

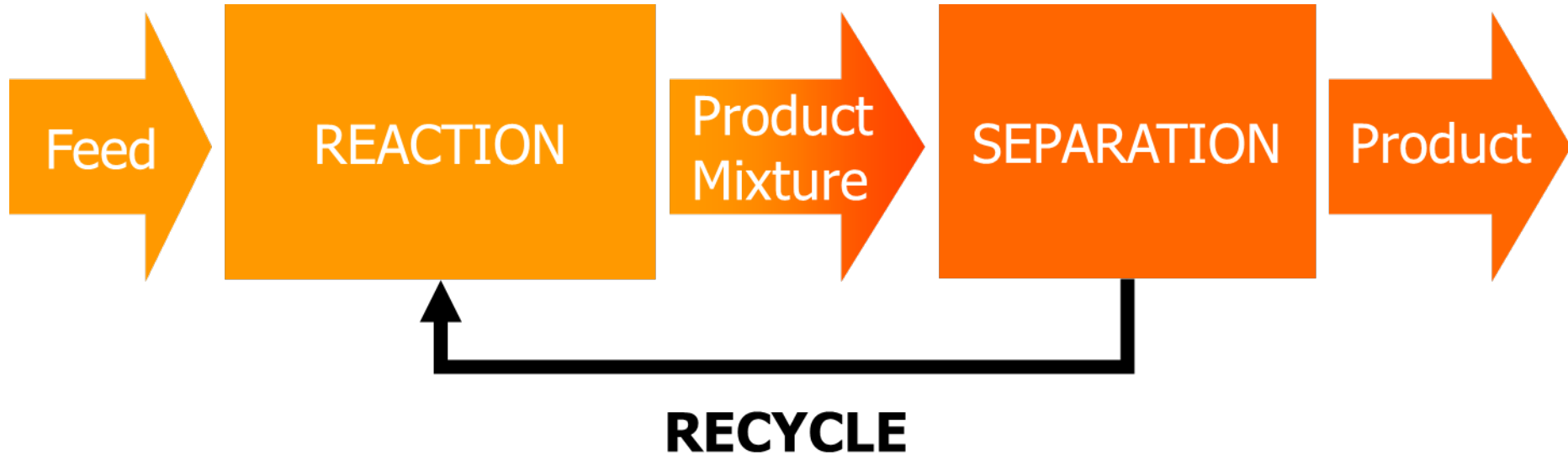
# Development of New Methodology for Making Catalysts with Application to Production of Ammonia, Ethylene, Methanol, and Related Chemicals

Kick-off Meeting  
Idaho National Laboratory  
March 8, 2017

Dr. Dickson Ozokwelu  
Advanced Manufacturing Office

# Why Catalysis? Possible Efficiency Gains

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Low conversions and selectivity rates in Reaction phase



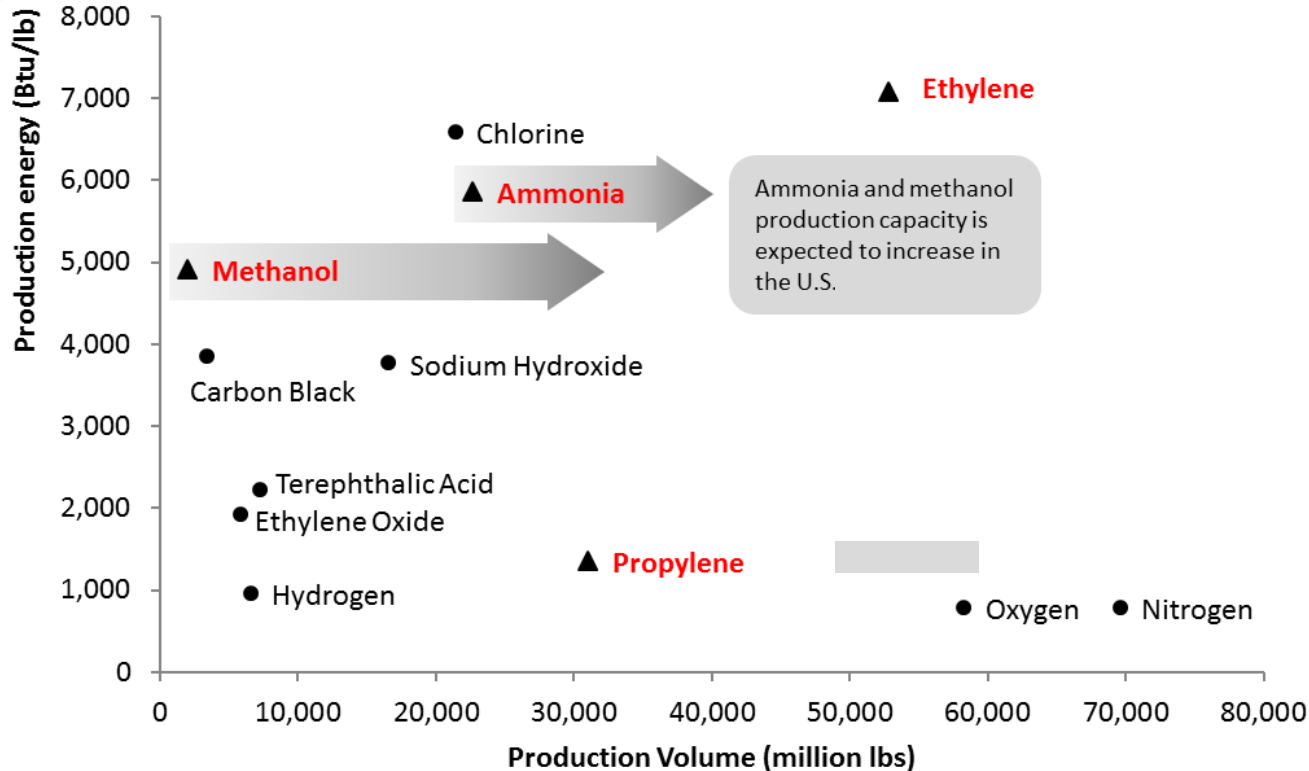
High energy intensity in Separations phase



