



ConocoPhillips test results and data analysis

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Overall Ignik Sikumi Project Goals

- North Slope reservoir-scale field trial to evaluate CO₂/CH₄ exchange
- Short-term test to demonstrate concepts at larger-than-lab scale
- Validate exchange mechanism results from laboratory work
 - Confirm injectivity into naturally occurring methane hydrates
 - Confirm methane release without production of water or sand
 - Obtain reaction rate data to facilitate reservoir-scale modeling
- Demonstrate stable production of natural gas hydrates by depressurization





Project History

2008 – 2010

- Select site and gain access
- Characterize reservoir

2011

- Drill, log, complete and suspend Ignik Sikumi #1
- Design field test

• 2012

- Re-enter well and perforate
- Perform exchange test
- Perform depressurization test
- P & A well and remediate site
- Prepare datasets
- Begin data analysis

2013

Data analysis and history matching





July 2012 Status

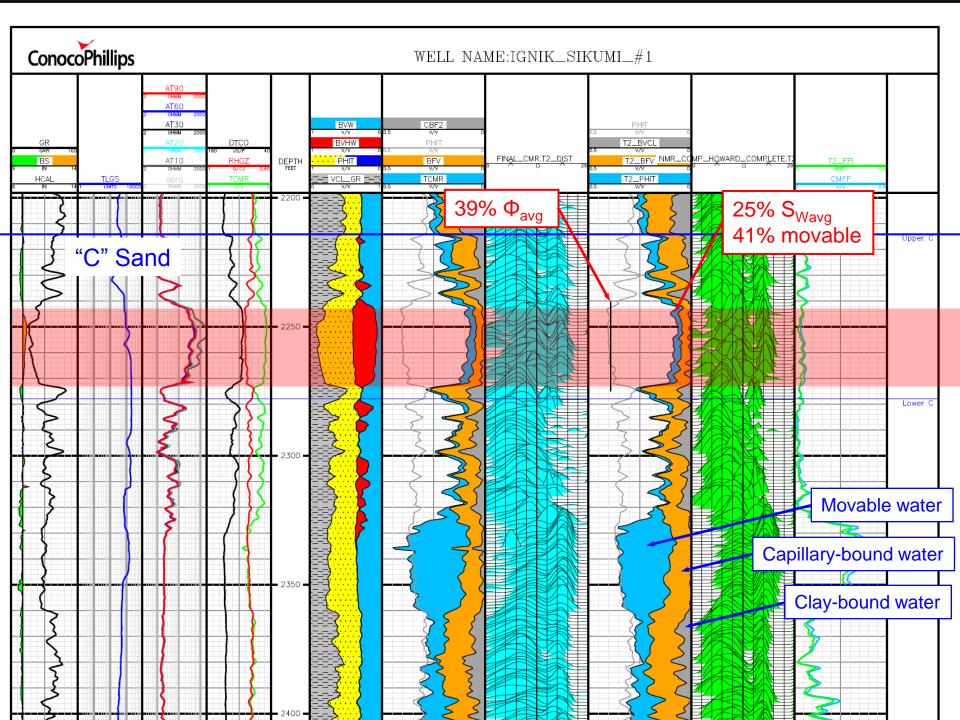
Data Correction/Reconsolidation in Progress

- Outliers/spikes removed
- Time stamps for each source corrected
- GC data reprocessed
- Three DTS data sets obtained
 - Un-normalized and 2 types of normalization
- Created 1 and 5 min time average datasets
- Adding corrections for dead volumes/wellbore storage

Path Forward

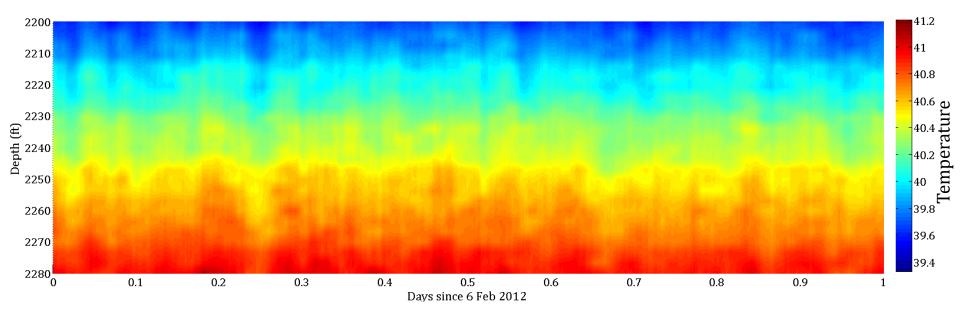
- Perform material balance of test
- Injectivity analysis, using simulation
 - Infer hydrate saturation changes
- Production analysis using cell-to-cell model
 - Gas phase composition history match
- Issue final database and report
- DOE workshop

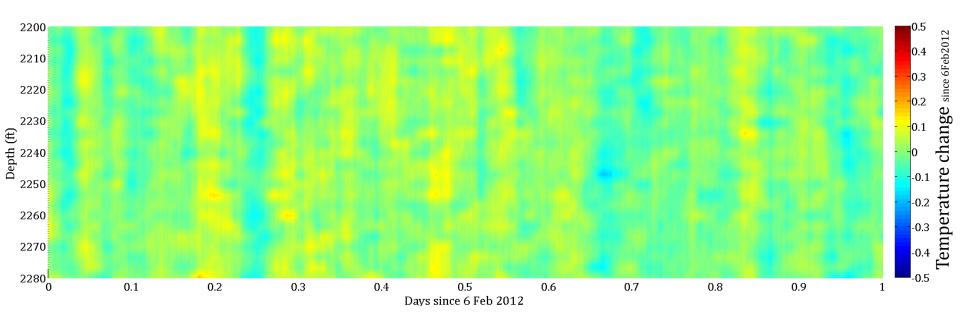




Initial Period before any well work

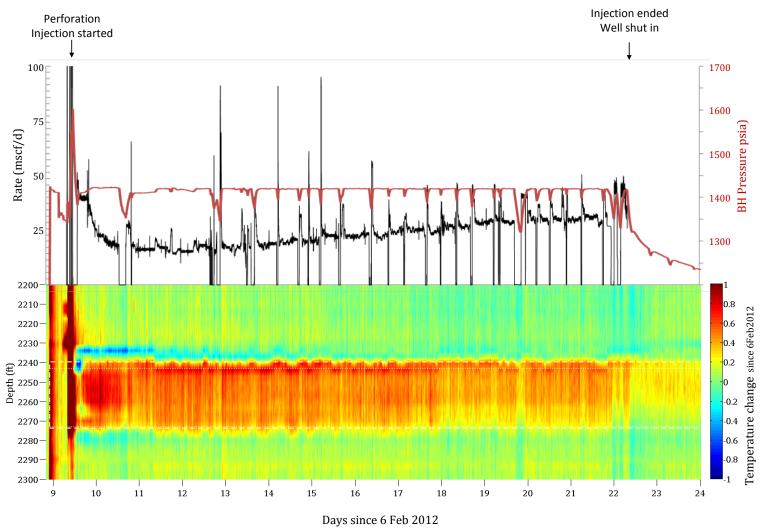
Temperature linear with geothermal gradient (~1.79°F/100ft), Temperature change ~0







