

Catalytic Nanofabrication for Improved Energy Storage





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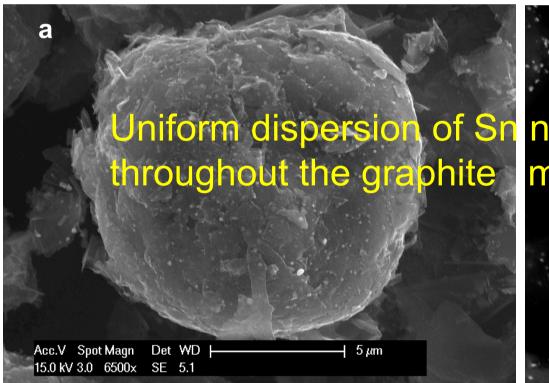
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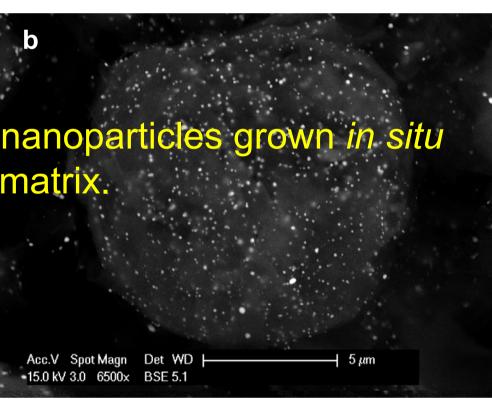


ABSTRACT

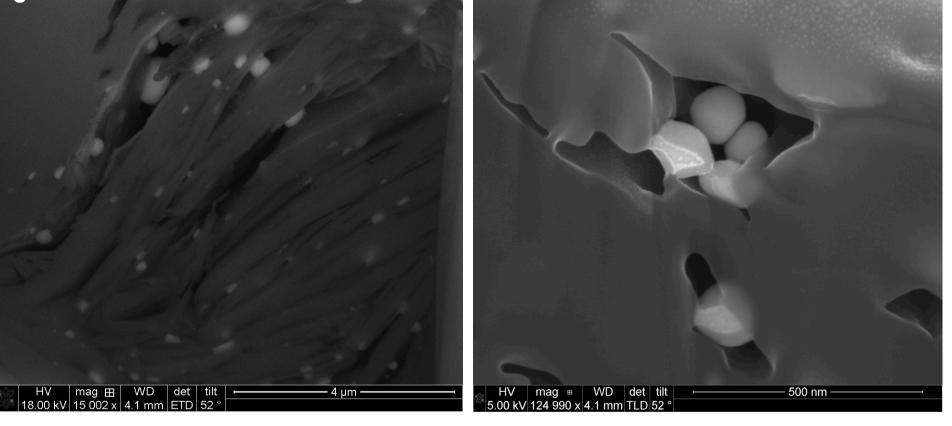
- We describe a new low cost method for kinetically controlled catalytic synthesis of nanocomposite anodes, cathodes and protectants for high-power Li ion batteries (LIBs).
- Anodes are nanocomposites of Sn or Si nanocrystals catalytically grown *in situ* and uniformly dispersed in compliant and conductive graphite or carbon nanotube matrices. They exhibit exceptionally high power, electrochemical capacity and stable cyclability.
- Cathodes are catalytically synthesized as nanocomposites of nanocrystalline metal oxides in highly conductive CNT matrices. They exhibit high voltage, high power and high energy density.
- BaTiO₃ and BaSrTiO₃ nanocrystals catalytically synthesized in high yield and purity yield fine-grained nanocrystalline ceramics with strong positive thermal coefficients of resistivity (PTCR) that can be used as internal protectants against thermal runaway.
- Syntheses of the anodes, cathodes and PTCR protectant are affordably scalable.