**Must Improve Reaction and Separation Efficiency**



# Catalysts: Global Impact



- Catalysts are involved in 90% of all chemical manufacturing
- In 2010, total U.S. primary energy consumption in the chemical industry was 4.3 Quads
- Domestically in 2010, 559 TBtu were consumed by the 4 chemicals highlighted in red
- Savings Potential of **112-224 TBtu/yr.** (20%-40%)

# **A Transient Kinetic Approach to Catalytic Materials For Energy-Efficient Routes to Ammonia, Ethylene and Related Chemicals**

**Chemical Catalysts Research and Testing  
Idaho National Laboratory, Argonne National Laboratory**

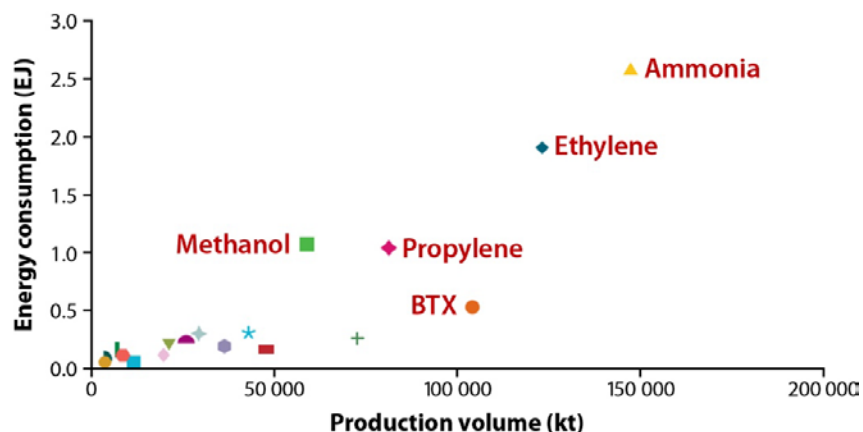
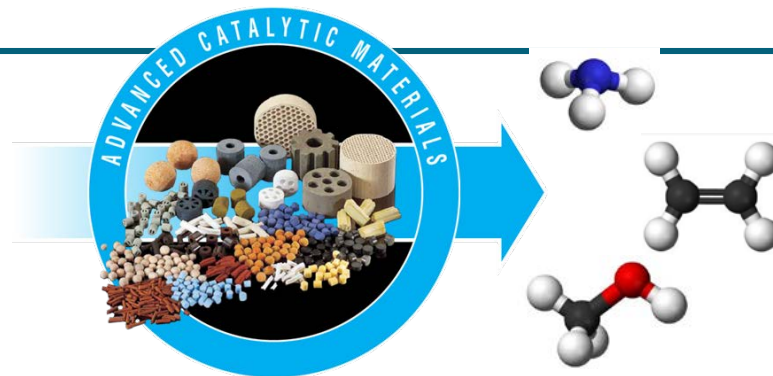
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Dr. Rebecca Fushimi, Idaho National Laboratory

U.S. DOE Advanced Manufacturing Office Technical Resources &  
Networking Forum  
Washington, D.C.  
June 15, 2017

# Project Objective

- A new paradigm for catalyst development based on transient kinetics
- Catalyst development is primarily trial-and-error to address:
  - Complex multistep reaction mechanism
  - Complex multicomponent, ill-defined materials
- New catalytic routes to ammonia, ethylene, etc.
  - Ammonia: 2% of the world's energy use
  - Ethylene: 30% energy saving with catalysis over current steam cracking practice



International Energy Agency, Technology Roadmap, Energy and GHG Reduction in the Chemical Industry via Catalytic Processes, 2013.

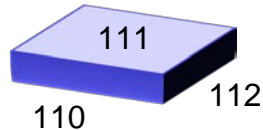


# Technical Innovation

- Current practice of catalyst development

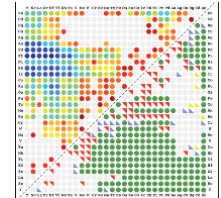
## Surface Science

- Detailed kinetics
- Model materials



## High-Throughput

- Basic kinetics
- Industrial materials



Curtarolo, Stefano, et al. *Nature materials* 12.3 (2013): 191-201.