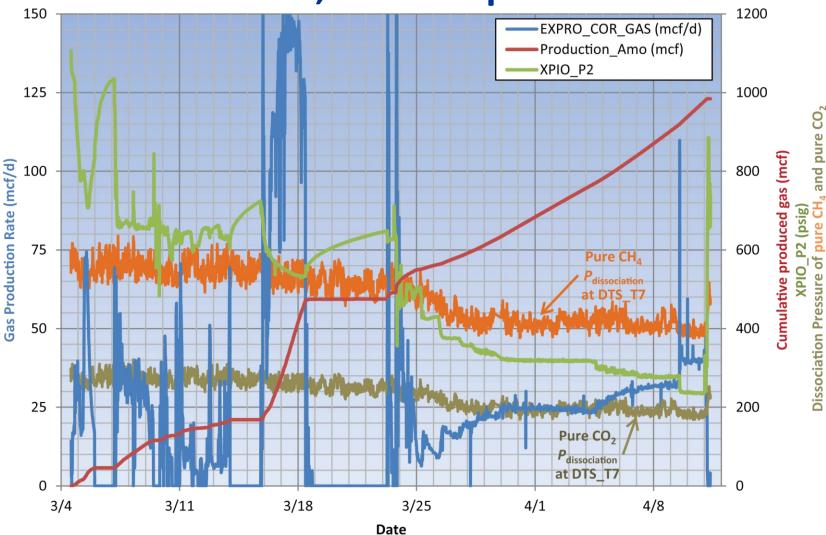
Injection





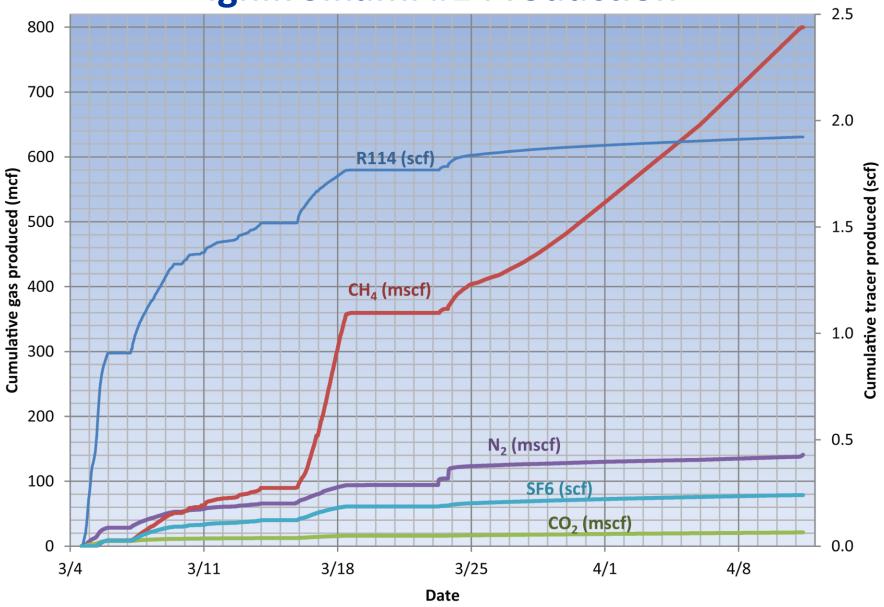


Overall Production – Gas rates, Pressure, and Temperature



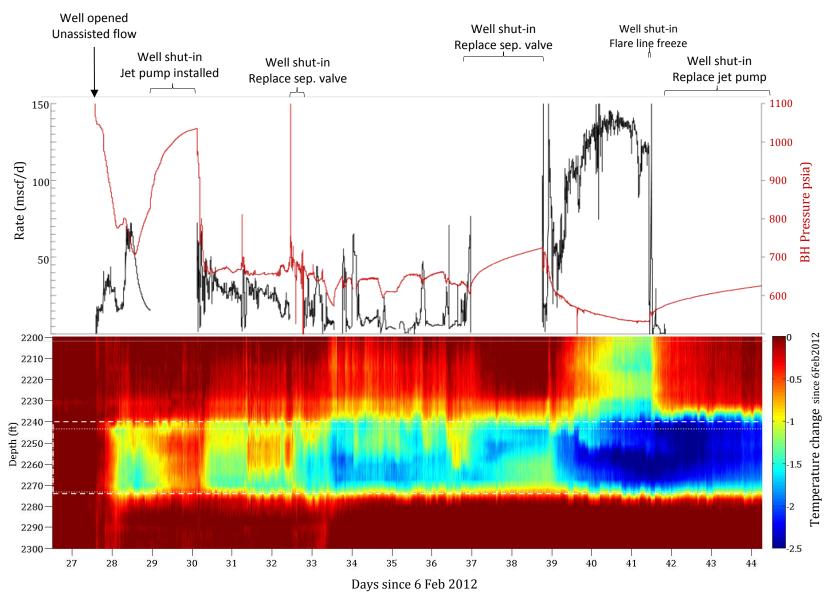


Ignik Sikumi #1 Production



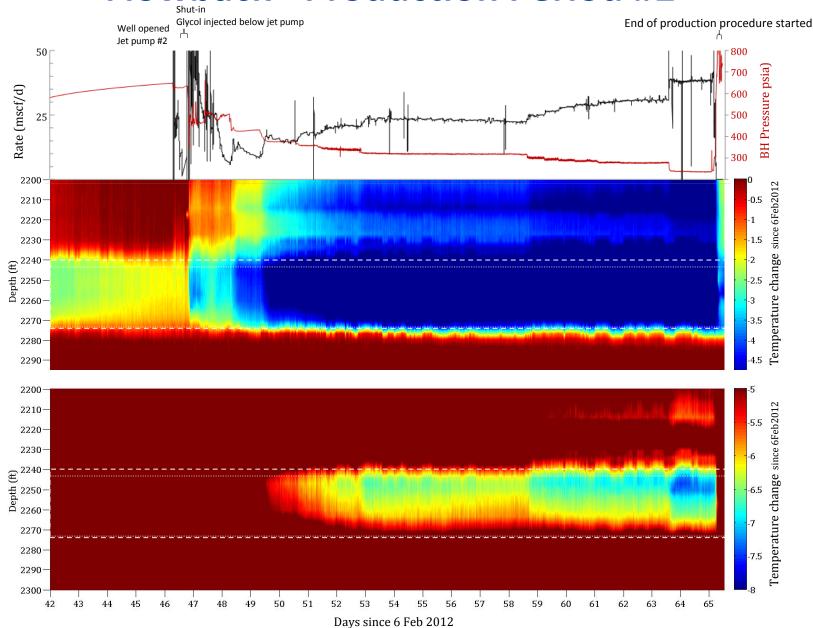


Flowback - Production Period #1





Flowback - Production Period #2





Summary Observations

- Successful injection of CO₂ mixture into hydrate reservoir
- Methane produced both above / below CH₄-stability pressure
 - CO₂ was retained in the reservoir compared with N₂
 - Indicates the possibility of CO₂ exchange
- Depressurization sustained below CH₄-stability pressure
 - Steady increase in production rate
 - Over 850 mscf (24,000 scm) of CH₄ produced in total
 - Low BHP achieved (~250 psi)
- Solids production significant
- Evidence for heterogeneous injection / production



Database Summary

Diagrams of the operations included

PI&D's + dead volumes of surface equipment and well

Master Variable List

- Where to go for complete info on any recorded variable
 - e.g., what instrument recorded the data, calibration, etc

Supporting Data Document

Where to go for notes on calculations and data corrections

Operation Event Log

- Where to go to see what was happening at every step of the test
- All raw data in MySQL and CSV format
- All final data available in MS SQL database format, CSV, Matlab
 - Clean, 1 min averaged, and 5 min averaged data



Data Streams

Composition

On-line GC (~15 min sampling int.)