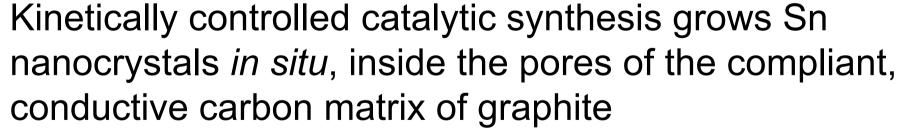
1. HIGH-POWER NANOCOMPOSITE ANODE

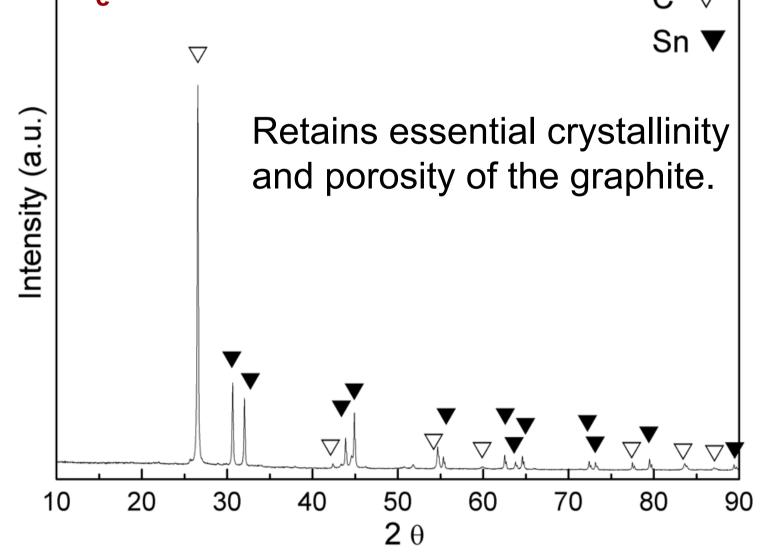




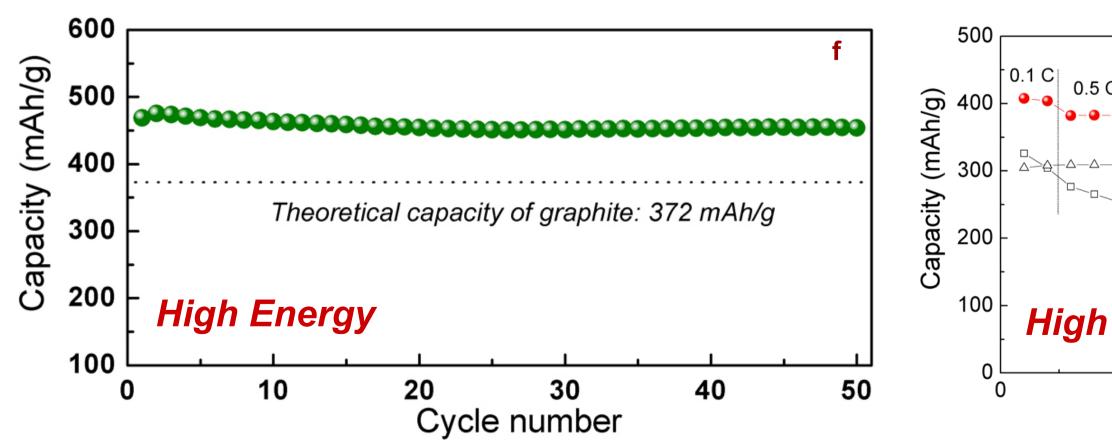


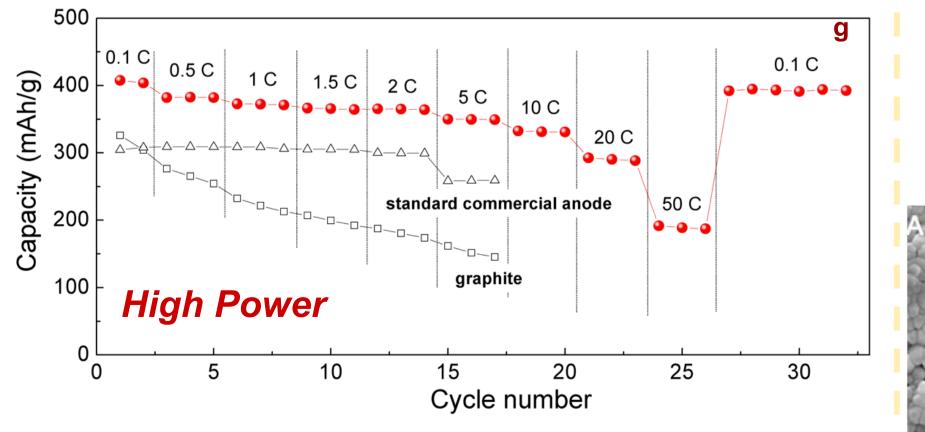






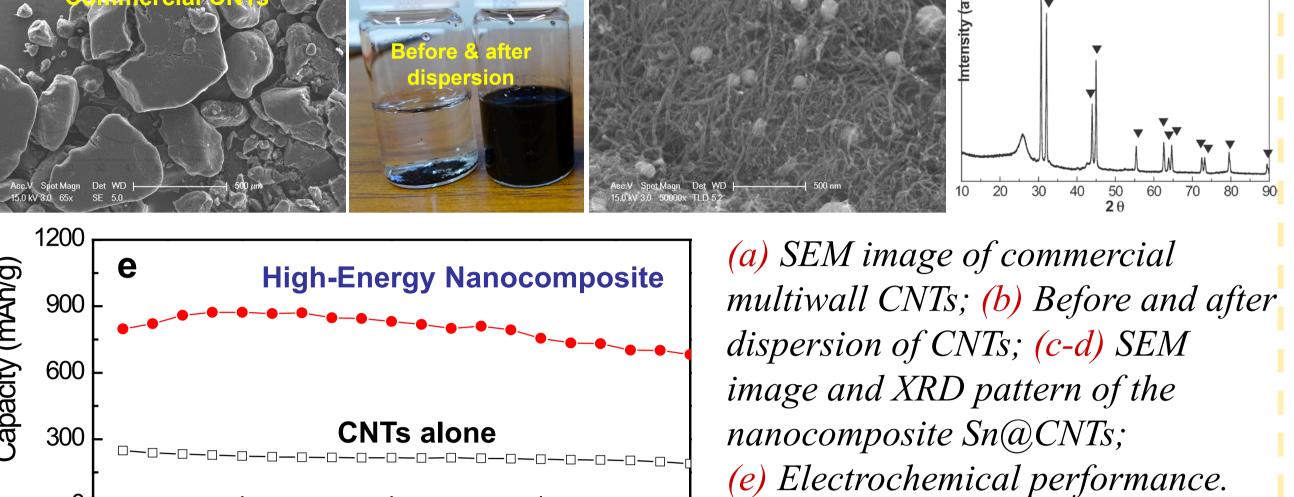
• The resilient graphite "breathes" to accommodate large volume changes accompanying each cycle of Li alloying and de-alloying on charging and discharging, yielding exceptionally high power, stability and cyclability:



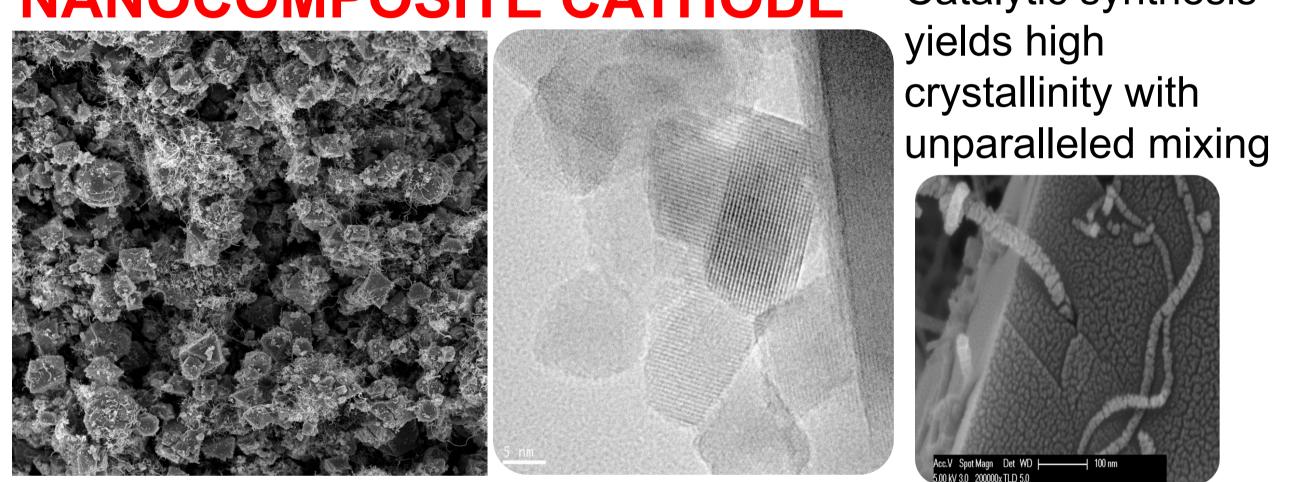


(a) SEM image of the Sn@graphite composite after surface modification; (b) Backscattered electron image of the composite showing nanocrystalline Sn; (c-d) Cross-section images of the Sn@graphite composite obtained with focused ion beam and TEM; (e) XRD pattern of the nanocomposite; (f-g) Cyclic performance and rate capability of the Sn@graphite composite, respectively.

 Catalytic growth of Sn nanocrystals inside the conductive and resilient matrix of dispersed multiwall carbon nanotubes:



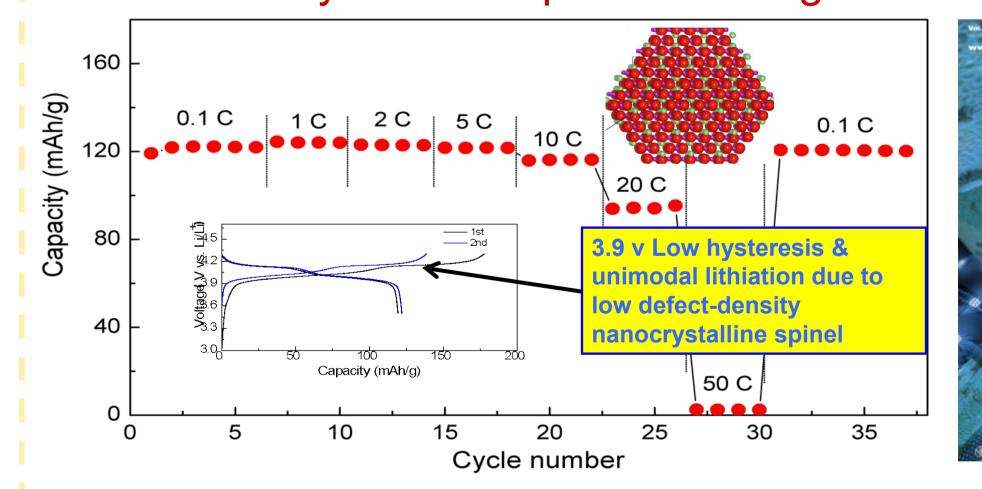
2. HIGH-POWER, HIGH VOLTAGE CNT-LiMn₂O₄
NANOCOMPOSITE CATHODE Catalytic synthesis
yields high



- 96% retention of original capacity after discharge at 10 C;
- >80% capacity retention at the exceptionally high rate of 20 C;
 Full recovery after complete discharge at 50C:

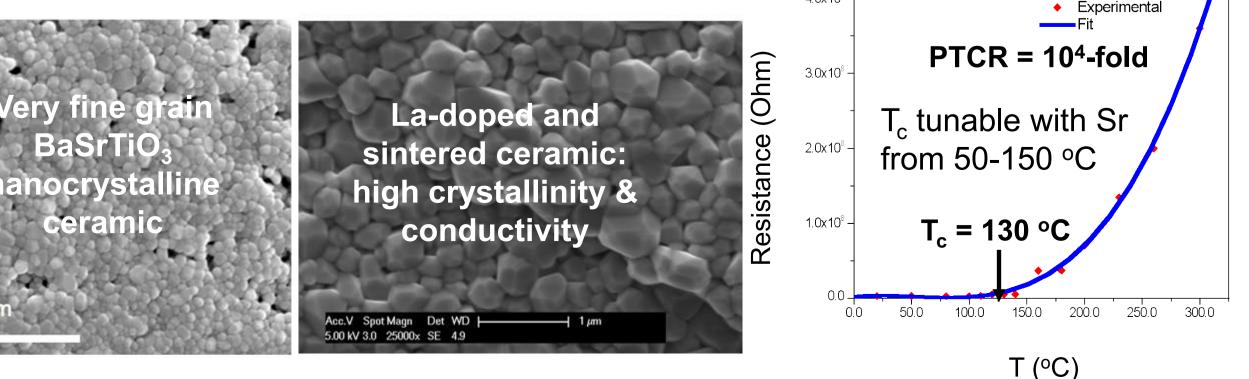
ADVANCED

MATERIALS



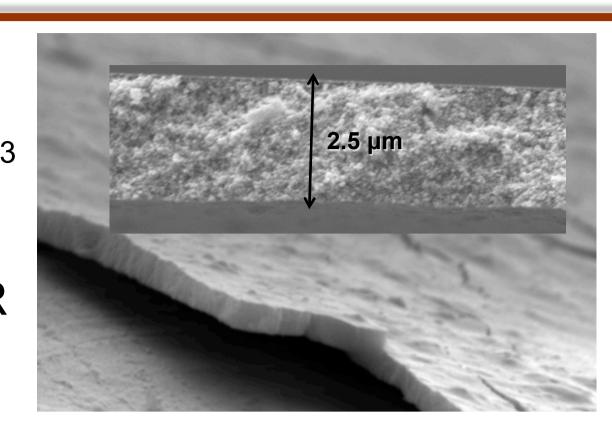
3. COLLABORATION WITH CERDEC, LIFECEL and QUALLION FOR BATTERY SAFETY:
Internal PTCR Protection from Thermal Runaway

Doped BaSrTiO₃ nanocrystalline ceramic coatings on electrodes or current collectors to provide internal PTCR protectant against fire and explosion in rechargeable LIBs

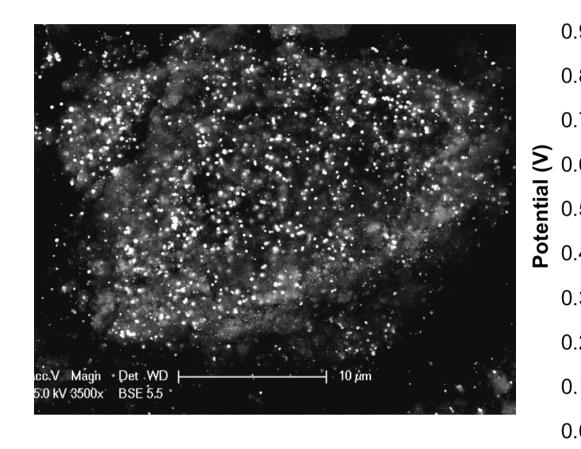


 Protection results from unique Positive Thermal Coefficient of Resistance (PTCR)

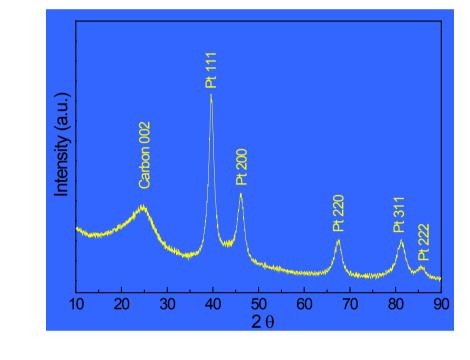
Electrophoretic deposition and double-annealing of doped BaSrTiO₃ nanocrystals yields smooth, uncracked films with low initial resistivity, lower T_c and strong PTCR protection.