- Transient kinetics

Temporal Analysis of Products (TAP) Reactor System

- Detailed kinetics : Complex mechanism
- Industrial materials : Complex materials

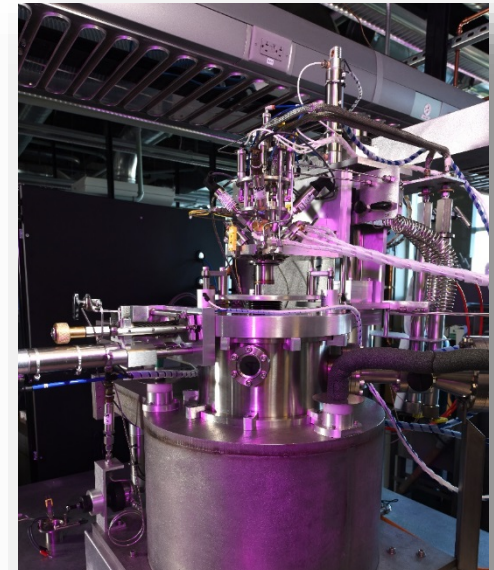
Understanding *how* and *why* materials function

- Kinetic-Centric Informatics

Exploiting data science tools around microkinetics

- Experimental & Theoretical

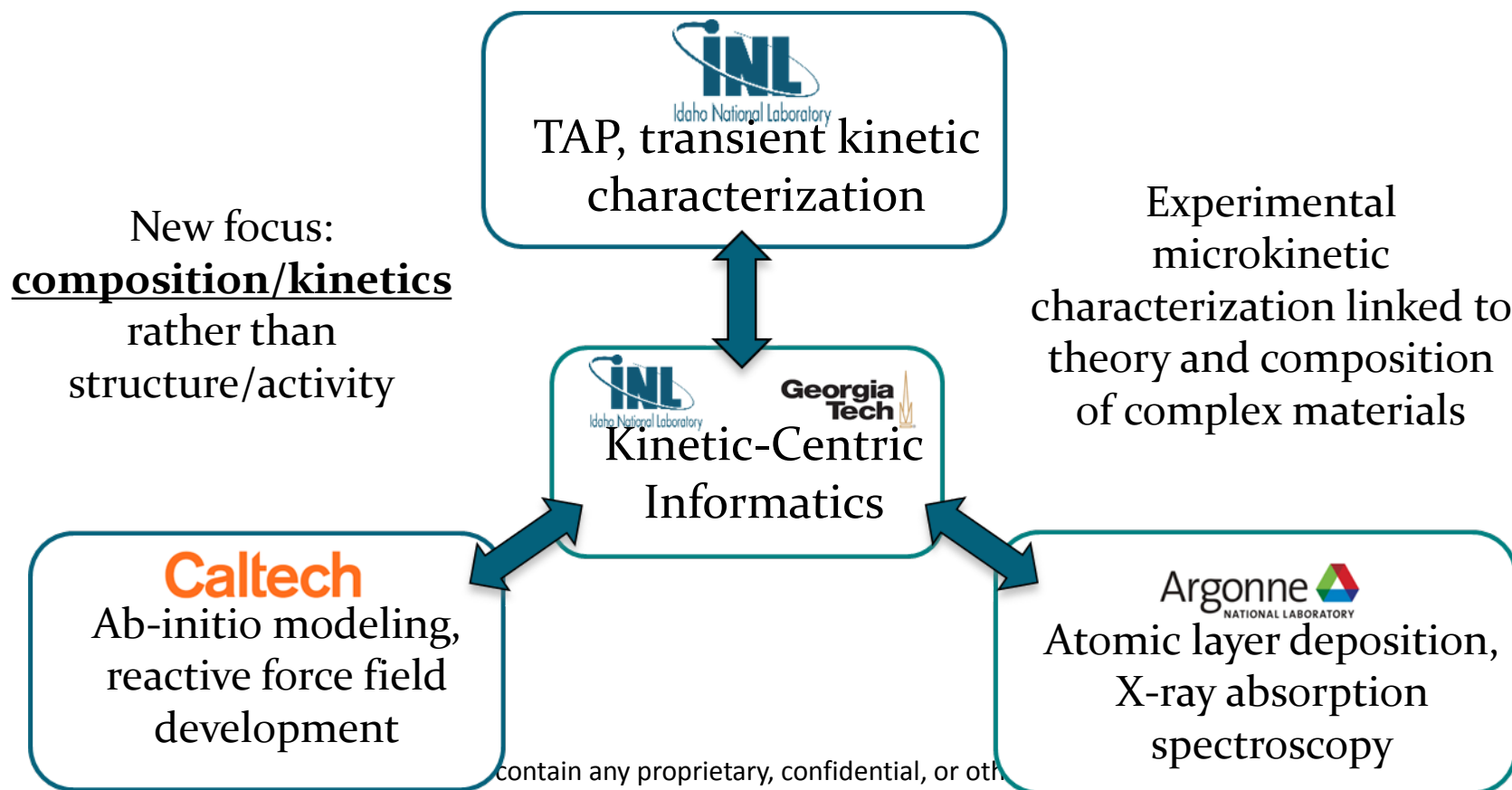
- Accelerating the catalyst development



INL: 2 of 3 TAP Reactors in the US

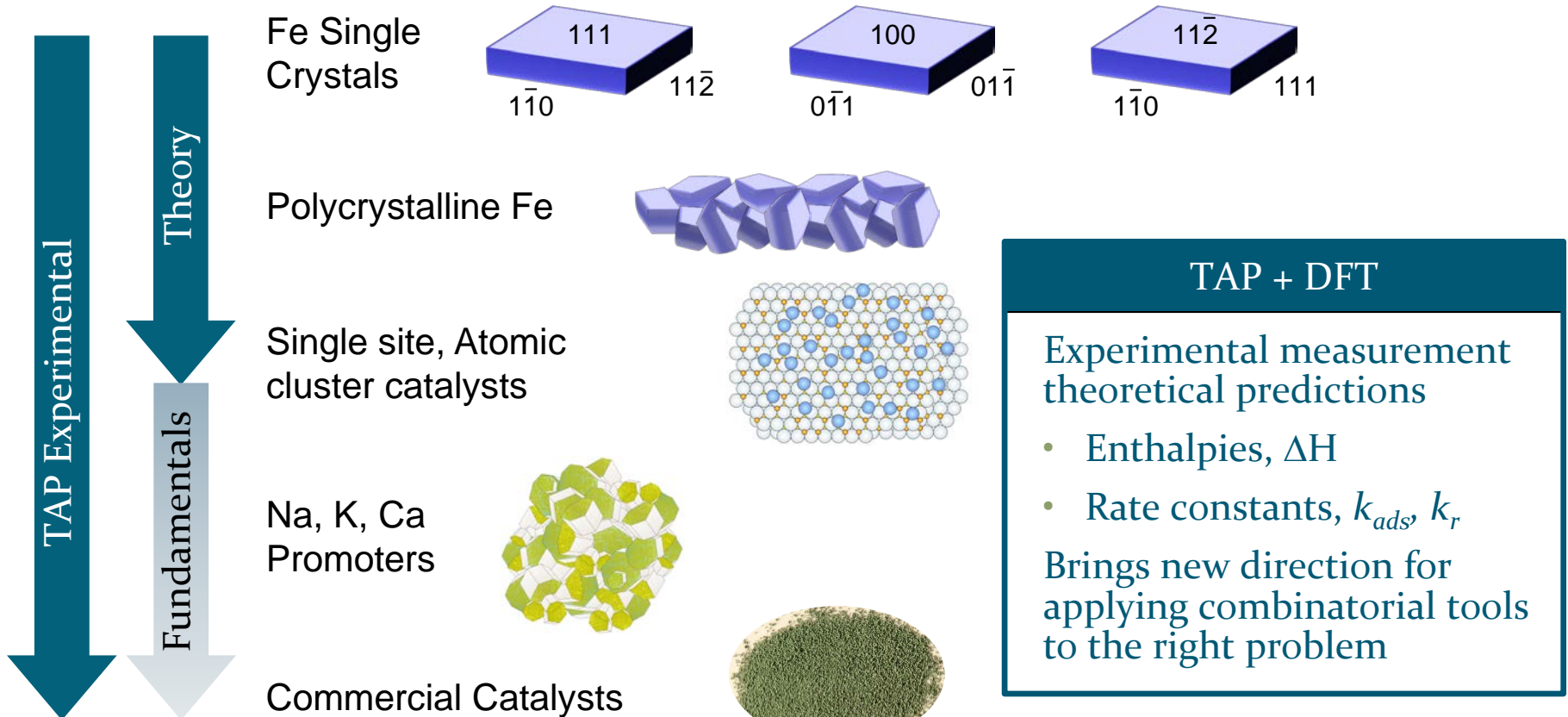
# Technical Approach & Resources

- A new paradigm for catalyst development:  
Transient kinetic experiments
- Kinetic-Centric Informatics tool:  
More meaningful connections from complex data sources



# Technical Approach

- Ammonia synthesis from 200 to 20 bar
  - Elementary reaction steps, N-N rupture, N-H formation
  - Incremental surface coverage change, N, NH, H, etc.
  - Material complexity





# Measure of Success

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- Moving catalyst development beyond trial-and-error
  - Enduring benefit
    - A unique capability to support industrial research