Continuous downhole conditions

- 3 downhole pressure gauges (P&T)
- Distributed Temperature Sensing (T per ft)

Continuous surface conditions

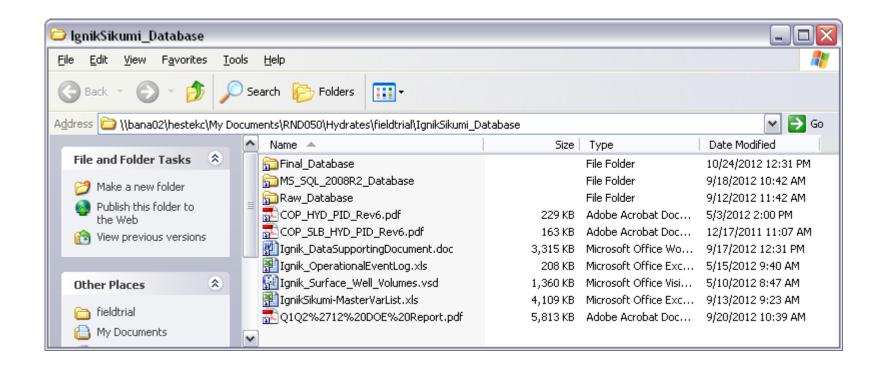
- Pump rates
- Flow rates (gas, jet pump fluid)
- Line pressures and temperatures
- Separator P&T

Produced fluid measurements

- Collected on regular intervals
- Water prod rate
 - Tank straps (~30min int.)
- Water (~1hr int.)
 - pH, salinity, SG
- Gas (~1hr int.)
 - Gas gravity

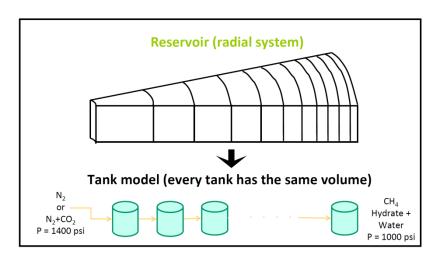


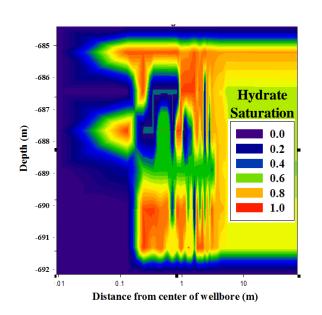
Database Folder



Modeling and Simulation Efforts

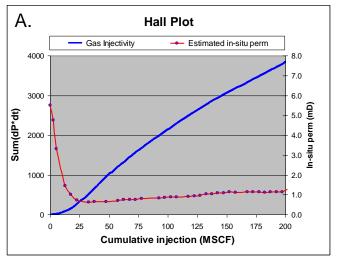
- Adiabatic CTC Model (ConocoPhillips)
 - Cell Volume (3.5 ft), $S_H = 65\%$, $P_i = 1000$ psi, $T_i = 40.5$ F
- Solids production
- Heterogeneous production
- History-match simulations of the Ignik Sikumi field test with newly-developed Mix3HRS software
- Complex pressure, temperature, and composition history
 - CO₂+N₂ injected into a CH₄ reservoir with all 3 gases produced
 - Competing thermodynamics for hydrate formation and dissociation in the reservoir

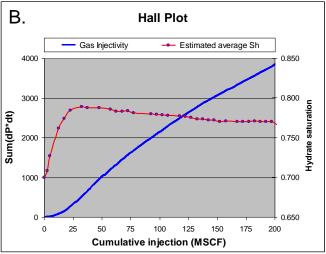




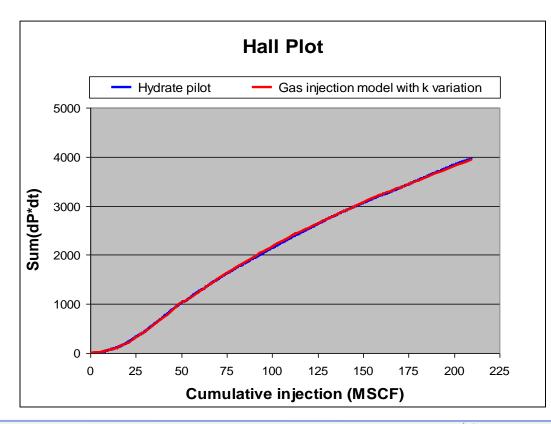


Hall Plot – Varying Permeability





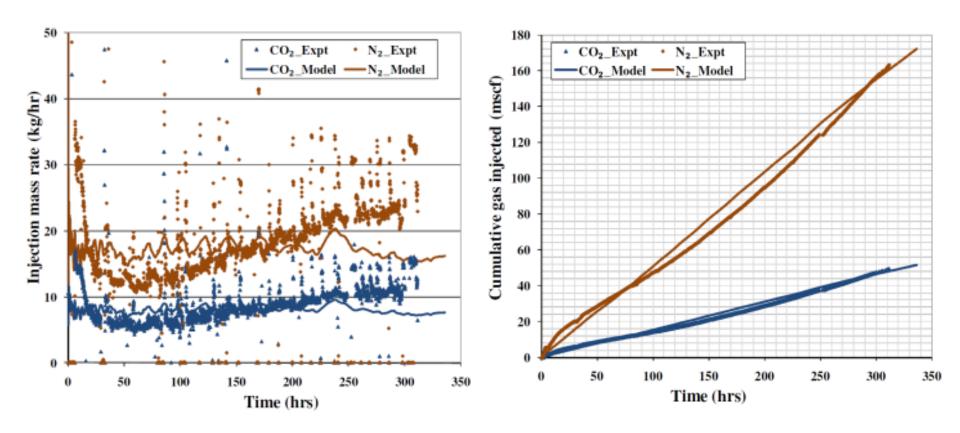
- Permeability adjusted over time
- ightharpoonup One possibility is Δ hydrate sat.
- Good match obtained



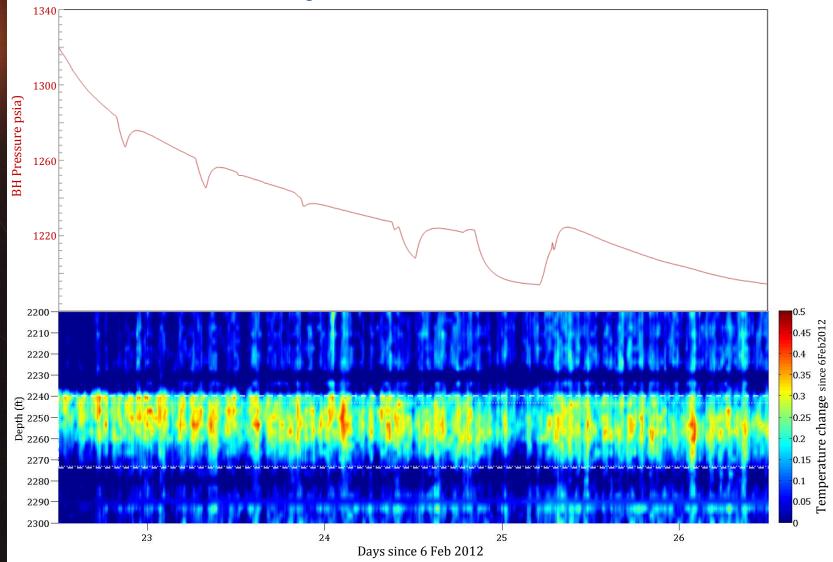


Injection matching

• The Injection flow rate and cumulative injection of CO_2 and N_2 into the reservoir are matched with the field data.

