

# Iron Based Flow Batteries

## for Low Cost Grid Level Energy Storage

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OFFICE OF  
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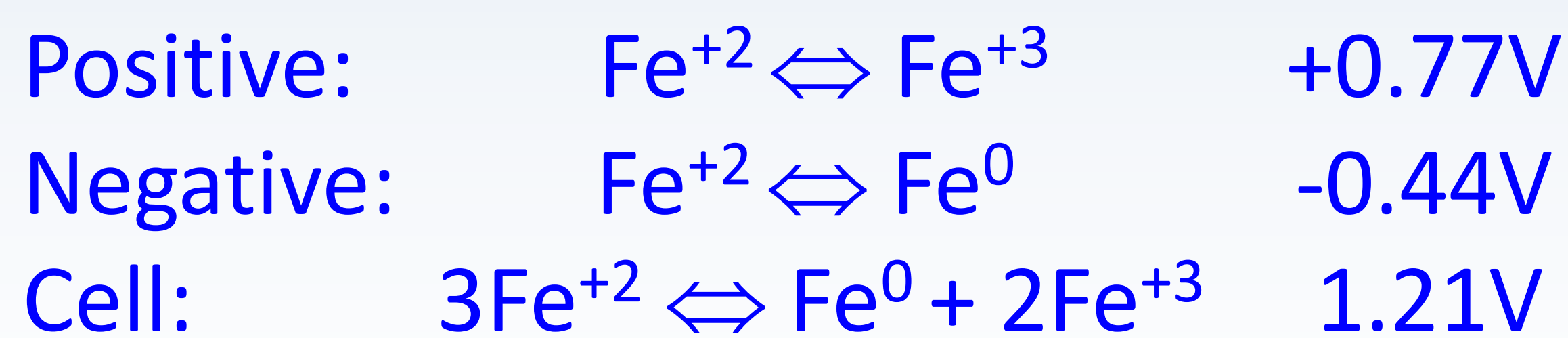
### Purpose

Develop efficient, cost-effective grid level storage capability based on iron.

### Goals of this Effort:

- Minimize Cost/Watt by increasing current density - Hardware Cost >> Electrolyte Cost
- Minimize Cost/Whr by increasing plating capacity
- Maximize Efficiency by minimizing current lost to hydrogen evolution

### Electrochemistry of the all-Iron system:



### Impact on Iron Based Batteries on the DOE OE Energy Storage Mission

Widespread grid level storage will require:

- Low Cost
  - All-Fe battery uses one low cost active element and inexpensive separators
- Environmental Acceptability
  - Mild pH, non-toxic electrolyte
- Geographic Flexibility
  - Iron is readily available from domestic sources

### Research Plan

Year 1: COMPLETE

- Ligand Screening – demonstrated  $[\text{Fe}^{+3}] > 0.5\text{M}$  @  $\text{pH} > 2$
- $\text{H}_2$  evolution suppression – effect of pH, anions evaluated