    - Changing the way materials are studied and designed
  - Measurement
    - Industry participation (CRADA and SPP Agreements)
    - Transition of early-research to industry
- 
- Wide-adoption of TAP and understanding of transient kinetics
  - Move beyond catalysis to other gas/solid reaction applications
    - Chemical sensors, advanced sorbents, photo and electrochemical systems, energetic reactions, materials characterization, etc.
  - Adoption of more-efficient transient chemical processes
    - Redox mode, transport bed, switched feed, chemical looping

# Program Engagement

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- Industry Advisory Board
- CRADA, Cooperative Research and Development Agreements
- SPP, Strategic Partnership Projects
- Minimum 6 month engagement suggested for R&D
- CAES: Center for Advanced Energy Studies
  - CAES & INL Technical Assistance Program
  - Short-term proof-of-concept



# Catalysis Development Tools, Research and Catalyst Technologies Network

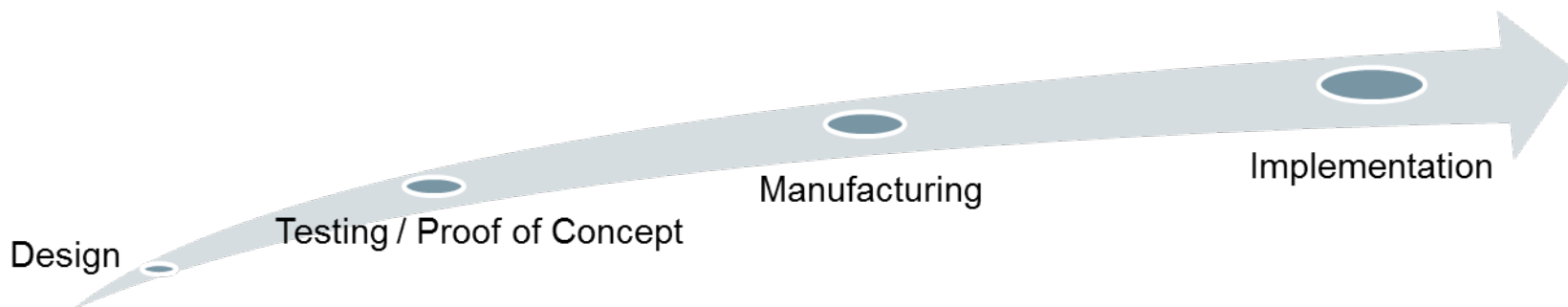
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Magdalena Ramirez/INL