4. FUEL CELLS



0.90
0.80
0.70
0.60
0.50
0.40
0.30
0.20
0.10
0.00
0 150
0.30
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0 150
0 300
0 450
0 600
0 750
Current Density (mA/cm²)

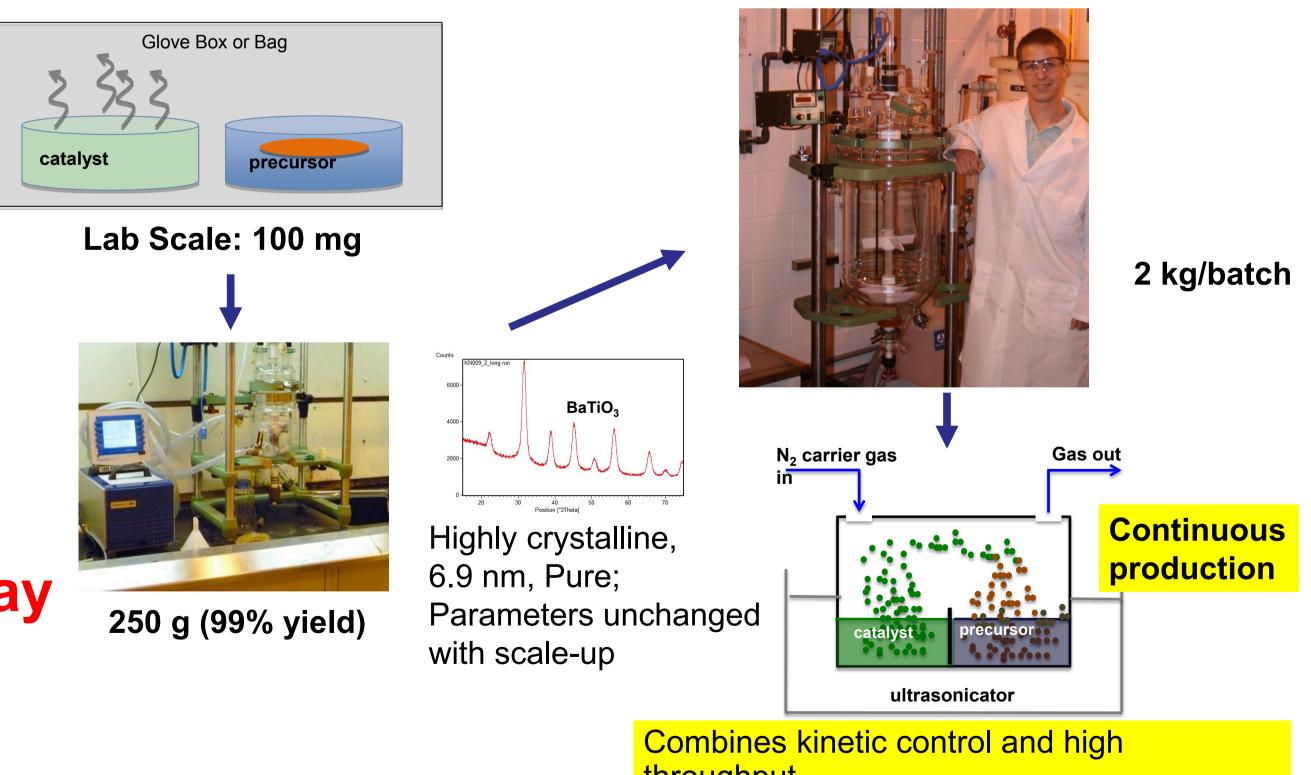


- Bio-inspired, vapor-diffusion catalytic synthesis of nanocrystalline Pt@Carbon Black nanocomposite.
- Unoptimized performance matches commercial, but with 30% less Pt.

Proof-of-principle at 1 kg/hr for BaTiO₃.

5. TECH TRANSFER: SCALE-UP SYNTHESIS

- Catalytic syntheses of Anode, Cathode and PTCR Protectant are "Chemical" (not "chip" technology); thus Readily Scalable
- Spun-out LifeCel Technology Inc. for manufacturing.



Acknowledgements

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- **References**1. HL Zhang, et al. J. Mater. Res. 2011, doi:10.1557/jmr.2011.397
- ¹2. HL Zhang, et al. J. Mater. Chem., 19(2009)9006
- 3. HL Zhang, et al. J. Phys. Chem. C, 114(2010)19550
- 4. J Fold, HL Zhang, DE Morse, Advanced Energy. Materials 2: 277-389.
- 5. T Ould-Ely, et al. Nature Protocols, 6(2011)97