Year 2: IN PROGRESS

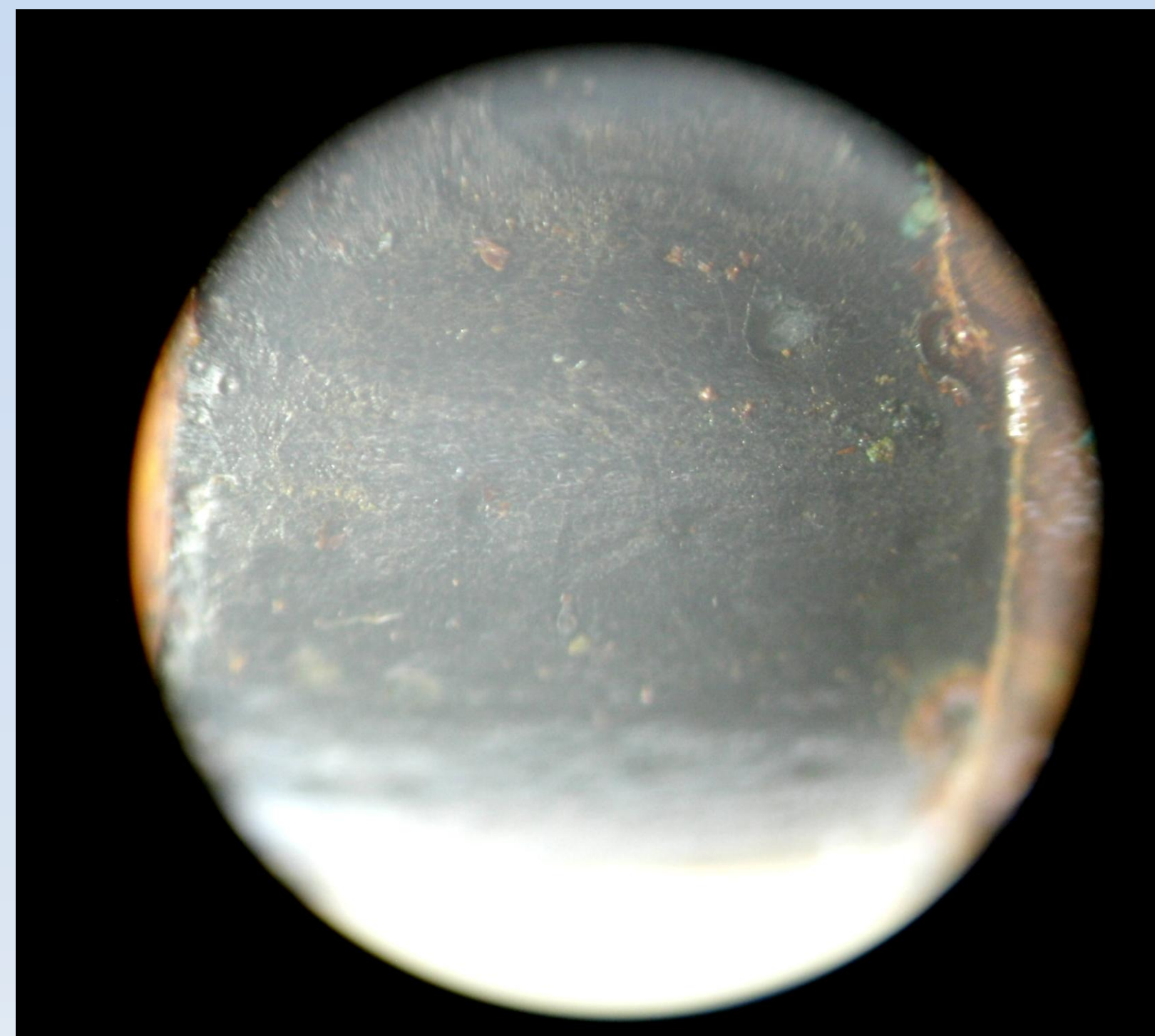
- Effect of Ligands on Fe plating efficiency, morphology
- Separator studies –  $\text{Fe}^{+3}$ , Ligand crossover

Year 3:

- Optimization of plating capacity, current density to maximize efficiency
- Scale up from  $50 \text{ cm}^2$  to  $250 \text{ cm}^2$