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# Program Introduction

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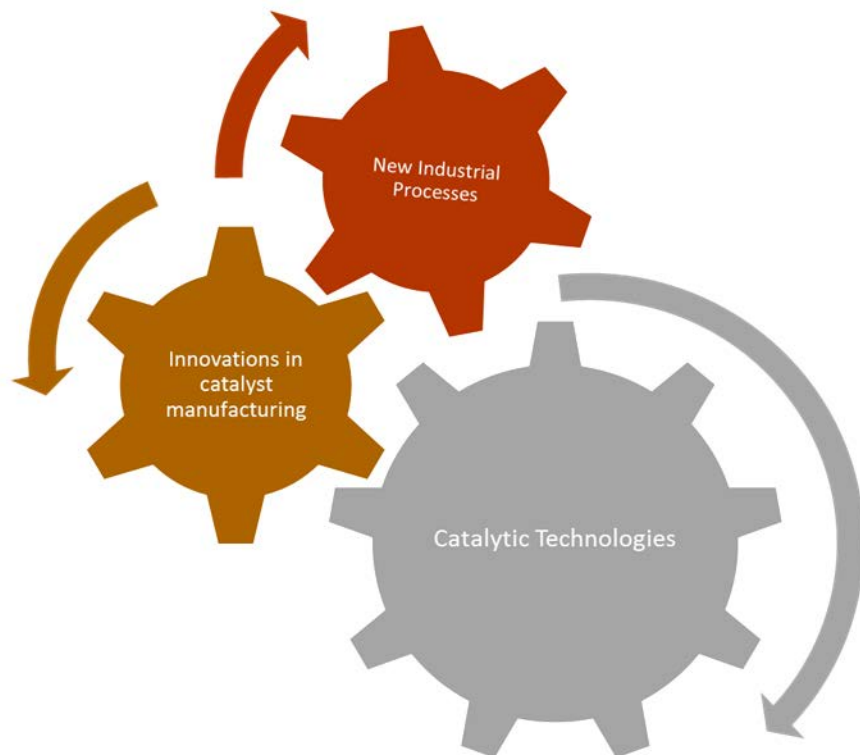


- For research purposes, catalytic materials are prepared at lab-scale using careful and controlled synthetic methods, which are not typically represented in industrial manufacturing processes.
- The missing link: Scientific basis of catalysts manufacturing processes
- The Program will accelerate the path of R&D outcomes to industrial uses

# Catalytic Technology Network, CTN

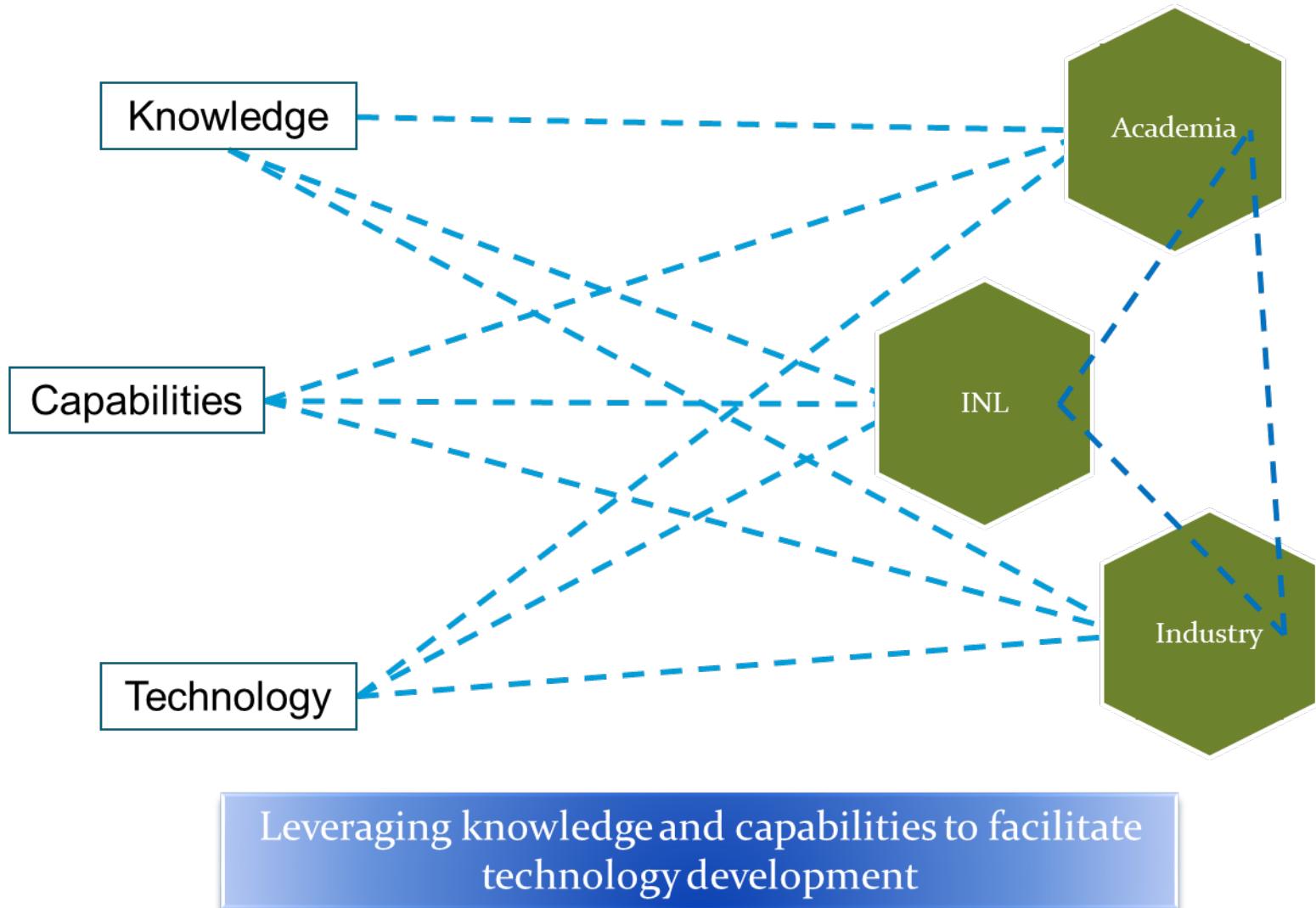
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Catalytic technologies to enable cleaner and more efficient manufacturing



