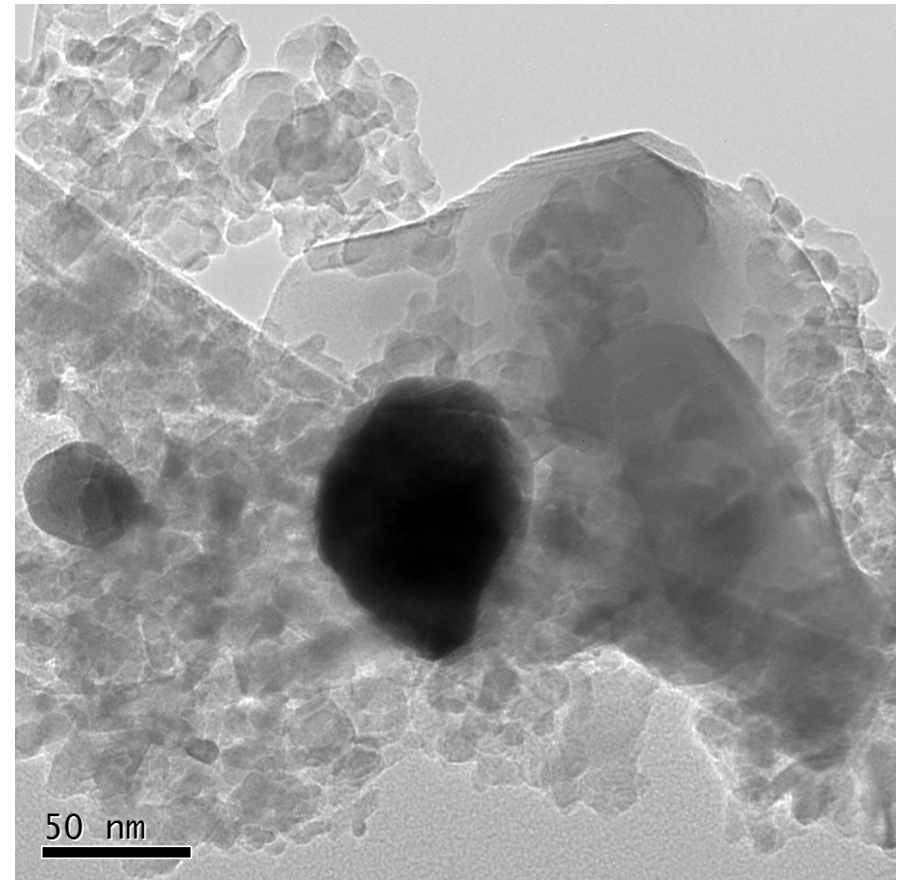


P2Z-700



Size of particles: 15 - 80 nm
Average Pt/Pd = 84/16

P2Z-900



Size of particles: 30 – 100 nm
Average Pt/Pd = 74/26

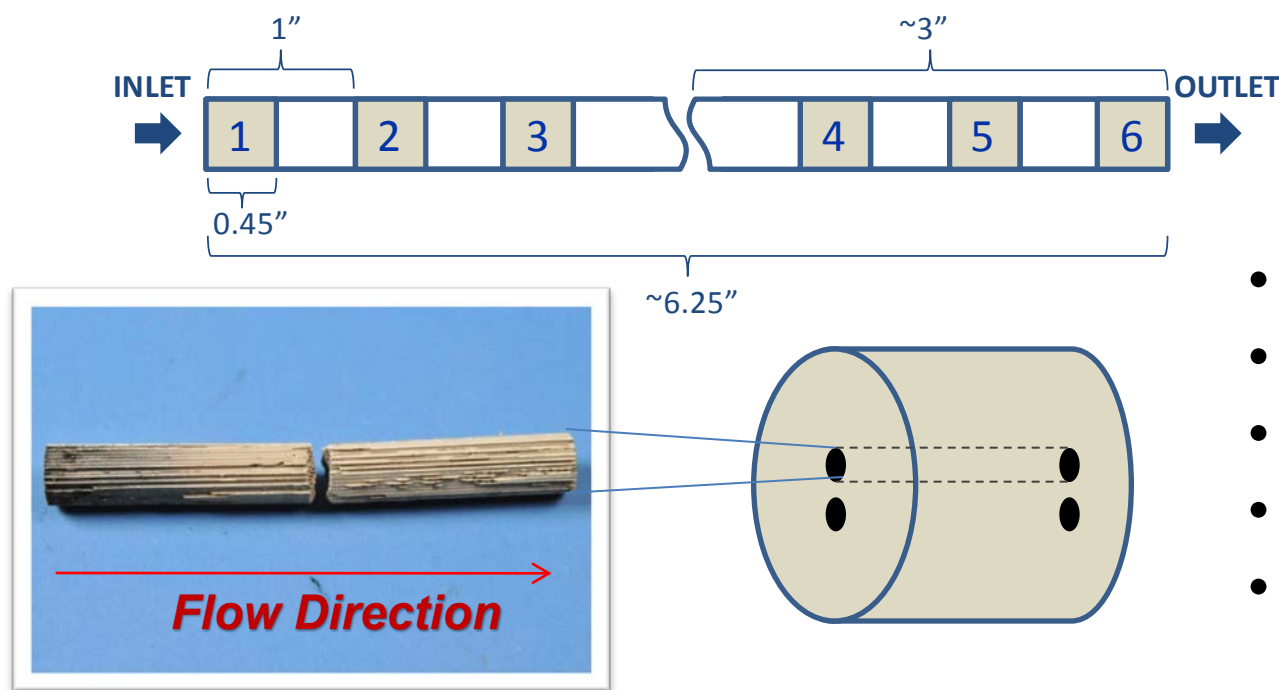
TEM/EDX show the more enrichment of Pd in the severely aged particles.