### Recent Results

**Demonstrated Adherent, Stress-Free, Dendrite-Free Plating**

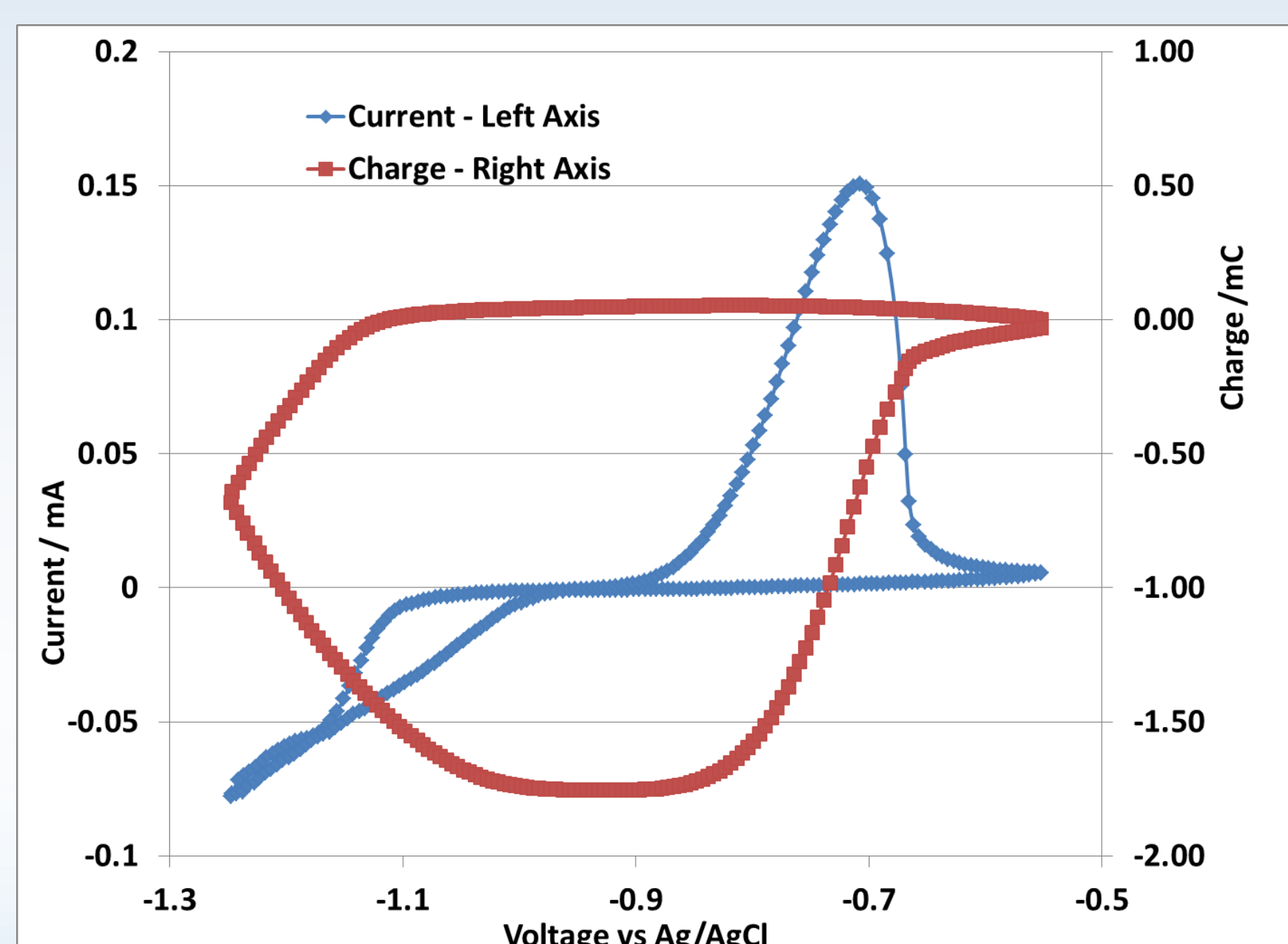


Deposit Thickness Shown is Equivalent to  $75 \text{ mAh/cm}^2$

Deposits up to  $150 \text{ mAh/cm}^2$  have been made

**Demonstrated Coulombic Efficiency >99% for Iron Plating**

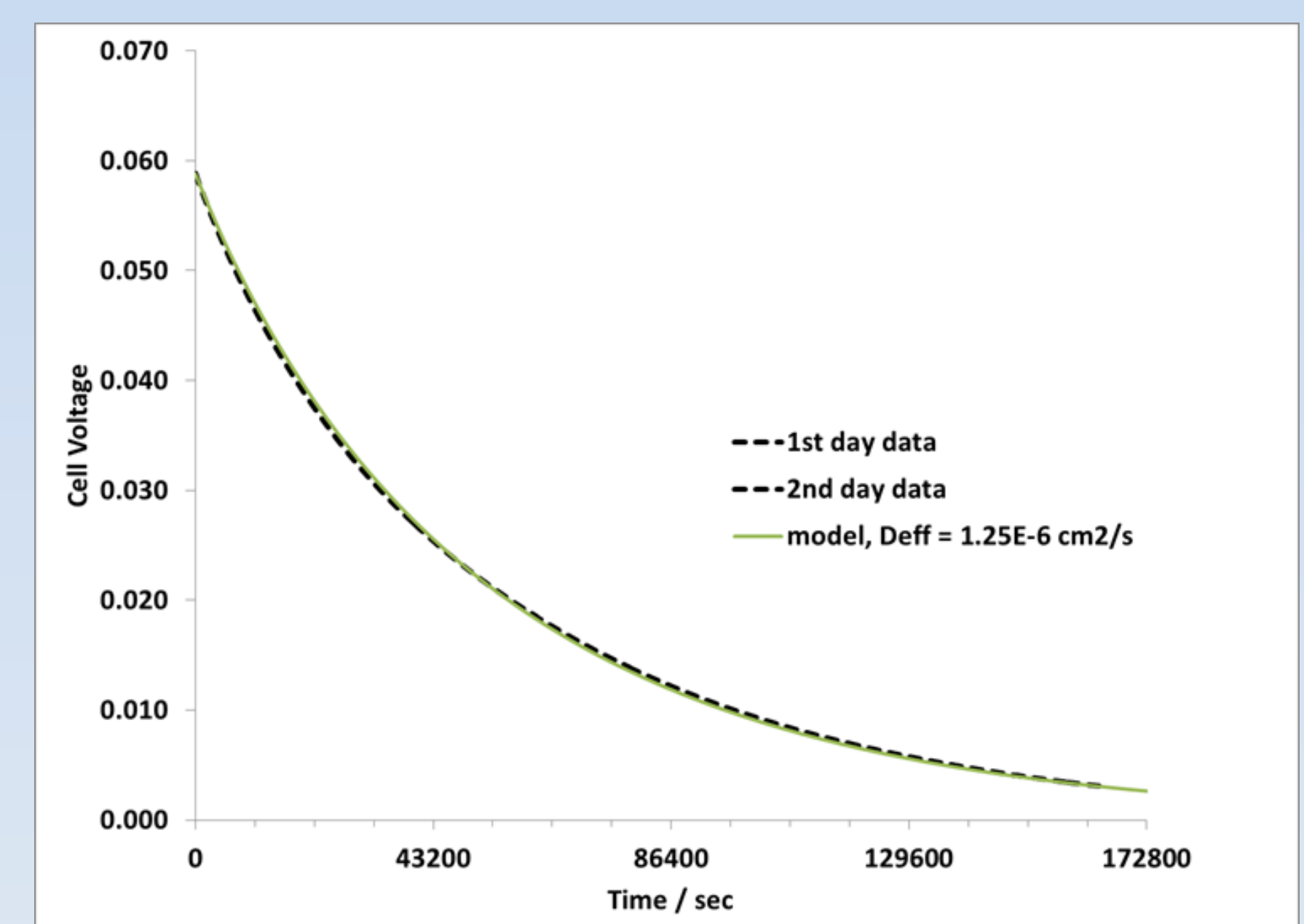
**T = 60C**



### Recent Results

**Measurement and model of  $\text{Fe}^{+3}$  crossover**

- Room Temperature
- Daramic Separator
- Equivalent to 0.5% Capacity Loss in 24 hrs with Electrolyte Circulating



**Measurement and model of  $\text{Fe(II)/Fe(III)}$  Overpotentials**

- Negligible Kinetic Loss
- Ohmic and Mass Transfer Overpotentials
- Total Overpotential <20 mV @  $0.1 \text{ A/cm}^2$