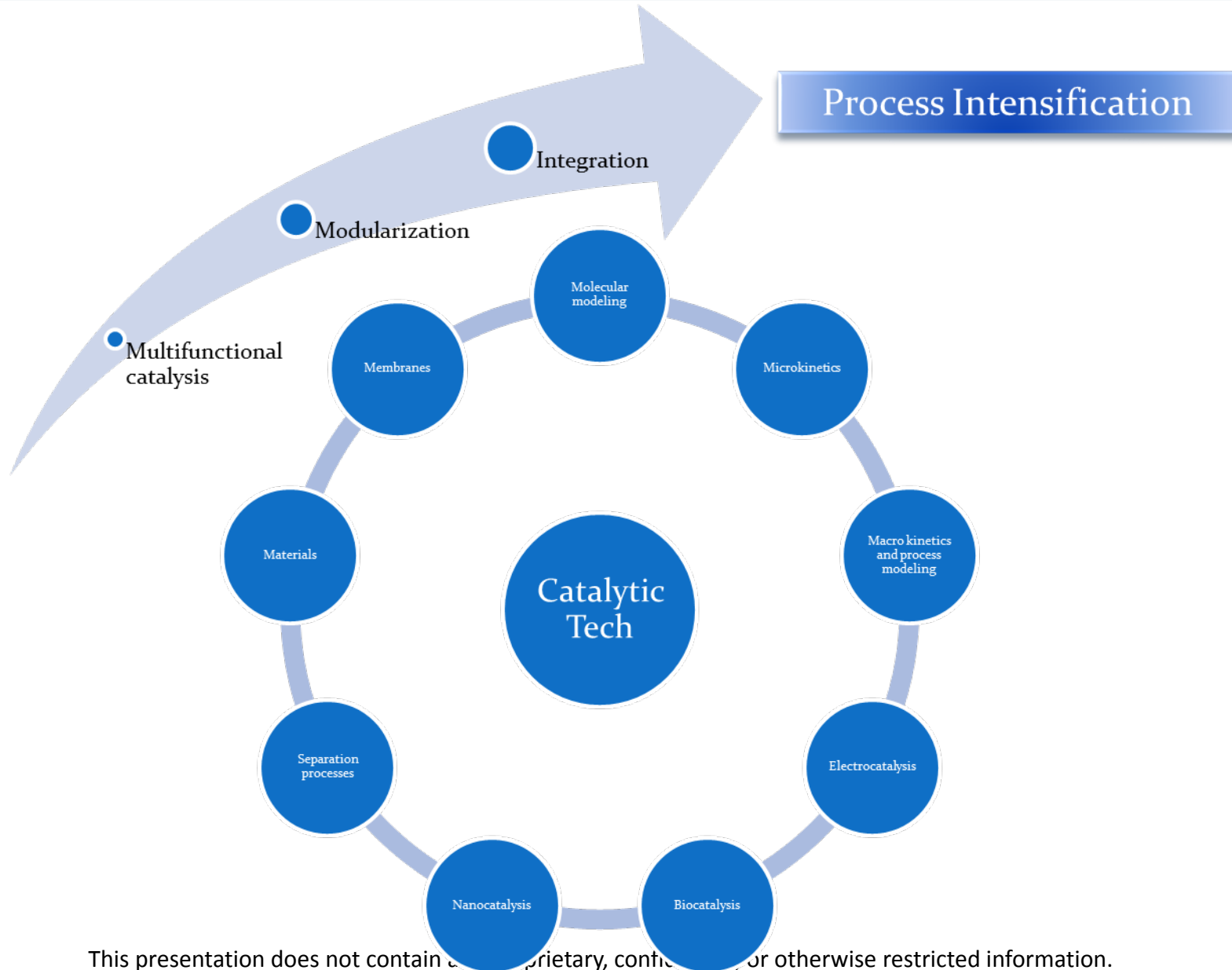
- A world-class **catalytic technologies network**,
- providing more energy efficient and cleaner manufacturing methods,
  - for catalytic materials
  - and for fuels/chemicals through catalytic processes.