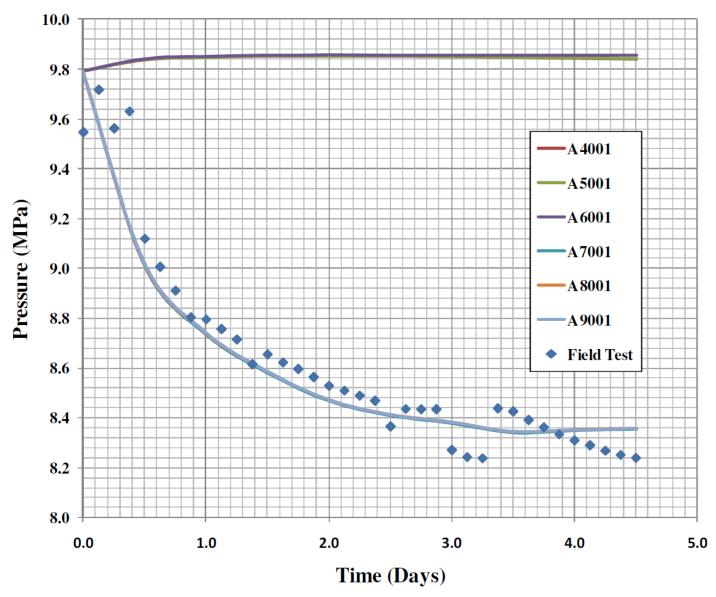


Post-Injection Period



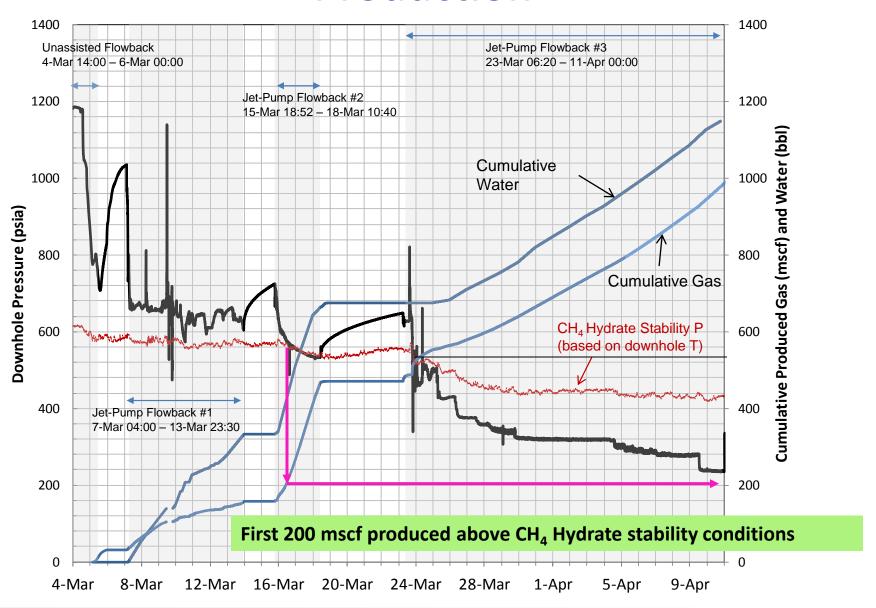


Post-Injection Period



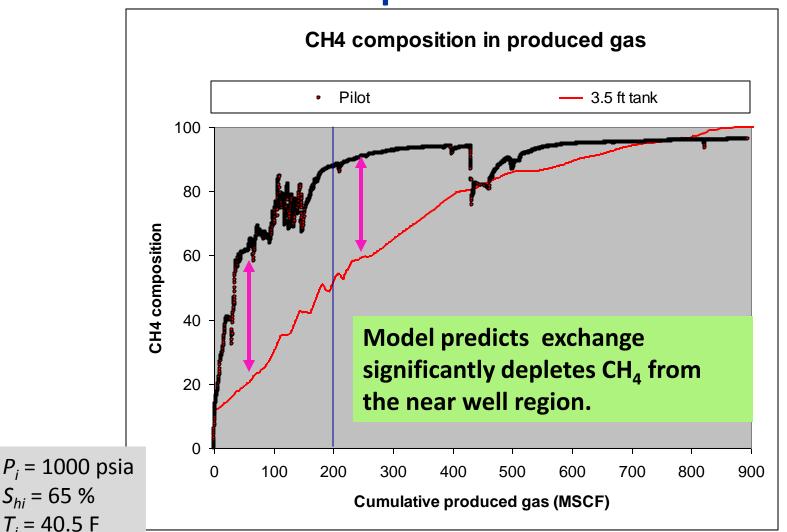


Production



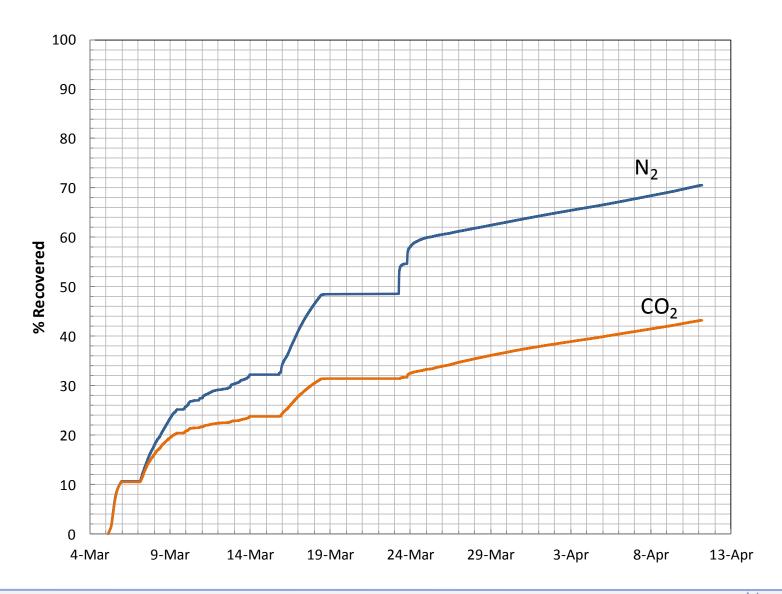


More CH₄ produced than Equilibrium Model predicts.