- Trends in performance were observed over DOC samples with different Pt-Pd ratios
- Hydrothermal stability improves when Pd is present
- Light-off performance significantly improved with presence of both Pt and Pd
- NO₂ yield depends directly on Pt-content
- These observations suggest that the Pt-Pd ratio of the DOC can be adjusted to maximize performance depending on the application



- Full-size DOC was taken from a vehicle driven for 135,000 miles.
- One 0.75" x 0.45" piece from each inch of the core was tested.
- Both HC and NO oxidation measured as before.



CHARACTERIZATION

- XRD (1,2,3,4,5,6)
- TEM/EDX (1,3,5)
- XPS (1,3,4,6)
- In depth SEM/EDX (1, 6)
- In depth XPS (1): before and after oxidation

- Activity of vehicle-aged parts falls between fresh (degreened, "F") and prior baseline ("OB") laboratory aging
- Exception – Inlet (sample 1) before soot removal
- Need an intermediate aging method in order to mimic vehicle aging

Time (h)	600°C	700°C	750°C	850°C	950°C
2.4			F		
4					
16					
24					
48					
72			OB		

All with 10% H₂O



Too Mild

Too Severe

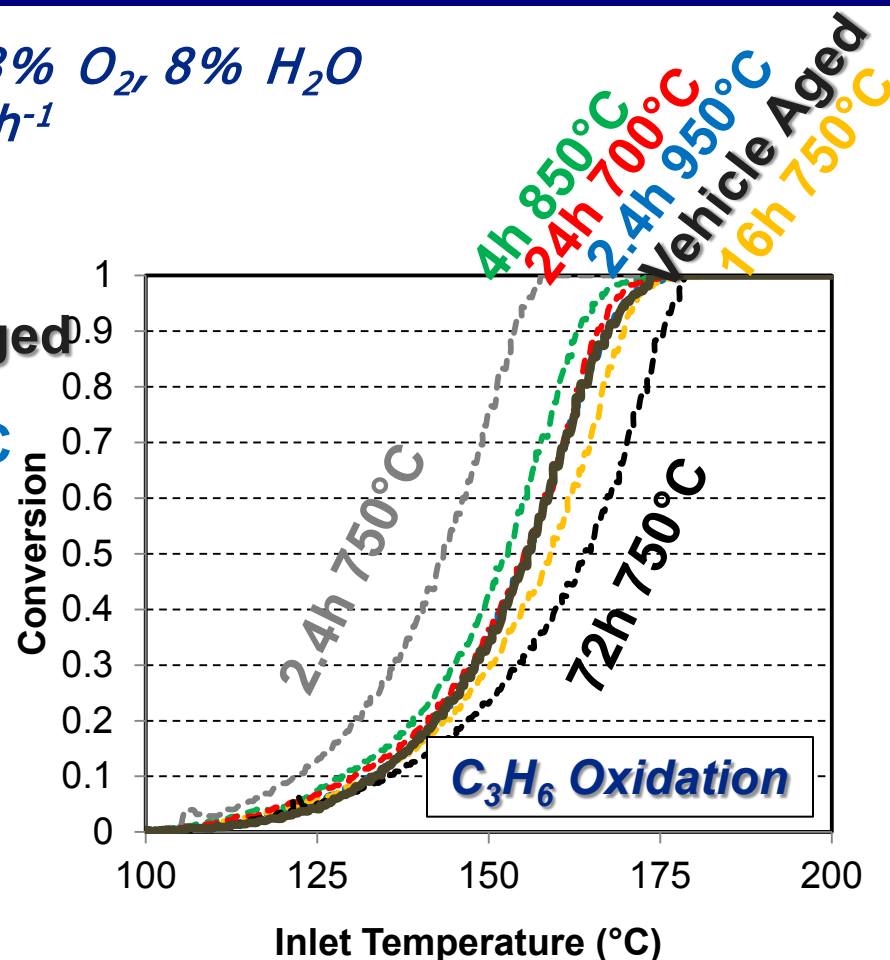
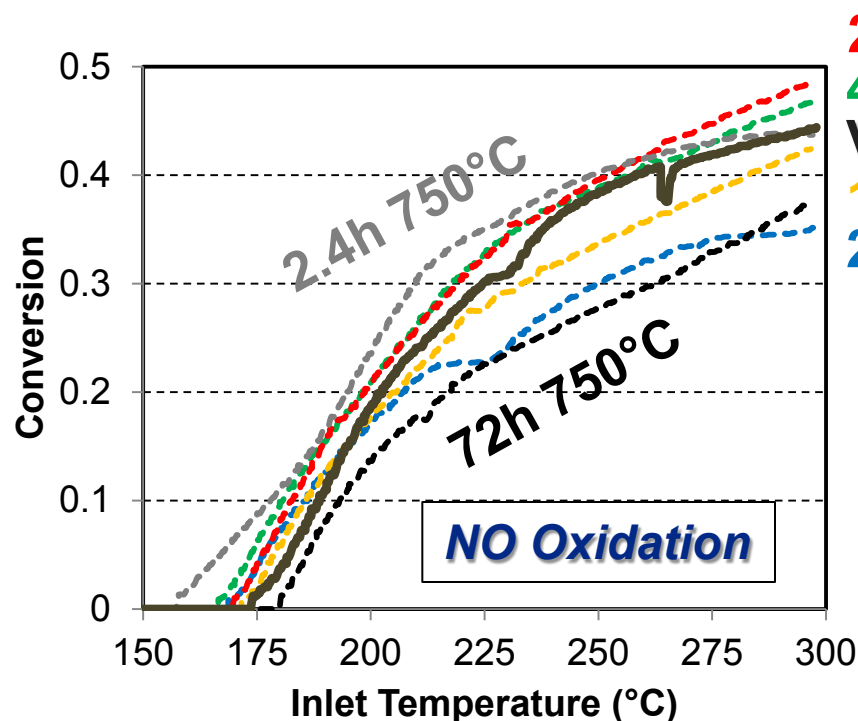
Possibly Comparable

Tested

- Shorten time for hydrothermal aging
- Improved approximation of deactivation due to vehicle aging