# The Network



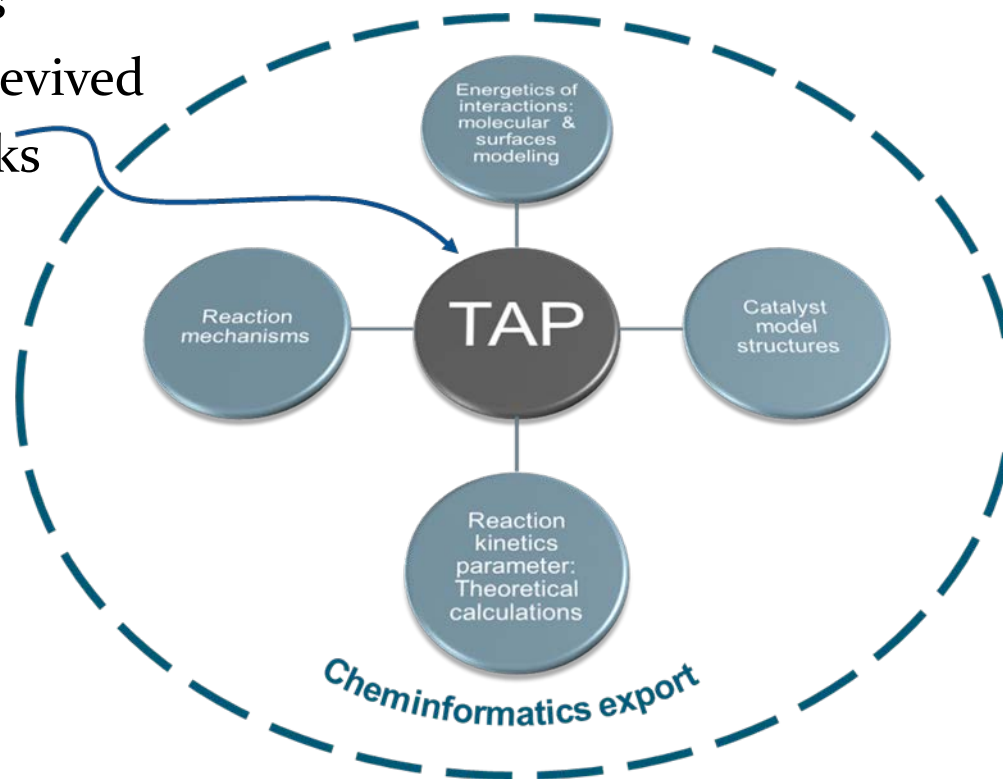


# Capabilities and Resources



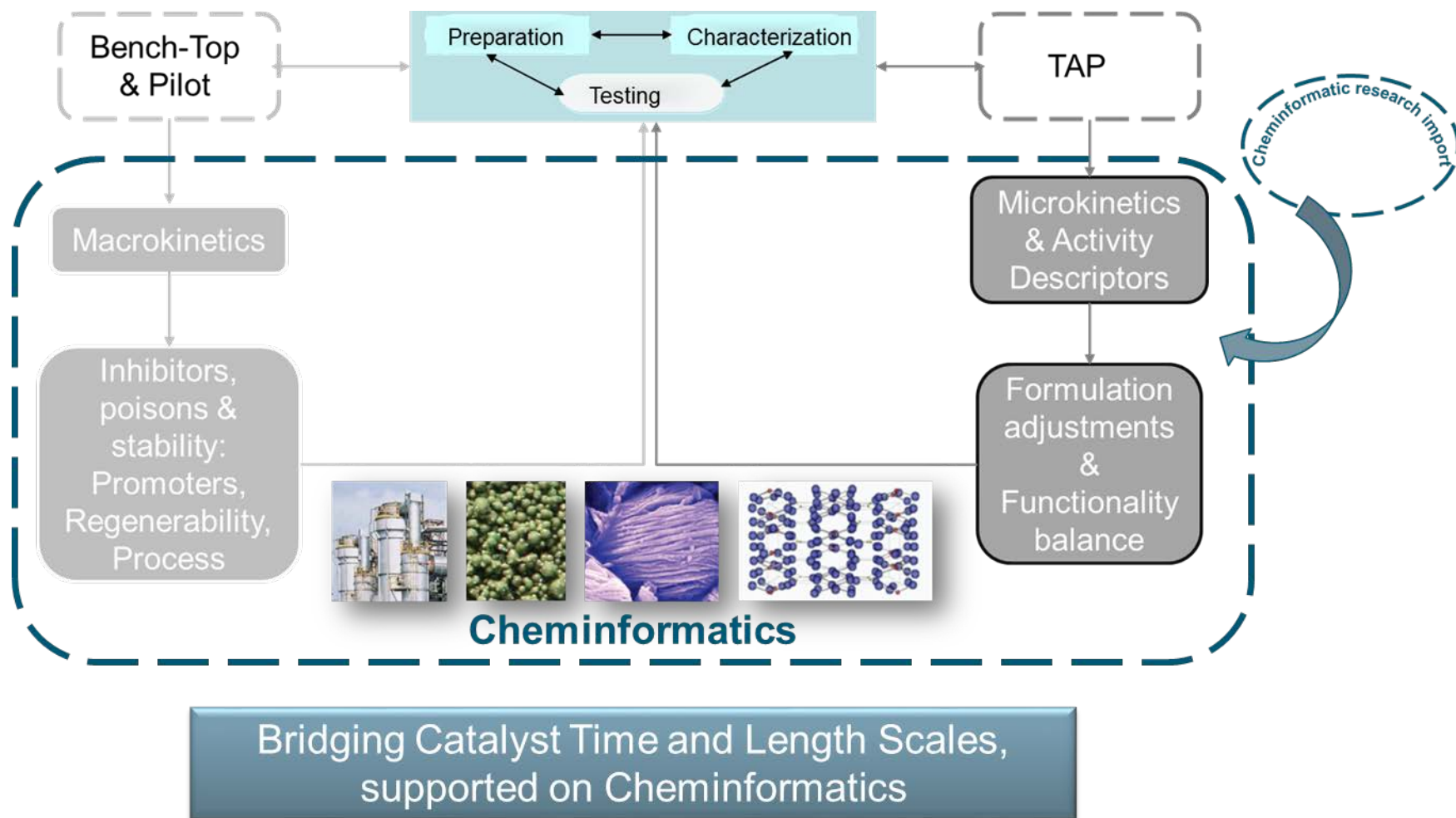
# Unique resource: TAP

- Activity: How much it does
- Selectivity: Where it takes the feed
- Stability: How long it lasts
- Regenerability: Can it be revived
- **Functionality**: How it works

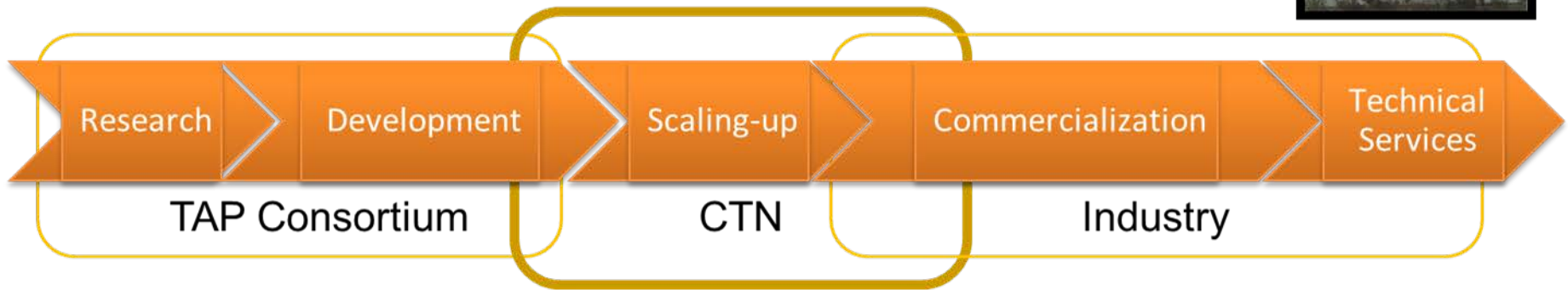
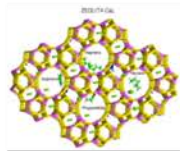


Basic Research Contributions to TAP R&D

# CTN will link TAP outcomes to catalyst development



# CTN within the Catalytic technologies supply chain



# Program Focus and Deliverables

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- Four major energy-intensive processes:
  - Ammonia
  - Ethylene
  - Propylene
  - Methanol
- Three types of outputs:
  - Catalysts
  - Catalyst manufacturing methods
  - More energy efficient catalytic manufacturing processes

# Program Success

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- The identified scientific basis for a new catalyst synthesized using conventional methods allows its reproduction through manufacturing processes, as proved by an industrial partner.
- The identified scientific basis for a new synthetic method allows its reproduction through a newly designed manufacturing process that rises industrial interest.
- The potential process that uses a catalyst developed through CTN rises industrial interest



# Program Application Process/Business Engagement

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## CTN Value Proposition

### Academia

- An accelerated mechanism to connect research outputs with market
- Guidance and advice from industrial views, needs and problems