 $S_{hi} = 65 \%$ $T_i = 40.5 \text{ F}$

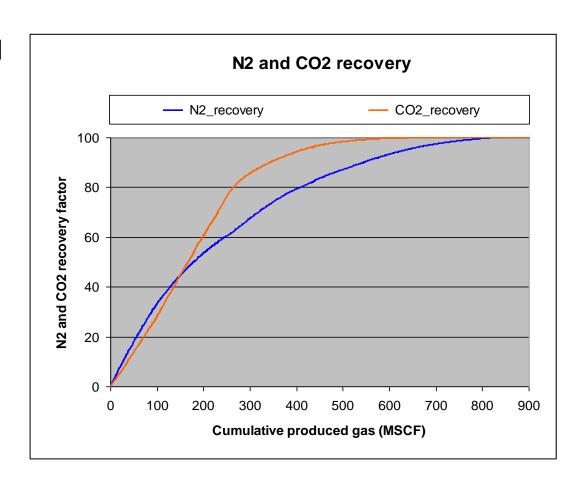
% Recovered based on Injected Amounts





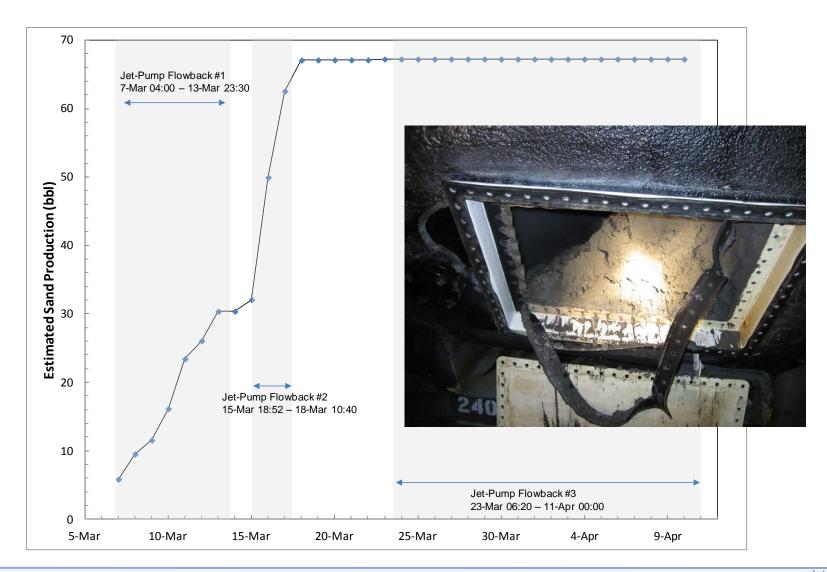
Observations: Field versus Model

- Not enough CH₄ from model
- Not enough water from model
- Temperature increase too high in model
- Recovery of N₂ to CO₂ reversed in model
- Examining potential mechanisms of gas production
 - 1. Dissociation in place w/o permeability enhancement
 - Dissociation in place w/sand migration + permeability enhancement
 - 3. Production of solid hydrate (< 200 µm) and subsequent dissociation in wellbore above the jetpump when contacted with warm power fluid





Sand Production





Mechanism 2 – Experience at Mallik

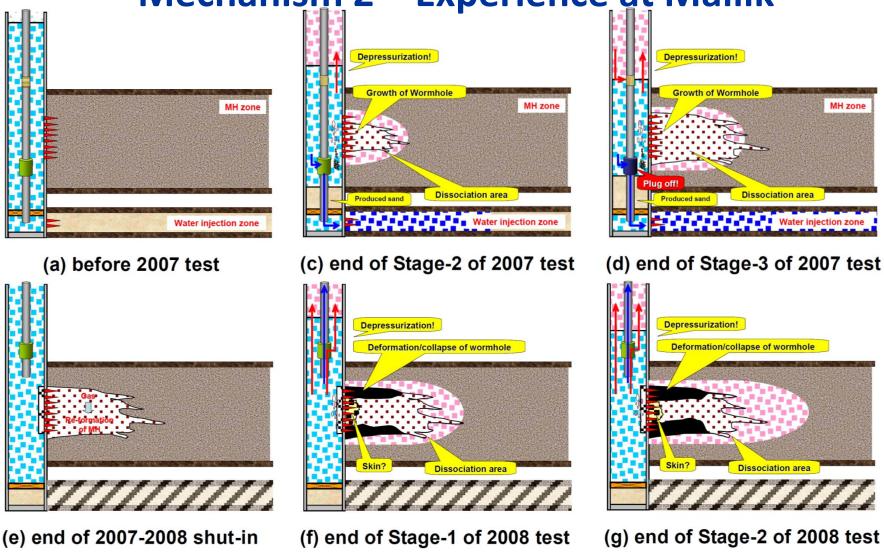


Figure 17 Schemata of reservoir performances through 2007 and 2008 tests inferred from history matching simulation From: Kurihara, et al., Proceedings of the 7th International Conference on Gas Hydrates (ICGH 2011), Edinburgh, Scotland, United Kingdom, July 17-21, 2011



Mechanism 2

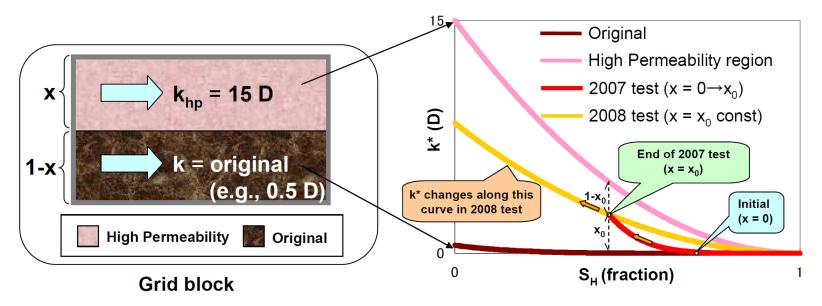


Figure 11 Concept expressing overall grid block permeability as a function of MH saturation with growth of high permeability conduits

$$k^* = xk_{hp}(1 - S_h)^2 + (1 - x)k_o(1 - S_h)^N$$

$$k_{eg} = k^*k_{rg}$$

$$k_{ew} = k^*k_{rw},$$

From: Kurihara, et al., Proceedings of the 7th International Conference on Gas Hydrates (ICGH 2011), Edinburgh, Scotland, United Kingdom, July 17-21, 2011

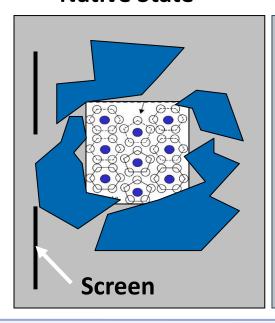


Mechanism 3: Solid CH₄ – Hydrate produced?

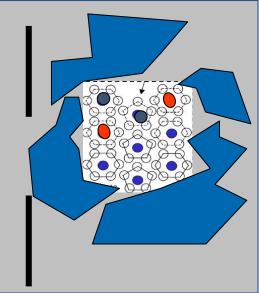
- \triangleright Largest source of CH_4 & water = CH_4 Hydrate
- > Solids (sand) were produced