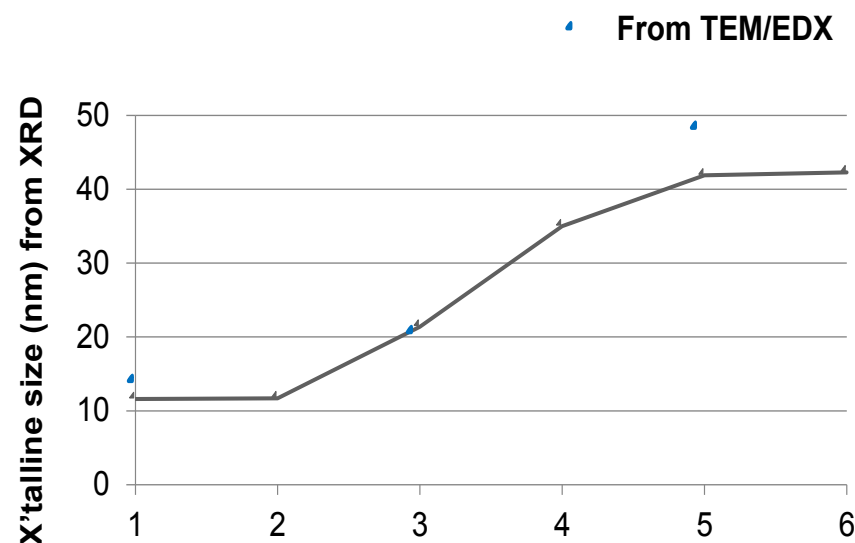
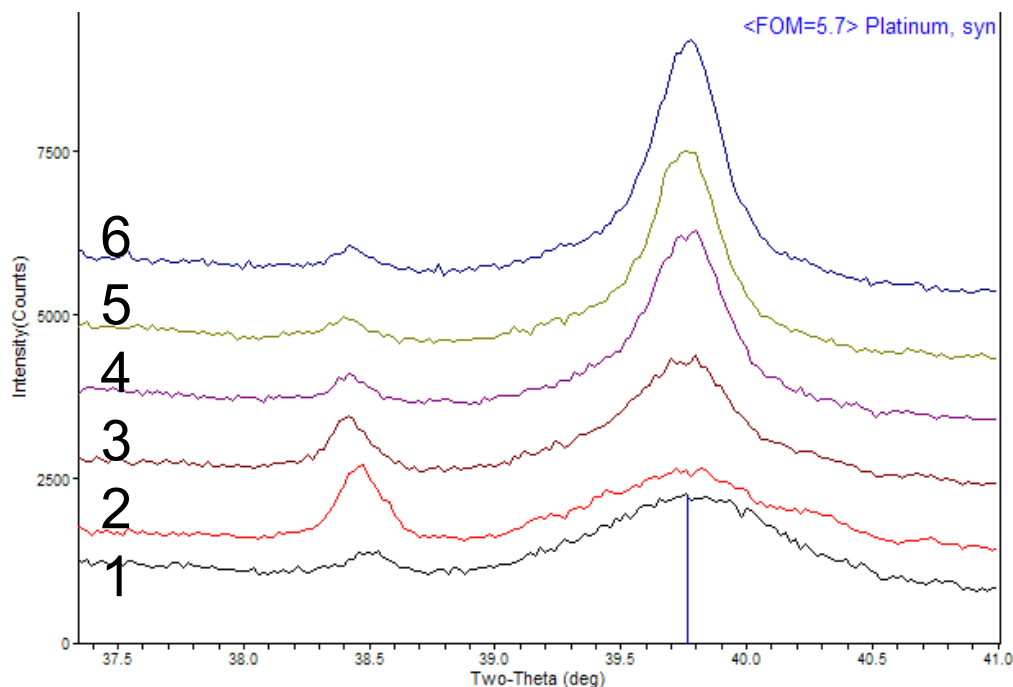


200 ppm NO, 260 ppm C₃H₆, 90 ppm C₃H₈, 8% O₂, 8% H₂O
Temp. Ramp Rate: 2 °C/min, GHSV: 30,000 h⁻¹



- 24h at 700°C give similar NO and HC oxidation performance





- XRD and TEM results are fully consistent with each other.
- PM particles are Pd rich crystallites that increase in size towards the back end (#s 5 and 6) of the DOC monolith.

- The cause of deactivation on the vehicle-aged DOC includes both thermal sintering as well as sulfur poisoning of the inlet. Thermal damage is more severe at the exit of the monolith, not the inlet.
- Deactivation Mechanism for NO oxidation
 - Accumulation of soot – reversible
 - Sintering of Pt-Pd particles – irreversible
- Deactivation Mechanism for CO/HC oxidation
 - Accumulation of sulfur – reversible
 - Sintering of Pt-Pd particles – irreversible
- Correlation was developed to accelerate the aging protocol, and a more accurate prediction of a 135,000 mile vehicle aged DOC was achieved.



- N/A – program completed



- The Urea SCR technology coupled with a DOC system is being implemented by GM as an effective path to meeting emission standards for 2010 and beyond for light-duty diesel vehicles.
- PNNL's role has been to obtain a fundamental understanding of DOC and SCR catalyst deactivation mechanisms through the use of state-of-the-art catalyst characterization methods, with the ultimate goal of developing a "damage equation" relating performance deterioration to the catalyst aging parameters..
- Technical highlights from this project to date have included:
 - The primary deactivation mode identified in aged DOCs is precious metal alloying and sintering.
 - The primary deactivation modes in SCR catalysts are the destruction of the zeolite structure and the agglomeration of the active metal.
 - Correlation between lab-aged and vehicle-aged samples has led to the development of laboratory rapid aging protocols.
- *This has been a highly interactive and collaborative program between GM and PNNL.*