- A path to catalysts industrialization and commercialization
- A smooth mode of industry – academia liaison

### Industry

- Direct translation of academia outputs to industry
- Access to and influencing mechanisms for world class R&D intellectual capital
- Access to potential results and early technologies
- Product slate diversification
- A smooth mode of industry – academia liaison

# Program Application Process/Business Engagement

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## Case-by-case options through different R&D project types

- AOP collaborator / partner: subject to cycle schedule, approval by DOE and budget.
- FOA cost-sharing projects: limited to the announcement and call for projects, in the area of interest.
- Cooperative Research and Development Agreements. CRADAs are agreement between federal laboratories and nonfederal entities that allow the parties to collaborate.
- Strategic Partnership Projects (SPP). work for federal agencies and the private sector. Activities can include research and development or applied engineering.
- Agreement to Commercialize Technology (ACT agreement): The agreement allows a private sector to access our personnel and/or facilities under terms that are not as restrictive as a SPP.

# Program Application Process/Business Engagement

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- Industry engagement in two steps:
  - Socialization of the basics
  - Listening day workshop
- Funding needs include:
  - Strategic definition, planning and launching: ~ \$300k
  - AOP for establishing the scientific basis of conventional manufacturing methods of new catalysts: ~ \$2M/y
  - Tools growth: ~ \$5M