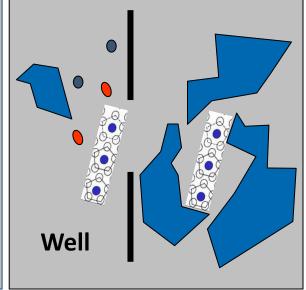
Native State



Exchange

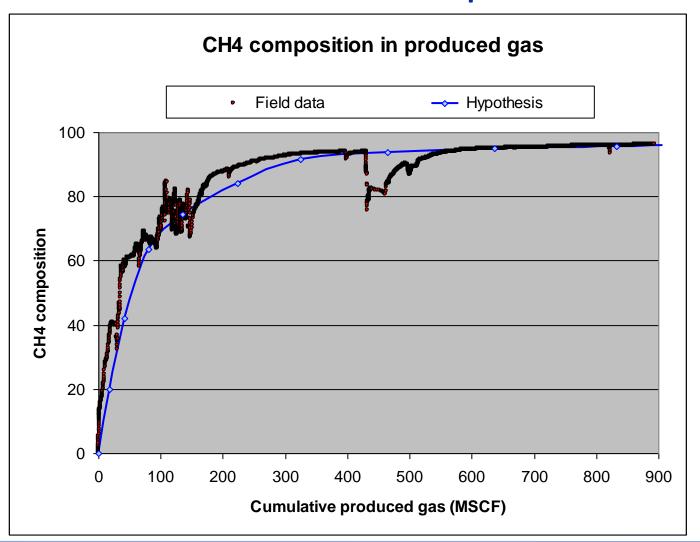


Depressurization



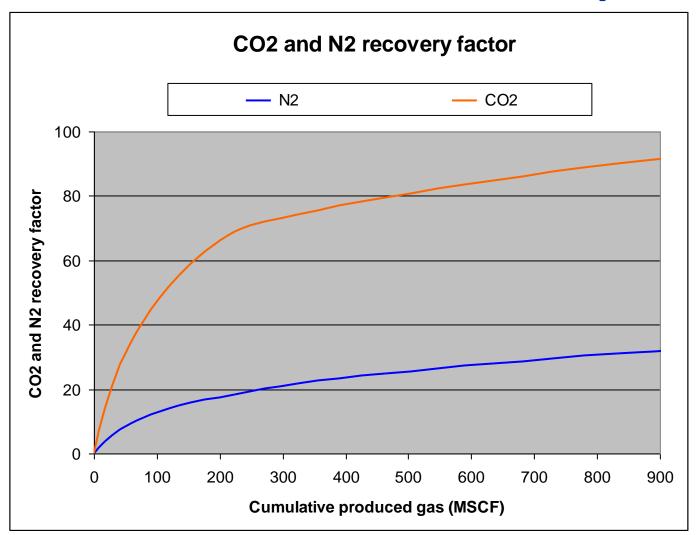


Mechanism 3: CTC Model Solids Recombination – CH₄ Match





Mechanism 3: CTC Model Solids Recombination – Recovery





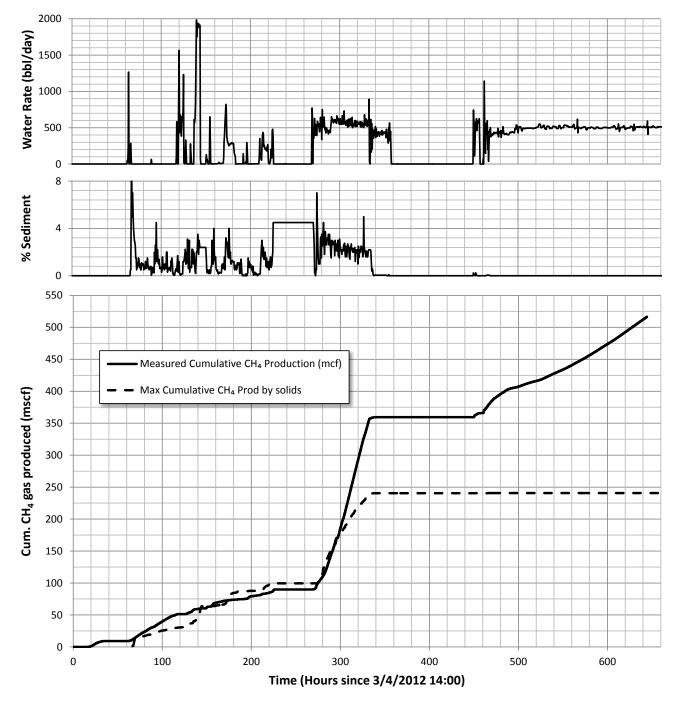
Mechanism 3

Method

- Use EXPRO water rate and %Sed measurements
- Scale sediment rates to match observed cumulative sand production

Worst-case Assumptions

- All sand produced had associated CH₄ hydrate that was produced
- S_H values from CMR log
- Gives upper limit to CH₄ from solids



Field trial likely a combination of mechanisms

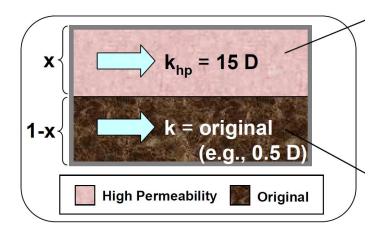
Mechanism 2

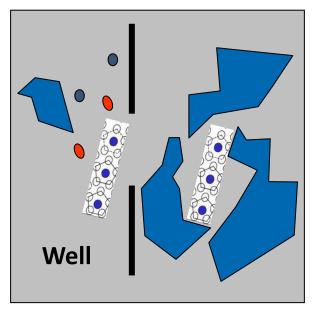
 Dissociation in place w/sand migration + permeability enhancement

Mechanism 3

 Production of solid hydrate (< 200 μm) and subsequent dissociation in wellbore above the jetpump when contacted with warm power fluid

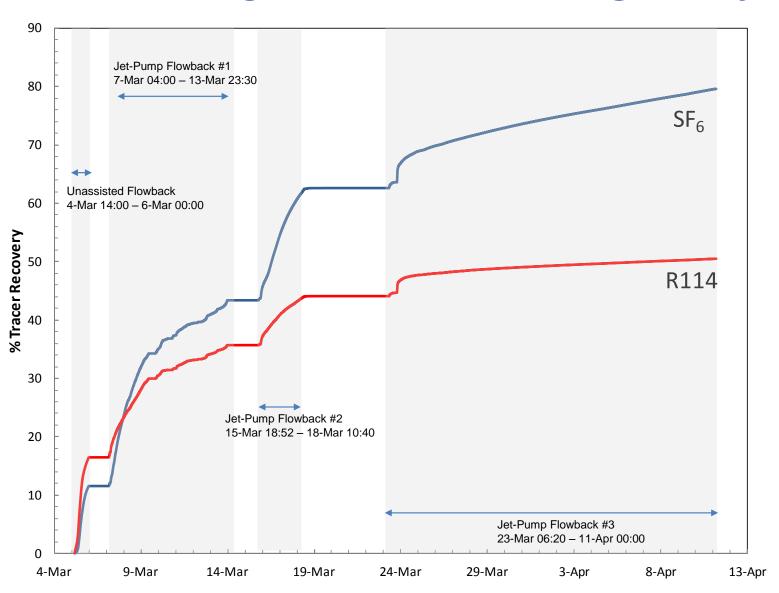
Reservoir heterogeneity





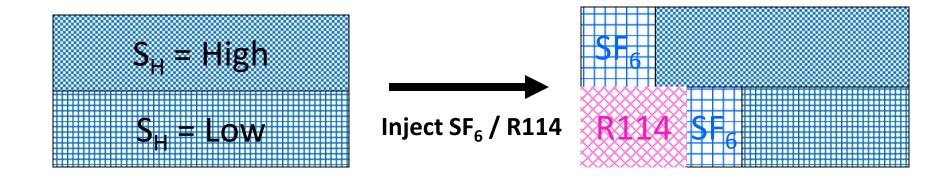


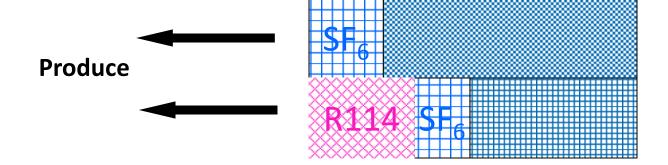
Tracer ... Argument for Heterogeneity?



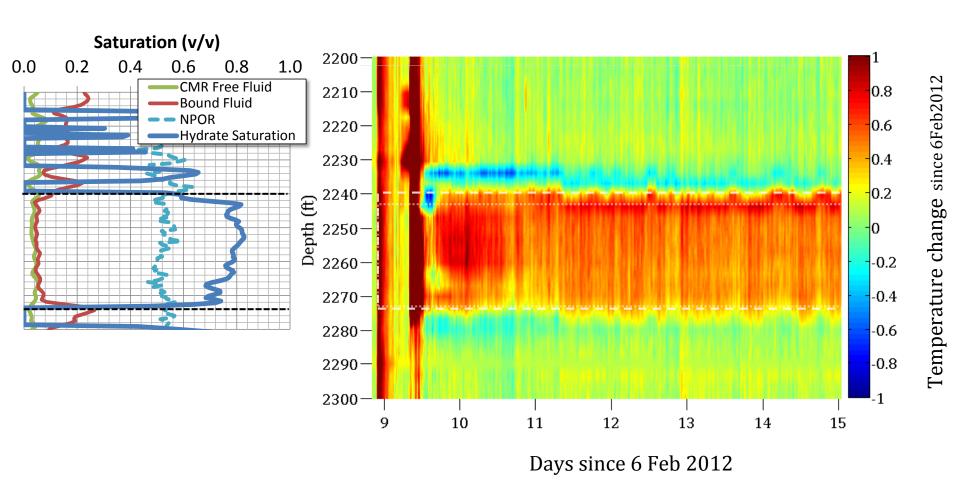


Heterogeneous Injection / Production



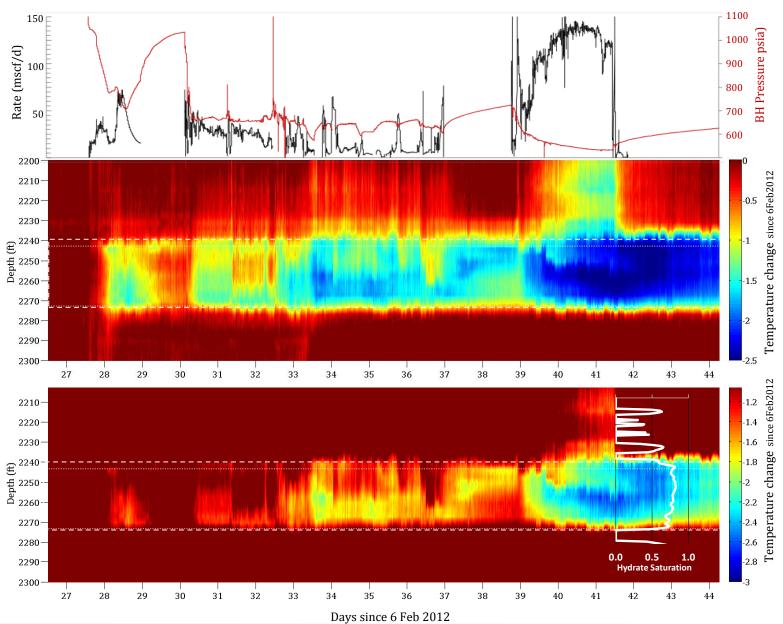


Heterogeneous Injection



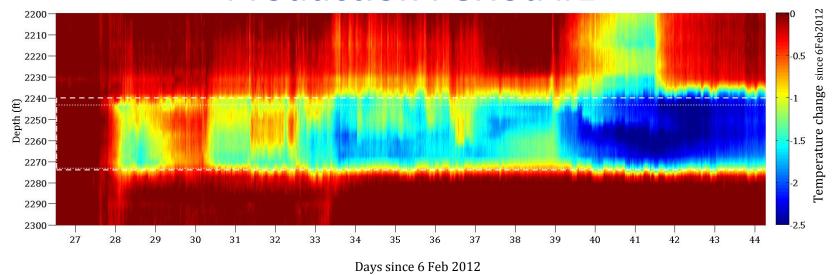


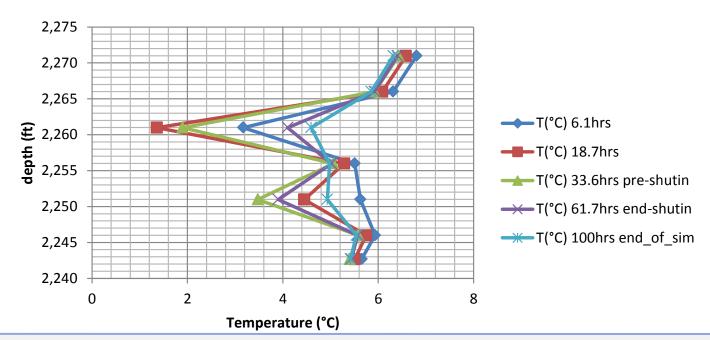
Flowback - Production Period #1





Production Period #1



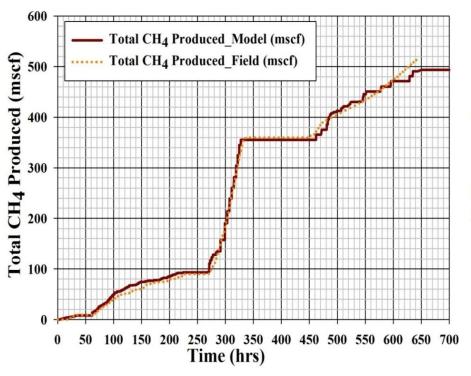


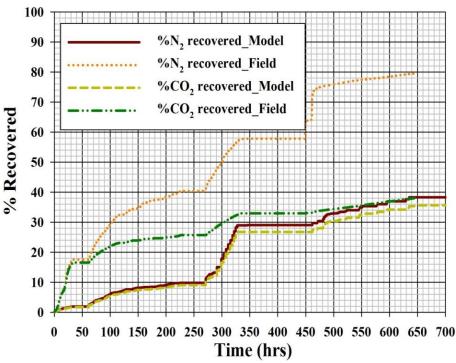


Production Simulations

Production

- Production phase is modeled by maintaining fixed-state boundary as aqueous phase at the bottom-hole pressure.
- Still attempting to match sand production and each gas rate (with recovery factors)







Tentative Conclusions

- Demonstrated injection of CO₂ mixture into water filled hydrate reservoir
 - Possibly some injection out-of-zone
- Confirmed mixture / CH₄-Hydrate Exchange
 - CH4 produced above CH4-hydrate stability pressure
 - Produced CO₂: N₂ ratios altered from injectant value
 - Injectivity decline consistent w hydrate exchange
- Low BHP are achievable during depressurization
 - Icing not observed @ 250 psi BHP
- Heterogeneous injection / production observed (DTS)
- Temperature record consistent w hydrate association / dissociation during injection / production cycles





Going Forward

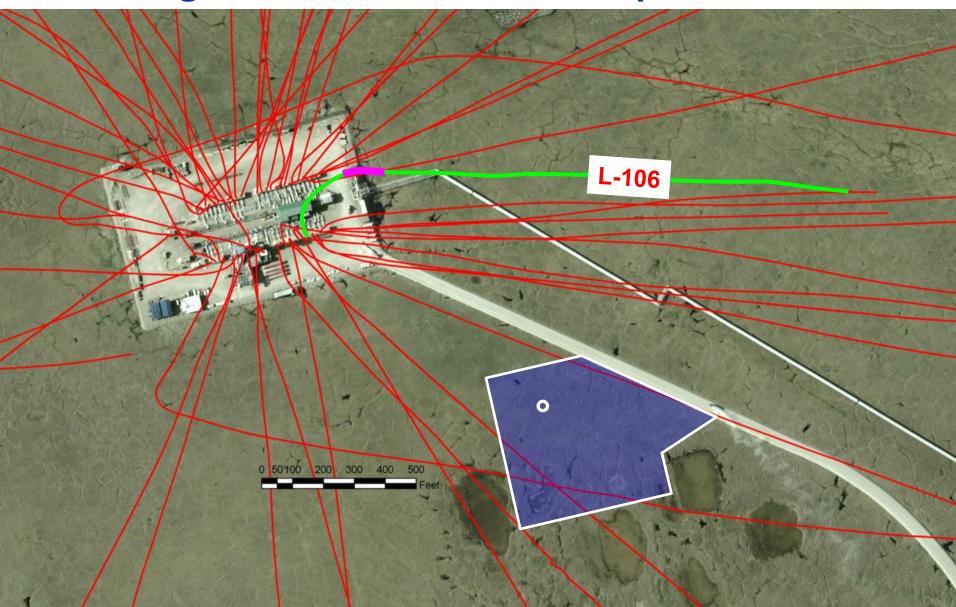
- Datasets and ConocoPhillips project reports can be downloaded from the NETL website.
 - http://www.netl.doe.gov/technologies/oilgas/FutureSupply/MethaneHydrates/rdprogram/ANSWell/co2 ch4exchange.html
 - google "ignik sikumi" or see the announcement in the latest
 Fire in the Ice
- Organizing a problem for the Code Comparison Project on the Ignik Sikumi Results
- DOE has previously facilitated creation of Special Volumes in peer-reviewed journals to consolidate reporting

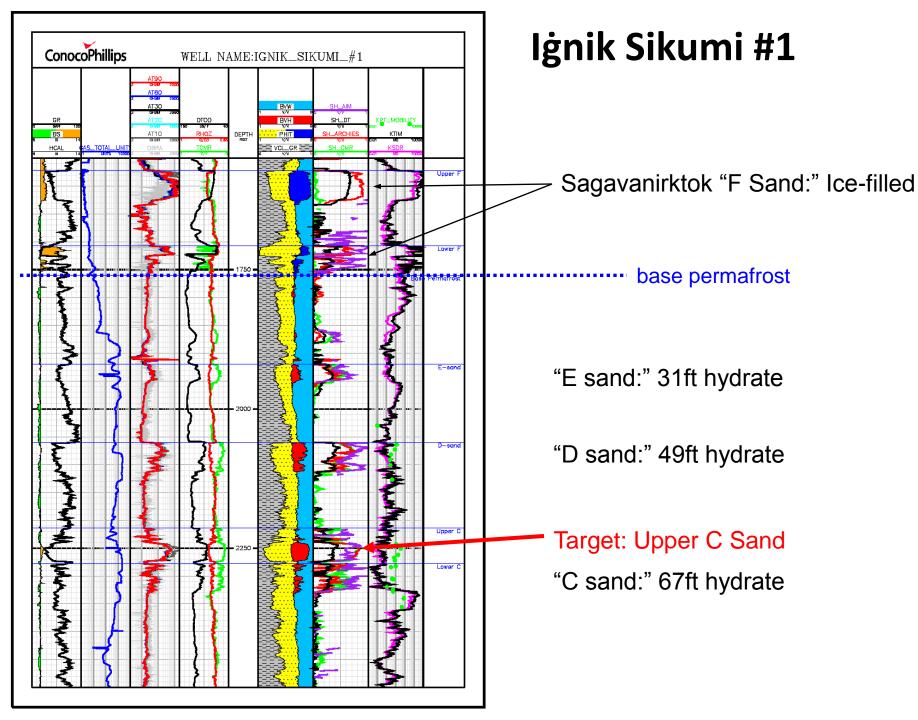




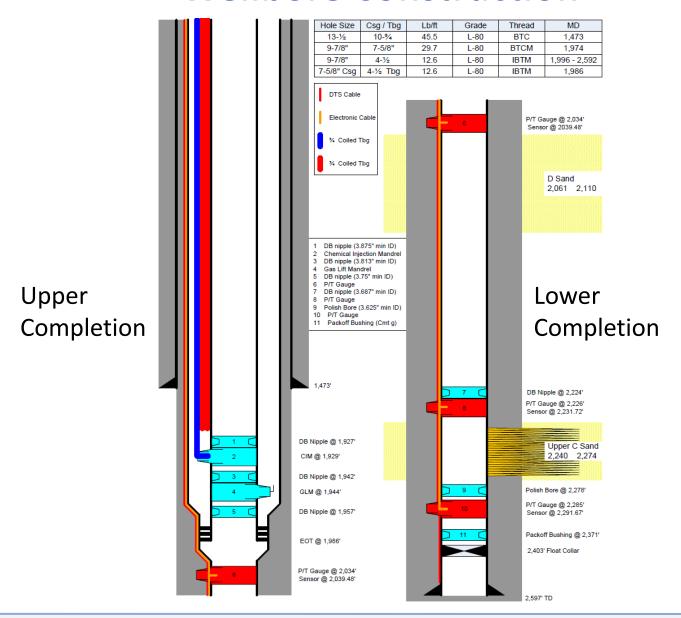
Backup Slides – do not print

CPAI - Ignik Sikumi #1 and PBU L-pad



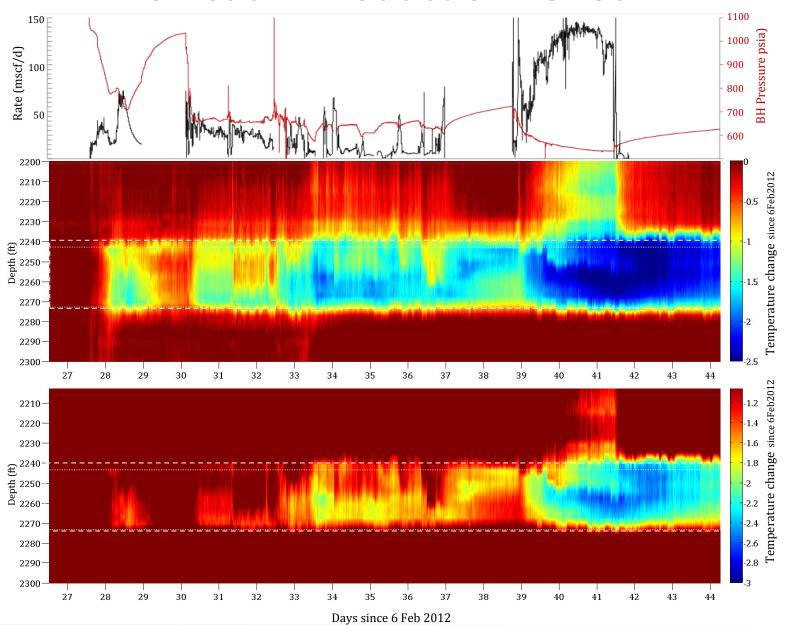


Wellbore Construction





Flowback - Production Period #1





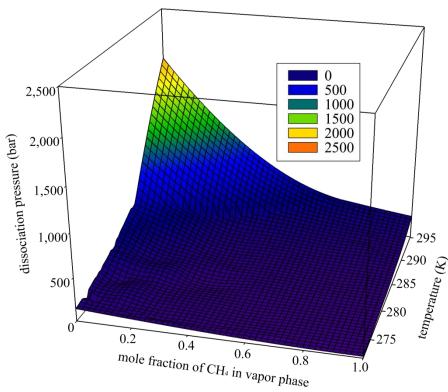
Ternary Hydrate Modeling

- The phase equilibrium data for a three-component (CH₄-CO₂-N₂) gas hydrate are incorporated using tri-linear interpolation, where in the code can interpolate data from a table containing stability pressure, temperature and composition of the hydrate phase
 - Based on predictions using our statistical mechanics model that has been validated against experimental data for 1-, 2-, and 3-component gas mixtures with low error
 - Two data files are incorporated into

$$T_{eq} = f(P, y_1, y_2)$$
 and $P_{eq} = f(T, y_1, y_2)$

where T is temperature (C), P is pressure (MPa), y_1 is CH_4 composition in gas phase and y_2 is CO_2 composition in gas phase (y_{N_2} is not independent)

- Two new primary variables for each phase state and two governing equations are added for the binary (CO₂) and ternary (N₂) gases
- Gas-Hydrate (GsH) system was added to consider the possibility of converting all available free water to form hydrate with injected gas



Prediction of stability pressure for the $\mathrm{CH_4\text{-}CO_2}$ mixed hydrate system